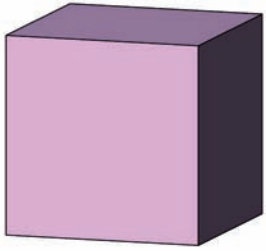
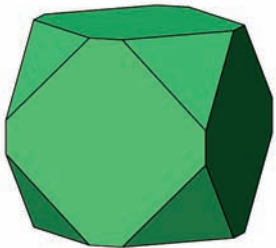


Fluorit:

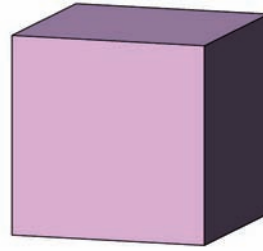


Hexaeder

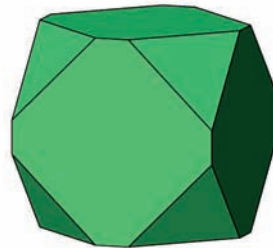


Hexaeder  
Oktaeder

Fluorita:



Hexaedro (cubo)



Hexaedro  
Octaedro

**HERBERT PÖLLMANN & MARCONDES LIMA DA COSTA**

SYMMETRIEN	SIMETRIA
DIE 32 PUNKTGRUPPEN	OS 32 GRUPOS DE PONTOS
MINERALE UND IHRE FORMEN	MINERAIS E SUAS FORMAS



**Mitarbeit**

DIPL.MIN. CHRIS STRAUB

JOÃO VICTOR BORGES FEIO

TIAGO MASCARENHAS AGUIAR

FRANZISKA SETZER

MATEUS MARCHADO MORGADO

FLAVIA SUYANE RODRIGUES SANTOS

**Colaboração**

PROF.DR. ROMULO SIMÕES ANGÉLICA

CAIO ALVES DE MORAES

CHIMEDNOROV OTGONBAYAR

LEONARDO BOIADEIRO AYRES NEGRAO

MAURICE PAWLIK

Titelbild / Imagem da capa

Fluorit / Fluorita

# HALLESCHES JAHRBUCH FÜR GEOWISSENSCHAFTEN

Herausgeber

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der Martin - Luther Universität Halle-Wittenberg

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Schriftleitung

D. MERTMANN T. DEGEN S. STÖBER

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## BEIHEFT 30

Halle (Saale) 2013

Institut für Geowissenschaften und Geographie  
der Martin - Luther Universität Halle-Wittenberg



**Anschrift von Herausgebern und Schriftleitung:**

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## Vorwort

Im vorliegenden zusammenfassenden Buch wird versucht, die Grundlagen der Kristallographie und insbesondere der Symmetriellehre in deutscher und portugiesischer Sprache darzustellen. Die Zusammenfassung der Symmetriellehre basierend auf den 32 Punktgruppen wird erweitert durch Anwendungsbeispiele der makroskopischen Form von wichtigen Mineralen sowie Hinweisen zu den Raumgruppen.

Die Zusammenarbeit zwischen deutschen und brasilianischen Wissenschaftlern über lange Jahre basiert hierbei nicht nur auf der Kooperation in wissenschaftlichen Projekten sondern wurde darüber hinaus auch durch einen regen Wissenschaftler- und Studentenaustausch weiter intensiviert. Die Unterstützung des Deutschen Akademischen Austauschdienstes (DAAD) auf deutscher Seite, bzw. Capes/Cnpq auf brasilianischer Seite soll an dieser Stelle besonders hervorgehoben werden.

Die Idee, Sprachbarrieren durch gemeinsame Grundlagenpublikationen abzubauen, ergab sich aus den Schwierigkeiten von Austauschstudenten, Vorlesungen und Übungen in einer fremden Sprache und zudem einer fremden Fachsprache zu verstehen. Oft werden fehlende Kenntnisse der deutschen oder portugiesischen Sprache dann über Englisch ausgeglichen.

Das Buch ist vor allem dahingehend ausgerichtet, zu helfen, erste Sprachbarrieren zu überwinden und entsprechendes Grundlagenwissen anschaulich darzustellen. Durch die Komplexität des Gesamtgebietes kann hier natürlich nur ein kleiner Ausschnitt aufgezeigt werden, der trotzdem Basisdaten in beiden Sprachen darstellt und als Einführung verwendet werden kann.

Hier soll bewusst der Versuch unternommen werden, die für den Austausch zwischen Brasilien und Deutschland notwendigen Sprachen in den Vordergrund zu stellen. Es ist geplant, diese Zusammenstellung in den kommenden Jahren auszuweiten, zu vervollständigen sowie auch eine entsprechende Zusammenstellung in englischer Sprache zur Verfügung zu stellen.

Besonderer Dank gebührt Frau PD Dr. Dorothee Mertmann für die geduldige und hilfreiche Umsetzung des Manuskriptes in die vorliegende Endform.

Herbert Pöllmann, Halle, September 2013

## Prefácio

O presente livro é uma introdução aos princípios da cristalografia, com ênfase principalmente ao ensino da simetria, usando para tal linguagem e apresentação gráfica a mais clara possível, em dois idiomas, alemão e português.

A cooperação científica duradoura, por décadas, entre pesquisadores alemães e brasileiros, não esteve centrada tão somente em projetos científicos, mas também na contínua e aprofundada formação de recursos humanos em quase todos os níveis, do graduado ao pós-doutorado. Para este sucesso continuado foi muito importante, e continua sendo, o apoio do DAAD pelo lado Alemão e do CNPq e CAPES pelo lado brasileiro.

A ideia para amenizar as barreiras de idiomas surgiu por ocasião das aulas ministrada e durante a discussão e o desenvolvimento de projetos de pesquisas conjuntos, que nos levaram a elaborar publicações introdutórias em conjunto. Em geral se tenta corrigir a falta de conhecimento da língua alemã ou portuguesa através da inglesa, o que nem sempre é óbvia.

Portanto o presente o livro tem também como objetivo auxiliar nesta empreitada voltada para o intercâmbio de estudantes e pesquisadores do Brasil e da Alemanha.

Simplificando o ensino de simetria centrado nos 32 pontos de simetria é enriquecido com desenhos das formas dos minerais mais importantes, além de informações sobre os grupos espaciais. Planejamos enriquecer e ampliar este livro nos próximos anos, em uma nova edição mais completa, considerando inclusive uma edição em língua inglesa.

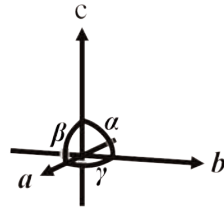
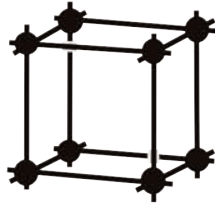
Este livro pensado para ajudar a amenizar as barreiras dos idiomas, também espera principalmente contribuir para uma visão geral do conhecimento sobre cristalografia e mineralogia. É claro que devido a complexidade e a amplitude do conhecimento na área de cristalografia e mineralogia, o presente livro é tão somente uma pequena amostra do mesmo, mesmo assim acreditamos que ao apresentá-lo nos dois idiomas, ele poderá ser utilizado como uma introdução a este vasto conhecimento.

Este livro pensado para ajudar a amenizar as barreiras dos idiomas, também espera principalmente contribuir para uma visão geral do conhecimento sobre cristalografia e mineralogia. É claro que devido a complexidade e a amplitude do conhecimento na área de cristalografia e mineralogia, o presente livro é tão somente uma pequena amostra do mesmo, mesmo assim acreditamos que ao apresentá-lo nos dois idiomas, ele poderá ser utilizado como uma introdução a este vasto conhecimento.

A tentativa em apresentar um texto resumido nos dois idiomas também deverá eliminar a lacuna existente nesta área do conhecimento, já que poderá ser um importante suporte aos estudantes.

Marcondes Lima da Costa, Belem, September 2013

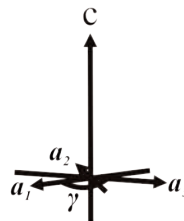
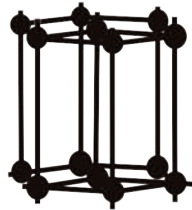
## Die 7 Kristallsysteme



*Kubisch:*

$$a_0 = b_0 = c_0$$

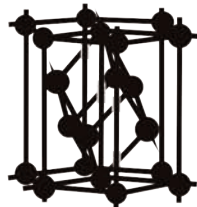
$$\alpha = \beta = \gamma = 90^\circ$$



*Hexagonal:*

$$a_1 = a_2 = a_3 \neq c_0$$

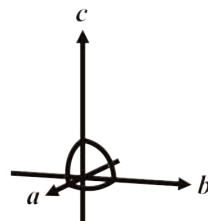
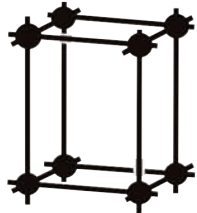
$$\alpha = \beta = 90^\circ; \gamma = 120^\circ$$



*Rhomboedrisch:*

$$a_1 = a_2 = a_3$$

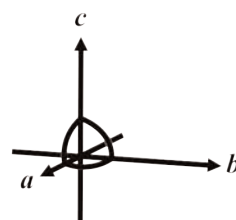
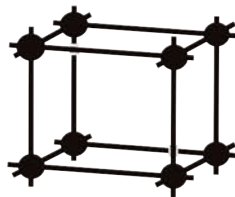
$$\alpha_1 = \alpha_2 = \alpha_3 \neq 90^\circ$$



*Tetragonal:*

$$a_0 = b_0 \neq c_0$$

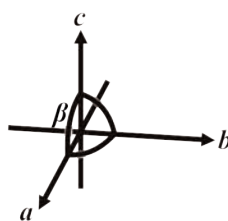
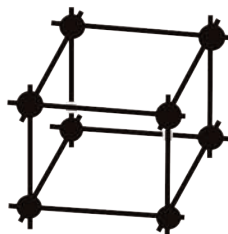
$$\alpha = \beta = \gamma = 90^\circ$$



*Orthorhombisch:*

$$a_0 \neq b_0 \neq c_0$$

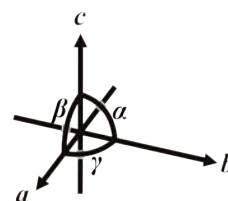
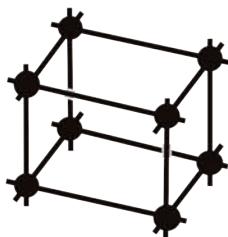
$$\alpha = \beta = \gamma = 90^\circ$$



*Monoklin:*

$$a_0 \neq b_0 \neq c_0$$

$$\alpha = \gamma = 90^\circ; \beta \neq 90^\circ$$



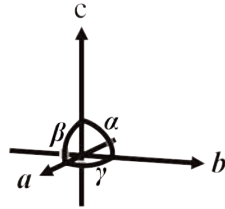
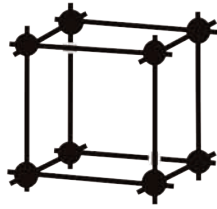
*Triklin:*

$$a_0 \neq b_0 \neq c_0$$

$$\alpha \neq \beta \neq \gamma \neq 90^\circ$$



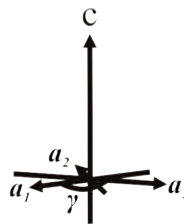
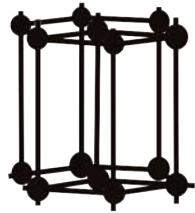
## Os 7 sistemas cristalinos



*Cúbico ou Isométrico:*

$$a_0 = b_0 = c_0$$

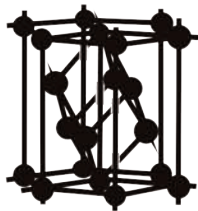
$$\alpha = \beta = \gamma = 90^\circ$$



*Hexagonal:*

$$a_1 = a_2 = a_3 \neq c_0$$

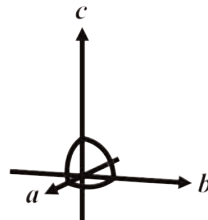
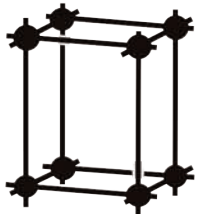
$$\alpha = \beta = 90^\circ; \gamma = 120^\circ$$



*Romboédrico:*

$$a_1 = a_2 = a_3$$

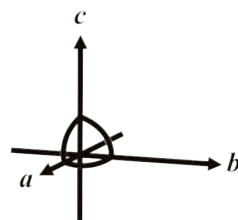
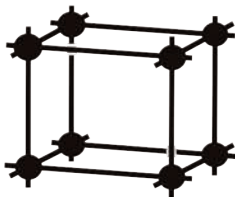
$$\alpha_1 = \alpha_2 = \alpha_3 \neq 90^\circ$$



*Tetragonal:*

$$a_0 = b_0 \neq c_0$$

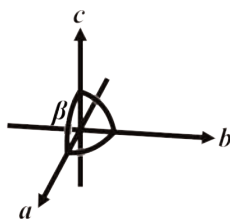
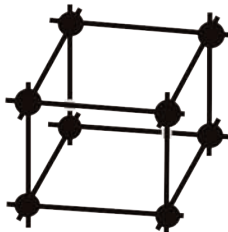
$$\alpha = \beta = \gamma = 90^\circ$$



*Ortorrombico:*

$$a_0 \neq b_0 \neq c_0$$

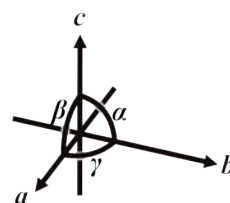
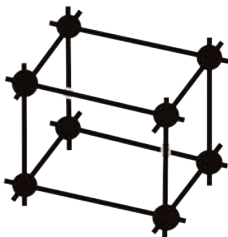
$$\alpha = \beta = \gamma = 90^\circ$$



*Monoclínico:*

$$a_0 \neq b_0 \neq c_0$$

$$\alpha = \gamma = 90^\circ; \beta \neq 90^\circ$$



*Triclínico:*

$$a_0 \neq b_0 \neq c_0$$

$$\alpha \neq \beta \neq \gamma \neq 90^\circ$$

## Häufigkeit von Mineralen

Von 3510 bekannten Mineralen (Stand Mai 1993) sind:

Kubisch	5 Kristallklassen	346	9,9%
Hexagonal	7 Kristallklassen	326	9,3%
Trigonal - Rhomboedrisch	5 Kristallklassen	295	8,4%
Tetragonal	7 Kristallklassen	278	7,9%
Orthorhombisch	3 Kristallklassen	774	22%
Monoklin	3 Kristallklassen	1129	32,2%
Triklin	2 Kristallklassen	332	9,5%

Es gibt 30 Arten amorpher Minerale, sie machen 0,8% der Mineralvorkommen aus.  
Aus „Minerale: Bestimmen nach äußeren Kennzeichen / HOCHLEITNER; PHILIPSBORN; WEINER“

## Wirkung der verschiedenen Drehachsen

Name	Symbol	Wirkung
Inversion	$* / i$	Inversion am Zentrum
Zweizählige Drehachse	$\mathbf{2} / 2$	Drehungen um $180^\circ$
Spiegelebene / inverse zweizählige Drehachse.	$m = \bar{2}$	Spiegelung an einer Ebene
Dreizählige Drehachse	$\mathbf{3} / 3$	Drehungen um $120^\circ$
Inverse dreizählige Drehachse.	$\mathbf{3} / \bar{3}$	Drehungen um $120^\circ$ und Inversion
Vierzählige Drehachse	$\mathbf{4} / 4$	Drehungen um $90^\circ$
Inverse vierzählige Drehachse	$\mathbf{4} / \bar{4}$	Drehungen um $90^\circ$ und Inversion
Sechszählige Drehachse	$\mathbf{6} / 6$	Drehungen um $60^\circ$
Inverse sechszählige Drehachse	$\mathbf{6} / \bar{6}$	Drehungen um $60^\circ$ und Inversion

## Freqüência dos Minerais

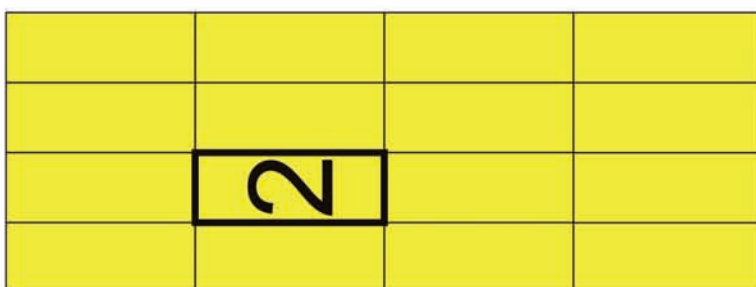
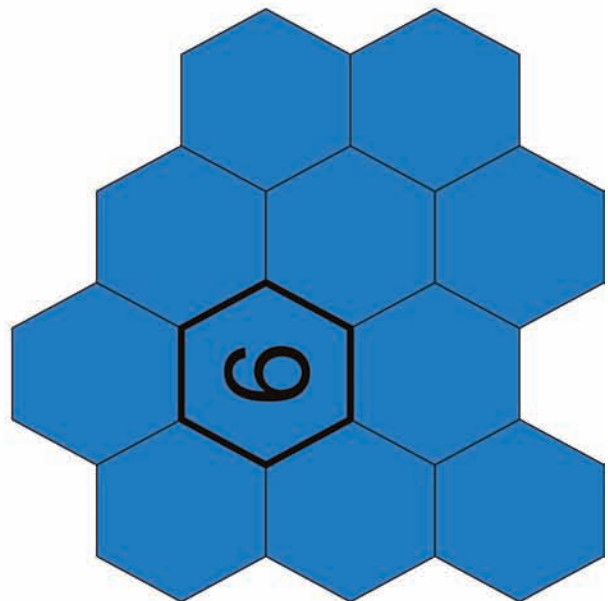
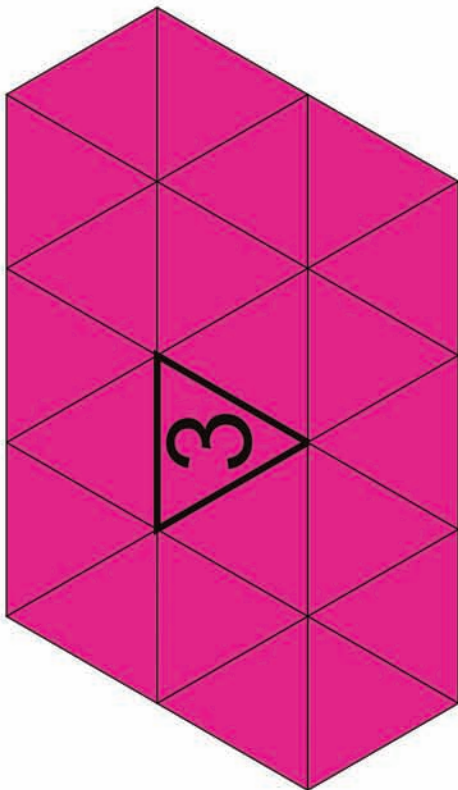
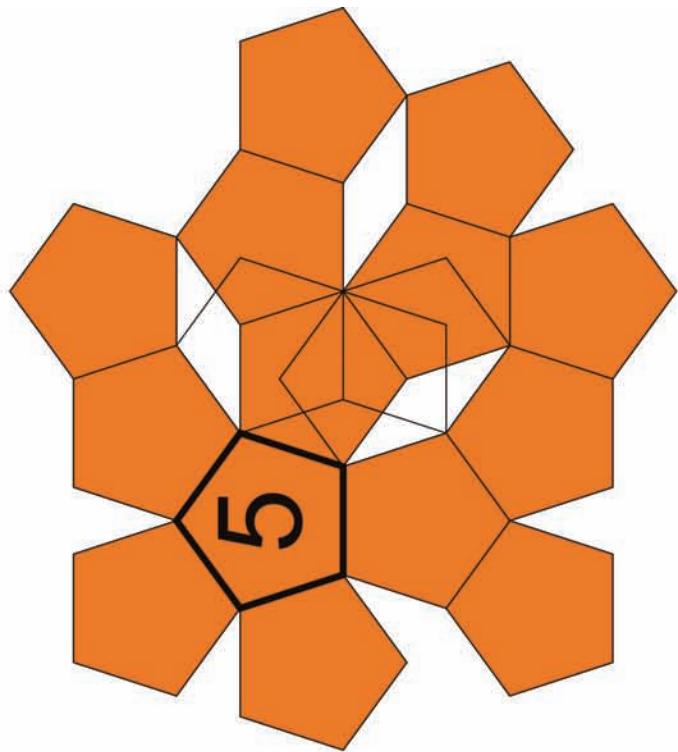
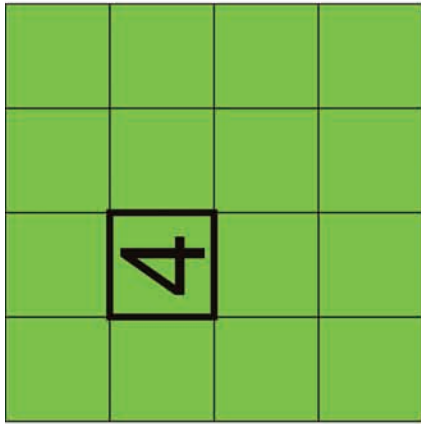
Os 3510 minerais conhecidos até maio de 1993 distribuem-se da seguinte forma:

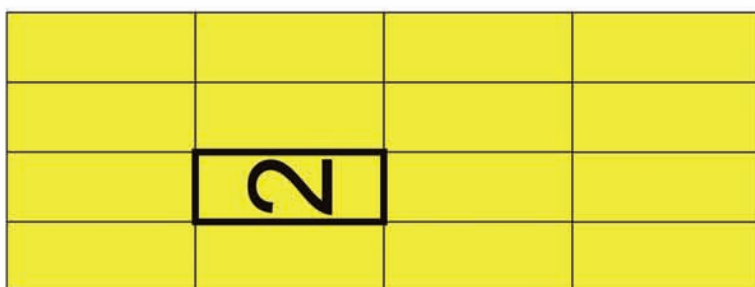
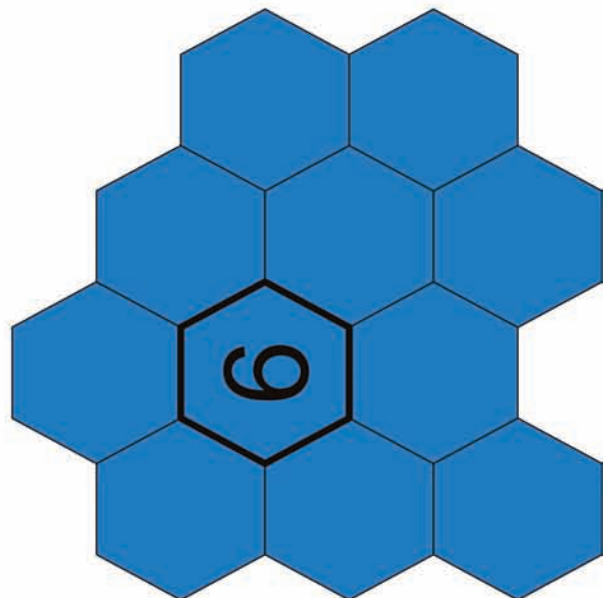
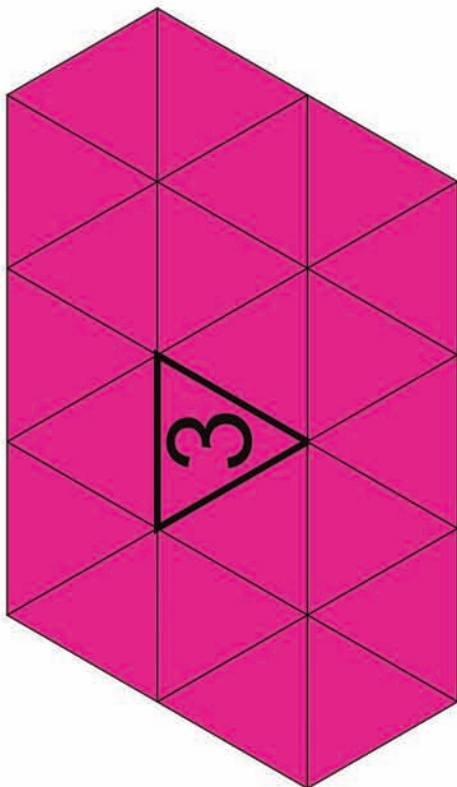
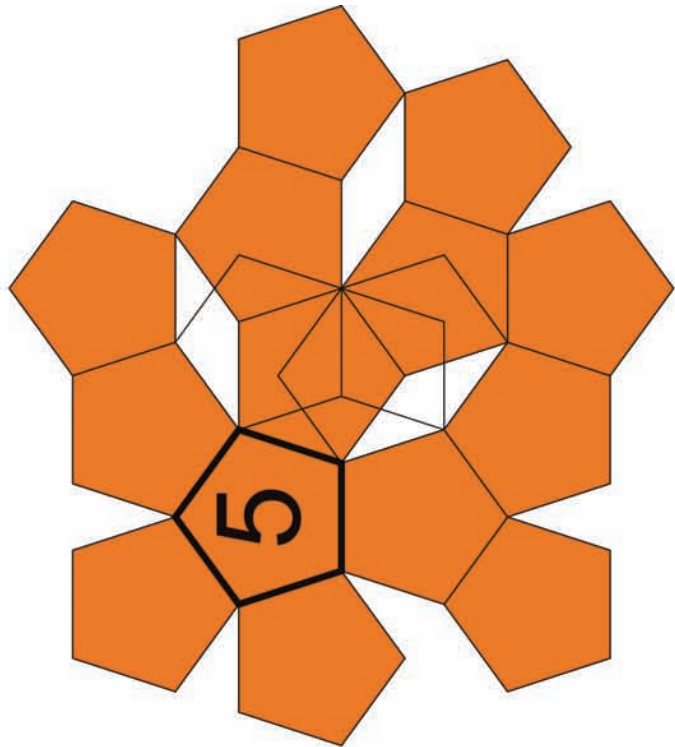
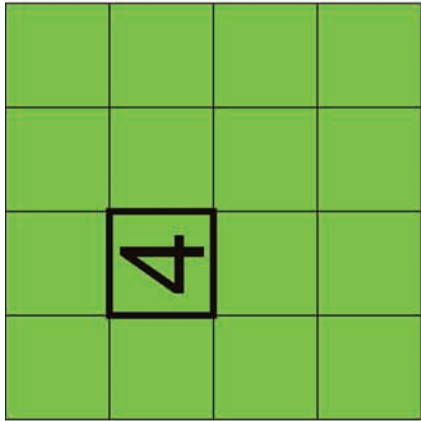
Sistemas	Grupo de Pontos/classes cristalinas	Espécies de minerais	Frequência em percentagem
Isométrico	5	346	9,9%
Hexagonal	7	326	9,3%
Romboédrico	5	295	8,4%
Tetragonal	7	278	7,9%
Ortorrômico	3	774	22%
Monoclínico	3	1129	32,2%
Triclínico	2	332	9,5%

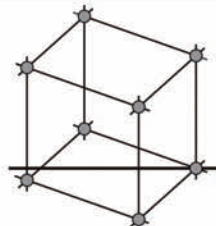
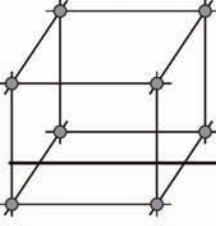
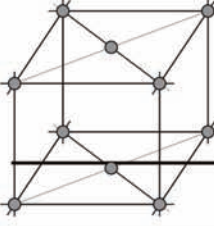
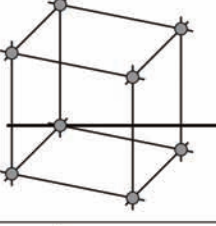
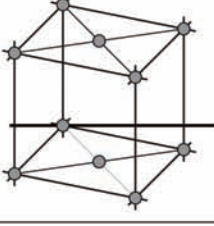
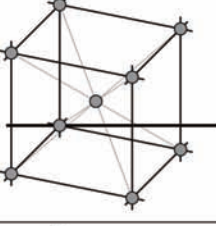
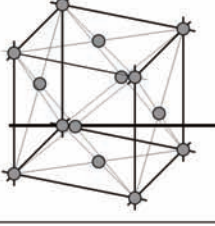
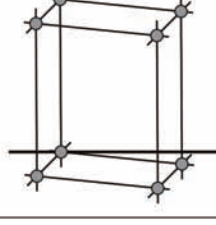
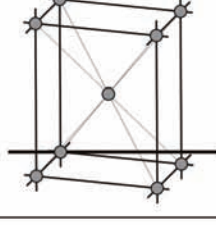
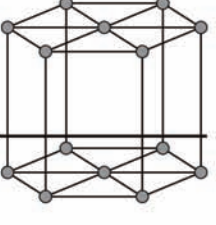
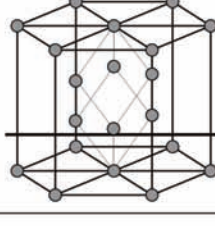
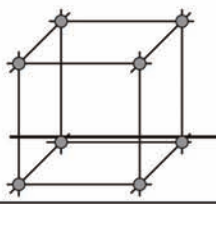
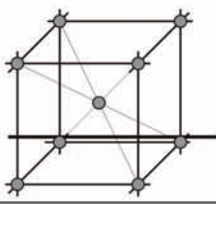
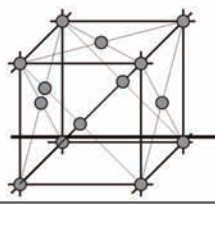
Os minerais amorfos envolvem 30 espécies, ou seja 0,8 % em frequência. Fonte: “Minerais: Identificação através de características externas / HOCHLEITNER; PHILIPSBORN; WEINER”

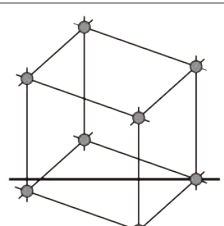
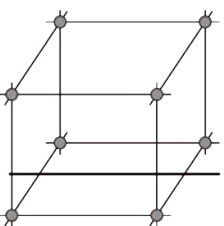
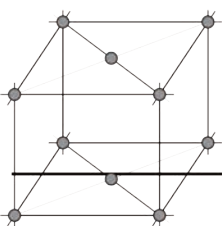
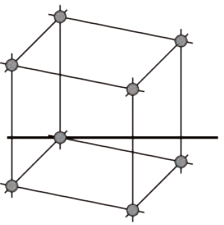
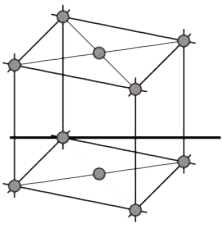
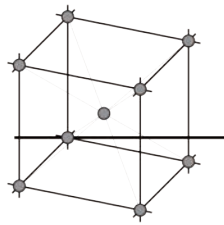
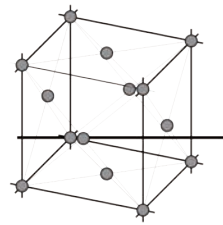
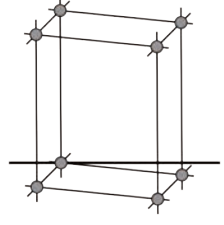
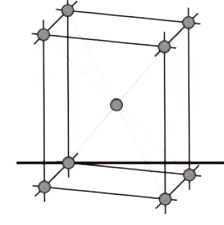
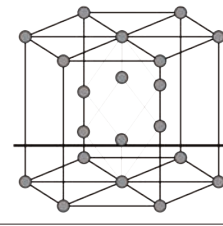
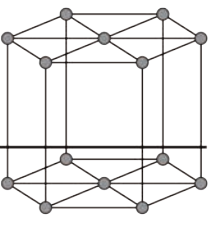
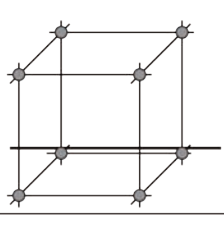
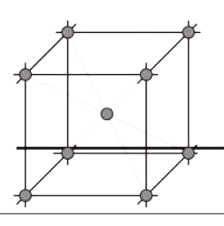
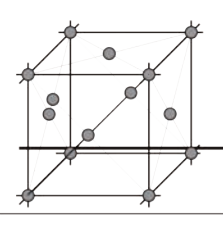
## Efeitos dos diversos eixos de rotação

Nome	Símbolo	Efeito
Inversão (eixo unitário de inversão)	* / i	Centro de inversão
Eixo de rotação binária	2 / 2	Rotação 180°
Plano de reflexão (Espelho)/ Eixo de rotação binária inversa	$m = \bar{2}$	Plano de Reflexão
Eixo de rotação ternária	3 / 3	Rotação a 120°
Eixo de rotação ternária inversa	$\bar{3} / 3$	Rotação a 120° mais inversão
Eixo de rotação quaternária	4 / 4	Rotação a 90°
Eixo de rotação quaternária inversa	$\bar{4} / 4$	Rotação a 90° mais inversão
Eixo de rotação senária	6 / 6	Rotação a 60°
Eixo de rotação senária inversa	$\bar{6} / 6$	Rotação a 60° mais inversão





	P	C	I	F
triklin				
monoklin			identisch mit C-Gitter	identisch mit C-Gitter
orthorhombisch				
tetragonal		identisch mit P-Gitter		identisch mit I-Gitter
trigonal				
hexagonal				
kubisch		unmöglich		

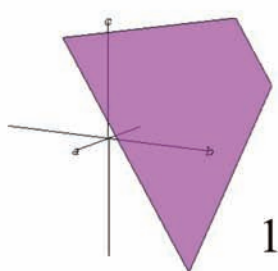
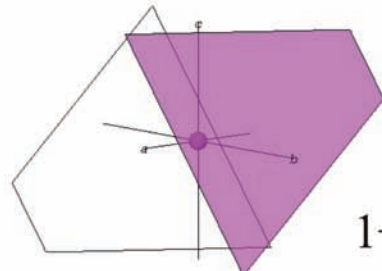
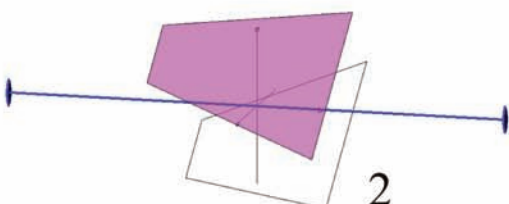
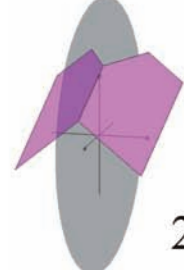
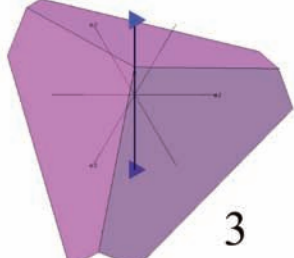
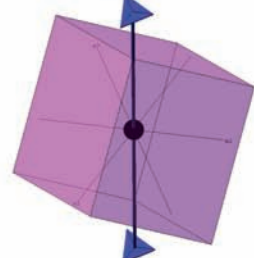
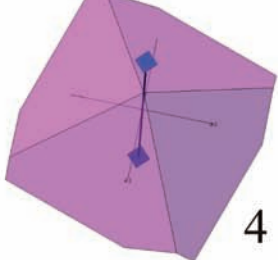

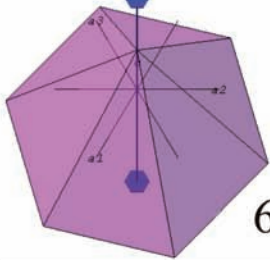
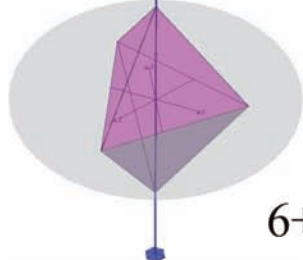
	P	C	I	F
triclínico				
monoclínico			idêntico como retículo tipo C	idêntico como retículo tipo C
ortorrômbo				
tetragonal		idêntico como retículo tipo P		idêntico como retículo tipo I
trigonal				
hexagonal				
cúbico		impossível		

# Die Symmetrieelemente

(normale Drehachsen, Drehachsen kombiniert mit Inversion)  
kombiniert mit  
Inversionszentrum

normal

Inversionszentrum

 <p>1</p>	<p>1</p>	 <p>1+i</p>
 <p>2</p>	<p>2</p>	 <p>2+i = m</p>
 <p>3</p>	<p>3</p>	 <p>3+i</p>
 <p>4</p>	<p>4</p>	 <p>4+i</p>
 <p>6</p>	<p>6</p>	 <p>6+i</p>

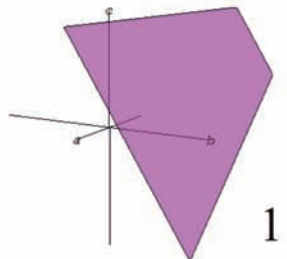
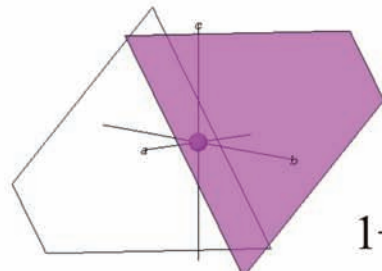
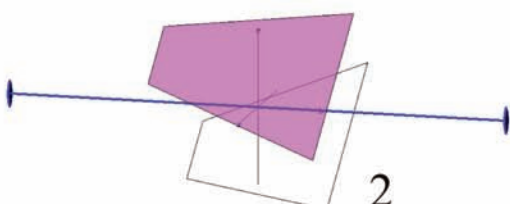
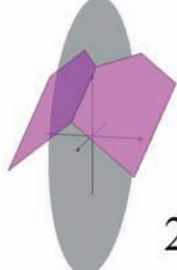
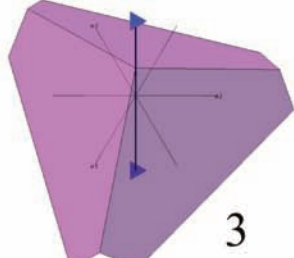
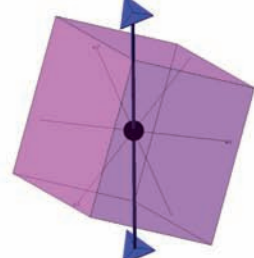
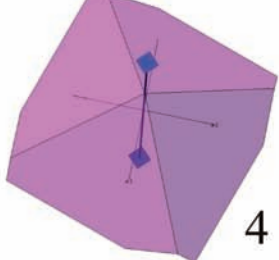
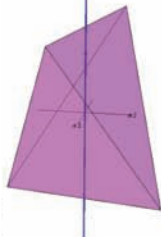
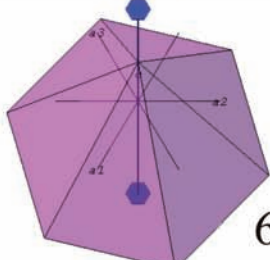
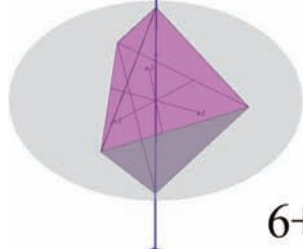


# Os Elementos de Simetria

(eixos normal, eixos com centro de inversão)

Combinado com  
centro de inversão

normal

 <p>1</p>	<p>1</p>	 <p>1+i</p>
 <p>2</p>	<p>2</p>	 <p>2+i = m</p>
 <p>3</p>	<p>3</p>	 <p>3+i</p>
 <p>4</p>	<p>4</p>	 <p>4+i</p>
 <p>6</p>	<p>6</p>	 <p>6+i</p>

Symmetrie	Symbole nach	Symbol				
		*	●	⊠	◆	⬢
	Schönflies	$C_1$	$C_2$	$C_3$	$C_4$	$C_6$
(polare Drehachse)	Hermann-Mauguin	1	2	3	4	6
Kombination von Drehachsen und horizontalen Symmetrieebenen	Schönflies		$C_{2h}$	$C_{3h}$	$C_{4h}$	$C_{6h}$
	Hermann-Mauguin		$2/m$	$3/m$	$4/m$	$6/m$
Kombination von Drehachsen und vertikalen Symmetrieebenen	Schönflies	$C_s$	$C_{2v}$	$C_{3v}$	$C_{4v}$	$C_{6v}$
	Hermann-Mauguin	$m$	$mm2$	$3m$	$4mm$	$6mm$
Kombination von Drehachsen, horizontalen und vertikalen Symmetrieebenen	Schönflies		$D_{2h}$	$D_{3h}$	$D_{4h}$	$D_{6h}$
	Hermann-Mauguin		$mmm$	$1\bar{2}2m$	$4/mmm$	$6/mmm$
Kombination mit zweizähligen Drehachsen	Schönflies		$D_2$	$D_3$	$D_4$	$D_6$
	Hermann-Mauguin		$222$	$32$	$422$	$622$
Symmetriezentrum und Inversionsachsen	Schönflies	$C_i$		$C_{3i}$	$S_4$	
	Hermann-Mauguin	$\bar{1}$	$1\bar{1}(m)$	$1\bar{1}c$	$1\bar{1}2m$	$1\bar{1}c(3/m)$ $1\bar{1}2m$

<b>Simetria</b>	<b>Simbologia segundo</b>	<b>Simbologia</b>				
		*	●	⊠	◆	⬢
<i>(Eixos de rotação polar)</i>	Schönflies	$C_1$	$C_2$	$C_3$	$C_4$	$C_6$
	Hermann-Mauguin	1	2	3	4	6
<i>Combinação de eixos de rotação e plano de simetria horizontal</i>	Schönflies		$C_{2h}$	$C_{3h}$	$C_{4h}$	$C_{6h}$
	Hermann-Mauguin		2/m	3/m ( $\bar{1}$ )	4/m	6/m
<i>Combinação de eixos de rotação com plano de simetria vertical</i>	Schönflies	$C_s$	$C_{2v}$	$C_{3v}$	$C_{4v}$	$C_{6v}$
	Hermann-Mauguin	m	mm2	3m	4mm	6mm
<i>Combinação de eixo de rotação com plano de simetria vertical e horizontal</i>	Schönflies		$D_{2h}$	$D_{3h}$	$D_{4h}$	$D_{6h}$
	Hermann-Mauguin		mmm	$\bar{1}2m$	4/mmm	6/mmm
<i>Combinação com eixos de simetria binário</i>	Schönflies		$D_2$	$D_3$	$D_4$	$D_6$
	Hermann-Mauguin		222	32	422	622
<i>Centro de simetria e eixo de inversão</i>	Schönflies	$C_i$		$C_{3i}$	$S_4$	
	Hermann-Mauguin	$\bar{1}$	$\bar{1}2(m)$	$\bar{1}2m$	$\bar{1}2m$	$\bar{1}2(m)$ $\bar{1}2m$

# Die 32 Punktgruppen

Kristallsystem	Triklin	Monoklin	Orthorhombisch	Trigonal	Hexagonal	Tetragonal	Kubisch
<b>Holoedrie</b>	$\bar{1}$	$\frac{2}{m}$	$\frac{2}{m} \frac{2}{m} \frac{2}{m}$	$\frac{3}{m}$	$\frac{6}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{3}{m} \frac{2}{m}$
<b>Blickrichtung</b>	- keine-	[010]	[100] [010] [001]	[001] [100]	[001] [100] [110]	[001] [100] [110]	[001] [111] [110]
<b>Punktgruppen</b>	1 $\bar{1}$	2 m $\frac{2}{m}$	2 2 2 m m 2 $\frac{2}{m} \frac{2}{m} \frac{2}{m}$	3 3 2 3 m $\bar{3}$ $\frac{3}{m}$	6 6 2 2 6 m m $\bar{6}$ $\bar{6} 2 m$ $\bar{6} m 2$ $\frac{6}{m}$ $\frac{6}{m} \frac{2}{m} \frac{2}{m}$	4 4 2 2 4 m m $\bar{4}$ $\bar{4} 2 m$ $\bar{4} m 2$ $\frac{4}{m}$ $\frac{4}{m} \frac{2}{m} \frac{2}{m}$	2 3 4 3 2 $\frac{2}{m} \frac{3}{m}$ $\bar{4} 3 m$ $\frac{4}{m} \frac{3}{m} \frac{2}{m}$

# Os 32 Grupos de Pontos

Sistema Cristalino	Triclínico	Monoclínico	Ortorrombico	Trigonal	Hexagonal	Tetragonal	Cubico
<b>Holoedria</b>	$\bar{1}$	$\frac{2}{m}$	$\frac{2}{m} \frac{2}{m} \frac{2}{m}$	$\frac{3}{m}$	$\frac{6}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{3}{m} \frac{2}{m}$
<b>Perspectiva</b>	- nenhum-	[010]	[100] [010] [001]	[001] [100]	[001] [100] [110]	[001] [100] [110]	[001] [111] [110]
<b>Grupos de Pontos</b>	1 $\bar{1}$	2 m $\frac{2}{m}$	2 2 2 m m 2 $\frac{2}{m} \frac{2}{m} \frac{2}{m}$	3 3 2 3 m $\bar{3}$ $\frac{3}{2} \frac{2}{m}$	6 6 2 2 6 m m $\bar{6}$ $\bar{6} 2 m$ $\bar{6} m 2$ $\frac{6}{m}$ $\frac{6}{m} \frac{2}{m} \frac{2}{m}$	4 4 2 2 4 m m $\bar{4}$ $\bar{4} 2 m$ $\bar{4} m 2$ $\frac{4}{m}$ $\frac{4}{m} \frac{2}{m} \frac{2}{m}$	2 3 4 3 2 $\frac{2}{m} \frac{3}{m}$ $\bar{4} 3 m$ $\frac{4}{m} \frac{3}{m} \frac{2}{m}$

## Die 32 Kristallklassen

Anzahl der Klassen	Bezeichnung nach Herman-Mauguin		Bezeichnung der Kristallklasse
	komplett	abgekürzt	
Triklines System			
2 Klassen	1	1	Triklin Pedial
	-1	-1	Triklin Pinakoidal
Monoklines System			
3 Klassen	$2/m$	$2/m$	Monoklin Prismatisch
	$m$	$m$	Monoklin Domatisch
	2	2	Monoklin Sphenoid
Orthorhombisches System			
3 Klassen	$2/m2 / m/m$	$mmm$	Orthorhombisch Dipyramidal
	$mm2$	$mm2$	Orthorhombisch Pyramidal
	$222$	$222$	Orthorhombisch Disphenoid
Tetraagonales System			
7 Klassen	$4/m2/m2/m$	$4/mmm$	Ditetragonal Dipyramidal
	$4mm$	$4mm$	Ditetragonal Pyramidal
	$422$	$422$	Tetraagonal Trapezoedrisch
	$4/m$	$4/m$	Tetraagonal Dipyramidal
	4	4	Tetraagonal Pyramidal
	$4-2 m$	$4-2 m$	Tetraagonal Skalenoedrisch
	-4	-4	Tetraagonal Disphenoidisch
Hexagonales System:			
7 Klassen	$6/m2/m2/m$	$6/mmm$	Dihexagonal Dipyramidal
	$6mm$	$6mm$	Dihexagonal Pyramidal
	$622$	$622$	Hexagonal Trapezoedrisch
	$6/m$	$6/m$	Hexagonal Dipyramidal
	6	6	Hexagonal Pyramidal
	$-6m2$	$-6m2$	Ditrigonal Dipyramidal
	-6	-6	Trigonal Dipyramidal
Trigonales System:			
5 Klassen	$-3 2m$	$-3m$	Ditrigonal Skalenoedrisch
	$3m$	$3m$	Ditrigonal Pyramidal
	$32$	$32$	Trigonal Trapezoedrisch
	3	3	Trigonal Pyramidal
	-3	-3	Trigonal Rhomboedrisch
Kubisches System			
5 Klassen	$4/m-3 2/m$	$m3m$	Hexakisoktaedrisch
	$432$	$432$	Pentagonikositetraedrisch
	$2/m -3$	$m3$	Disdodekaedrisch
	$-4 3m$	$-4 3m$	Hexakistetraedrisch
	$23$	$23$	Tetraedrisch Pentagondodekaedrisch

## As 32 classes de simetria

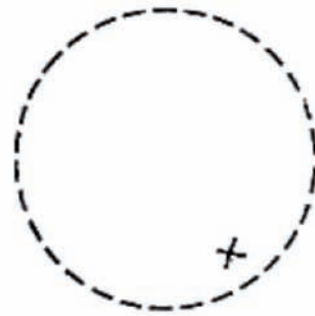
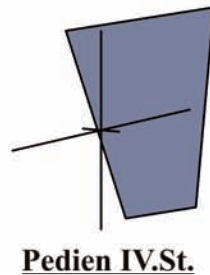
Número de classes	Índices de Hermann-Mauguin		Denominação da classe
	completos	abreviados	
Sistema triclínico:			
2 classes	1	1	classe pinacoidal
	-1	-1	classe pedial
Sistema monoclinico:			
3 classes	$2/m$	$2/m$	classe prismática
	$m$	$m$	classe domática
	2	2	classe esfenóidica
Sistema ortorrômbico:			
3 classes	$2/m2/m/m$	$mmm$	classe dipiramidal ortorrômbica
	$mm2$	$mm2$	classe piramidal ortorrômbica
	222	222	classe diesfenóidica ortorrômbica
Sistema tetragonal:			
7 classes	$4/m2/m2/m$	$4/mmm$	classe dipiramidal ditetragonal
	$4mm$	$4mm$	classe piramidal ditetragonal
	422	422	classe trapezoédrica tetragonal
	$4/m$	$4/m$	classe dipiramidal tetragonal
	4	4	classe piramidal tetragonal
	$4-2m$	$4-2m$	classe escalenoédrica tetragonal
	-4	-4	classe diesfenóidica tetragonal
Sistema hexagonal: -1 -2 -4 -6 -3			
7 classes	$6/m2/m2/m$	$6/mmm$	classe dipiramidal dihexagonal
	$6mm$	$6mm$	classe piramidal- dihexagonal
	622	622	classe trapezoédrica-hexagonal
	$6/m$	$6/m$	classe dipiramidal-hexagonal
	6	6	classe piramidal-hexagonal
	$-6m2$	$-6m2$	classe dipiramidal ditrigonal
	-6	-6	classe dipiramidal trigonal
Sistema trigonal:			
5 classes	$-3 2m$	$-3m$	classe escalenoédrica ditrigonal
	$3m$	$3m$	classe piramidal ditrigonal
	32	32	classe trapezoédrica trigonal
	3	3	classe piramidal trigonal
	-3	-3	classe romboédrica
Sistema cúbico ou isométrico:			
5 classes	$4/m-3 2/m$	$m3m$	classe hexaocáédrica
	432	432	classe pentagontriaocáédrica
	$2/m -3$	$m3$	classe didodecaédrica
	$-4 3m$	$-4 3m$	classe hexatetraédrica
	23	23	classe pentagontriaedrica

# Triklines Kristallsystem

Triklone Hemiedrie  
Triklin-pediale Klasse

Symbol: 1 oder  $C_1$

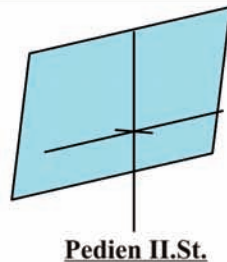
Allgemeine Form:  
{hkl}



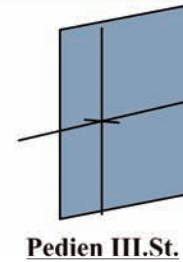
{0kl}



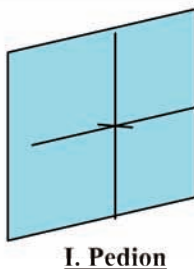
{h0l}



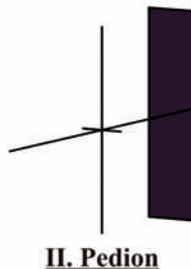
{hk0}



{100}



{010}



{001}



## Mineral-Bsp.:

Sinnerite  $Cu_6As_4S_9$

Hartite  $C_{20}H_{34}$

Nekoite  $Ca_3[Si_6O_{15}] \cdot 7H_2O$

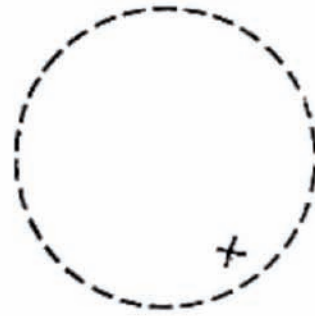
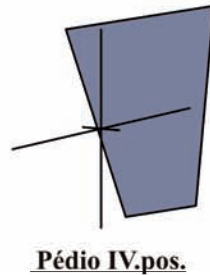


# Sistema Cristalino Triclínico

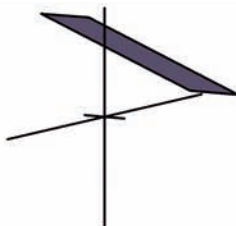
Hemiedria Triclínico  
Classe pedial triclínica

Símbolo: 1 ou  $C_1$

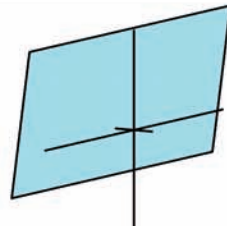
Forma geral:  
{hkl}



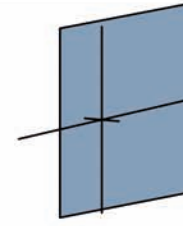
{0kl}



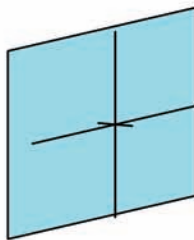
{h0l}



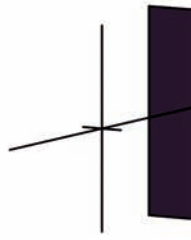
{hk0}



{100}



{010}



{001}



## Exemplos de minerais:

Sinnerita  $Cu_6As_4S_9$

Hartita  $C_{20}H_{34}$

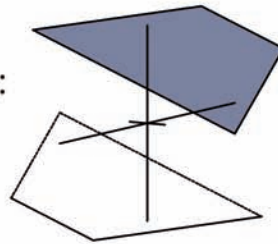
Nekoita  $Ca_3[Si_6O_{15}] \cdot 7H_2O$

# Triklines Kristallsystem

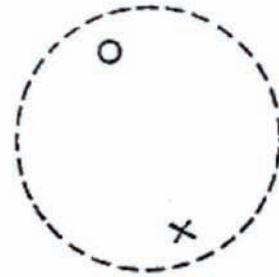
Triklone Holoedrie  
Triklin-pinakoidale Klasse

Symbol:  $\bar{1}$  oder  $C_i$

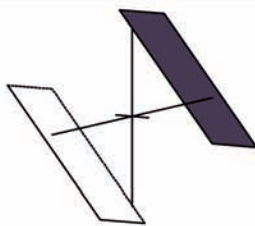
Allgemeine Form:  
{hkl}



Pinakoide IV.St.

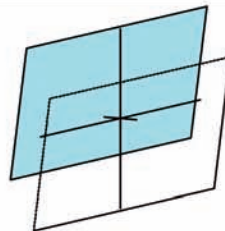


{0kl}



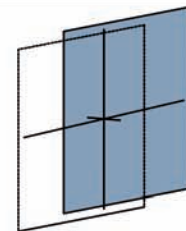
Pinakoide I.St.

{h0l}



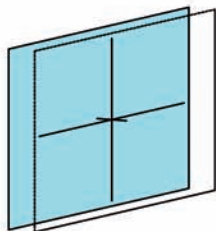
Pinakoide II.St.

{hk0}



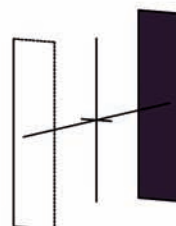
Pinakoide III.St.

{100}



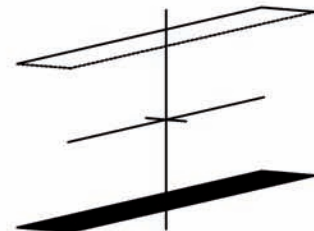
I. Pinakoid

{010}



II. Pinakoid

{001}



III. Pinakoid

## Mineral-Bsp.:

Albit  $\text{Na}[\text{AlSi}_3\text{O}_8]$

Kyanit/Disthen  $\text{Al}_2[\text{O}/\text{SiO}_4]$

Wollastonit  $\text{Ca}[\text{SiO}_3]$

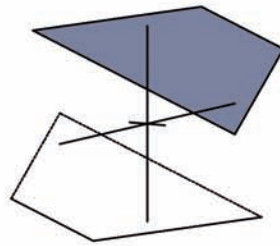
Axinit  $\text{Ca}_2(\text{Fe},\text{Mn})\text{Al}_2[\text{BO}_3\text{OH}/\text{Si}_4\text{O}_{12}]$

# Sistema Cristalino Triclínico

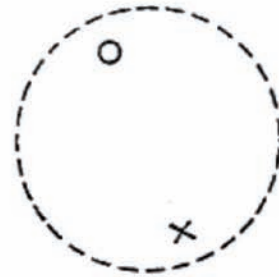
Holoedria Triclínico  
Classe pinacoidal triclínica

Símbolo:  $\bar{1}$  ou  $C_i$

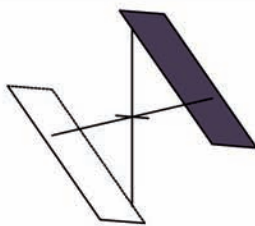
Forma geral:  
{hkl}



Pinacóide IV. pos.

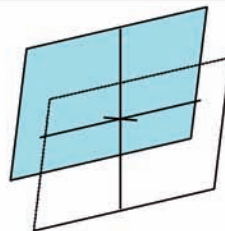


{0kl}



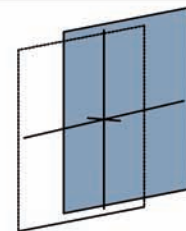
Pinacóide I. pos.

{h0l}



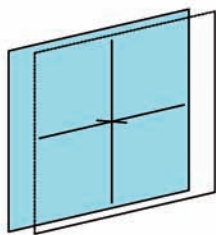
Pinacóide II. pos.

{hk0}



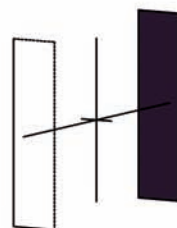
Pinacóide III. pos.

{100}



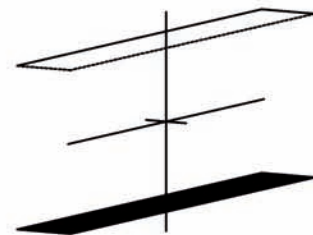
I. Pinacóide

{010}



II. Pinacóide

{001}



III. Pinacóide

## Exemplos de Minerais:

Albita  $\text{Na}[\text{AlSi}_3\text{O}_8]$

Cianita/Distênio  $\text{Al}_2[\text{O}/\text{SiO}_4]$

Wollastonita  $\text{Ca}[\text{SiO}_3]$

Axinita  $\text{Ca}_2(\text{Fe},\text{Mn})\text{Al}_2[\text{BO}_3\text{OH}/\text{Si}_4\text{O}_{12}]$

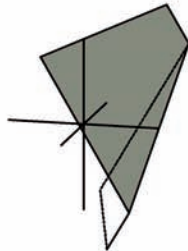
# Monoklines Kristallsystem

Monokline Hemimorphie

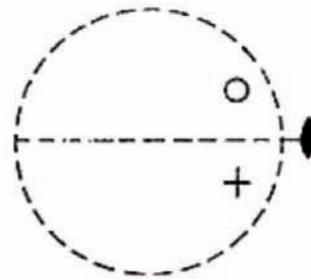
Symbol: 2 oder  $C_2$

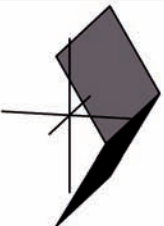
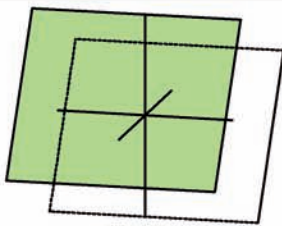
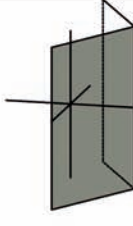
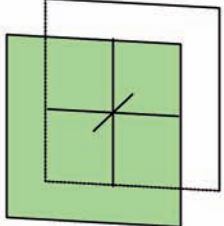

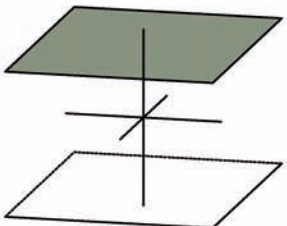
Monoklin-sphenoidische Klasse

Allgemeine Form:  
{hkl}



Sphenoide IV.St.



{0kl}	{h0l}	{hk0}
 <p data-bbox="296 1240 475 1272"><u>Sphenoide I.St.</u></p>	 <p data-bbox="708 1240 887 1272"><u>Pinakoide II.St.</u></p>	 <p data-bbox="1114 1240 1310 1272"><u>Sphenoide III.St.</u></p>
{100}	{010}	{001}
 <p data-bbox="320 1644 448 1675"><u>I. Pinakoid</u></p>	 <p data-bbox="740 1644 868 1675"><u>II. Pedion</u></p>	 <p data-bbox="1139 1644 1299 1675"><u>III. Pinakoid</u></p>

## Mineral-Bsp.:

(Rohr-/Kandis-)Zucker  $C_{12}H_{22}O_{11}$

Weinsäure  $C_4H_6O_6$

Latiumite  $(Ca,K)_8(Al,Mg,Fe)[(Si,Al)_{10}O_{25}/(SO_4)]$

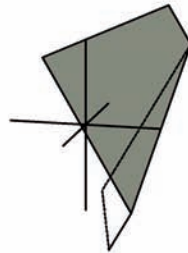
# Sistema Cristalino Monoclínico

Monoclínico hemimorfite

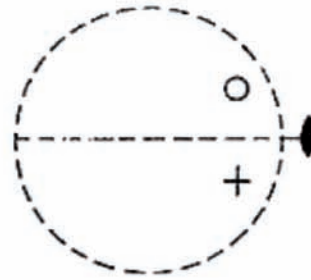
Símbolo: 2 ou  $C_2$

Classe monoclínica-esfenoidal

Forma geral:  
{hkl}



Esfenoide IV.pos.



{0kl}	{h0l}	{hk0}
<u>Esfenoide I.pos.</u>	<u>Pinacoide II.pos.</u>	<u>Esfenoide III.pos.</u>
{100}	{010}	{001}
<u>I. Pinacoide</u>	<u>II. Pédion</u>	<u>III. Pinacoide</u>

## Exemplos de minerais:

Açúcar de cana  $C_{12}H_{22}O_{11}$

Àcido tartárico  $C_4H_6O_6$

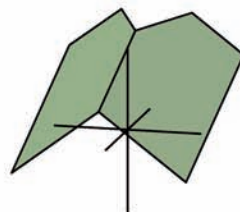
Latiumita  $(Ca,K)_8(Al,Mg,Fe)[(Si,Al)_{10}O_{25}/(SO_4)]$

# Monoklines Kristallsystem

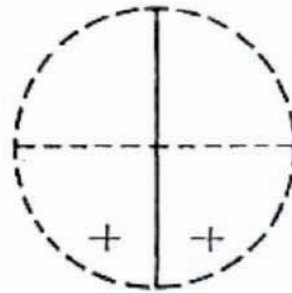
Monokline Hemiedrie II. Art  
Monoklin-domatische Klasse

Symbol:  $m$  oder  $C_s$

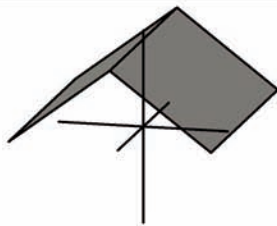
Allgemeine Form:  
 $\{hkl\}$



**Domen IV.St.**

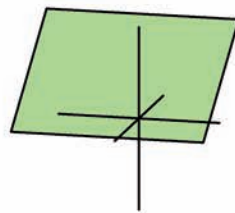


$\{0kl\}$



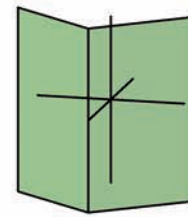
**Domen I.St.**

$\{h0l\}$



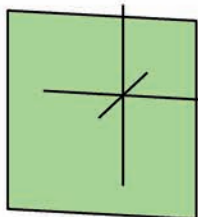
**Pedien II.St.**

$\{hk0\}$



**Domen III.St.**

$\{100\}$



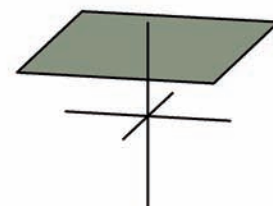
**I. Pedion**

$\{010\}$



**II. Pinakoid**

$\{001\}$



**III. Pedion**

## Mineral-Bsp.:

Posnjakite  $Cu_4[(OH)_6/SO_4] \cdot H_2O$

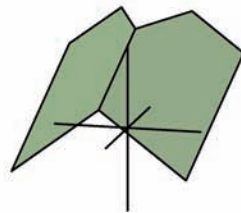
Hilgardit  $Ca_2[Cl/B_5O_8(OH)_2]$

Skolezit  $Ca[Al_2Si_3O_{10}] \cdot 3H_2O$

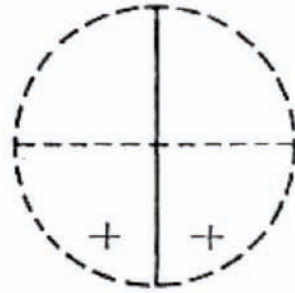
# Sistema Cristalino Monoclínico

Monoclínico Hemiédrico II. Art    Símbolo:  $m$  ou  $C_s$   
 Monoclínico- Classe domática

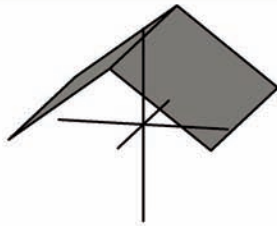
Forma geral:  
 $\{hkl\}$



**Domo IV.pos.**

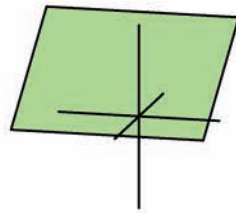


$\{0kl\}$



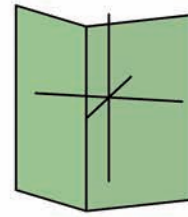
**Domo I.pos.**

$\{h0l\}$



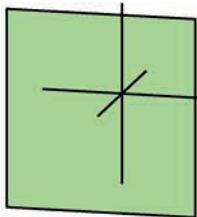
**Pédion II.pos.**

$\{hk0\}$



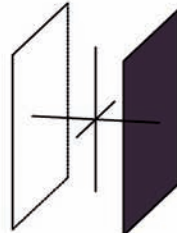
**Domo III.pos.**

$\{100\}$



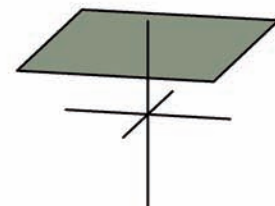
**I. Pédio**

$\{010\}$



**II. Pinacoide**

$\{001\}$



**III. Pédio**

## Exemplos de minerais:

Posnjaquita  $Cu_4[(OH)_6/SO_4]*H_2O$

Hilgardita  $Ca_2[Cl/B_5O_8(OH)_2]$

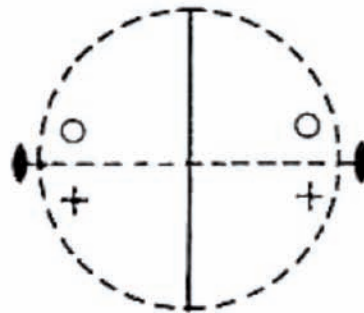
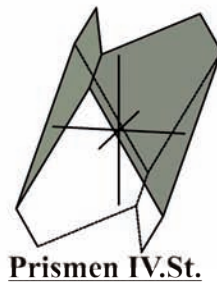
Escolecita  $Ca[Al_2Si_3O_{10}]*3H_2O$

# Monoklines Kristallsystem

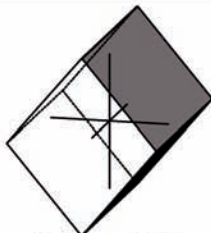
Monokline Holoedrie  
Monoklin-prismatische Klasse

Symbol:  $2/m$  oder  $C_{2h}$

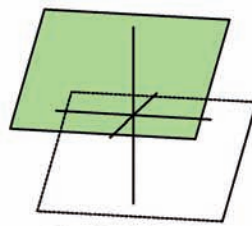
Allgemeine Form:  
{hkl}



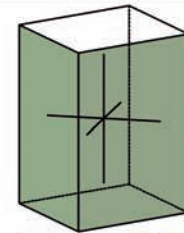
{0kl}



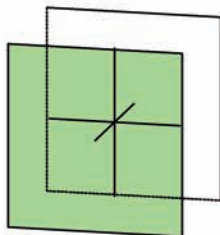
{h0l}



{hk0}



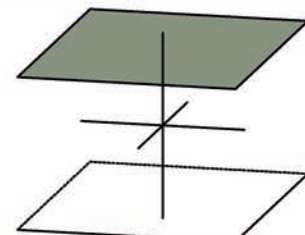
{100}



{010}



{001}



## Mineral-Bsp.:

- häufigste Kristallklasse
- siehe Zusatzseite

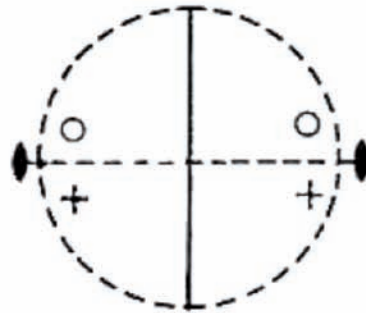
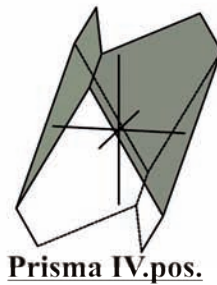


# Sistema Cristalino Monoclínico

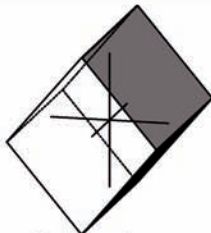
Monoclínico Holoedro  
Classe monoclínico-prismático

Símbolo:  $2/m$  ou  $C_{2h}$

Forma geral:  
{hkl}

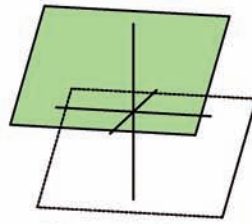


{0kl}



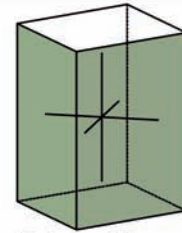
**Prisma I.pos.**

{h0l}



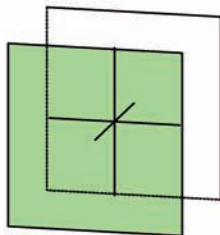
**Pinacoide II.pos.**

{hk0}



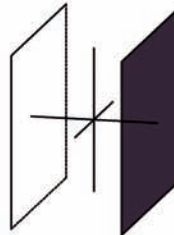
**Prisma III.pos.**

{100}



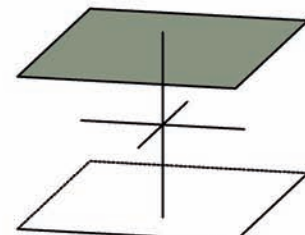
**I. Pinacoide**

{010}



**II. Pinacoide**

{001}



**III. Pinacoide**

## Exemplos de minerais:

- classe cristalina mais frequente
- ver a página adicional

**Mineral-Bsp. für 2/m:** (häufigste Kristallklasse)

Gips  $\text{Ca}[\text{SO}_4] \cdot 2\text{H}_2\text{O}$

Diopsid  $\text{Ca}(\text{Mg}, \text{Fe})[\text{Si}_2\text{O}_6]$

Orthoklas  $\text{K}[\text{AlSi}_3\text{O}_8]$

Soda  $\text{Na}_2[\text{CO}_3] \cdot 10\text{H}_2\text{O}$

Arsenopyrit  $\text{FeAsS}$

Azurit  $\text{Cu}_3[(\text{CO}_3)_2/(\text{OH})_2]$

Biotit  $\text{K}(\text{Mg}, \text{Fe}^{2+}, \text{Mn}^{2+})_3[(\text{OH}, \text{F})_2/\text{Al}, \text{Fe}^{3+}, \text{Ti}^{3+}]\text{Si}_3\text{O}_{10}]$

Borax  $\text{Na}_2[\text{B}_4\text{O}_5/(\text{OH})_4] \cdot 8\text{H}_2\text{O}$

Chalkosin  $\text{Cu}_2\text{S}$

Epidot  $\text{Ca}_2(\text{Fe}, \text{Al})_3[(\text{Si}_2\text{O}_7)/(\text{SiO}_4)_3/(\text{OH})_2]$

Hornblende  $\text{Ca}_2(\text{Mg}, \text{Fe}, \text{Al})_5[(\text{Al}, \text{Si})_8\text{O}_{22}/(\text{OH})_2]$

Kryolith  $\text{Na}_3\text{AlF}_6$

Malachit  $\text{Cu}_2 [(\text{CO}_3)/(\text{OH})_2]$

Muskovit  $\text{KAl}_2[\text{Si}_3\text{AlO}_{10}/(\text{OH}, \text{F})_2]$

Phlogopit  $\text{KMg}_3[\text{Si}_3\text{AlO}_{10}/(\text{F}, \text{OH})_2]$

Realgar  $\text{AsS}$

Sanidin  $(\text{K}, \text{Na})[\text{Al}_3\text{SiO}_8]$

Staurolith  $(\text{Fe}^{2+}, \text{Mg}, \text{Zn})_2\text{Al}_9[(\text{Si}, \text{Al})_4/\text{O}_{22}/(\text{OH})_2]$

Titanit  $\text{CaTi}[\text{O}/\text{SiO}_4]$

**Exemplos de minerais 2/m:** (classe cristalográfica mais frequente)

Gipso  $\text{Ca}[\text{SO}_4] \cdot 2\text{H}_2\text{O}$

Diopsídio  $\text{Ca}(\text{Mg}, \text{Fe})[\text{Si}_2\text{O}_6]$

Ortoclásio  $\text{K}[\text{AlSi}_3\text{O}_8]$

Sodalita  $\text{Na}_2[\text{CO}_3] \cdot 10\text{H}_2\text{O}$

Arsenopirita  $\text{FeAsS}$

Azurita  $\text{Cu}_3[(\text{CO}_3)_2/(\text{OH})_2]$

Biotita  $\text{K}(\text{Mg}, \text{Fe}^{2+}, \text{Mn}^{2+})_3[(\text{OH}, \text{F})_2/\text{Al}, \text{Fe}^{3+}, \text{Ti}^{3+}]\text{Si}_3\text{O}_{10}]$

Bórax  $\text{Na}_2[\text{B}_4\text{O}_5/(\text{OH})_4] \cdot 8\text{H}_2\text{O}$

Calcocita  $\text{Cu}_2\text{S}$

Epidoto  $\text{Ca}_2(\text{Fe}, \text{Al})_3[(\text{Si}_2\text{O}_7)/(\text{SiO}_4)_3/(\text{OH})_2]$

Hornblenda  $\text{Ca}_2(\text{Mg}, \text{Fe}, \text{Al})_5[(\text{Al}, \text{Si})_8\text{O}_{22}/(\text{OH})_2]$

Criolita  $\text{Na}_3\text{AlF}_6$

Malaquita  $\text{Cu}_2 [(\text{CO}_3)/(\text{OH})_2]$

Muscovita  $\text{KAl}_2[\text{Si}_3\text{AlO}_{10}/(\text{OH}, \text{F})_2]$

Flogopita  $\text{KMg}_3[\text{Si}_3\text{AlO}_{10}/(\text{F}, \text{OH})_2]$

Realgar  $\text{AsS}$

Sanidina  $(\text{K}, \text{Na})[\text{Al}_3\text{SiO}_8]$

Estaurolita  $(\text{Fe}^{2+}, \text{Mg}, \text{Zn})_2\text{Al}_9[(\text{Si}, \text{Al})_4\text{O}_{22}/(\text{OH})_2]$

Titanita  $\text{CaTi}[\text{O}/\text{SiO}_4]$

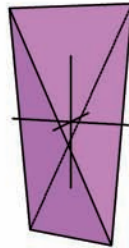
# Orthorhombisches Kristallsystem

Rhombische Hemiedrie

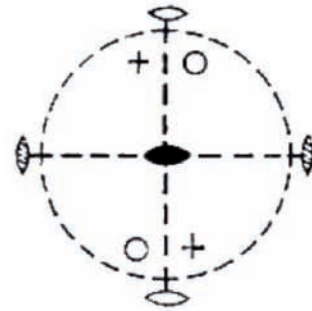
Symbol:  $222$  oder  $D_2$

Rhombisch-disphenoidische Klasse

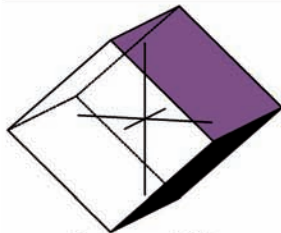
Allgemeine Form:  
 $\{hkl\}$



Rhombische Disphenoide

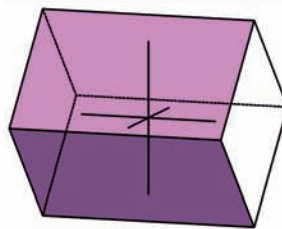


$\{0kl\}$



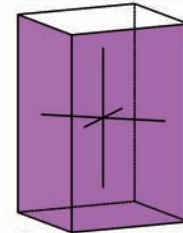
Prismen I.St.

$\{h0l\}$



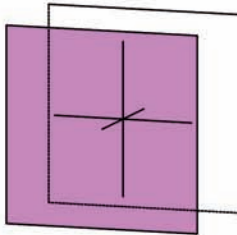
Prismen II.St.

$\{hk0\}$



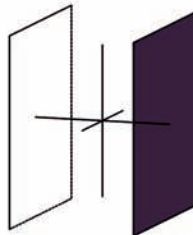
Prismen III.St.

$\{100\}$



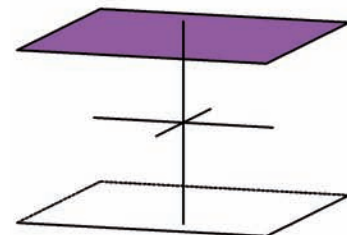
I. Pinakoid

$\{010\}$



II. Pinakoid

$\{001\}$



III. Pinakoid

## Mineral-Bsp.:

Epsomit (Bittersalz)  $MgSO_4 \cdot 7H_2O$

Zinkvitriol/Goslarit  $ZnSO_4 \cdot 7H_2O$

Austinit  $CaZn[(AsO_4)/(OH)]$

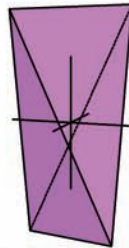
Arsenoclasite  $Mn_5[(AsO_4)_2/(OH)_4]$

# Sistema Cristalino Ortorrômbico

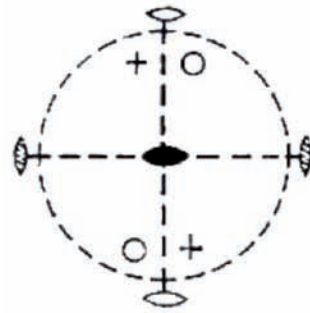
Hemiedro rômboico  
Classe Disfenóide rômboico

Símbolo:  $222$  ou  $D_2$

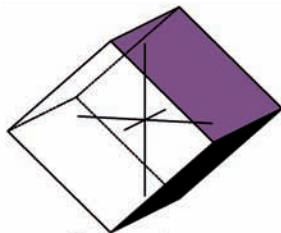
Forma geral:  
 $\{hkl\}$



Disfenóide rômboico

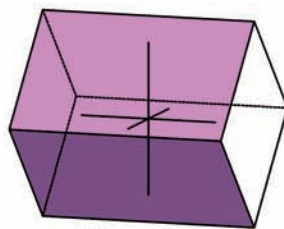


$\{0kl\}$



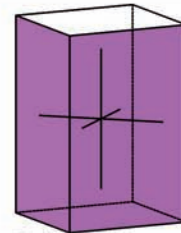
Prisma I.pos.

$\{h0l\}$



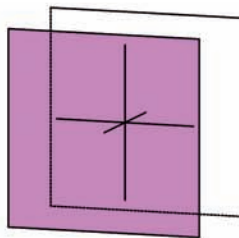
Prisma II.pos.

$\{hk0\}$



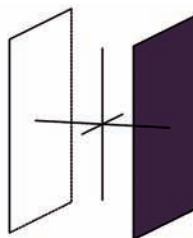
Prisma III.pos.

$\{100\}$



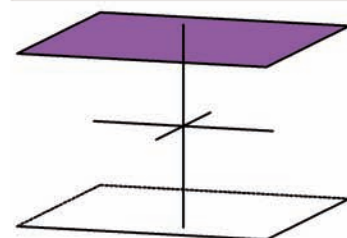
I. Pinacóide

$\{010\}$



II. Pinacóide

$\{001\}$



III. Pinacóide

## Exemplos de minerais:

Epsomita  $MgSO_4 \cdot 7H_2O$

Goslarita  $ZnSO_4 \cdot 7H_2O$

Austinita  $CaZn[(AsO_4)/(OH)]$

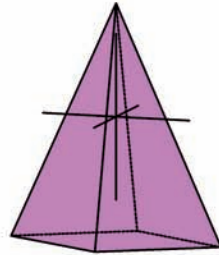
Arsenoclosita  $Mn_5[(AsO_4)_2/(OH)_4]$

# Orthorhombisches Kristallsystem

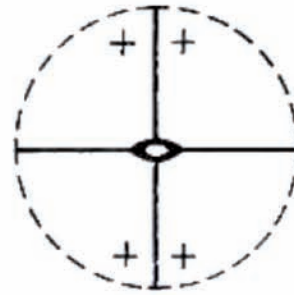
Rhombische Hemimorphie  
Rhombisch-pyramidale Klasse

Symbol:  $mm2$  oder  $C_{2v}$   
(mm)

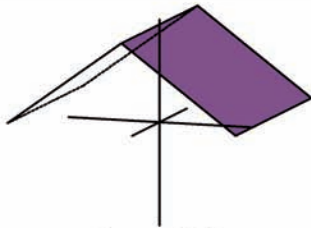
Allgemeine Form:  
 $\{hkl\}$



Rhombische Pyramiden

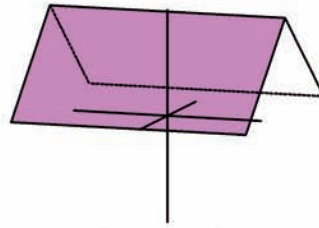


$\{0kl\}$



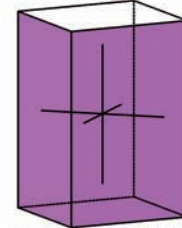
Domen I.St.

$\{h0l\}$



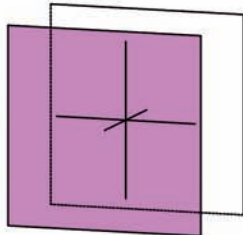
Domen II.St.

$\{hk0\}$



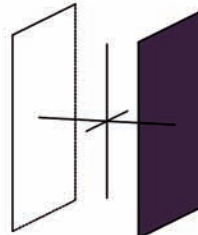
Prismen III.St.

$\{100\}$



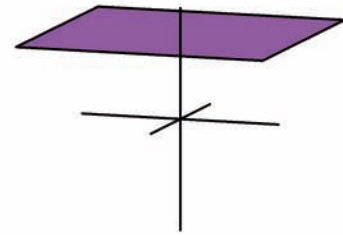
I. Pinakoid

$\{010\}$



II. Pinakoid

$\{001\}$



III. Pedion

## Mineral-Bsp.:

Hemimorphit/Kieselzinkerz  $Zn_4[(OH)_2/Si_2O_7]*H_2O$

Struvit  $MgNH_4[PO_4]*6H_2O$

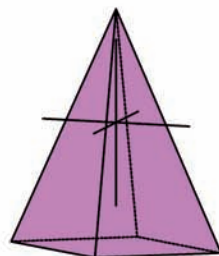
Prehinit  $Ca_2Al_2[Si_3O_{10}/(OH)_2]$

# Sistema Cristalino Ortorrômbico

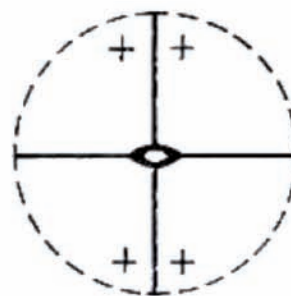
Hemimorfia rômbrica  
Classe piramidal rômbrica

Símbolo:  $mm2$  ou  $C_{2v}$   
(mm)

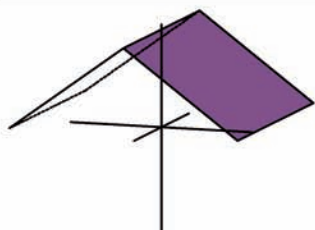
Forma geral:  
 $\{hkl\}$



Pirâmide rômbrica

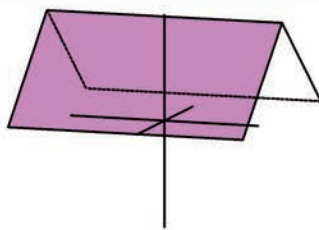


$\{0kl\}$



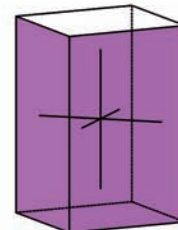
Domo I.pos.

$\{h0l\}$



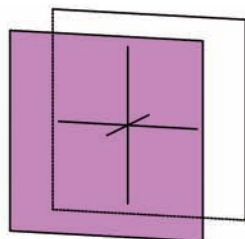
Domo II.pos.

$\{hk0\}$



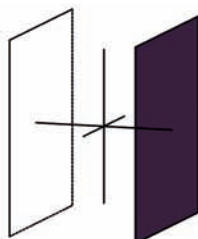
Prisma III.pos.

$\{100\}$



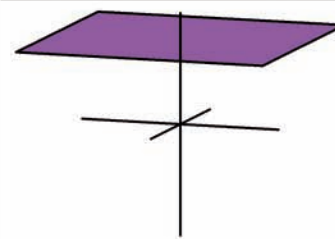
I. Pinacóide

$\{010\}$



II. Pinacóide

$\{001\}$



III. Pédio

## Exemplos de minerais:

Hemimorfita/Calamina  $Zn_4[(OH)_2/Si_2O_7]*H_2O$

Estrovita  $MgNH_4[PO_4]*6H_2O$

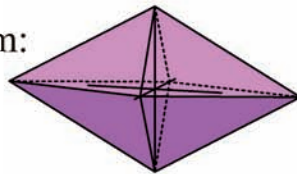
Prehnita  $Ca_2Al_2[Si_3O_{10}/(OH)_2]$

# Orthorhombisches Kristallsystem

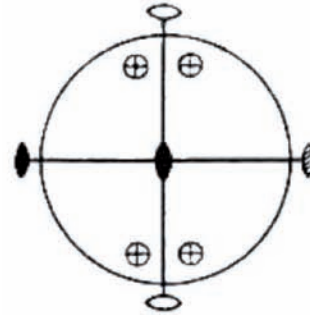
Rhombische Holoedrie  
Rhombisch-dipyramidale Klasse

Symbol:  $2/m\ 2/m\ 2/m$   
(mmm) oder  $D_{2h}$

Allgemeine Form:  
{hkl}



Rhombische Dipyramiden



{0kl}	{h0l}	{hk0}
<p><u>Prismen I.St.</u></p>	<p><u>Prismen II.St.</u></p>	<p><u>Prismen III.St.</u></p>
{100}	{010}	{001}
<p><u>I. Pinakoid</u></p>	<p><u>II. Pinakoid</u></p>	<p><u>III. Pinakoid</u></p>

## Mineral-Bsp.:

Schwefel S

Aragonit  $\text{Ca}[\text{CO}_3]$

Anhydrit  $\text{Ca}[\text{SO}_4]$

Baryt  $\text{Ba}[\text{SO}_4]$

Topas  $\text{Al}_2[\text{F}_2/\text{SiO}_4]$

Olivin  $(\text{Mg,Fe})_2[\text{SiO}_4]$

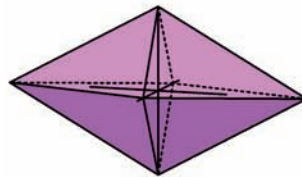


# Sistema Cristalino Ortorrômbico

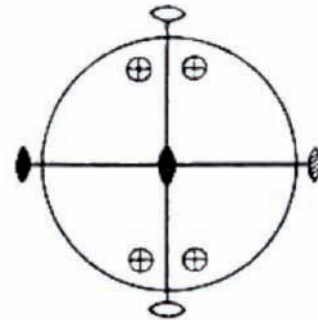
Holoedro rômboico  
Classe dipiramidal ortorrômbica

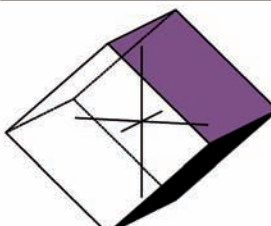
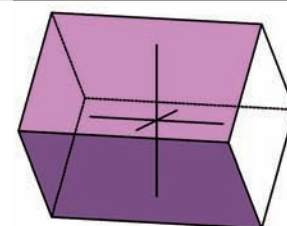
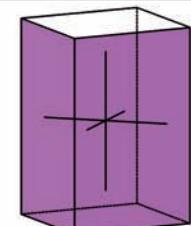
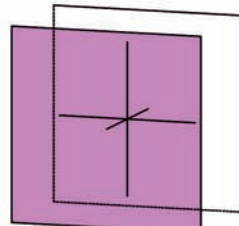
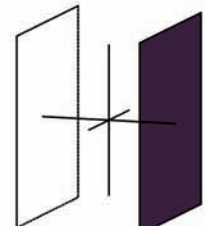
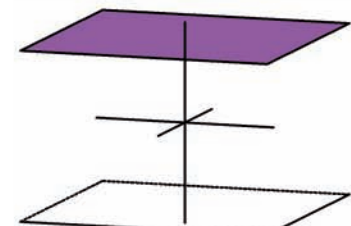
Símbolo:  $2/m\ 2/m\ 2/m$   
(mmm) ou  $D_{2h}$

Forma geral:  
{hkl}



Dipiramidal ortorrômbica



{0kl}	{h0l}	{hk0}
		
<u>Prisma I.pos.</u>	<u>Prisma II.pos.</u>	<u>Prisma III.pos.</u>
{100}	{010}	{001}
		
<u>I. Pinacóide</u>	<u>II. Pinacóide</u>	<u>III. Pinacóide</u>

## Exemplos de minerais:

Enxofre S

Barita  $Ba[SO_4]$

Aragonita  $Ca[CO_3]$

Topázio  $Al_2[F_2/SiO_4]$

Anidrita  $Ca[SO_4]$

Olivina  $(Mg,Fe)_2[SiO_4]$

# Trigonales Kristallsystem

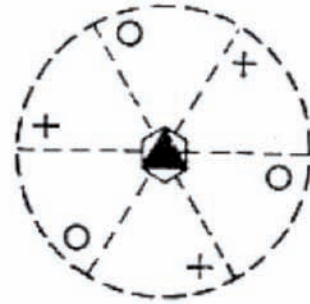
Hexagonal-rhomboedrische  
Tetartoedrie  
Rhomboedrische Klasse

Symbol:  $\bar{3}$  oder  $C_{3i}$

Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



**Rhomboeder III.St.**



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<b>Rhomboeder I.St.</b>	<b>Rhomboeder II.St.</b>	<b>Hex. Prismen II.St.</b>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<b>Hex. Prisma I.St.</b>	<b>Hex. Prisma II.St.</b>	<b>Basispinakoid</b>

## Mineral-Bsp.:

Dolomit  $\text{CaMg}[(\text{CO}_3)_2]$

Diopas  $\text{Cu}[\text{Si}_6\text{O}_{18}]$

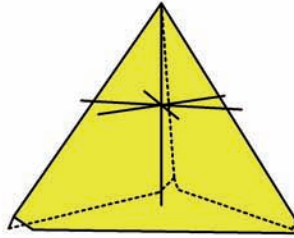
Phenakit  $\text{Be}_2\text{SiO}_4$

# Sistema Cristalino Trigonal

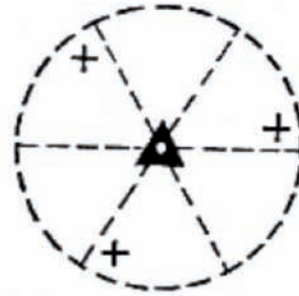
Hemimorfo Trigonal Tetraedro  
Classe Trigonal-Piramidal

Símbolo: 3 ou  $C_3$

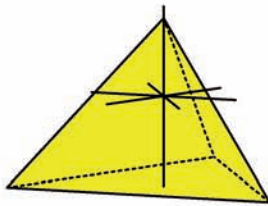
Forma geral:  
 $\{hkl\} / \{hkil\}$



**Piramide Trigonal III. pos.**

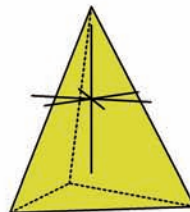


$\{h0l\} / \{h0\bar{h}l\}$



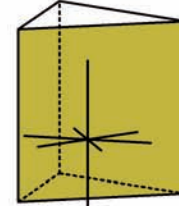
**Piramide Trigonal I.pos.**

$\{hhl\} / \{hh2\bar{h}l\}$



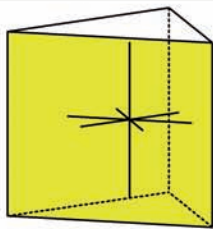
**Piramide Trigonal II.pos.**

$\{hk0\} / \{hki0\}$



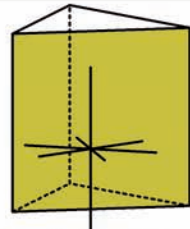
**Prisma Trigonal III.pos.**

$\{100\} / \{10\bar{1}0\}$



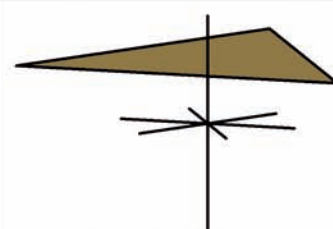
**Prisma Trigonal I.pos.**

$\{110\} / \{112\bar{0}\}$



**Prisma Trigonal II.pos.**

$\{001\} / \{0001\}$



**Base de Pedium**

## Exemplos de minerais:

Susannita  $Pb_4[(SO_4)/(CO_3)_2/(OH)_2]$

Carlinita  $TlS_2$

Beligermanita (Modificação de baixa temperatura)

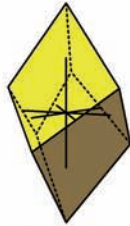
$Pb_5Ge_3O_{11}$

# Trigonales Kristallsystem

Hexagonal-trapezoedrische  
Tetartoedrie  
Trigonal-trapezoedrische Klasse

Symbol:  $32$  oder  $D_3$

Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Trig. Trapezoeder



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh2\bar{h}l\}$	$\{hk0\} / \{hki0\}$
<u>Rhomboeder I.St.</u>	<u>Trig. Dipyramiden II.St.</u>	<u>Ditrig. Prismen II.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{112\bar{0}\}$	$\{001\} / \{0001\}$
<u>Hex. Prisma I.St.</u>	<u>Trig. Prismen II.St.</u>	<u>Basispinakoid</u>

## Mineral-Bsp.:

Zinnober/Cinnabarit  $HgS$

Tiefquarz  $SiO_2$

Selen  $Se$

Tellur  $Te$

# Sistema Cristalino Trigonal

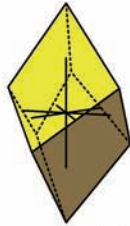
Hexagonal trapezoedral

Tetardoedro

Classe trigonal trapezoédrica

Símbolo:  $32$  ou  $D_3$

Forma geral:  
 $\{hkl\} / \{hkil\}$



Trapezoedro trigonal



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Romboedro I.pos.</u>	<u>Dipiramide Trigonal II.pos.</u>	<u>Prisma Ditrigonal II.pos.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Prisma Hexagonal I.pos.</u>	<u>Prisma Trigonal II.pos.</u>	<u>Base pinacoidal</u>

## Exemplos de minerais:

Cinábrio HgS

Quartzo baixa Temperatura SiO<sub>2</sub>

Selênio Se

Telúrio Te

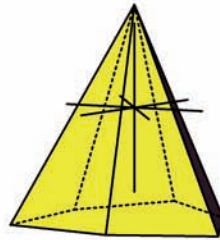
# Trigonales Kristallsystem

Hemimorphie d. rhomboedrischen  
Hemiedrie

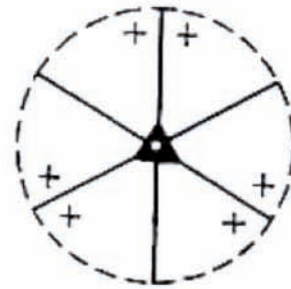
Symbol:  $3m$  oder  $C_{3v}$

Ditrigonal-pyramidale Klasse

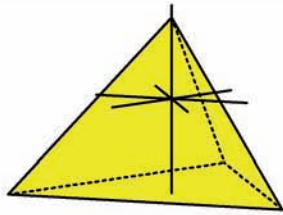
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Ditrig. Pyramiden I.St.

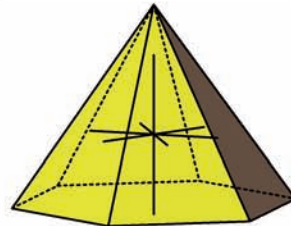


$\{h0l\} / \{h0\bar{h}l\}$



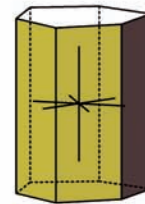
Trig. Pyramiden I.St.

$\{hhl\} / \{hh\bar{2}hl\}$



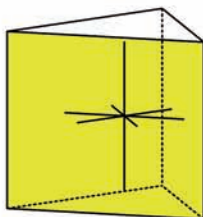
Hex. Pyramiden II.St.

$\{hk0\} / \{hki0\}$



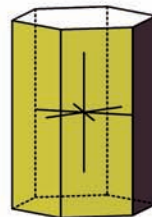
Ditrig. Prismen II.St.

$\{100\} / \{10\bar{1}0\}$



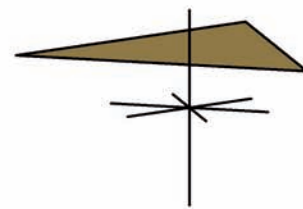
Trig. Prismen I.St.

$\{110\} / \{11\bar{2}0\}$



Hex. Prisma II.St.

$\{001\} / \{0001\}$



Basispedien

## Mineral-Bsp.:

Turmalin  $(X)(Y_3)(Z_6)[Si_6O_{18}/(BO_3)_3/(V_3)(W)]$

Proustite  $Ag_3AsS_3$

Lithiumniobat  $LiNbO_3$

# Sistema Cristalino Trigonal

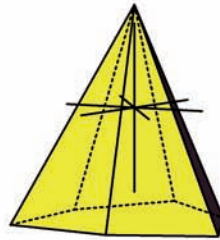
Hemimorfia Romboédrica

Hemiedria

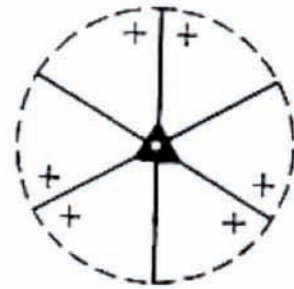
Classe Ditrigonal-Piramidal

Símbolo:  $3m$  ou  $C_{3v}$

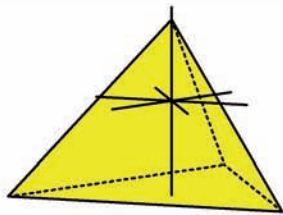
Forma geral:  
 $\{hkl\} / \{hkil\}$



Ditrigonal Piramidal I.pos.

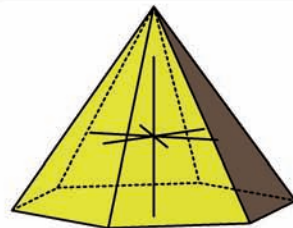


$\{h0l\} / \{h0\bar{h}l\}$



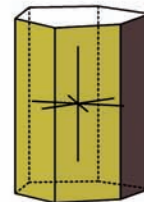
Piramide Trigonal I.pos.

$\{hhl\} / \{hh\bar{2}hl\}$



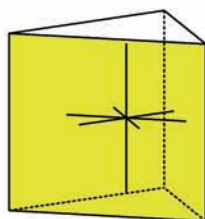
Piramide Hexagonal II.pos.

$\{hk0\} / \{hki0\}$



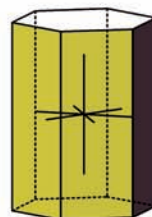
Prisma Ditrigonal II.pos.

$\{100\} / \{10\bar{1}0\}$



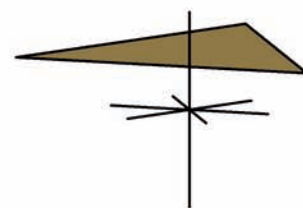
Prisma Trigonal I.pos.

$\{110\} / \{11\bar{2}0\}$



Prisma Hexagonal II.pos.

$\{001\} / \{0001\}$



Base de Pedium

## Exemplos de minerais e compostos inorgânicos:

Turmalina  $(X)(Y_3)(Z_6)[Si_6O_{18}/(BO_3)_3/(V_3)(W)]$

Proustita  $Ag_3AsS_3$

Niobato de lítico  $LiNbO_3$

# Trigonales Kristallsystem

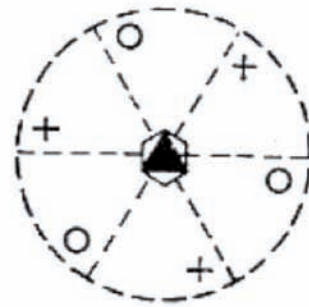
Hexagonal-rhomboedrische  
Tetartoedrie  
Rhomboedrische Klasse

Symbol:  $\bar{3}$  oder  $C_{3i}$

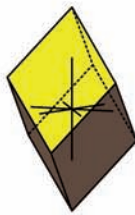
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



**Rhomboeder III.St.**



$\{h0l\} / \{h0\bar{h}l\}$



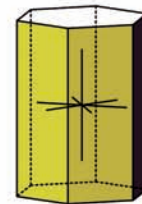
**Rhomboeder I.St.**

$\{hhl\} / \{hh\bar{2}hl\}$



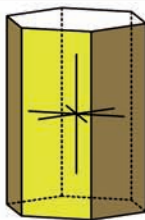
**Rhomboeder II.St.**

$\{hk0\} / \{hki0\}$



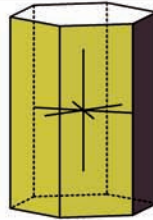
**Hex. Prismen II.St.**

$\{100\} / \{10\bar{1}0\}$



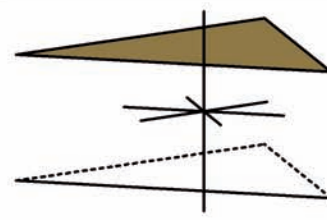
**Hex. Prisma I.St.**

$\{110\} / \{11\bar{2}0\}$



**Hex. Prisma II.St.**

$\{001\} / \{0001\}$



**Basispinakoid**

## Mineral-Bsp.:

Dolomit  $\text{CaMg}[(\text{CO}_3)_2]$

Diopas  $\text{Cu}[\text{Si}_6\text{O}_{18}]$

Phenakit  $\text{Be}_2\text{SiO}_4$

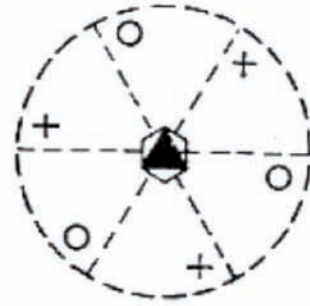
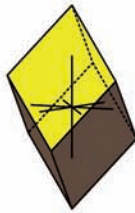
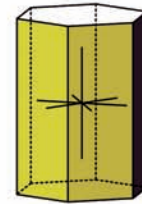
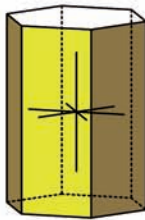
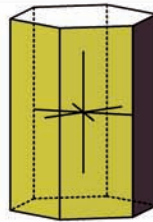
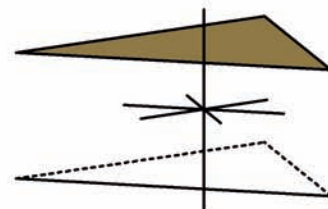


# Sistema Cristalino Trigonal

Hexagonal-romboédrica

Tetartoédrica

Classe Romboédrica

Símbolo:  $\bar{3}$  ou  $C_{3i}$ Forma geral:  
 $\{hkl\} / \{hkil\}$ **Romboedro III.pos.** $\{h0l\} / \{h0\bar{h}l\}$ **Romboedro I.pos.** $\{hhl\} / \{hh\bar{2}hl\}$ **Romboedro II.pos.** $\{hk0\} / \{hki0\}$ **Prisma Hexagonal II.pos.** $\{100\} / \{10\bar{1}0\}$ **Prisma Hexagonal I.pos.** $\{110\} / \{11\bar{2}0\}$ **Prisma Hexagonal II.pos.** $\{001\} / \{0001\}$ **Bases de Pinacóide**

## Exemplos de minerais:

Dolomita  $\text{CaMg}[(\text{CO}_3)_2]$ Dioptásio  $\text{Cu}[\text{Si}_6\text{O}_{18}]$ Fenacita  $\text{Be}_2\text{SiO}_4$

# Trigonales Kristallsystem

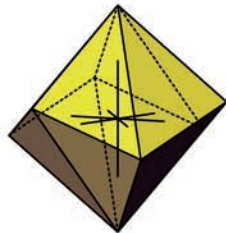
Hexagonal-rhomboedrische

Hemiedrie

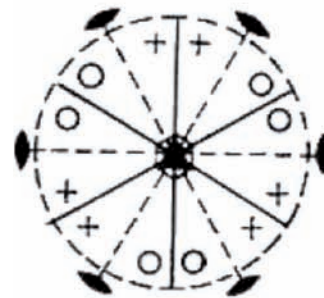
Ditrigonal-skalenoedrische Klasse

Symbol:  $\bar{3} 2/m$  oder  $D_{3d}$   
( $\bar{3}m$ )

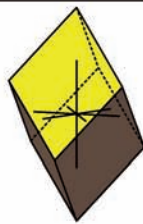
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Ditrig. Skalenoeder

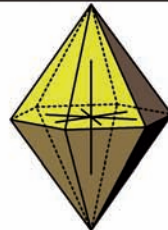


$\{h0l\} / \{h0\bar{h}l\}$



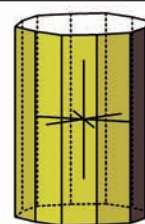
Rhomboeder I.St.

$\{hhl\} / \{hh\bar{2}hl\}$



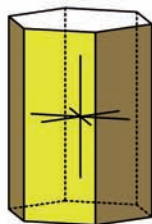
Hex. Dipyramiden II.St.

$\{hk0\} / \{hki0\}$



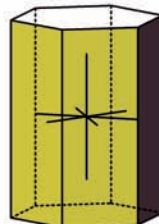
Dihex. Prismen

$\{100\} / \{10\bar{1}0\}$



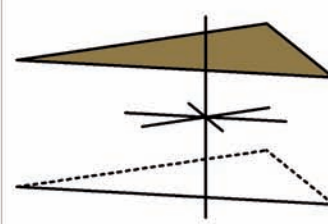
Hex. Prisma I.St.

$\{110\} / \{11\bar{2}0\}$



Hex. Prisma II.St.

$\{001\} / \{0001\}$



Basispinakoid

## Mineral-Bsp.:

Calcit  $\text{Ca}[\text{CO}_3]$

Korund  $\text{Al}_2\text{O}_3$

Hämatit  $\text{Fe}_2\text{O}_3$

Siderit  $\text{Fe}[\text{CO}_3]$

Rhodochrosit  $\text{Mn}[\text{CO}_3]$

# Sistema Cristalino Trigonal

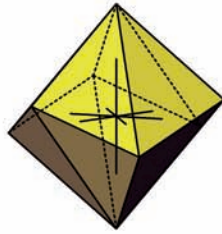
Hexagonal-romboédrica

Hemiedro

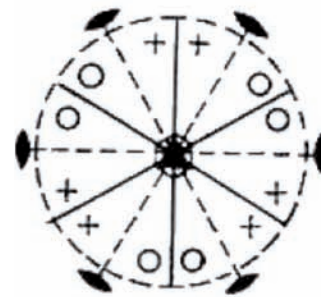
Classe ditrigonal escalonétrica

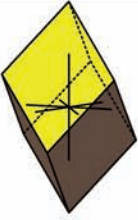
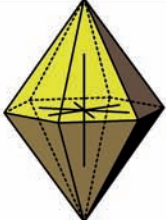
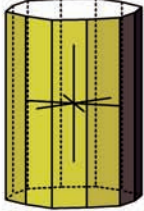
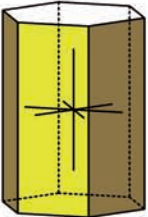
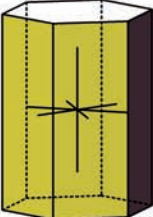
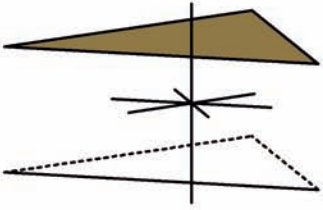
Símbolo:  $\bar{3} 2/m$  ou  $D_{3d}$   
( $\bar{3}m$ )

Forma geral:  
 $\{hkl\} / \{hkil\}$



Escalenoedro ditrigonal



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
Romboedral I.pos.	Dipiramide Hexagonal II.pos.	Prisma Dihexagonal
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
Prisma Hexagonal I.pos.	Prisma Hexagonal II.pos.	<u>Base Pinacoidal</u>

## Exemplos de minerais:

Calcita  $\text{Ca}[\text{CO}_3]$  Siderita  $\text{Fe}[\text{CO}_3]$

Coríndon  $\text{Al}_2\text{O}_3$  Rodocrosita  $\text{Mn}[\text{CO}_3]$

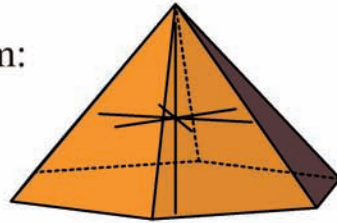
Hematita  $\text{Fe}_2\text{O}_3$

# Hexagonales Kristallsystem

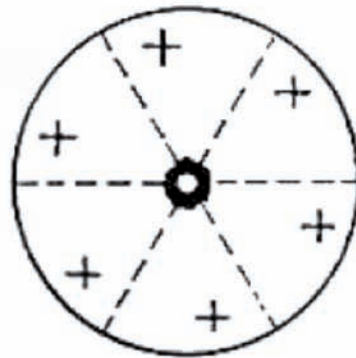
Hexagonale Tetartoedrie  
Hexagonal-pyramidale Klasse

Symbol: 6 oder  $C_6$

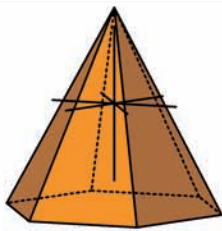
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



**Hexagonale Pyramiden III. St.**

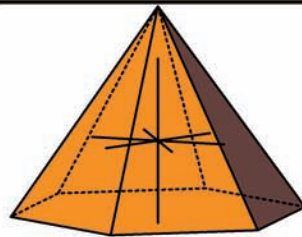


$\{h0l\} / \{h0\bar{h}l\}$



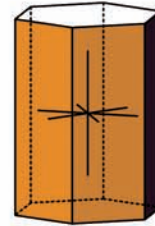
**Hex.Pyramiden I.St.**

$\{hhl\} / \{hh\bar{2}hl\}$



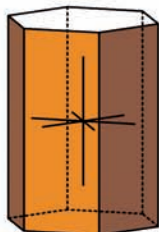
**Hex. Pyramiden II.St.**

$\{hk0\} / \{hki0\}$



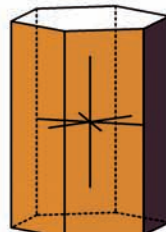
**Hex.Prismen III.St.**

$\{100\} / \{10\bar{1}0\}$



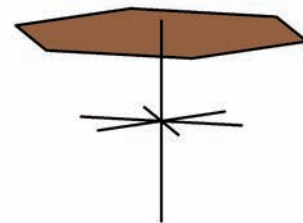
**Hex.Prisma I.St.**

$\{110\} / \{11\bar{2}0\}$



**Hex.Prisma II.St.**

$\{001\} / \{0001\}$



**Basispedien**

## Mineral-Bsp.:

Lithiumiodat  $\alpha$ -LiIO<sub>3</sub>

Nephelin (Na,K)[AlSiO<sub>4</sub>]

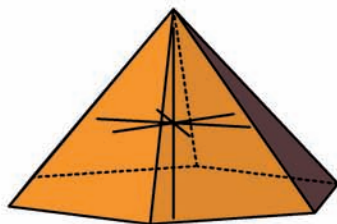
Thaumasit Ca<sub>3</sub>[SO<sub>4</sub>/CO<sub>3</sub>/Si(OH)<sub>6</sub>]\*12H<sub>2</sub>O

# Sistema Cristalino Hexagonal

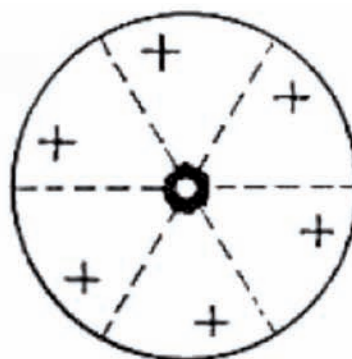
Tetraedro Hexagonal  
Classe piramidal hexagonal

Símbolo: 6 ou  $C_6$

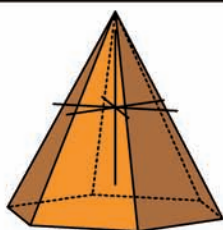
Forma geral:  
 $\{hkl\} / \{hkil\}$



**Pirâmide hexagonal III. pos.**

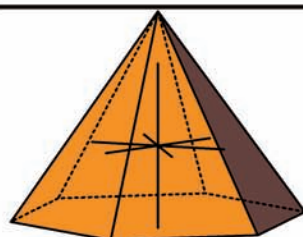


$\{h0l\} / \{h0\bar{h}l\}$



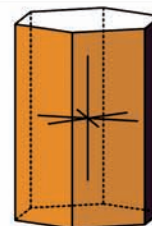
**Pirâmide hexagonal I. pos.**

$\{hhl\} / \{hh\bar{2}hl\}$



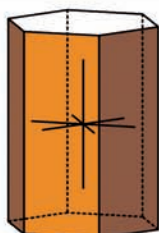
**Pirâmide hexagonal II. pos.**

$\{hk0\} / \{hki0\}$



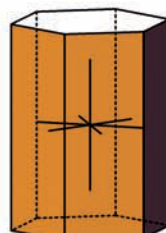
**Prisma Hexagonal III. pos.**

$\{100\} / \{10\bar{1}0\}$



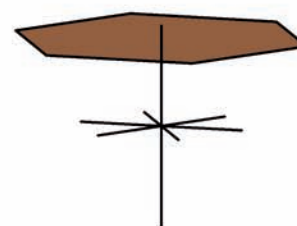
**Prisma Hexagonal I. pos.**

$\{110\} / \{11\bar{2}0\}$



**Prisma Hexagonal II. pos.**

$\{001\} / \{0001\}$



**Base pedal**

## Exemplos de minerais e compostos inorgânicos:

Iodato de lítico  $\alpha\text{-LiIO}_3$

Nefelina  $(\text{Na},\text{K})[\text{AlSiO}_4]$

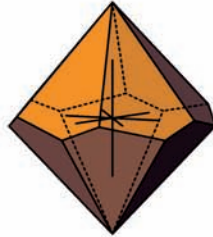
Thaumasita  $\text{Ca}_3[\text{SO}_4/\text{CO}_3/\text{Si}(\text{OH})_6] \cdot 12\text{H}_2\text{O}$

# Hexagonales Kristallsystem

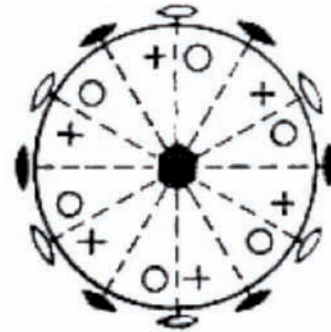
Hexagonal-trapezoedrische  
Hemiedrie  
Hexagonal-trapezoedrische Klasse

Symbol:  $622$  oder  $D_6$   
(6 2)

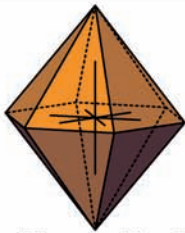
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Hexagonaler Trapezoeder

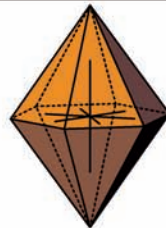


$\{h0l\} / \{h0\bar{h}l\}$



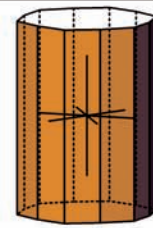
Hex.Dipyramiden I.St.

$\{hhl\} / \{hh\bar{2}hl\}$



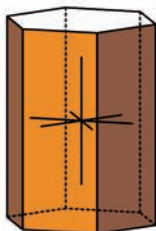
Hex. Dipyramiden II.St.

$\{hk0\} / \{hki0\}$



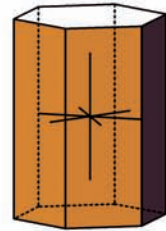
Dihex. Prismen

$\{100\} / \{10\bar{1}0\}$



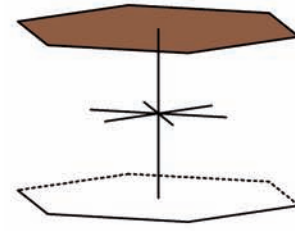
Hex.Prisma I.St.

$\{110\} / \{11\bar{2}0\}$



Hex. Prisma II.St.

$\{001\} / \{0001\}$



Basispinakoid

## Mineral-Bsp.:

Hochquarz  $\text{SiO}_2$

Virgilit  $\text{LiAlSi}_2\text{O}_6$

Rhabdophan  $(\text{Ce},\text{La})\text{PO}_4 \cdot \text{H}_2\text{O}$

# Sistema Cristalino Hexagonal

Trapezoedro Hexagonal

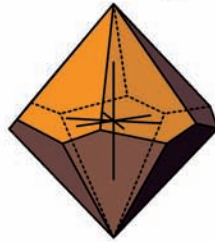
Hemiedria

Classe Trapezoédrica hexagonal

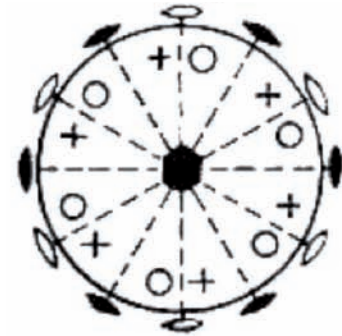
Símbolo:  $622$  ou  $D_6$

(6 2)

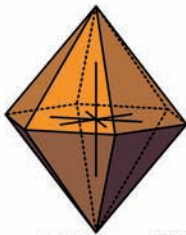
Forma geral:  
 $\{hkl\} / \{hkil\}$



Trapezoedro Hexagonal

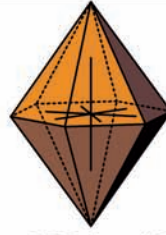


$\{h0l\} / \{h0\bar{h}l\}$



Hexagonal Bipiramidal I.pos.

$\{hhl\} / \{hh\bar{2}hl\}$



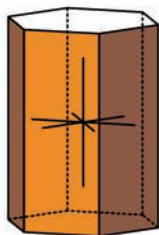
Hexagonal Bipiramidal II.pos.

$\{hk0\} / \{hki0\}$



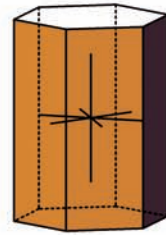
Prisma Dihexagonal

$\{100\} / \{10\bar{1}0\}$



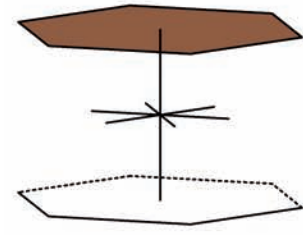
Prisma Hexagonal I.pos.

$\{110\} / \{11\bar{2}0\}$



Prisma Hexagonal II.pos.

$\{001\} / \{0001\}$



Base Pinacoidal

## Exemplos de minerais:

Quartzo beta  $\text{SiO}_2$

Virgilita  $\text{LiAlSi}_2\text{O}_6$

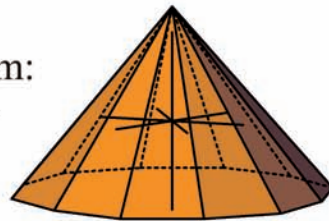
Rhabdofana  $(\text{Ce},\text{La})\text{PO}_4 \cdot \text{H}_2\text{O}$

# Hexagonales Kristallsystem

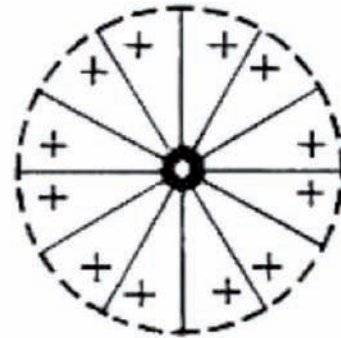
Hexagonale Hemimorphie  
Dihexagonal-pyramidale Klasse

Symbol:  $6mm$  oder  $C_{6v}$

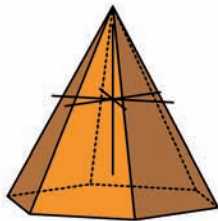
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



**Dihexagonale Pyramide**

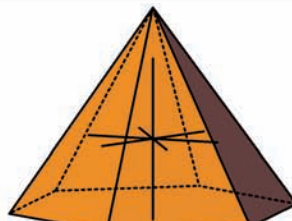


$\{h0l\} / \{h0\bar{h}l\}$



**Hex.Pyramiden I.St.**

$\{hhl\} / \{hh2\bar{h}l\}$



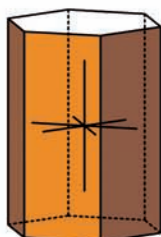
**Hex. Pyramiden II.St.**

$\{hk0\} / \{hki0\}$



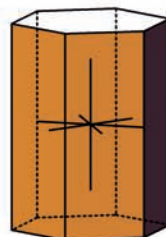
**Dihex.Prismen**

$\{100\} / \{10\bar{1}0\}$



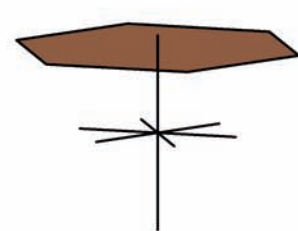
**Hex.Prisma I.St.**

$\{110\} / \{11\bar{2}0\}$



**Hex.Prisma II.St.**

$\{001\} / \{0001\}$



**Basispedien**

## Mineral-Bsp.:

Wurzit ZnS

Zinkit ZnO

Greenockit CdS

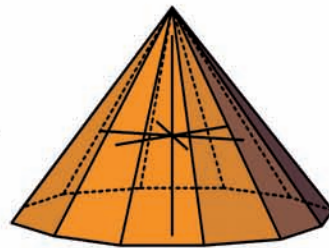


# Sistema Cristalino Hexagonal

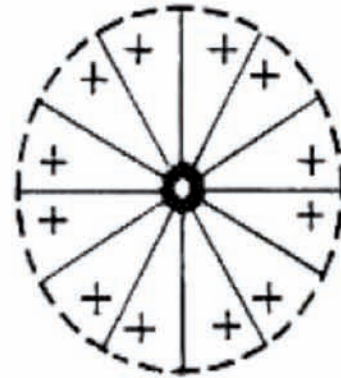
Hemimorfia Hexagonal  
Classe Dihexagonal Piramidal

Símbolo:  $6mm$  ou  $C_{6v}$

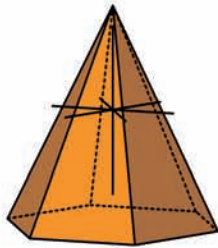
Forma geral:  
 $\{hkl\} / \{hkil\}$



**Dihexagonal Piramidal**

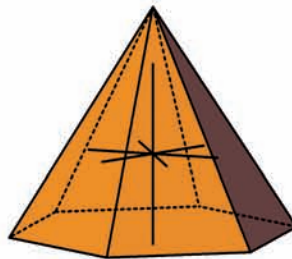


$\{h0l\} / \{h0\bar{h}l\}$



**Hexagonal Piramidal I.pos.**

$\{hhl\} / \{hh\bar{2}hl\}$



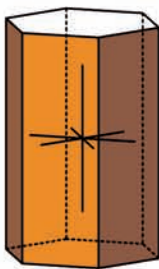
**Hexagonal Piramidal II.pos.**

$\{hk0\} / \{hki0\}$



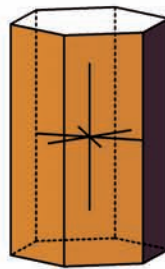
**Prisma Dihexagonal**

$\{100\} / \{10\bar{1}0\}$



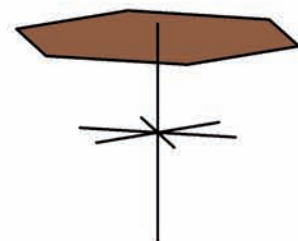
**Prisma Hexagonal I.pos.**

$\{110\} / \{11\bar{2}0\}$



**Prisma Hexagonal II.pos.**

$\{001\} / \{0001\}$



**Base de Pediem**

## Exemplos de minerais:

Wurzita ZnS

Zincita ZnO

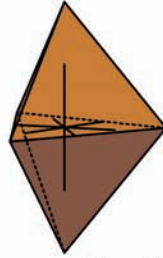
Greenoquita CdS

# Hexagonales Kristallsystem

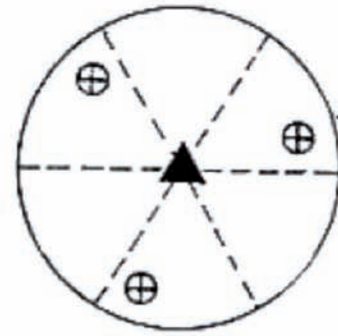
Trigonale Tetartoedrie  
Trigonal-dipyramidale Klasse

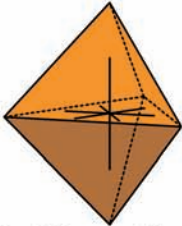
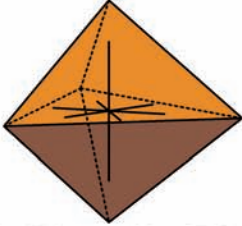
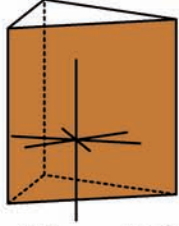
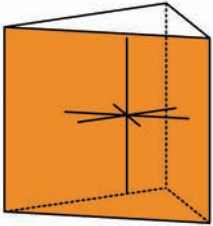
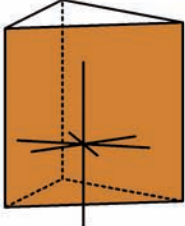
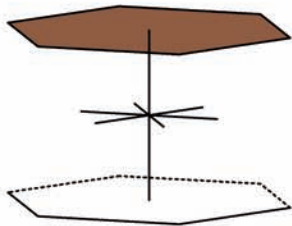
Symbol:  $\bar{6}$  oder  $C_{3h}$

Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Trig. Dipyramiden III.St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Trig. Dipyramiden I.St.</u>	<u>Trig. Dipyramiden II.St.</u>	<u>Trig. Prismen III.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Trig. Prismen I.St.</u>	<u>Trig. Prismen II.St.</u>	<u>Basispinakoid</u>

## Mineral-Bsp.:

Bleigermanat (Hochtemp. Modifikation)  $Pb_5[Ge_3O_{11}]$

Penfieldit  $Pb_2[Cl_3/OH]$

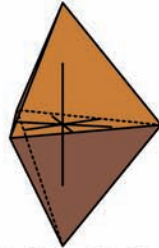
Ganomalit  $Pb_9Ca_5Mn[(Si_2O_7)_3/(SiO_4)_3]$

# Sistema Cristalino Hexagonal

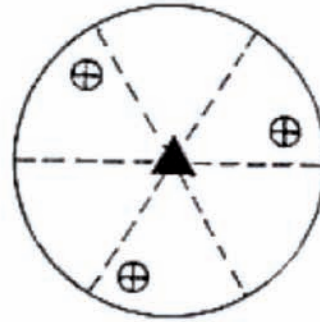
Tetraedro Trigonal  
Classe Trigonal Bipiramidal

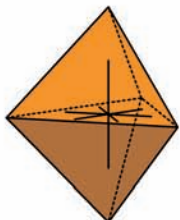
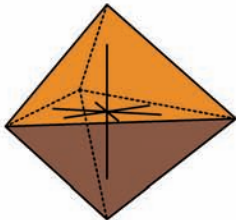
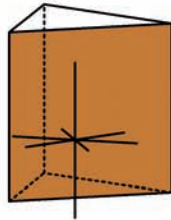
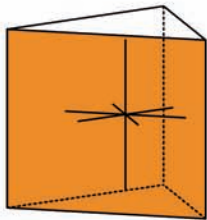
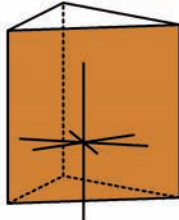
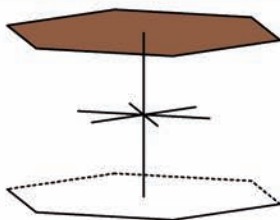
Símbolo:  $\bar{6}$  ou  $C_{3h}$

Forma geral:  
 $\{hkl\} / \{hkil\}$



**Trigonal Bipiramidal III.pos.**



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<b>Trigonal Bipiramidal I.pos.</b>	<b>Trigonal Bipiramidal II.pos.</b>	<b>Prisma Trigonal III.pos.</b>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<b>Prisma Trigonal I.pos.</b>	<b>Prisma Trigonal II.pos.</b>	<b>Base de Pinacoide</b>

## Exemplos de minerais e compostos inorgânicos:

Germanato de chumbo, modificação de baixa temperatura



Penfieldita  $Pb_2[Cl_3/OH]$

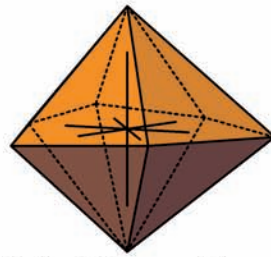
Ganomalita  $Pb_9Ca_5Mn[(Si_2O_7)_3/(SiO_4)_3]$

# Hexagonales Kristallsystem

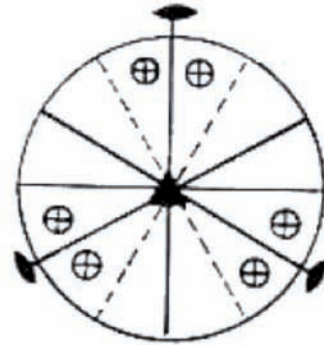
Trigonale Hemiedrie  
Ditrigonal-dipyramidale Klasse

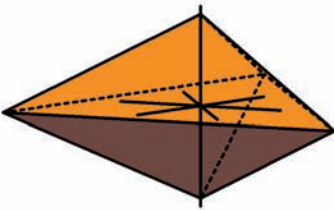
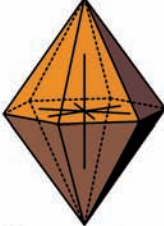
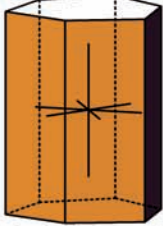
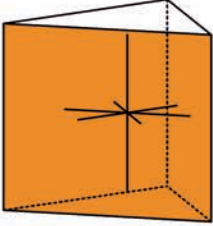
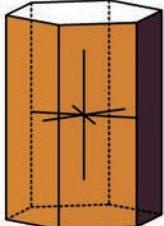
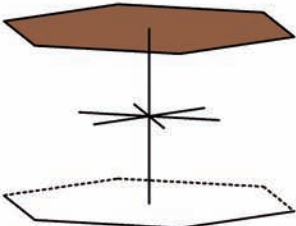
Symbol:  $\bar{6}2m$  oder  $D_{3h}$

Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



**Ditrig. Dipyramiden**



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<b>Trig. Dipyramiden I.St.</b>	<b>Hex. Dipyramiden II.St.</b>	<b>Ditrig.Prismen III.St.</b>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<b>Trig.Prismen I.St.</b>	<b>Hex.Prismen II.St.</b>	<b>Basispinakoid</b>

## Mineral-Bsp.:

Benitoit  $BaTi[Si_3O_9]$

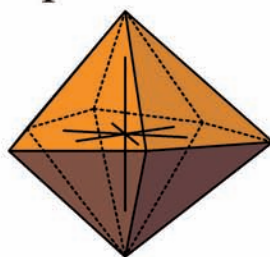
Belkovit  $Ba_3(Nb,Ti)_6(Si_2O_7)_2O_{12}$

# Sistema Cristalino Hexagonal

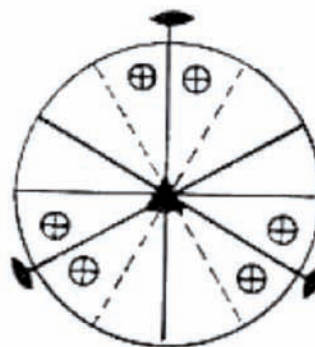
Hemiedro Trigonal  
Classe Ditrigonal Bipiramidal

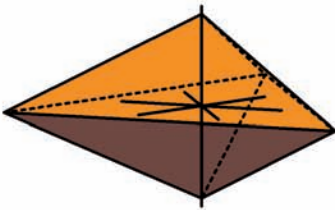
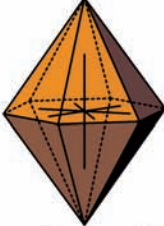
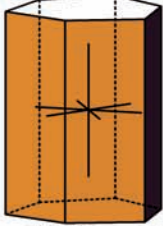
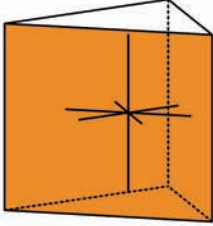
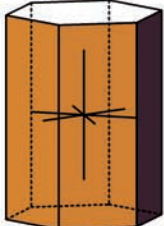
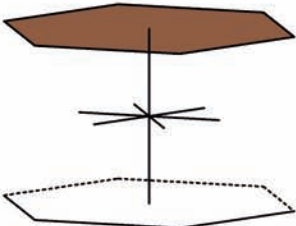
Símbolo:  $\bar{6}2m$  ou  $D_{3h}$

Forma geral:  
 $\{hkl\} / \{hkil\}$



**Bipirâmide ditrigonal**



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<b>Trigonal Bipiramidal I.pos.</b>	<b>Hexagonal Bipiramidal II.pos.</b>	<b>Prisma Ditrigonal III.pos.</b>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<b>Prisma Trigonal I.pos.</b>	<b>Prisma Hexagonal II.pos.</b>	<b>Base de Pinacóide</b>

## Exemplos de minerais:

Benitoíta  $BaTi[Si_3O_9]$

Belcovita  $Ba_3(Nb,Ti)_6(Si_2O_7)_2O_{12}$

# Hexagonales Kristallsystem

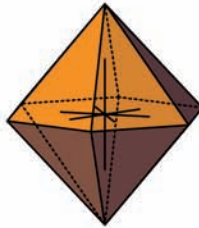
Hexagonal-pyramidale

Hemiedrie

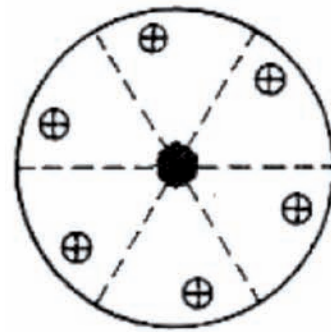
Hexagonal-dipyramidale Klasse

Symbol:  $6/m$  oder  $C_{6v}$

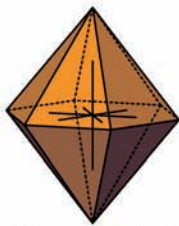
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Hexagonale Dipyramide III.St.

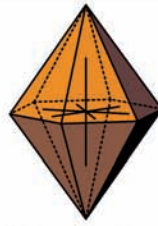


$\{h0l\} / \{h0\bar{h}l\}$



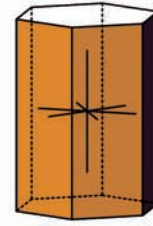
Hex.Dipyramiden I.St.

$\{hhl\} / \{hh\bar{2}hl\}$



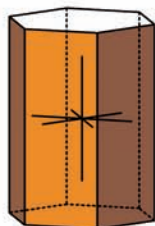
Hex. Dipyramiden II.St.

$\{hk0\} / \{hki0\}$



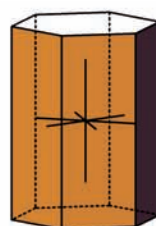
Hex.Prismen III.St.

$\{100\} / \{10\bar{1}0\}$



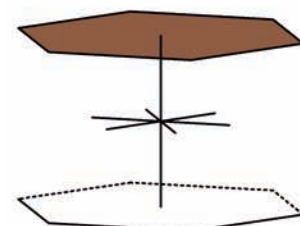
Hex.Prisma I.St.

$\{110\} / \{11\bar{2}0\}$



Hex.Prisma II.St.

$\{001\} / \{0001\}$



Basispinakoid

## Mineral-Bsp.:

Apatit  $(Ca, Ba, Pb, Sr, etc.)_5[(PO_4, CO_3)_3/(F, Cl, OH)]$

Pyromorphit  $Pb_5[(PO_4)_3/Cl]$

Vanadinit  $Pb_5[(VO_4)_3/Cl]$

# Sistema Cristalino Hexagonal

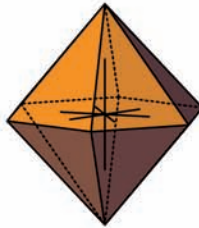
Hexagonal Piramidal

Hemiedria

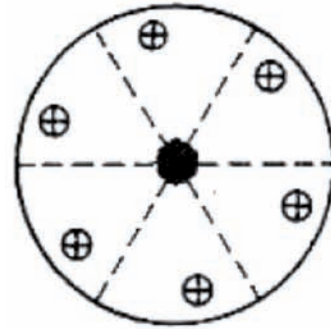
Classe Hexagonal-Dipiramidal

Símbolo:  $6/m$  ou  $C_{6v}$

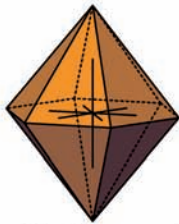
Forma geral:  
 $\{hkl\} / \{hkil\}$



Dipirâmide hexagonal III.pos.

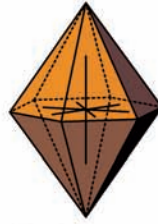


$\{h0l\} / \{h0\bar{h}l\}$



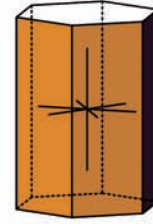
Dipiramide Hexagonal I.pos.

$\{hhl\} / \{hh\bar{2}hl\}$



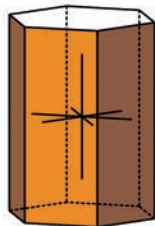
Dipiramide Hexagonal II.pos.

$\{hk0\} / \{hki0\}$



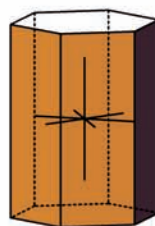
Prisma Hexagonal III.pos.

$\{100\} / \{10\bar{1}0\}$



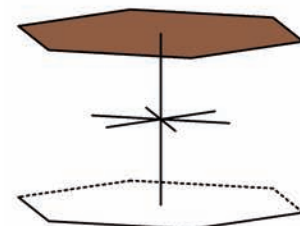
Prisma Hexagonal I.pos.

$\{110\} / \{11\bar{2}0\}$



Prisma Hexagonal II.pos.

$\{001\} / \{0001\}$



Bases Pinacóide

## Exemplos de minerais:

Apatita  $(Ca, Ba, Pb, Sr, etc.)_5[(PO_4, CO_3)_3/(F, Cl, OH)]$

Piromorfita  $Pb_5[(PO_4)_3/Cl]$

Vanadinita  $Pb_5[(VO_4)_3/Cl]$

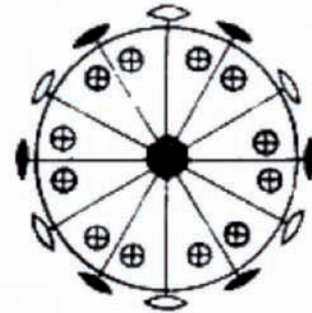
# Hexagonales Kristallsystem

Hexagonale Holoedrie      Symbol:  $6/mmm$  oder  $D_{6h}$   
 Dihexagonal-dipyramidale Klasse      ( $6/m \ 2/m \ 2/m$ )

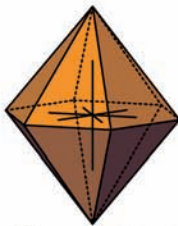
Allgemeine Form:  
 $\{hkl\} / \{hkil\}$



Dihexagonale Dipyramiden

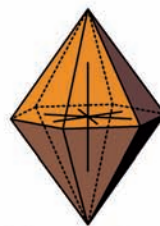


$\{h0l\} / \{h0\bar{h}l\}$



Hex.Dipyramiden I.St.

$\{hhl\} / \{hh\bar{2}hl\}$



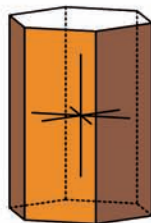
Hex.Dipyramiden II.St.

$\{hk0\} / \{hki0\}$



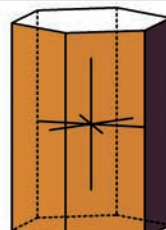
Dihex.Prismen

$\{100\} / \{10\bar{1}0\}$



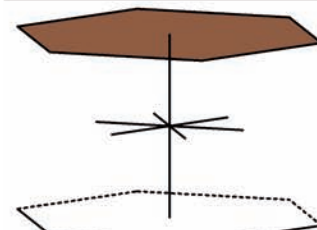
Hex.Prisma I.St.

$\{110\} / \{11\bar{2}0\}$



Hex.Prisma II.St.

$\{001\} / \{0001\}$



Basispinakoid

## Mineral-Bsp.:

Beryll  $\text{Be}_3\text{Al}_2[\text{Si}_6\text{O}_{18}]$       Covellin  $\text{CuS}$

Graphit  $\text{C}$       Eis  $\text{H}_2\text{O}$

Hochtridymit  $\text{SiO}_2$



# Sistema Cristalino Hexagonal

Holoedria hexagonal

Símbolo:  $6/mmm$  ou  $D_{6h}$

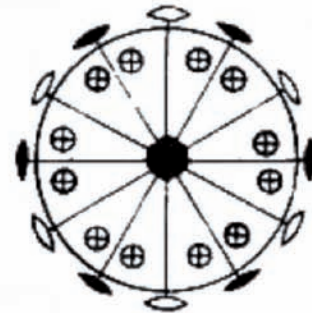
Classe bipiramidal dihexagonal

( $6/m\ 2/m\ 2/m$ )

Forma geral:  
 $\{hkl\} / \{hkil\}$



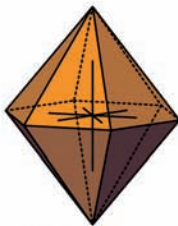
Bipirâmide dihexagonal



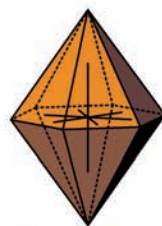
$\{h0l\} / \{h0\bar{h}l\}$

$\{hhl\} / \{hh\bar{2}hl\}$

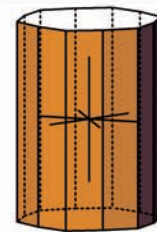
$\{hk0\} / \{hki0\}$



Bipirâmide hexagonal I.pos.



Bipirâmide hexagonal II.pos.

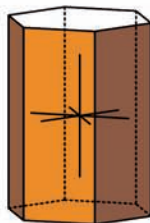


Prisma dihexagonal

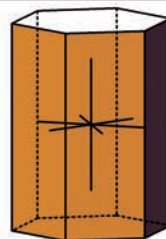
$\{100\} / \{10\bar{1}0\}$

$\{110\} / \{11\bar{2}0\}$

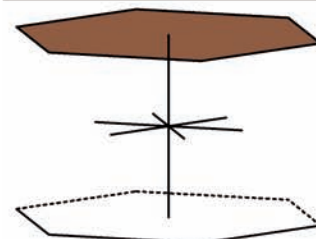
$\{001\} / \{0001\}$



Prisma hexagonal I.pos.



Prisma hexagonal II.pos.



Pinacoide Basal

## Exemplos de minerais:

Berilo  $Be_3Al_2[Si_6O_{18}]$  Covellita  $CuS$

Grafita  $C$

Gelo  $H_2O$

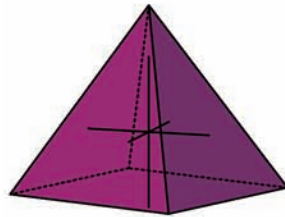
Tridimita  $SiO_2$

# Tetragonales Kristallsystem

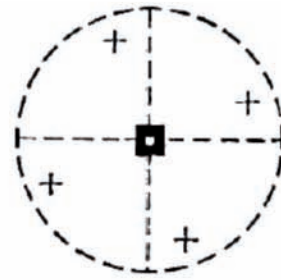
Hemimorphie d. tetragonal-  
pyramidalen Hemiedrie  
Tetragonal-pyramidale Klasse

Symbol: 4 oder  $C_4$

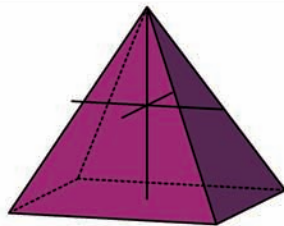
Allgemeine Form:  
 $\{hkl\}$



**Tetrag. Pyramiden III.St.**

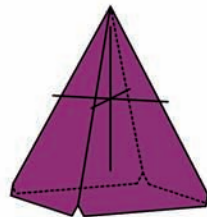


$\{h0l\}$



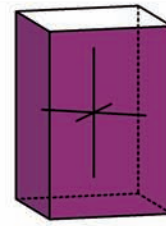
**Tetrag. Pyramiden II.St.**

$\{hhl\}$



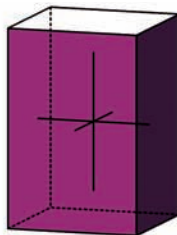
**Tetrag. Pyramiden I.St.**

$\{hk0\}$



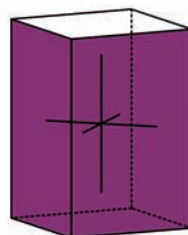
**Tetrag. Prismen III.St.**

$\{100\}$



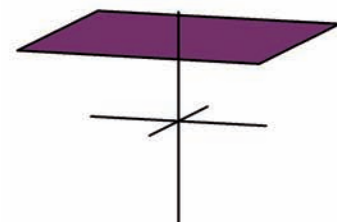
**Tetrag. Prisma II.St.**

$\{110\}$



**Tetrag. Prisma I.St.**

$\{001\}$



**Basispedien**

## Mineral-Bsp.:

Piypite  $K_2Cu_2[O/(SO_4)_2]$

Pinnoite  $Mg[B_2O_4]*3H_2O$

Percleveite-(Ce)  $(Ce,La,Nd)_2[Si_2O_7]$

# Sistema Cristalino Tetragonal

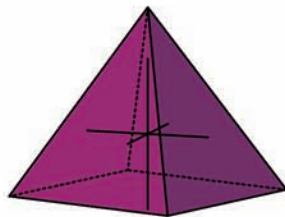
Ditetragonal Hemimorfo

Hemiedro Piramidal

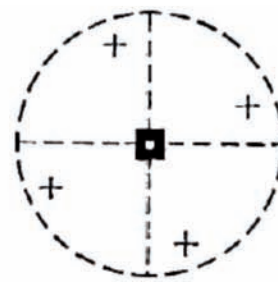
Símbolo: 4 ou  $C_4$

Classe tetragonal piramidal

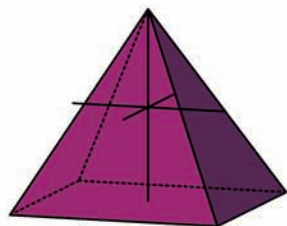
Forma geral:  
 $\{hkl\}$



**Pirâmide tetragonal III.pos.**

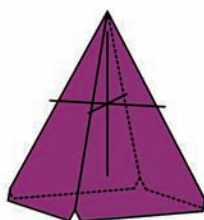


$\{h0l\}$



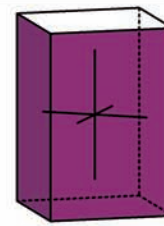
**Pirâmide tetragonal II.pos.**

$\{hhl\}$



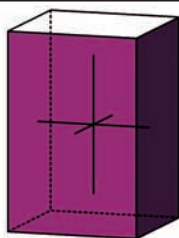
**Pirâmide tetragonal I.pos.**

$\{hk0\}$



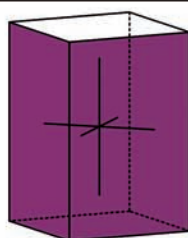
**Prisma tetragonal III.pos.**

$\{100\}$



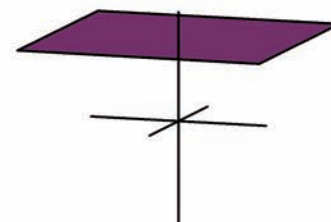
**Prisma tetragonal II.pos.**

$\{110\}$



**Prisma tetragonal I.pos.**

$\{001\}$



**Base Pedien**

## Exemplos de minerais:

Piypita  $K_2Cu_2[O/(SO_4)_2]$

Pinnoíta  $Mg[B_2O_4]*3H_2O$

Percleveíta-(Ce)  $(Ce,La,Nd)_2[Si_2O_7]$

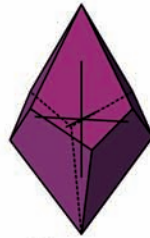
# Tetragonales Kristallsystem

Tetragonal-trapezoedrische  
Hemiedrie

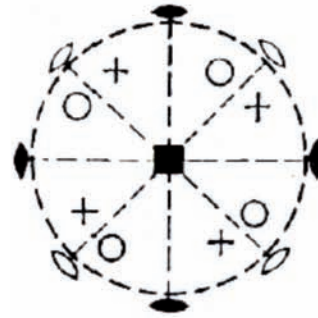
Symbol: 422 oder  $D_4$

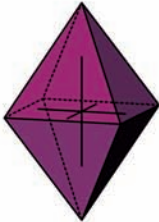
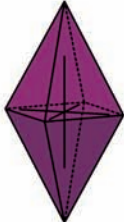
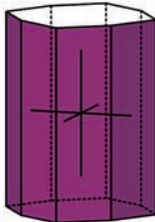
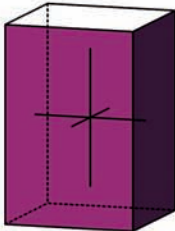
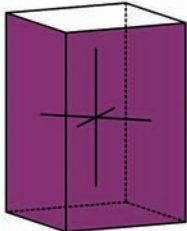
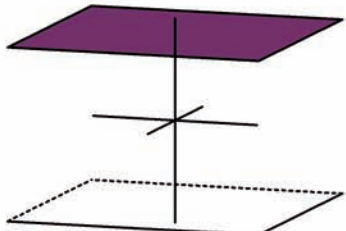
Tetragonal-trapezoedrische Klasse

Allgemeine Form:  
 $\{hkl\}$



Tetrag. Trapezoeder



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag. Dipyramiden II. St.</u>	<u>Tetrag. Dipyramiden I. St.</u>	<u>Ditetrag. Prismen</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag. Prisma II. St.</u>	<u>Tetrag. Prisma I. St.</u>	<u>Basispinakoid</u>

**Mineral-Bsp.:**

Retgersit  $\text{Ni}[\text{SO}_4] \cdot 6\text{H}_2\text{O}$

Maucherit  $\text{Ni}_{11}\text{As}_8$

Cristobalit  $\text{SiO}_2$

# Sistema Cristalino Tetragonal

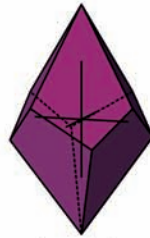
Trapezoedral tetragonal

Hemiedrio

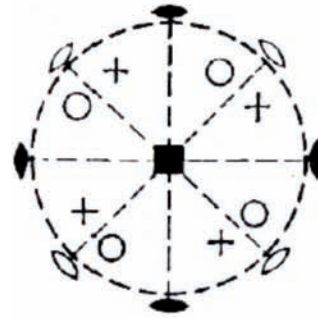
Símbolo: 422 ou  $D_4$

Classe tetragonal trapezoédrica

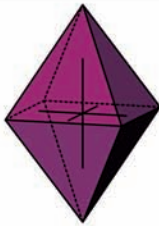
Forma geral:  
 $\{hkl\}$



Trapezoedro tetragonal

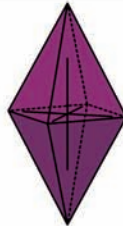


$\{h0l\}$



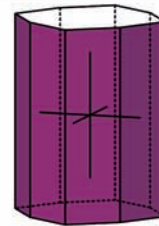
Dipiramide tetragonal II.pos.

$\{hhl\}$



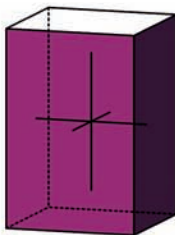
Dipiramide tetragonal I.pos.

$\{hk0\}$



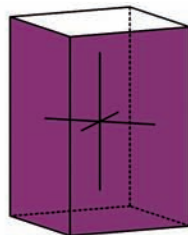
Prisma ditetragonal

$\{100\}$



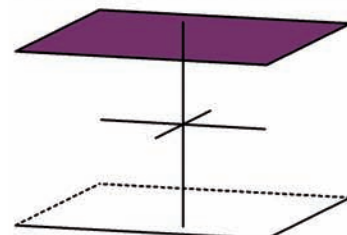
Prisma tetragonal II.pos.

$\{110\}$



Prisma tetragonal I.pos.

$\{001\}$



Base pinacoide

## Exemplos de minerais:

Retgersita  $Ni[SO_4] \cdot 6H_2O$

Maucherita  $Ni_{11}As_8$

Cristobalita  $SiO_2$

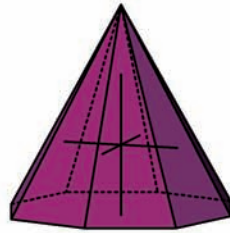
# Tetragonales Kristallsystem

Hemimorphie d. tetragonalen  
Holoedrie

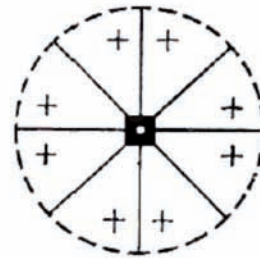
Symbol:  $4mm$  oder  $C_{4v}$

Ditetragonal-pyramidale Klasse

Allgemeine Form:  
 $\{hkl\}$



**Ditetrag. Pyramiden**



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
<b>Tetrag. Pyramiden II. St.</b>	<b>Tetrag. Pyramiden I. St.</b>	<b>Ditetrag. Prismen</b>
$\{100\}$	$\{110\}$	$\{001\}$
<b>Tetrag. Prisma II. St.</b>	<b>Tetrag. Prisma I. St.</b>	<b>Basispedien</b>

## Mineral-Bsp.:

Diaboleit  $Pb_2Cu[Cl_2/(OH)_4]$

Macedonite  $PbTiO_3$

Nielsenite  $PdCu_3$

# Sistema Cristalino Tetragonal

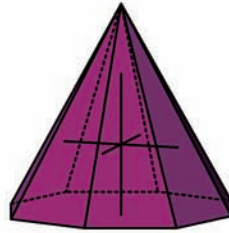
Tetragonal Hemimorfo

Holoedria

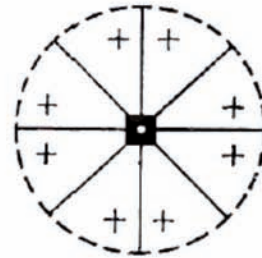
Classe ditetragonal piramidal

Símbolo:  $4mm$  ou  $C_{4v}$

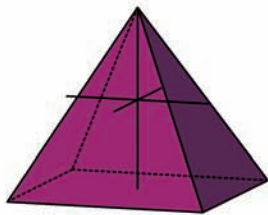
Forma geral:  
 $\{hkl\}$



**Pirâmide ditetragonal**

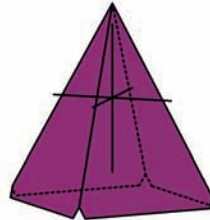


$\{h0l\}$



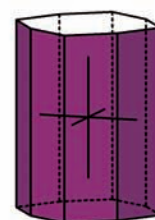
**Pirâmide tetragonal II.pos.**

$\{hhl\}$



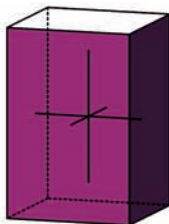
**Pirâmide tetragonal I.pos.**

$\{hk0\}$



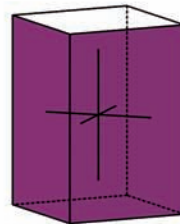
**Prisma Ditetragonal**

$\{100\}$



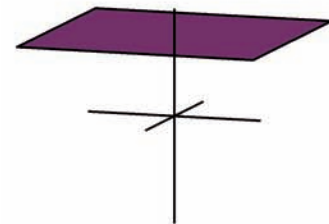
**Prisma tetragonal II.pos.**

$\{110\}$



**Prisma tetragonal I.pos.**

$\{001\}$



**Base Pedien**

## Exemplos de minerais:

Diaboleita  $Pb_2Cu[Cl_2/(OH)_4]$

Macedonita  $PbTiO_3$

Nielsenita  $PdCu_3$

# Tetragonales Kristallsystem

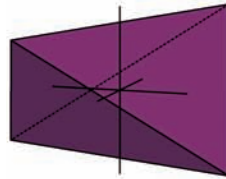
Tetragonal-sphenoidische

Tetartoedrie

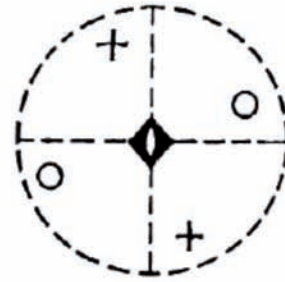
Tetragonal-disphenoidische Klasse

Symbol:  $\bar{4}$  oder  $S_4$

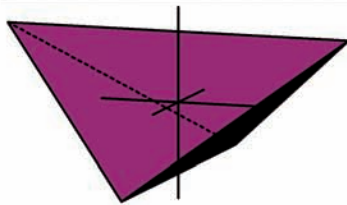
Allgemeine Form:  
 $\{hkl\}$



Tetrag.Disphenoide III.St.

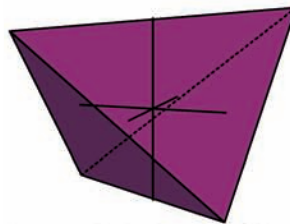


$\{h0l\}$



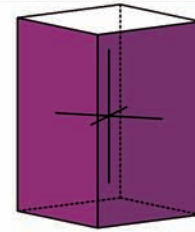
Tetrag.Disphenoide II.St.

$\{hhl\}$



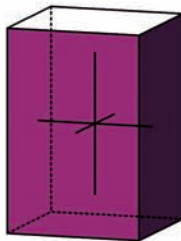
Tetrag.Disphenoide I.St.

$\{hk0\}$



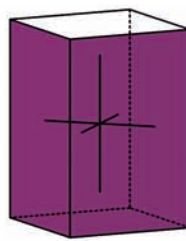
Tetrag.Prismen III.St.

$\{100\}$



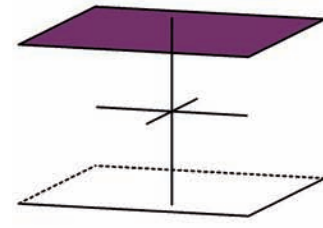
Tetrag.Prisma II.St.

$\{110\}$



Tetrag.Prisma I.St.

$\{001\}$



**Basispinakoid**

## Mineral-Bsp.:

Cahnit  $\text{Ca}_2[\text{AsO}_4/\text{B}(\text{OH})_4]$

Tugtupit  $\text{Na}_8[(\text{AlBeSi}_4\text{O}_{12})_2/\text{Cl}_2]$

Schreibersit  $(\text{Fe},\text{Ni})_3\text{P}$



# Sistema Cristalino Tetragonal

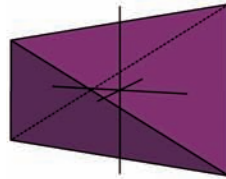
Esfenoide tetragonal

Tetardoedro

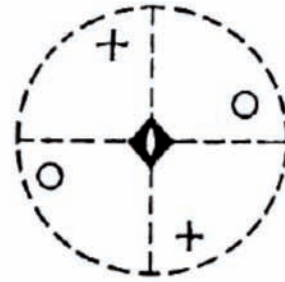
Classe tetragonal disfenoidal

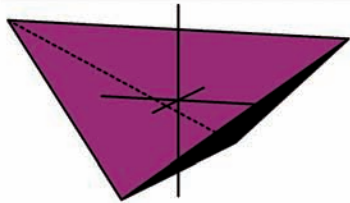
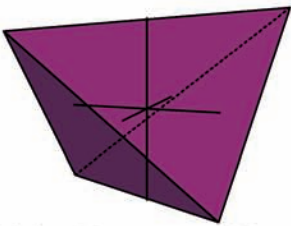
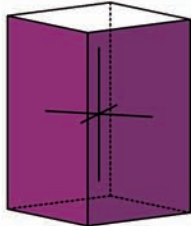
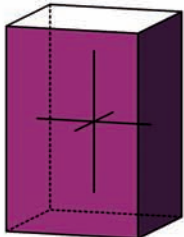
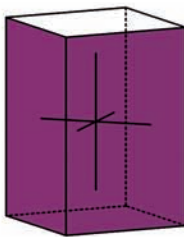
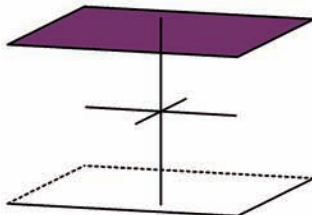
Símbolo:  $\bar{4}$  ou  $S_4$

Forma geral:  
 $\{hkl\}$



**Disfenóide tetragonal III.pos.**



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
 <p><b>Disfenóide tetragonal II.pos.</b></p>	 <p><b>Disfenóide tetragonal I.pos.</b></p>	 <p><b>Prisma tetragonal III.pos.</b></p>
$\{100\}$	$\{110\}$	$\{001\}$
 <p><b>Prisma tetragonal II.pos.</b></p>	 <p><b>Prisma tetragonal I.pos.</b></p>	 <p><b>Base Pinacóide</b></p>

## Exemplos de minerais:

Canita  $\text{Ca}_2[\text{AsO}_4/\text{B}(\text{OH})_4]$

Tugtupita  $\text{Na}_8[(\text{AlBeSi}_4\text{O}_{12})_2/\text{Cl}_2]$

Schreibersita  $(\text{Fe},\text{Ni})_3\text{P}$

# Tetragonales Kristallsystem

Tetragonal-sphenoidische  
Hemiedrie

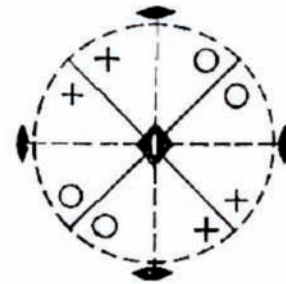
Symbol:  $\bar{4}2m$  oder  $D_{2d}$

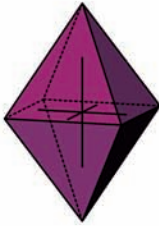
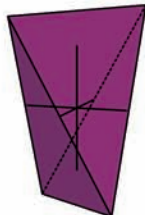
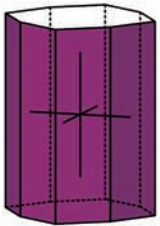
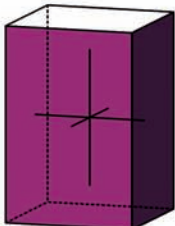
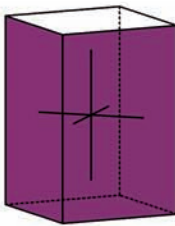
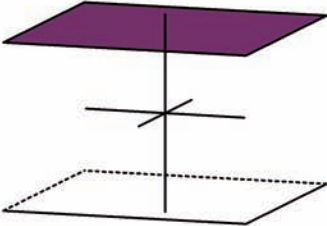
Tetragonal-skalenoedrische Klasse

Allgemeine Form:  
{hkl}



Tetrag. Skalenoeder



{h0l}	{hhl}	{hk0}
 <p data-bbox="229 1238 539 1272"><u>Tetrag. Dipyramiden II. St.</u></p>	 <p data-bbox="655 1238 943 1272"><u>Tetrag. Disphenoide I. St.</u></p>	 <p data-bbox="1106 1238 1318 1272"><u>Ditetrag. Prismen</u></p>
{100}	{110}	{001}
 <p data-bbox="264 1653 501 1686"><u>Tetrag. Prisma II. St.</u></p>	 <p data-bbox="687 1653 908 1686"><u>Tetrag. Prisma I. St.</u></p>	 <p data-bbox="1129 1653 1299 1686"><u>Basispinakoid</u></p>

## Mineral-Bsp.:

Chalkopyrit  $\text{CuFeS}_2$

Mooihoekite  $\text{Cu}_9\text{Fe}_9\text{S}_{16}$

Stannit  $\text{Cu}_2\text{FeSnS}_4$

# Sistema Cristalino Tetragonal

Esfenóide tetragonal

Hemiedria

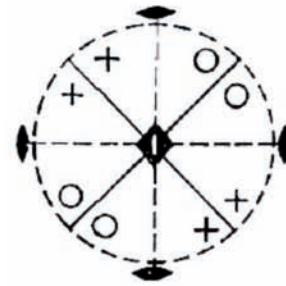
Símbolo:  $\bar{4}2m$  ou  $D_{2d}$

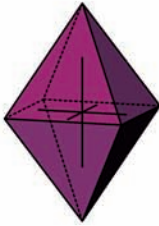
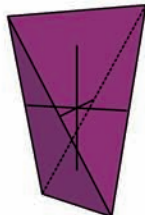
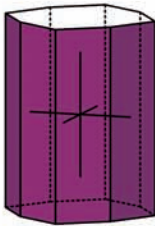
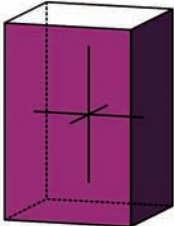
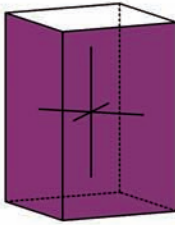
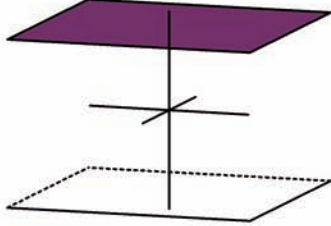
Classe escalenoédrica tetragonal

Forma geral:  
 $\{hkl\}$



Escalenoedro tetragonal



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Dipirâmide tetragonal II.pos.</u>	<u>Diesfenóide tetragonal I.pos.</u>	<u>Prisma ditetragonal</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Prisma tetragonal II.pos.</u>	<u>Prisma tetragonal I.pos.</u>	<u>Bases de Pinacóide</u>

## Exemplos de minerais:

Calcopirita  $CuFeS_2$

Mooihoequita  $Cu_9Fe_9S_{16}$

Estanita  $Cu_2FeSnS_4$

# Tetragonales Kristallsystem

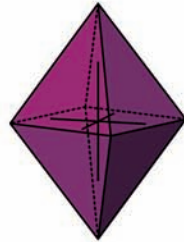
Tetragonal-dipyramidale

Hemiedrie

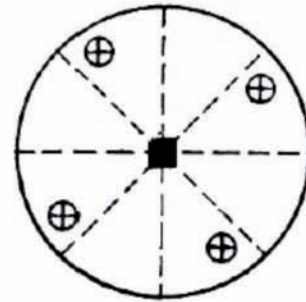
Symbol:  $4/m$  oder  $C_{4h}$

Tetragonal-dipyramidale Klasse

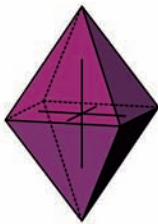
Allgemeine Form:  
 $\{hkl\}$



Tetrag. Dipyramiden III. St.

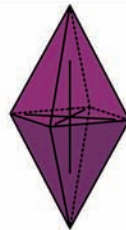


$\{h0l\}$



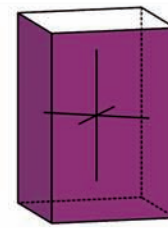
Tetrag. Dipyramiden II. St.

$\{hhl\}$



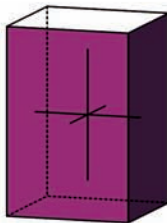
Tetrag. Dipyramiden I. St.

$\{hk0\}$



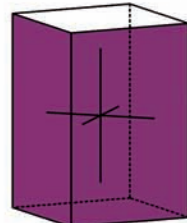
Tetrag. Prismen III. St.

$\{100\}$



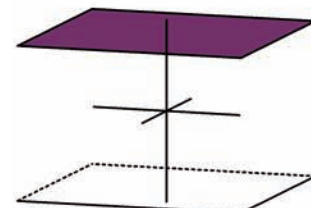
Tetrag. Prisma II. St.

$\{110\}$



Tetrag. Prisma I. St.

$\{001\}$



Basispinakoid

## Mineral-Bsp.:

Scheelit  $\text{Ca}[\text{WO}_4]$

Vesuvian  $\text{Ca}_{19}(\text{Mg}, \text{Fe}, \text{Ti})_4 \text{Al}_9 [(\text{OH}, \text{F})_{10} / (\text{SiO}_4)_{10} / (\text{Si}_2\text{O}_7)_4]$

Fergusonit  $\text{YNbO}_4$

# Sistema Cristalino Tetragonal

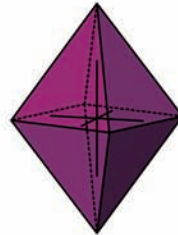
Tetragonal dipiramidal

Hemiedria

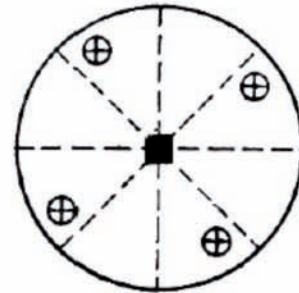
Símbolo:  $4/m$  ou  $C_{4h}$

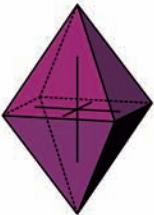
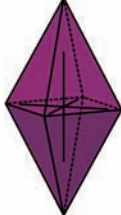
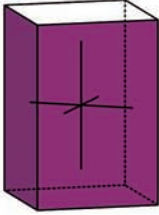
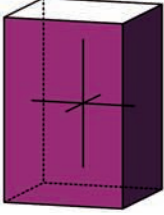
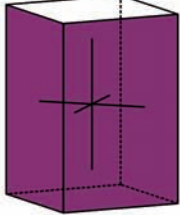
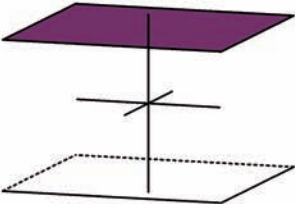
Classe Tetragonal dipiramidal

Forma geral:  
 $\{hkl\}$



Dipirâmide tetragonal III.pos.



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Dipirâmide Tetragonal II.pos.</u>	<u>Dipirâmide Tetragonal I.pos.</u>	<u>Prisma Tetragonal III.pos.</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Prisma Tetragonal II.pos.</u>	<u>Prisma Tetragonal I.pos.</u>	<u>Base de Pinacóide</u>

## Exemplos de minerais:

Scheelita  $Ca[WO_4]$

Vesuvianita  $Ca_{19}(Mg,Fe,Ti)_4Al_9[(OH,F)_{10}/(SiO_4)_{10}/(Si_2O_7)_4]$

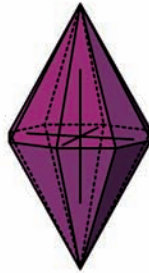
Fergusonita  $YNbO_4$

# Tetragonales Kristallsystem

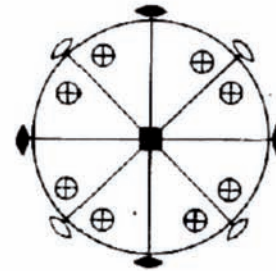
Tetragonale Holoedrie  
Ditetragonal-dipyramidale  
Klasse

Symbol:  $4/m\ 2/m\ 2/m$   
( $4/m\ m\ m$ ) oder  $D_{4h}$

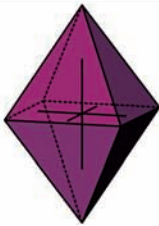
Allgemeine Form:  
 $\{hkl\}$



Ditetrag. Dipyramiden

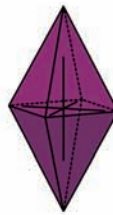


$\{h0l\}$



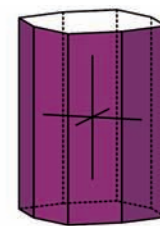
Tetrag. Dipyramiden II. St.

$\{hhl\}$



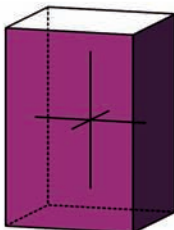
Tetrag. Dipyramiden I. St.

$\{hk0\}$



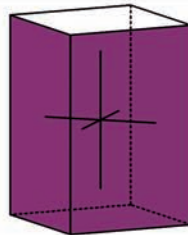
Ditetrag. Prismen

$\{100\}$



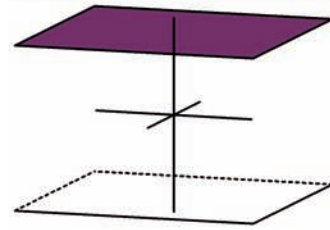
Tetrag. Prisma II. St.

$\{110\}$



Tetrag. Prisma I. St.

$\{001\}$



Basispinakoid

## Mineral-Bsp.:

Rutil  $\text{TiO}_2$

Anatas  $\text{TiO}_2$

Zirkon  $\text{Zr}[\text{SiO}_4]$

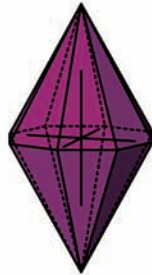
# Sistema Cristalino Tetragonal

Holoedria Tetragonal  
Classe dipiramidal  
ditetragonal

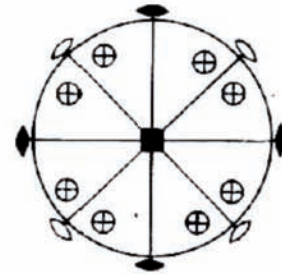
Símbolo:  $4/m\ 2/m\ 2/m$

$(4/m\ m\ m)$  ou  $D_{4h}$

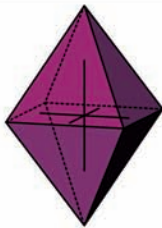
Forma geral:  
 $\{hkl\}$



Dipirâmide ditetragonal

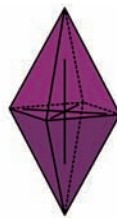


$\{h0l\}$



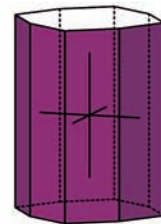
Dipirâmide Tetragonal II.pos.

$\{hhl\}$



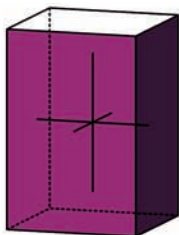
Dipirâmide Tetragonal I.pos.

$\{hk0\}$



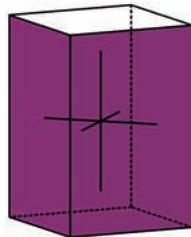
Prisma Ditetragonal

$\{100\}$



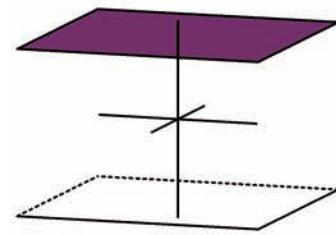
Prisma Tetragonal II.pos.

$\{110\}$



Prisma Tetragonal I.pos.

$\{001\}$



Base de Pinacóide

## Exemplos de minerais:

Rutilo  $TiO_2$

Anatásio  $TiO_2$

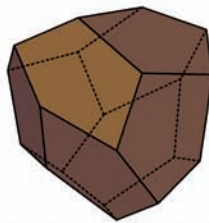
Zircão  $Zr[SiO_4]$

# Kubisches Kristallsystem

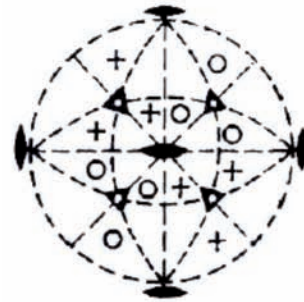
Kubische Tetartoedrie  
Tetraedisch-pentagon-  
doekaedrische Klasse

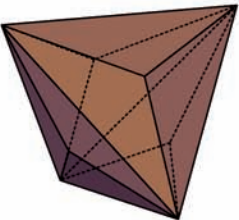
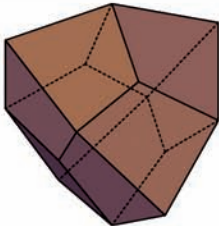
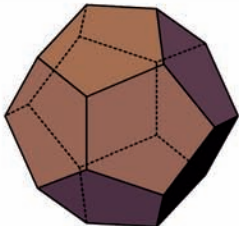
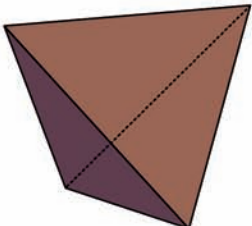
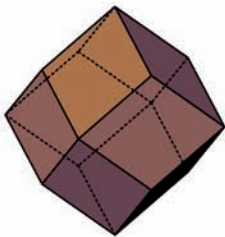
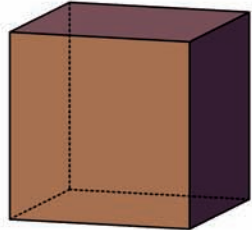
Symbol: 23 oder T

Allgemeine Form:  
{hkl}



Tetraedr.Pentagondodekaeder



{hll}	{hhl}	{hk0}
 <p><u>Pyramidentetraeder</u></p>	 <p><u>Deltoiddodekaeder</u></p>	 <p><u>Pentagondodekaeder</u></p>
{111}	{110}	{100}
 <p><u>Tetraeder</u></p>	 <p><u>Rhombendodekaeder</u></p>	 <p><u>Würfel/ Hexaeder</u></p>

## Mineral-Bsp.:

Langbeinit  $K_2Mg_2[(SO_4)_3]$

Ullmannit NiSbS

Gersdorffit NiAsS

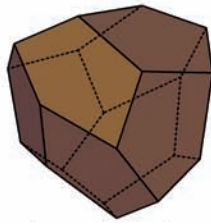


# Sistema Cristalino Isométrico

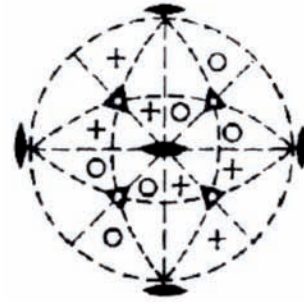
Tetartoedro Cúbico

Classe pentagontritetraedrica Símbolo: 23 ou T

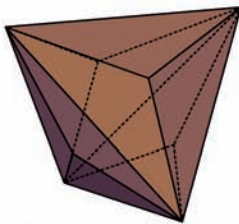
Forma geral:  
{hkl}



Pentagontritetraedro

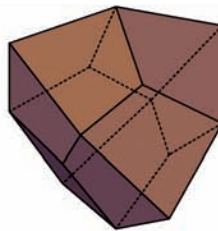


{hll}



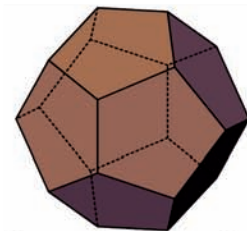
Tetraedro piramital

{hhl}



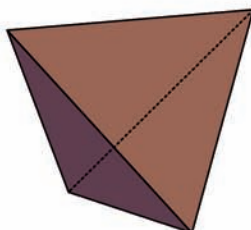
Deltoidecaedro

{hk0}



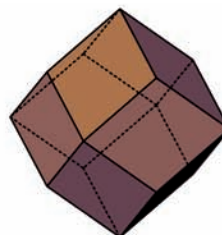
Pentagonododecaedro

{111}



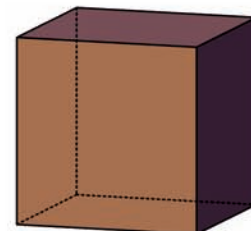
Tetraedro

{110}



Rombododecaedro

{100}



Cubo

## Exemplos de minerais:

Langbeinita  $K_2Mg_2[(SO_4)_3]$

Ullmanita NiSbS

Gersdorffita NiAsS

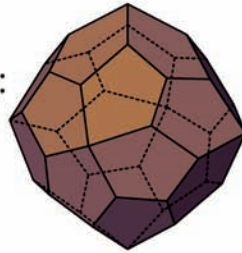
# Kubisches Kristallsystem

Gyroedrische Hemiedrie

Symbol: 432 oder O

Pentagonikositetraedrische Klasse

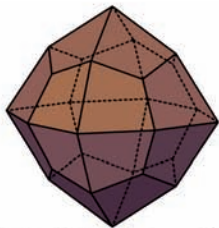
Allgemeine Form:  
{hkl}



Pentagonikositetraeder

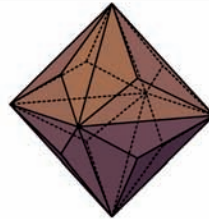


{hll}



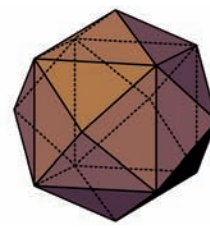
Deltoidikositetraeder

{hhl}



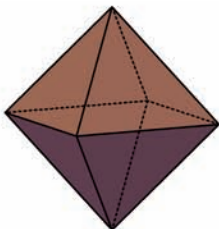
Pyramidenoktaeder

{hk0}



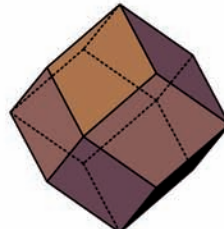
Pyramidenwürfel

{111}



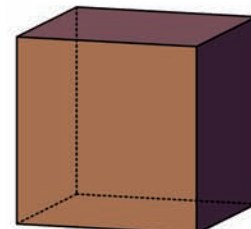
Oktaeder

{110}



Rhombendodekaeder

{100}



Würfel/ Hexaeder

## Mineral-Bsp.:

Maghemit  $\text{Gamma-Fe}_2\text{O}_3$

Choloalit  $\text{CuPb}[(\text{TeO}_3)_2]$

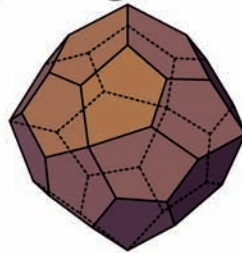
Petzit  $\text{Ag}_3\text{AuTe}_2$

# Sistema Cristalino Isométrico

Hemiedro Girodedral  
Classe tetraedro pentagonal

Símbolo: 432 ou O

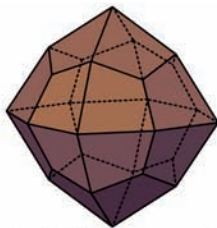
Forma geral:  
 $\{hkl\}$



Pentagonositetraedro

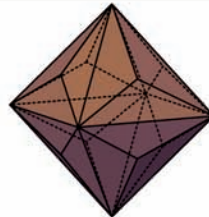


$\{hll\}$



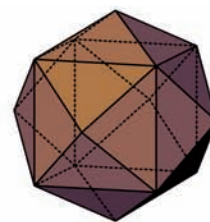
Deltoideicositetraedro

$\{hhl\}$



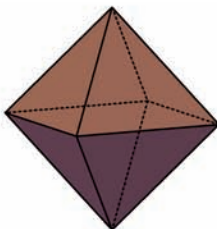
Trisoctaedro

$\{hk0\}$



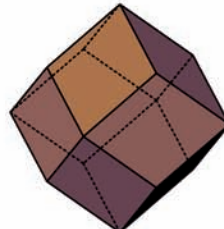
Tetrahexaedro

$\{111\}$



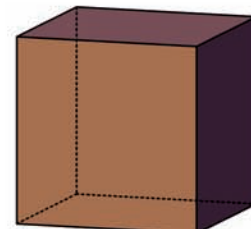
Oktaedro

$\{110\}$



Rombododecaedro

$\{100\}$



Ou Hexaedro

## Exemplos de minerais:

Maghemita  $\text{Gamma-Fe}_2\text{O}_3$

Choloalita  $\text{CuPb}[(\text{TeO}_3)_2]$

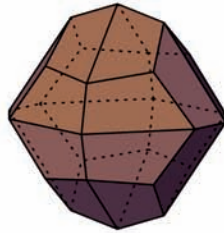
Petzita  $\text{Ag}_3\text{AuTe}_2$

# Kubisches Kristallsystem

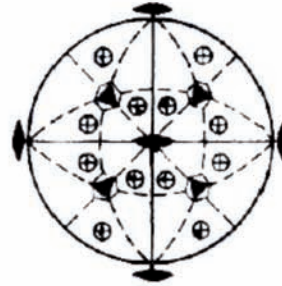
Parallelflächige Hemiedrie  
Disdodekaedrische Klasse

Symbol:  $2/m\bar{3}$  oder  $T_h$   
( $m\bar{3}$ )

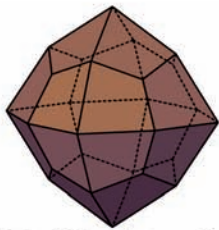
Allgemeine Form:  
{hkl}



Disdodekaeder

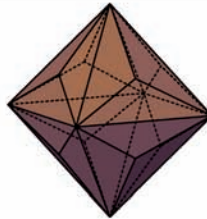


{hll}



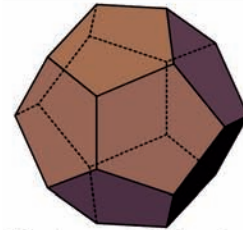
Deltoidikositetraeder

{hhl}



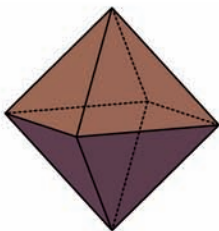
Pyramidenoktaeder

{hk0}



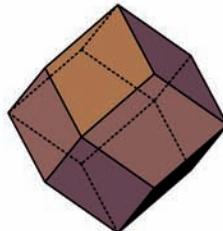
Pentagondodekaeder

{111}



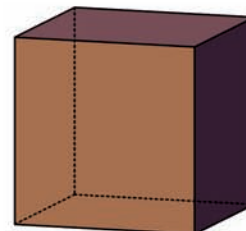
Oktaeder

{110}



Rhombendodekaeder

{100}



Würfel/Hexaeder

## Mineral-Bsp.:

Pyrit  $FeS_2$

Cobaltin  $CoAsS$

Alaune, z.B.  $KAl[SO_4]_2 \cdot 12H_2O$

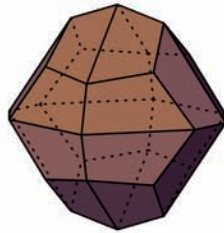
Bixbyit  $(Mn,Fe)_2O_3$

# Sistema Cristalino Ou Isométrico

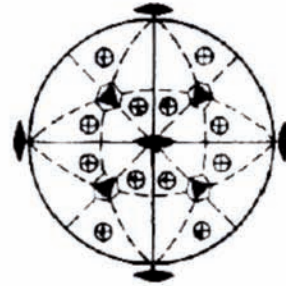
Hemiedro planer-paralelo  
Classe disdodecaédrica

Símbolo:  $2/m \bar{3}$  ou  $T_h$   
(m3)

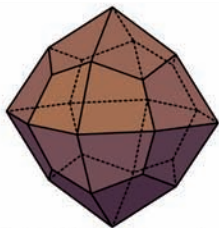
Forma geral:  
{hkl}



Disdodecaedro

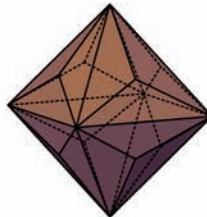


{hll}



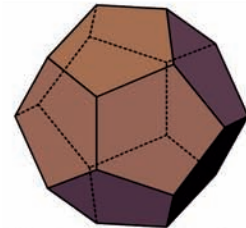
Deltoidicositetrahedro

{hhl}



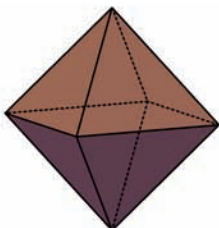
Trisoctaedro

{hk0}



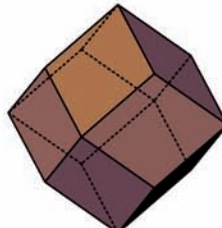
Pentagondodecaedro

{111}



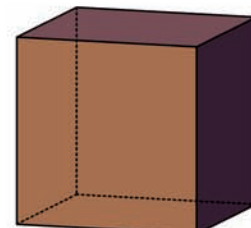
Octaedro

{110}



Rombododecaedro

{100}



Ou hexaedro

## Exemplos de minerais:

Pirita  $FeS_2$

Cobaltita  $CoAsS$

Alunita, z.B.  $KAl[SO_4]_2 \cdot 12H_2O$

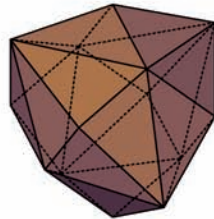
Bisbyita  $(Mn,Fe)_2O_3$

# Kubisches Kristallsystem

Geneigtflächige Hemiedrie  
Hexakistetraedrische Klasse

Symbol:  $\bar{4}3m$  oder  $T_d$

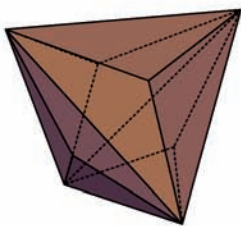
Allgemeine Form:  
{hkl}



Hexakistetraeder

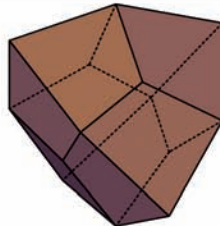


{hll}



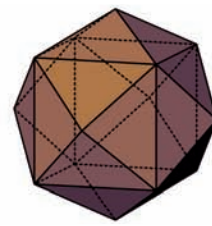
Pyramidentetraeder

{hhl}



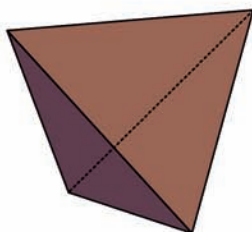
Deltoiddodekaeder

{hk0}



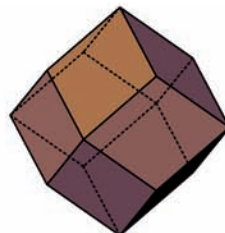
Pyramidenwürfel

{111}



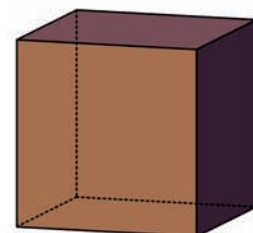
Tetraeder

{110}



Rhombendodekaeder

{100}



Würfel/ Hexaeder

## Mineral-Bsp.:

Sphalerit/Zinkblende  $ZnS$

Tetraedrit  $(Cu,Fe)_{12}[Sb_4S_{13}]$

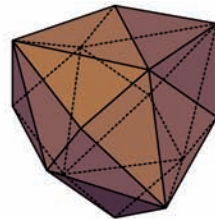
Sodalith  $Na_8[(Cl)_2(AlSiO_4)_6]$

Mayenit  $Ca_{12}Al_{14}O_{33}$

# Sistema Cristalino Ou isométrico

Hemiedria de fície inclinada      Símbolo:  $\bar{4}3m$  ou  $T_d$   
 Classe hexatetraédrica

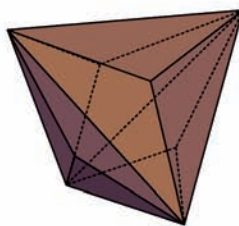
Forma geral:  
 $\{hkl\}$



Hexatetraedro

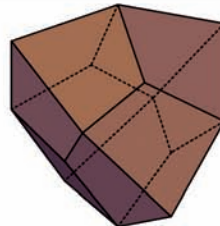


$\{hll\}$



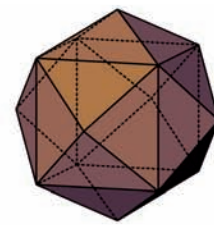
Tristetrahedro

$\{hhl\}$



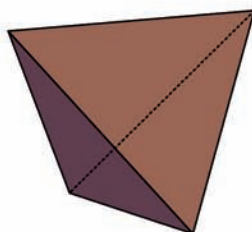
Tetraedro deltoide

$\{hk0\}$



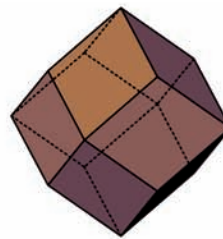
Tetrahexaedro

$\{111\}$



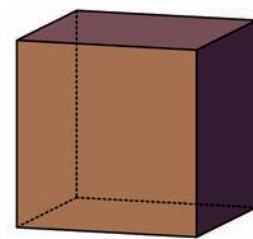
Tetraedro

$\{110\}$



Rombododecaedro

$\{100\}$



Ou hexaedro

## Exemplos de minerais:

Esfalerita ou blenda  $ZnS$

Tetraedrota  $(Cu,Fe)_{12}[Sb_4S_{13}]$

Sodalita  $Na_8[(Cl)_2(AlSiO_4)_6]$

Mayenita  $Ca_{12}Al_{14}O_{33}$

# Kubisches Kristallsystem

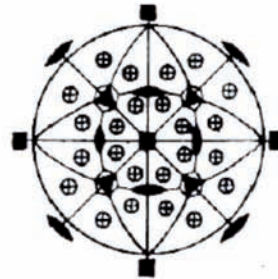
Kubische Holoedrie  
Hexakisoktaedrische Klasse

Symbol:  $4/m \bar{3} 2/m$  oder  $O_h$   
( $m\bar{3}m$ )

Allgemeine Form:  
{hkl}



Hexakisoktaeder



{hll}	{hhl}	{hk0}
<u>Deltoidikositetraeder</u>	<u>Pyramidenoktaeder</u>	<u>Pyramidenwürfel</u>
{111}	{110}	{100}
<u>Oktaeder</u>	<u>Rhombendodekaeder</u>	<u>Würfel/Hexaeder</u>

## Mineral-Bsp.:

Elemente: Au, Ag, Cu, Pt,  
Pb, Fe, W, Si, C(Diamant)  
Halt/Steinsalz NaCl

Galenit/Bleiglanz PbS  
Fluorit/Flußspat  $CaF_2$   
Spinelle, z.B.  $MgAl_2O_4$   
Granate  $Me^{II}_3Me^{III}_2[SiO_4]_3$

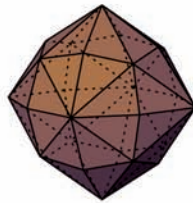


# Sistema Cristalino Cúbico

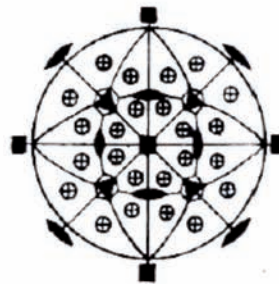
Holoedria Cubica  
Classe Hexaoctaédrica

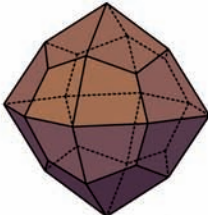
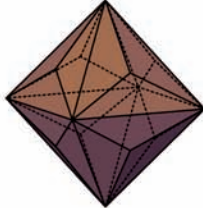
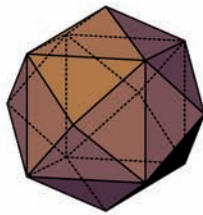
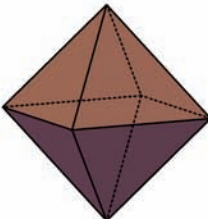
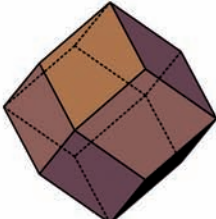
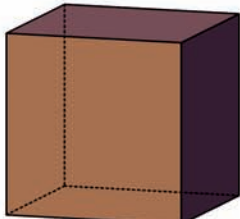
Símbolo:  $4/m \bar{3} 2/m$  ou  $O_h$   
( $m\bar{3}m$ )

Forma geral:  
{hkl}



Hexaoctaedro



{hll}	{hhl}	{hk0}
		
<u>Octaedro deltóide</u>	<u>Trisoctaedro</u>	<u>Tetrahexaedro</u>
{111}	{110}	{100}
		
<u>Oktaedro</u>	<u>Rombododecaedro</u>	<u>Ou hexaedro</u>

**Exemplos de minerais:**

Elementos nativos: Au, Ag, Cu,

Pt, Pb, Fe, W, Si, diamante C

Halita NaCl

Galena/Galenita PbS

Fluorita CaF<sub>2</sub>

Espinélio MgAl<sub>2</sub>O<sub>4</sub>

Granada Me<sup>II</sup><sub>3</sub>Me<sup>III</sup><sub>2</sub>[SiO<sub>4</sub>]<sub>3</sub>

## **Beispiele für korrele Kristallformen**

### **Formenausbildung**

**Positiv – Negativ**

**Rechts – Links**

**Oben – Unten**

**Vorn - Hinten**

## **Exemplos de formas cristalográficas correlatas**

### **Descrição de Formas**

**Positiva – Negativa**

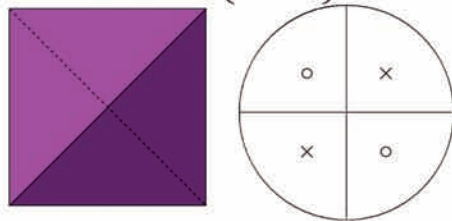
**Direita – Esquerda**

**Acima – Abaixo**

**A Frente - Atrás**

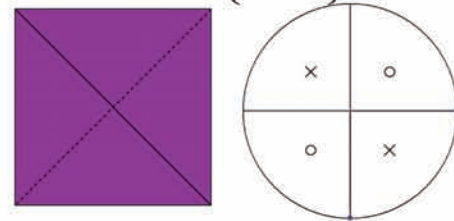
# Kubisches Kristallsystem

Tetraeder  $\{\bar{1}\bar{1}\bar{1}\}$



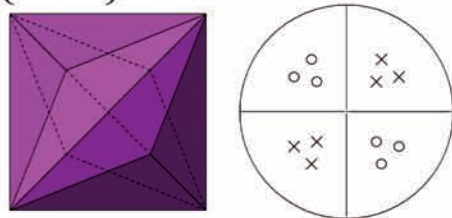
negativ

Tetraeder  $\{111\}$



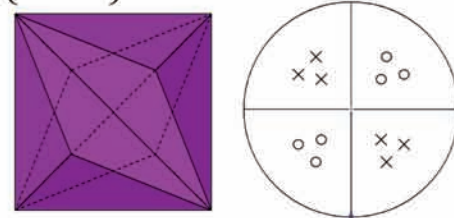
positiv

Pyramidentetraeder  $\{223\}$



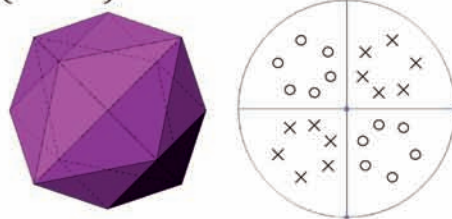
positiv

Pyramidentetraeder  $\{223\}$



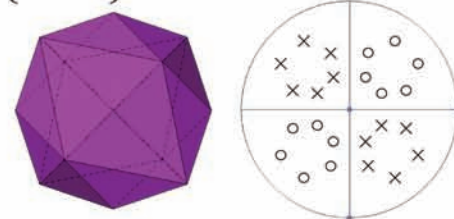
negativ

Hexakistetraeder  $\{123\}$



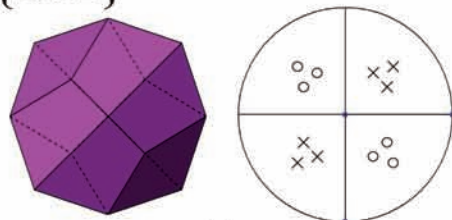
positiv

Hexakistetraeder  $\{\bar{1}23\}$



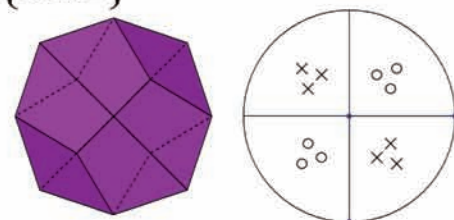
negativ

Deltoiddodekaeder  $\{332\}$



positiv

Deltoiddodekaeder  $\{\bar{3}\bar{3}\bar{2}\}$

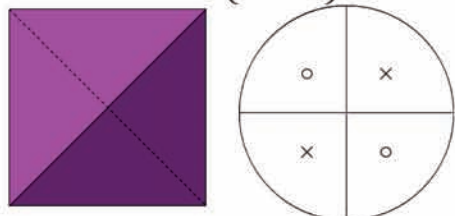


negativ

Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

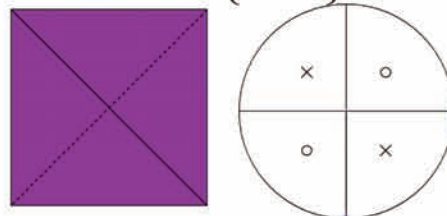
# Sistema Cristalino Cúbico

Tetraedro  $\{\bar{111}\}$



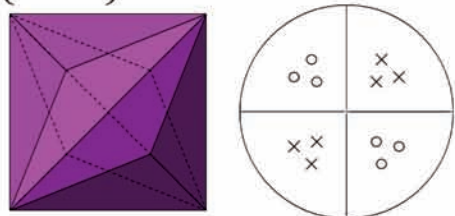
negativo

Tetraedro  $\{111\}$



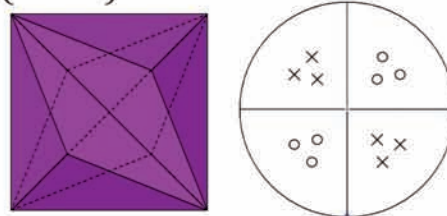
positivo

Pirâmide Tetraedro  $\{223\}$



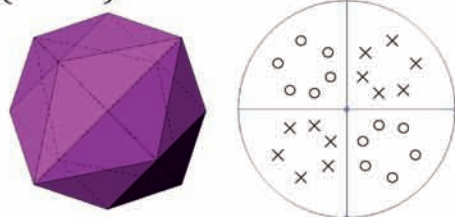
positivo

Pirâmide Tetraedro  $\{\bar{2}23\}$



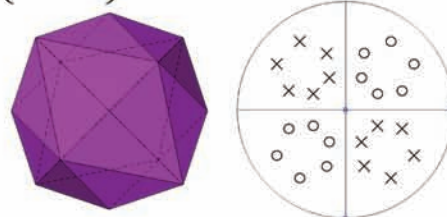
negativo

Hexatetraedro  $\{123\}$



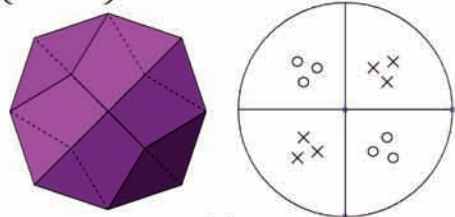
positivo

Hexatetraedro  $\{\bar{1}23\}$



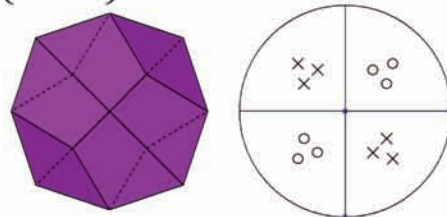
negativo

Dodecaedro deltóide  $\{332\}$



positivo

Dodecaedro deltóide  $\{\bar{3}32\}$

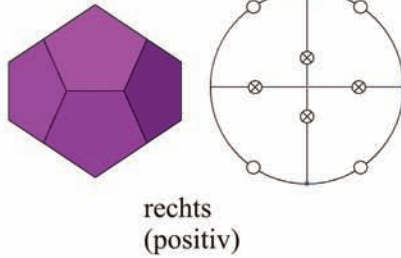


negativo

Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

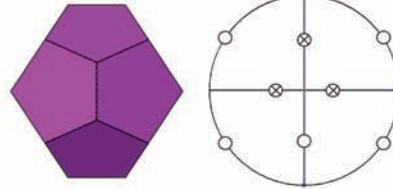
# Kubisches Kristallsystem

Pentagondodekaeder {230}



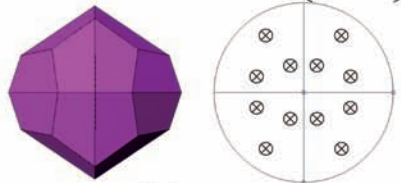
rechts  
(positiv)

Pentagondodekaeder {320}



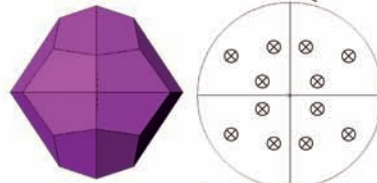
links  
(negativ)

Disdodekaeder {213}



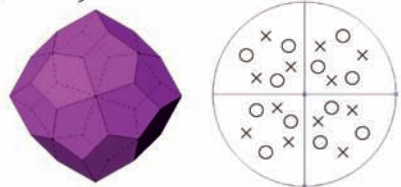
links  
(positiv)

Disdodekaeder {123}

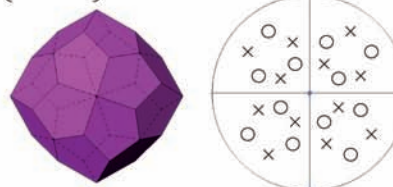


rechts  
(negativ)

Pentagonikositetraeder {123} rechts

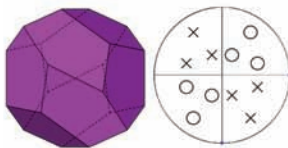


Pentagonikositetraeder {213} links



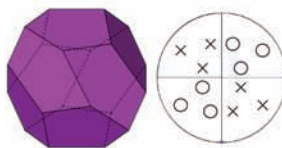
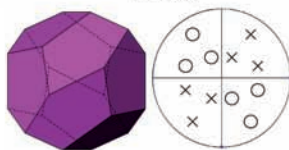
Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente; es handelt sich hier lediglich um Bezugsflächen

## Tetraedrischer Pentagondodekaeder



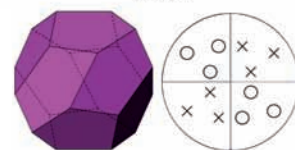
{123} positiv  
rechts

{ $\bar{1}23$ } negativ  
rechts



{213} positiv  
links

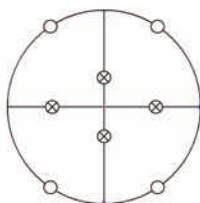
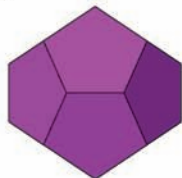
{ $\bar{2}13$ } negativ  
links



# Sistema Cristalino Ou Isométrico

Pentadodecaedro

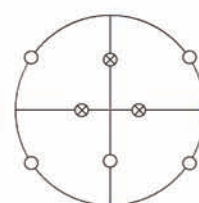
{230}



direito  
(positivo)

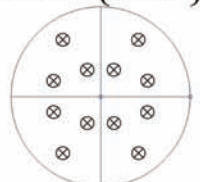
Pentadodecaedro

{320}



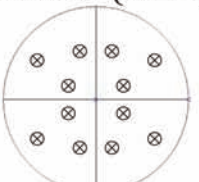
esquerdo  
(negativo)

Disdodecaedro {213}



esquerdo  
(positivo)

Disdodecaedro {123}



direito  
(negativo)

Icositetraedro pentagonal

{123}

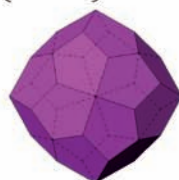
direito



Icositetraedro pentagonal

{213}

esquerdo



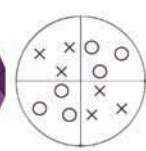
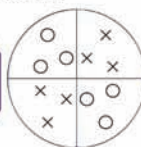
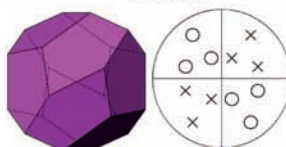
Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

## Pentadodecaedro Tetraédrico



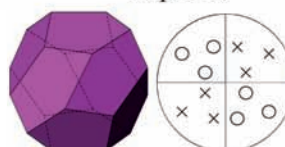
{123} positivo  
direito

{ $\bar{1}23$ } negativo  
direito



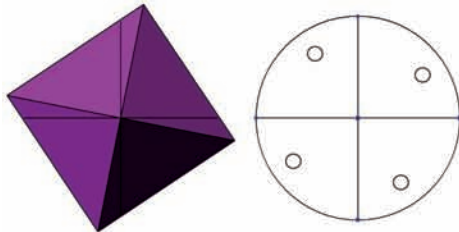
{213} positivo  
esquerdo

{ $\bar{2}13$ } negativo  
esquerdo



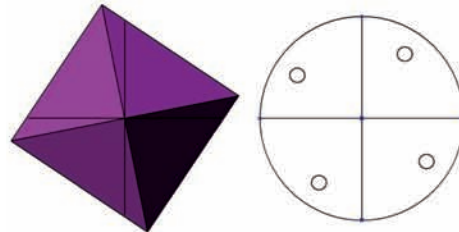
# Tetragonales Kristallsystem

Pyramide  $\{321\}$



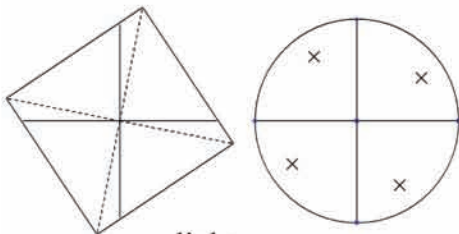
links  
oben

Pyramide  $\{231\}$



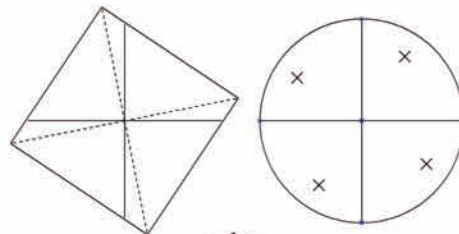
rechts  
oben

Pyramide  $\{32\bar{1}\}$



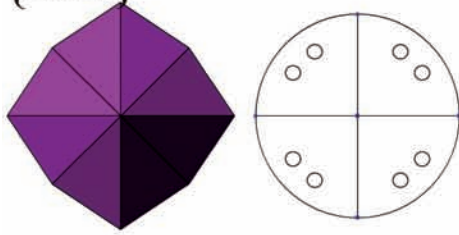
links  
unten

Pyramide  $\{23\bar{1}\}$



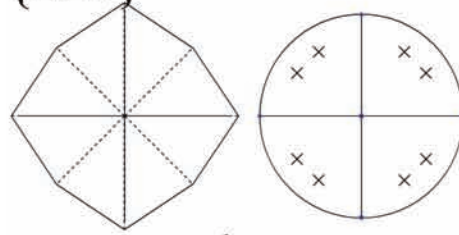
rechts  
unten

Ditetrag. Pyramide  
 $\{321\}$



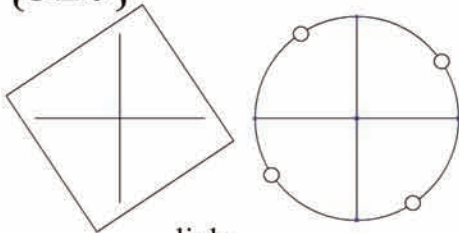
unten

Ditetrag. Pyramide  
 $\{32\bar{1}\}$



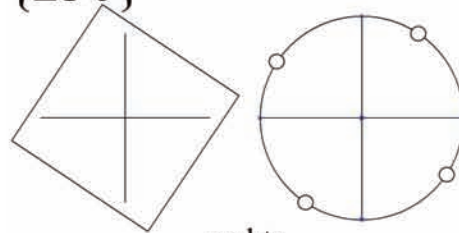
oben

Prisma III.Stellung  
 $\{320\}$



links

Prisma III.Stellung  
 $\{230\}$



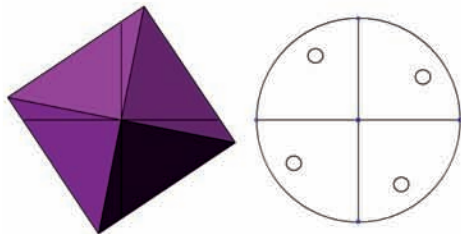
rechts

Diese stereographischen Projektionen behalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien



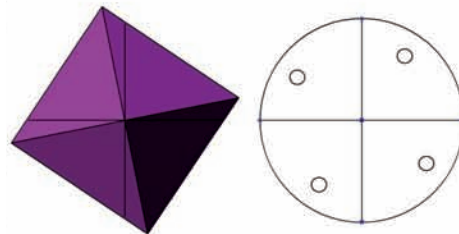
# Sistema Cristalino Tetragonal

Pirâmide {321}



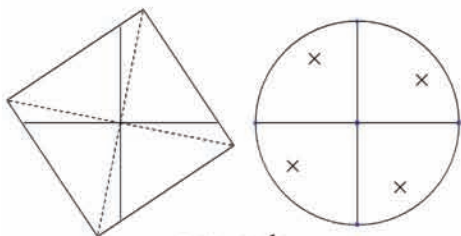
esquerda  
acima

Pirâmide {231}



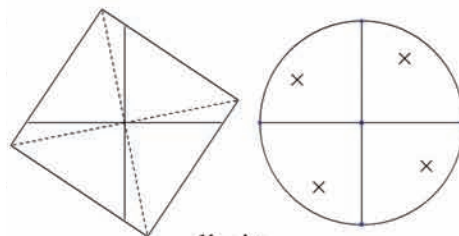
direita  
acima

Pirâmide {321̄}



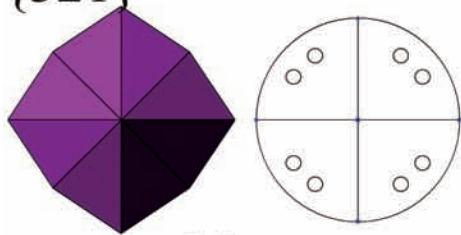
esquerda  
abaixo

Pirâmide {231̄}



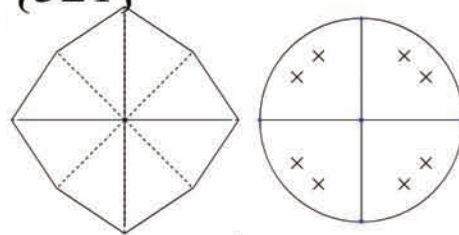
direita  
acima

Pirâmide Ditetragonal  
{321}



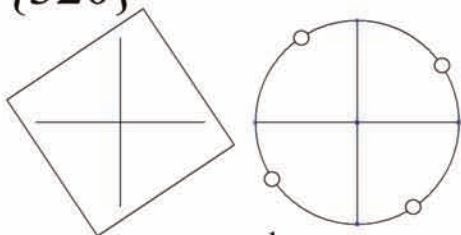
abaixo

Pirâmide Ditetragonal  
{321̄}



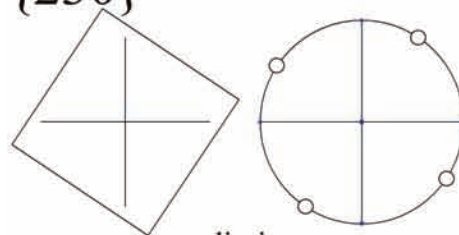
acima

Posição III Prisma  
{320}



esquerda

Posição III Prisma  
{230}

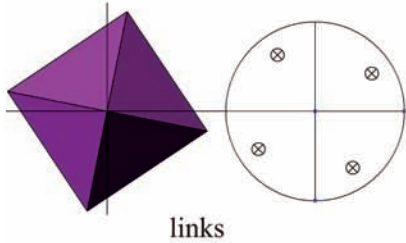


direita

Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

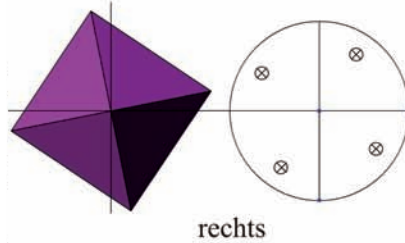
# Tetragonales Kristallsystem

Dipyramide {321}



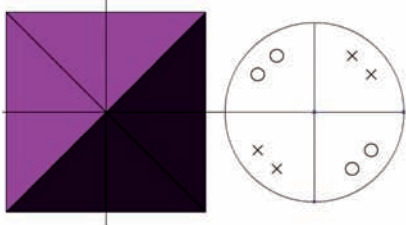
links

Dipyramide {231}



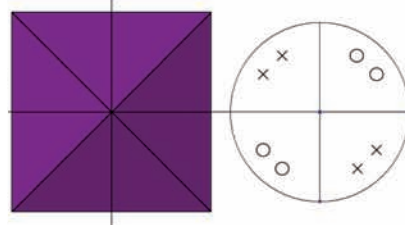
rechts

Skalenoeder {321}



positiv

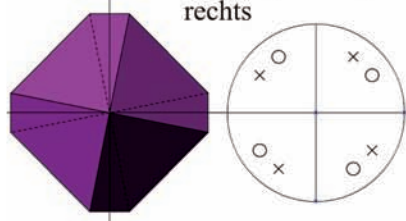
Skalenoeder {3̄21}



negativ

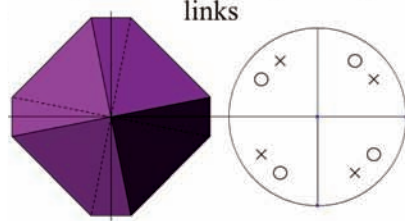
Trapezoeder {321}

rechts



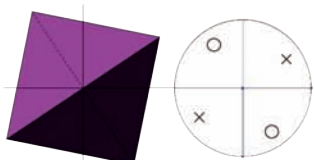
Trapezoeder {231}

links



Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente; es handelt sich hier lediglich um Bezugsflächen

## Tetragonales Disphenoid

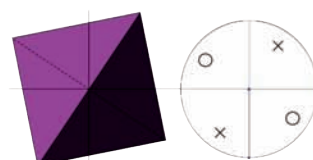
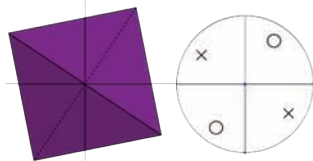


links  
positiv

{321}

links  
negativ

{3̄21}

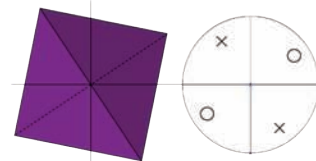


rechts  
positiv

{231}

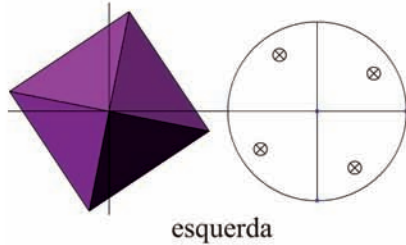
rechts  
negativ

{2̄31}

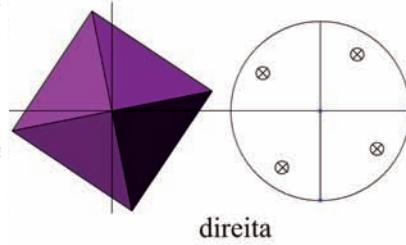


# Sistema Cristalino Tetragonal

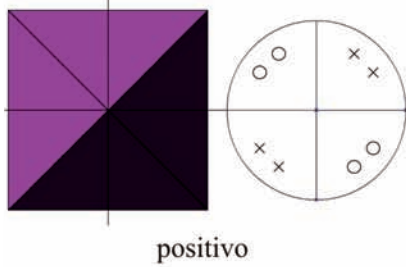
Dipirâmide  $\{321\}$



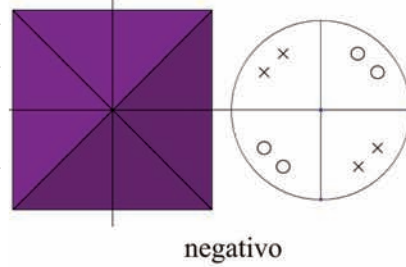
Dipirâmide  $\{231\}$



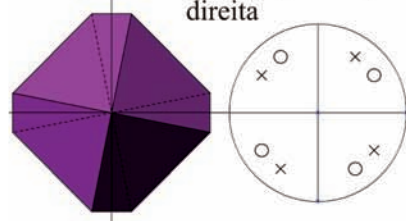
Escalenoedro  $\{321\}$



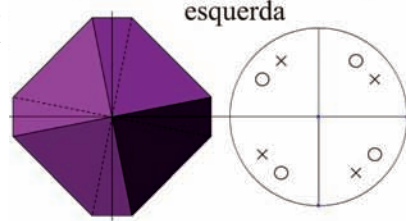
Escalenoedro  $\{\bar{3}21\}$



Trapezoedro  $\{321\}$



Trapezoedro  $\{231\}$



Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

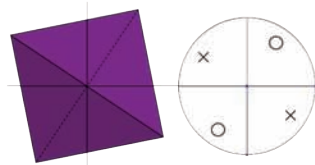
## Difenoide tetragonal



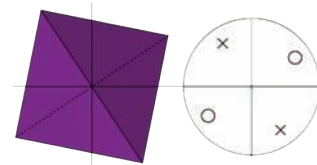
esquerda negativa  
 $\{321\}$



direita negativa  
 $\{\bar{2}31\}$

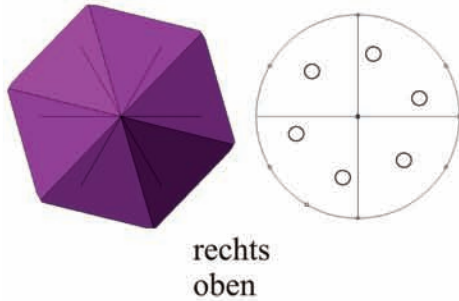


direita positiva  
 $\{231\}$

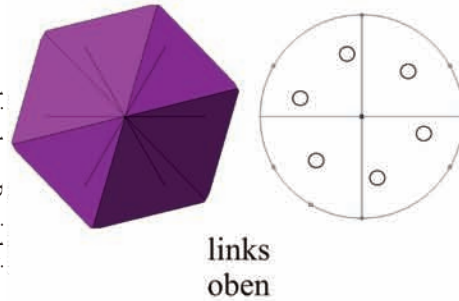


# Hexagonales Kristallsystem

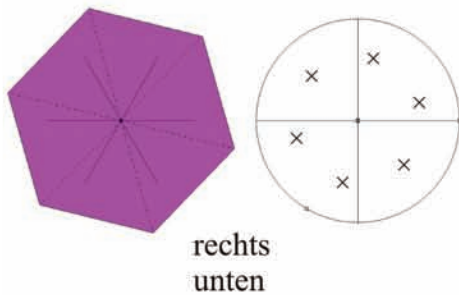
Pyramide  $\{132\}$



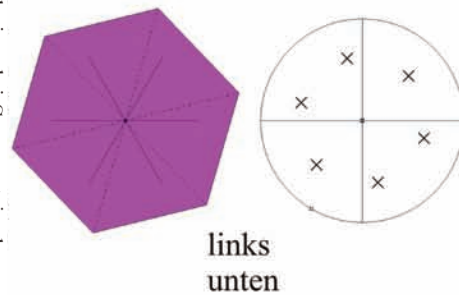
Pyramide  $\{312\}$



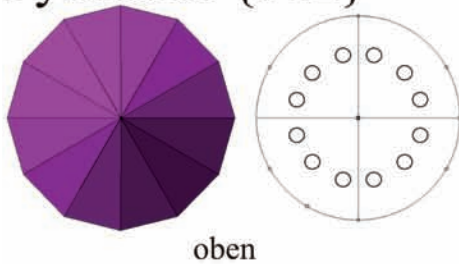
Pyramide  $\{13\bar{2}\}$



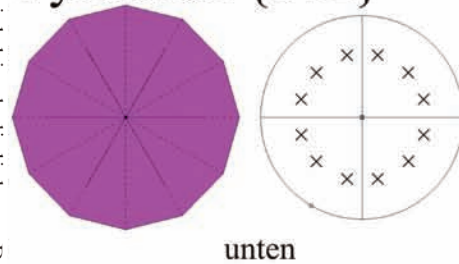
Pyramide  $\{31\bar{2}\}$



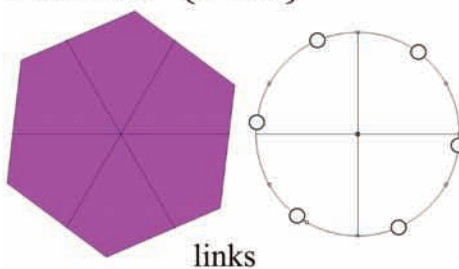
Dihexagonale  
Pyramide  $\{312\}$



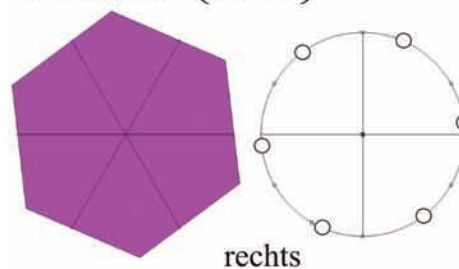
Dihexagonale  
Pyramide  $\{31\bar{2}\}$



Prisma  $\{320\}$



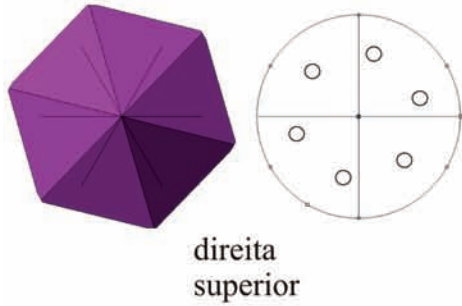
Prisma  $\{230\}$



Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

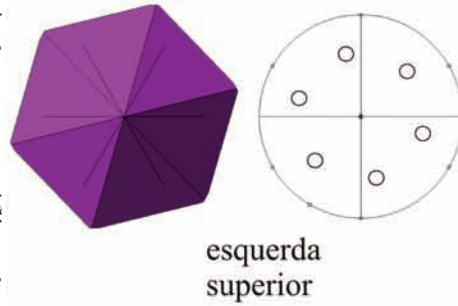
# Sistema Cristalino Hexagonal

Pirâmide  $\{132\}$



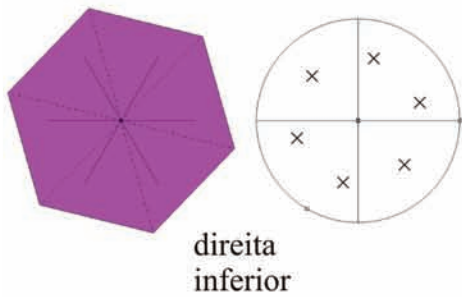
direita superior

Pirâmide  $\{312\}$



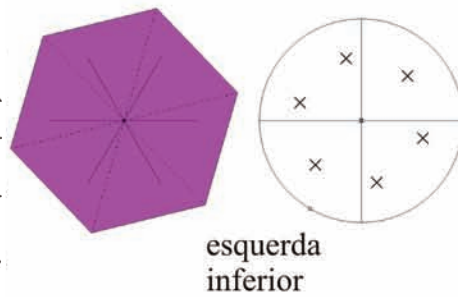
esquerda superior

Pirâmide  $\{13\bar{2}\}$



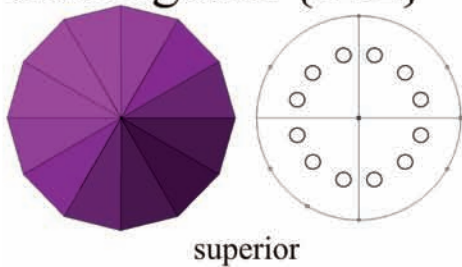
direita inferior

Pirâmide  $\{31\bar{2}\}$



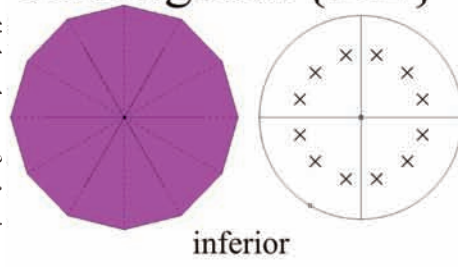
esquerda inferior

Pirâmide dihexagonal  $\{312\}$



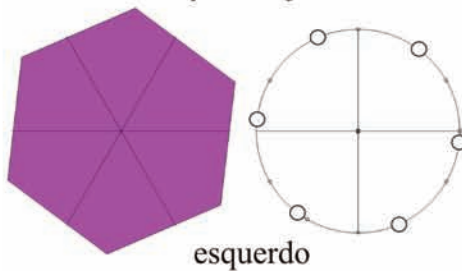
superior

Pirâmide dihexagonal  $\{31\bar{2}\}$



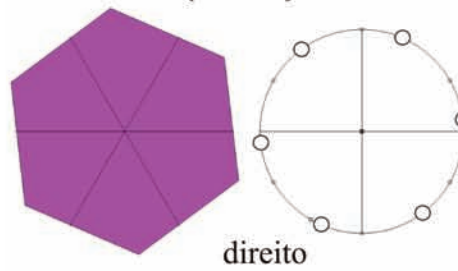
inferior

Prisma  $\{320\}$



esquerdo

Prisma  $\{230\}$

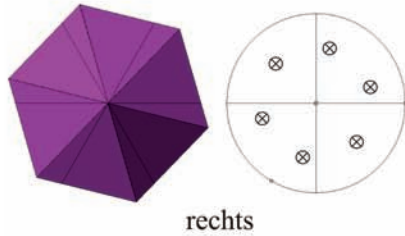


direito

Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

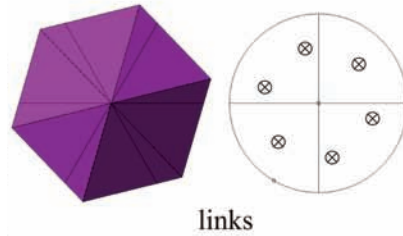
# Hexagonales Kristallsystem

Dipyramide {132}



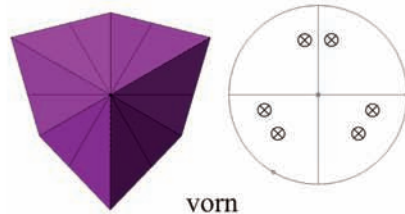
rechts

Dipyramide {312}



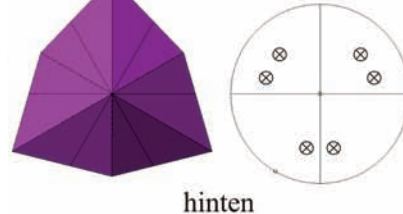
links

Ditrigonale Dipyramide {132}



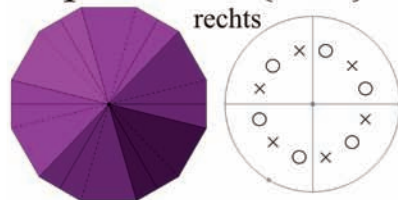
vorn

Ditrigonale Dipyramide {312}



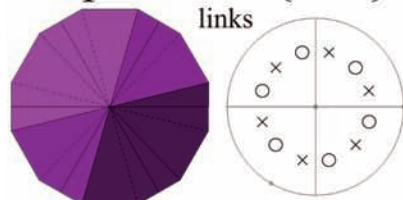
hinten

Trapezoeder {132}



rechts

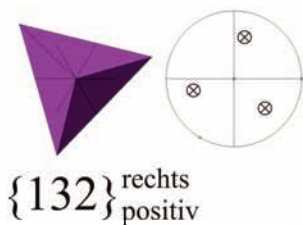
Trapezoeder {312}



links

Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente; es handelt sich hier lediglich um Bezugslinien

## Trigonale Dipyramide



{132} rechts positiv

$\bar{1}\bar{3}2$  rechts negativ



{312} links positiv

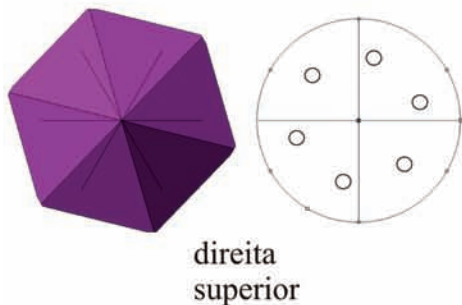


$\bar{3}\bar{1}2$  links negativ



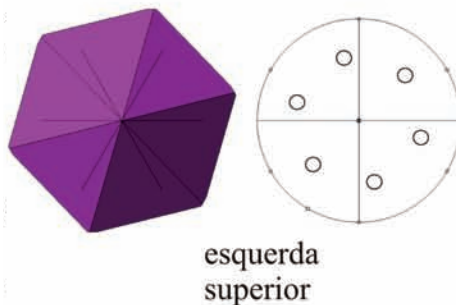
# Sistema Cristalino Hexagonal

Pirâmide {132}



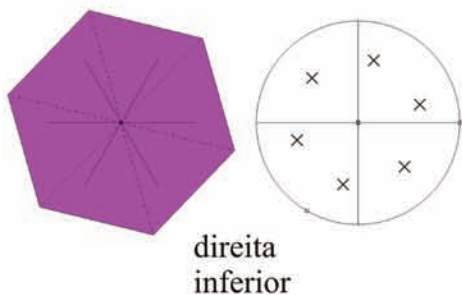
direita superior

Pirâmide {312}



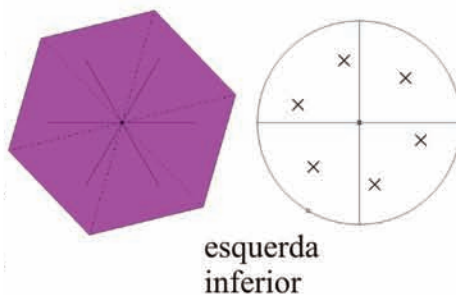
esquerda superior

Pirâmide {13 $\bar{2}$ }



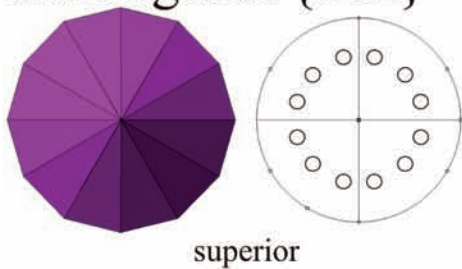
direita inferior

Pirâmide {31 $\bar{2}$ }



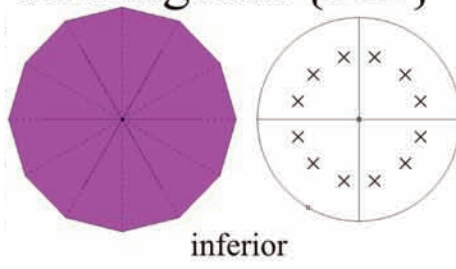
esquerda inferior

Pirâmide dihexagonal {312}



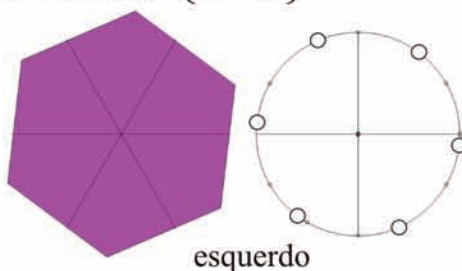
superior

Pirâmide dihexagonal {31 $\bar{2}$ }



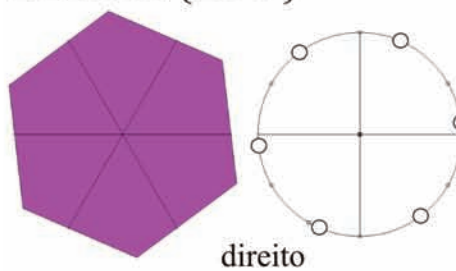
inferior

Prisma {320}



esquerdo

Prisma {230}

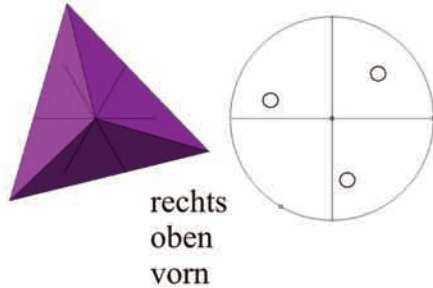


direito

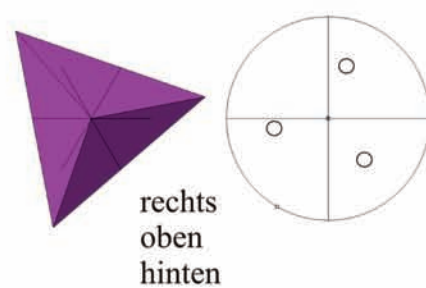
Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

# Trigonales Kristallsystem

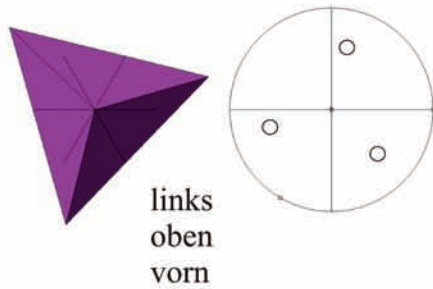
Pyramide  $\{312\}$



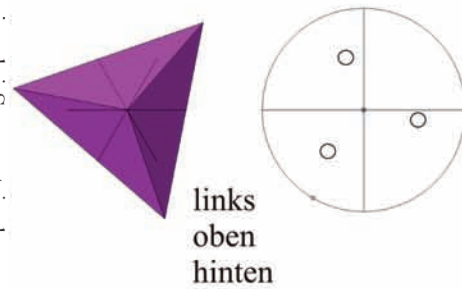
Pyramide  $\{\bar{3}12\}$



Pyramide  $\{132\}$

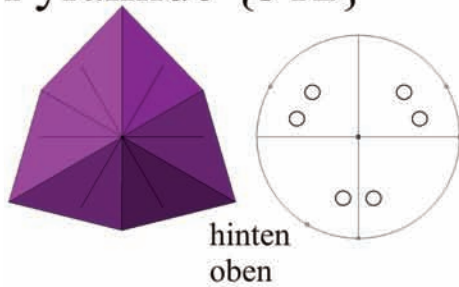


Pyramide  $\{\bar{1}32\}$

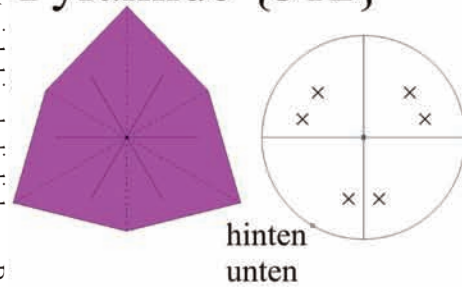


analog  
dazu  
„unten“

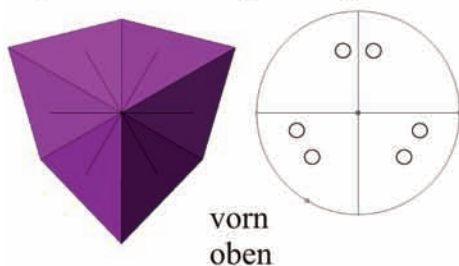
Ditrigonale  
Pyramide  $\{312\}$



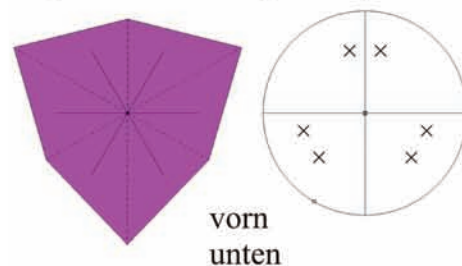
Ditrigonale  
Pyramide  $\{31\bar{2}\}$



Ditrigonale  
Pyramide  $\{132\}$



Ditrigonale  
Pyramide  $\{13\bar{2}\}$

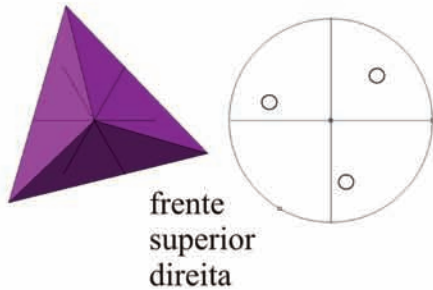


Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

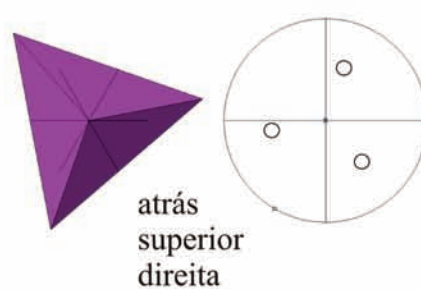


# Sistema Cristalino Trigonal

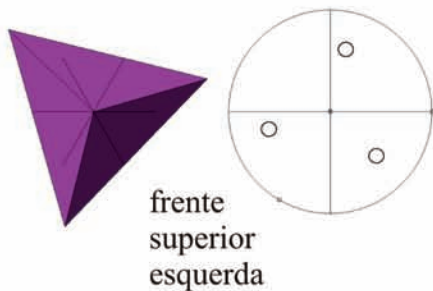
Pirâmide  $\{312\}$



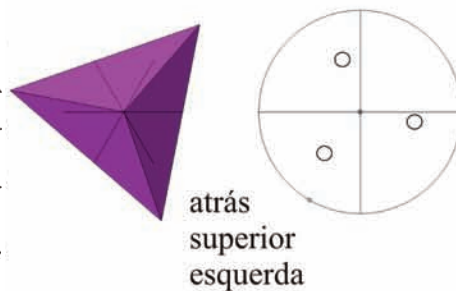
Pirâmide  $\{\bar{3}12\}$



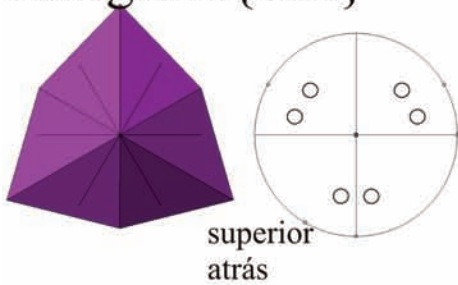
Pirâmide  $\{132\}$



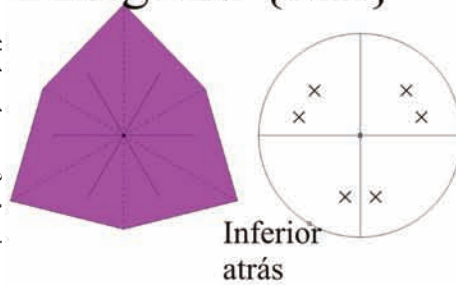
Pirâmide  $\{\bar{1}32\}$



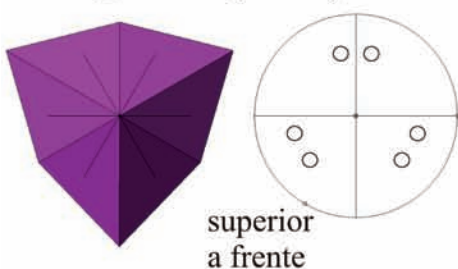
Pirâmide Ditrigonal  $\{312\}$



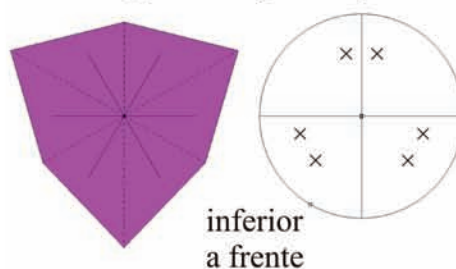
Pirâmide Ditrigonal  $\{31\bar{2}\}$



Pirâmide ditrigonal  $\{132\}$



Pirâmide Ditrigonal  $\{13\bar{2}\}$

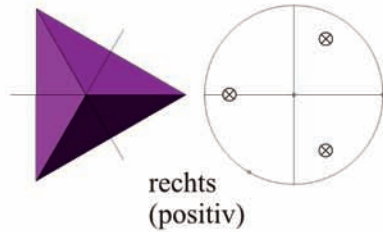


Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

análogo a parte inferior

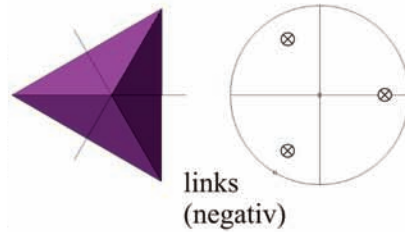
# Trigonales Kristallsystem

Dipyramide  $\{332\}$



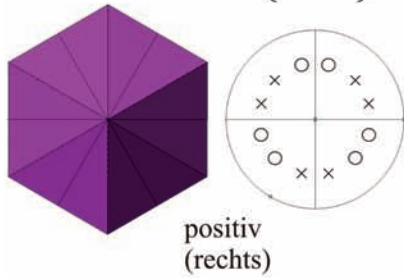
rechts  
(positiv)

Dipyramide  $\{\bar{3}32\}$



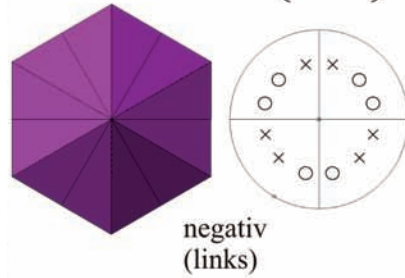
links  
(negativ)

Ditrigonaler  
Skalenoeder  $\{132\}$



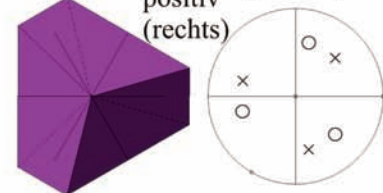
positiv  
(rechts)

Ditrigonaler  
Skalenoeder  $\{312\}$



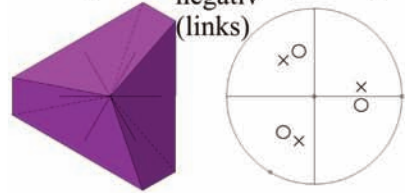
negativ  
(links)

Trapezoeder  $\{132\}$



positiv  
(rechts)

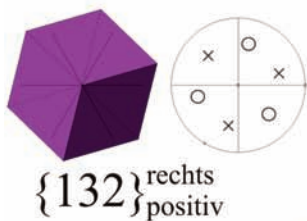
Trapezoeder  $\{\bar{1}32\}$



negativ  
(links)

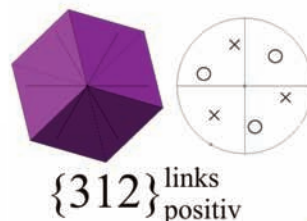
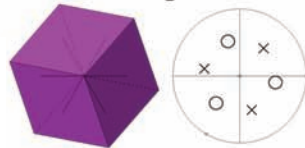
Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente; es handelt sich hier lediglich um Bezugsformen

## Trigonaler Rhomboeder



$\{132\}$  rechts  
positiv

$\{\bar{1}32\}$  rechts  
negativ



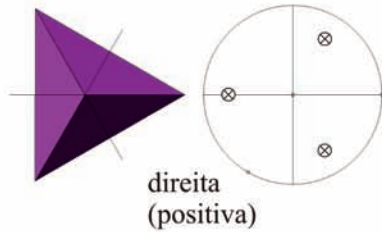
$\{312\}$  links  
positiv

$\{\bar{3}12\}$  links  
negativ

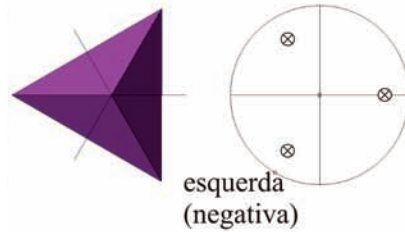


# Sistema Cristalino Trigonal

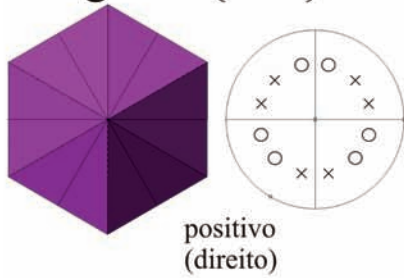
Bipirâmide  $\{332\}$



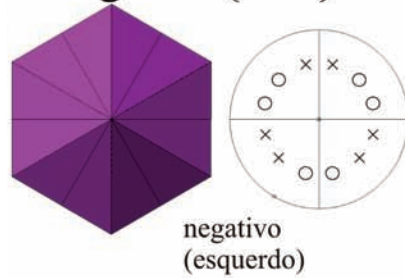
Bipirâmide  $\{\bar{3}\bar{3}2\}$



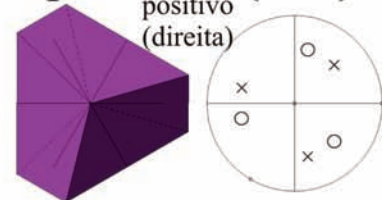
Escalenoedro ditrigonal  $\{132\}$



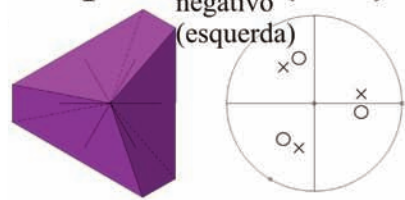
Escalenoedro ditrigonal  $\{312\}$



Trapezoedro  $\{132\}$



Trapezoedro  $\{\bar{1}\bar{3}2\}$

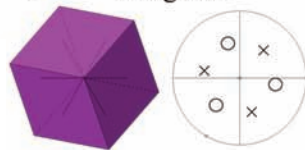


Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência

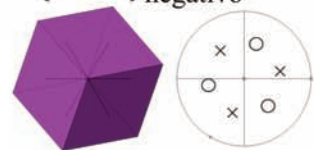
## Romboedro Trigonal



$\{\bar{1}\bar{3}2\}$  direito  
negativo

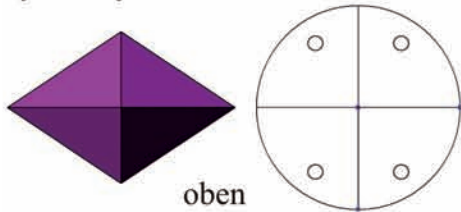


$\{\bar{3}\bar{1}2\}$  esquerdo  
negativo

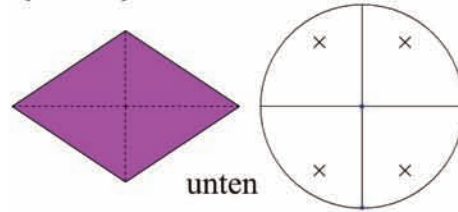


# Orthorhombisches Kristallsystem

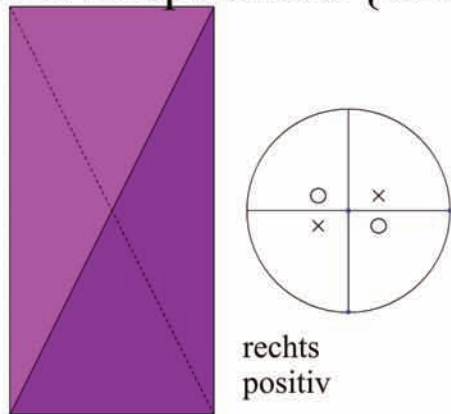
Rhombische Pyramide  
{321}



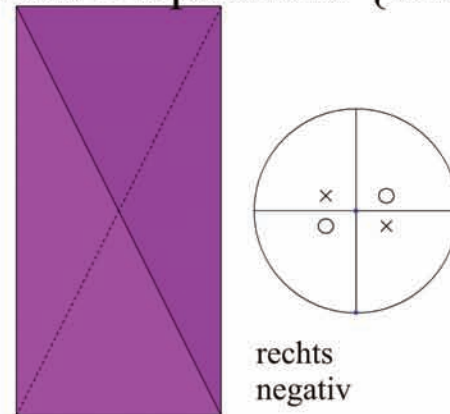
Rhombische Pyramide  
{321}



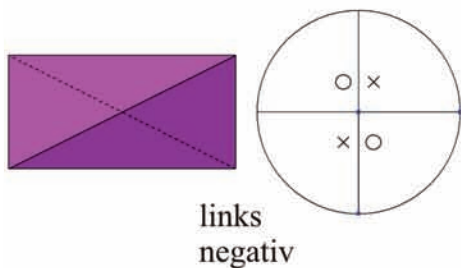
Rh. Disphenoid {123}



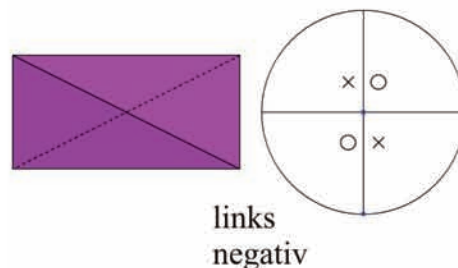
Rh. Disphenoid {1̄23}



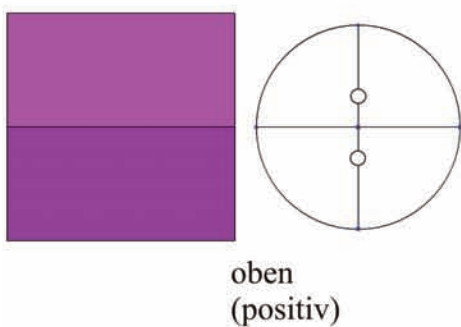
Rh. Disphenoid {213}



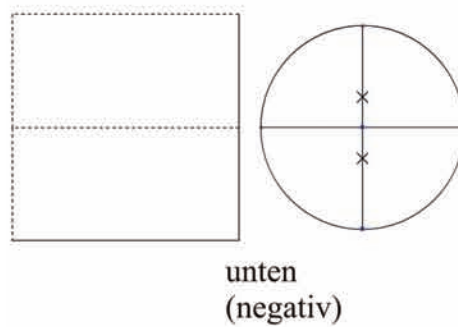
Rh. Disphenoid {2̄13}



Doma II.Stellung  
{203}



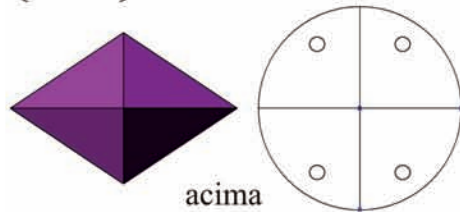
Doma II.Stellung  
{2̄03}



Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

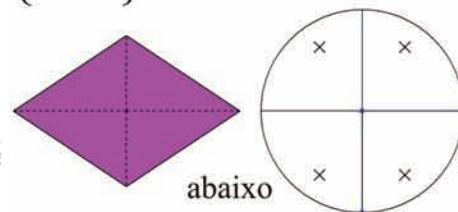
# Sistema Cristalino Ortorrômbo

Pirâmide ortorrômboica  
{321}



acima

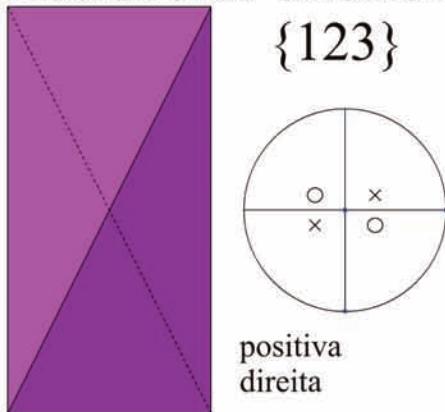
Pirâmide ortorrômboica  
{321}



abaixo

Diesfenóide ortorrômboica

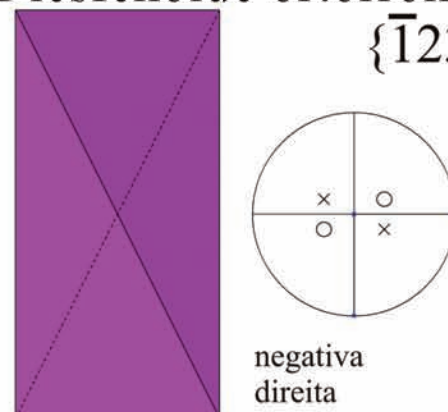
{123}



positiva  
direita

Diesfenóide ortorrômboica

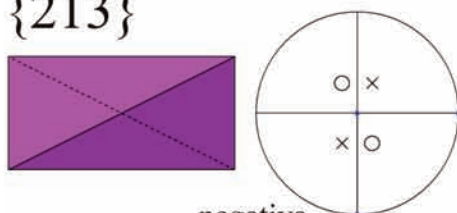
{1̄23}



negativa  
direita

Diesfenóide ortorrômboica

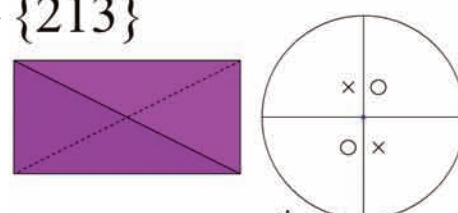
{213}



negativa  
esquerda

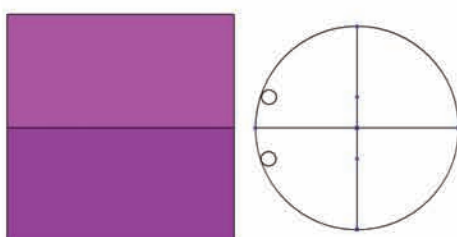
Diesfenóide ortorrômboica

{2̄13}



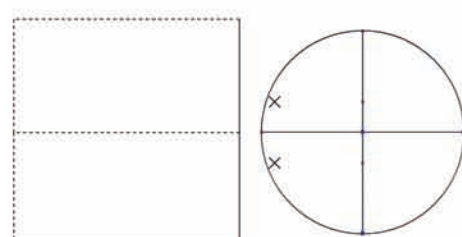
negativa  
esquerda

Posição II Domo  
{203}



acima  
(positivo)

Posição II Domo  
{2̄03}



abaixo  
(negativo)

Essas projeções estereográficas não envolvem elementos de simetria, estão aqui apenas para linha de preferência



Sistemas	Classes	Elementos de Si-	Direções polares no cristal	Direções apolares no cristal
monoclínico				
tetragonal				
trigonal				
hexagonal				
	$I\bar{2}m$	$1\ 2\ 3\ 4\ 6\ 8$	todas $\bullet$	$a_1, a_2, a_3$ e todas zonas que se situam $\square$
Cúbico ou iso-métrico				

Outras classes sem centro de simetria mas com direções polares: 222, 4, 422, 42m, 622, 432

\* = Classes com eixos de simetria polar único, para as quais não existe um único eixo equivalente: materiais ferroelétricos

## Die Kombination von Formen

Zwei Formen:

Würfel ↔ Oktaeder

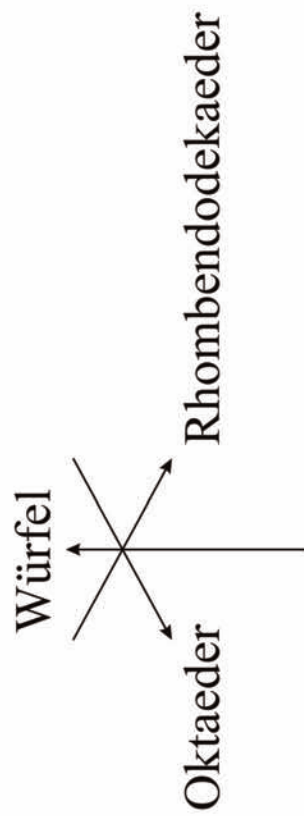
Rhombendodekaeder ↔  
 Würfel  
 Oktaeder

Tetraeder ↔  
 Würfel  
 Rhombendodekaeder

positiver Tetraeder ↔ negativer Tetraeder

Drei Formen:

Würfel  
 Oktaeder  
 Rhombendodekaeder

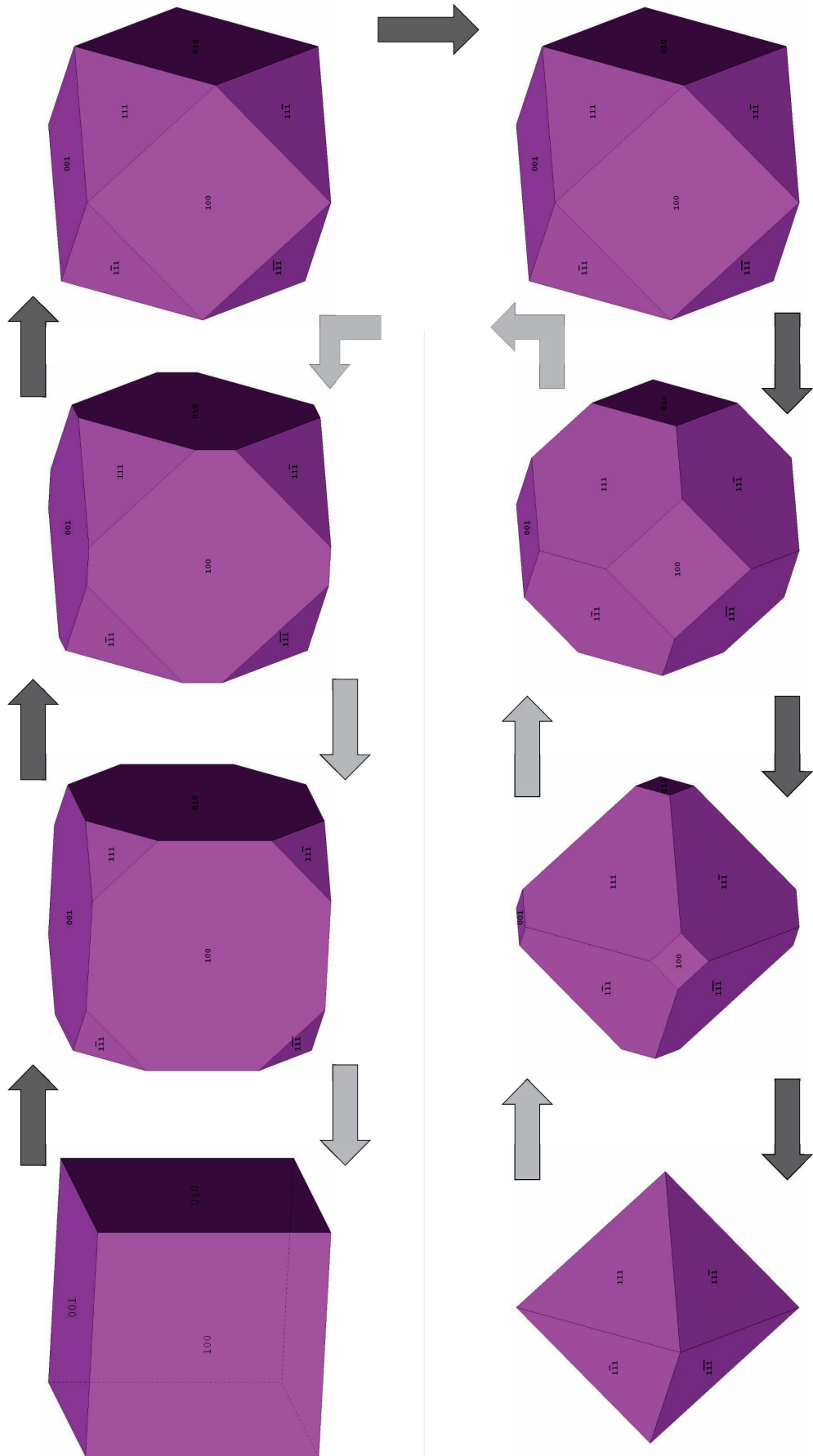


Kombinationen von Flächen und Steilen und Steilen Rhomboedern

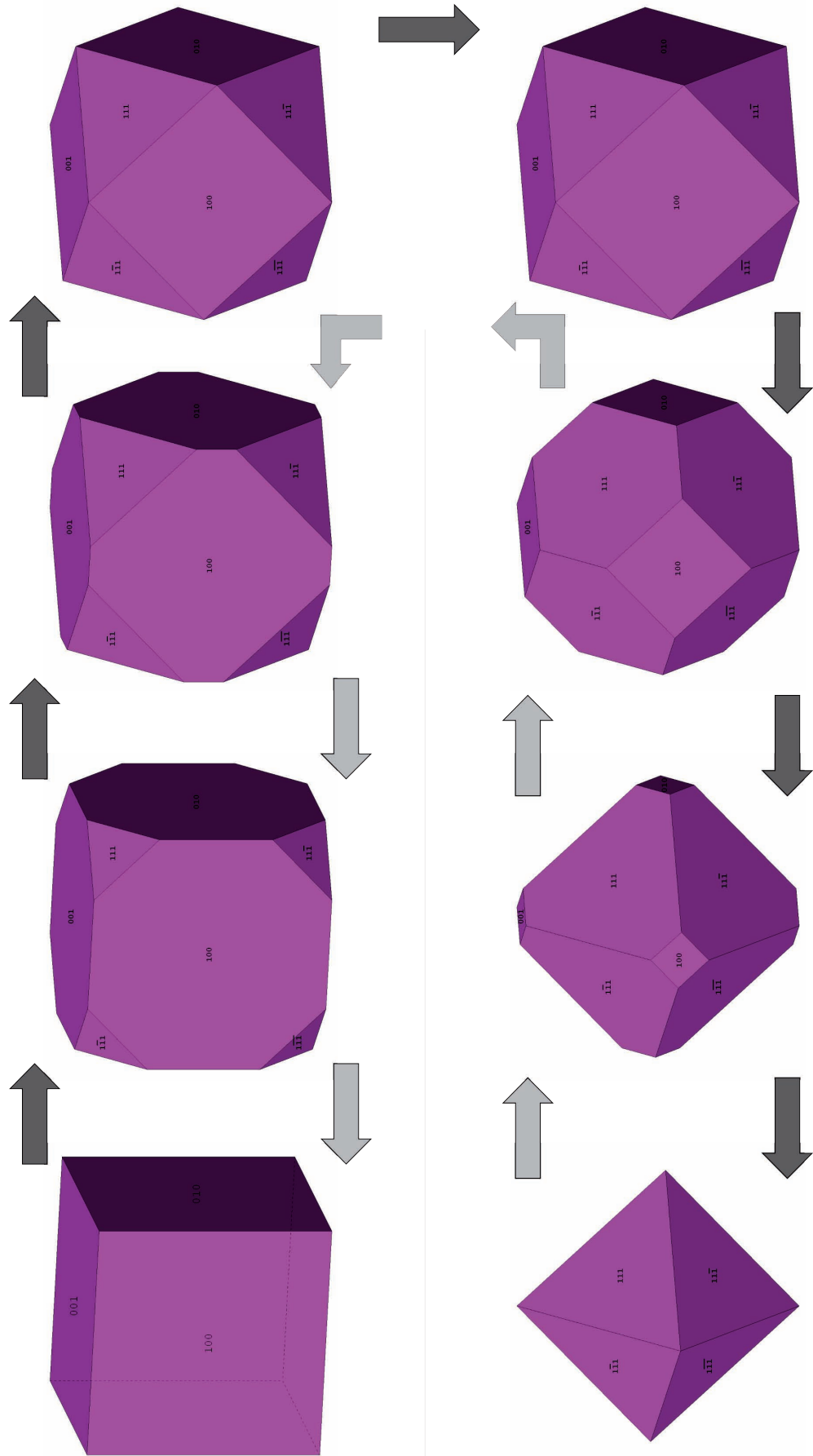




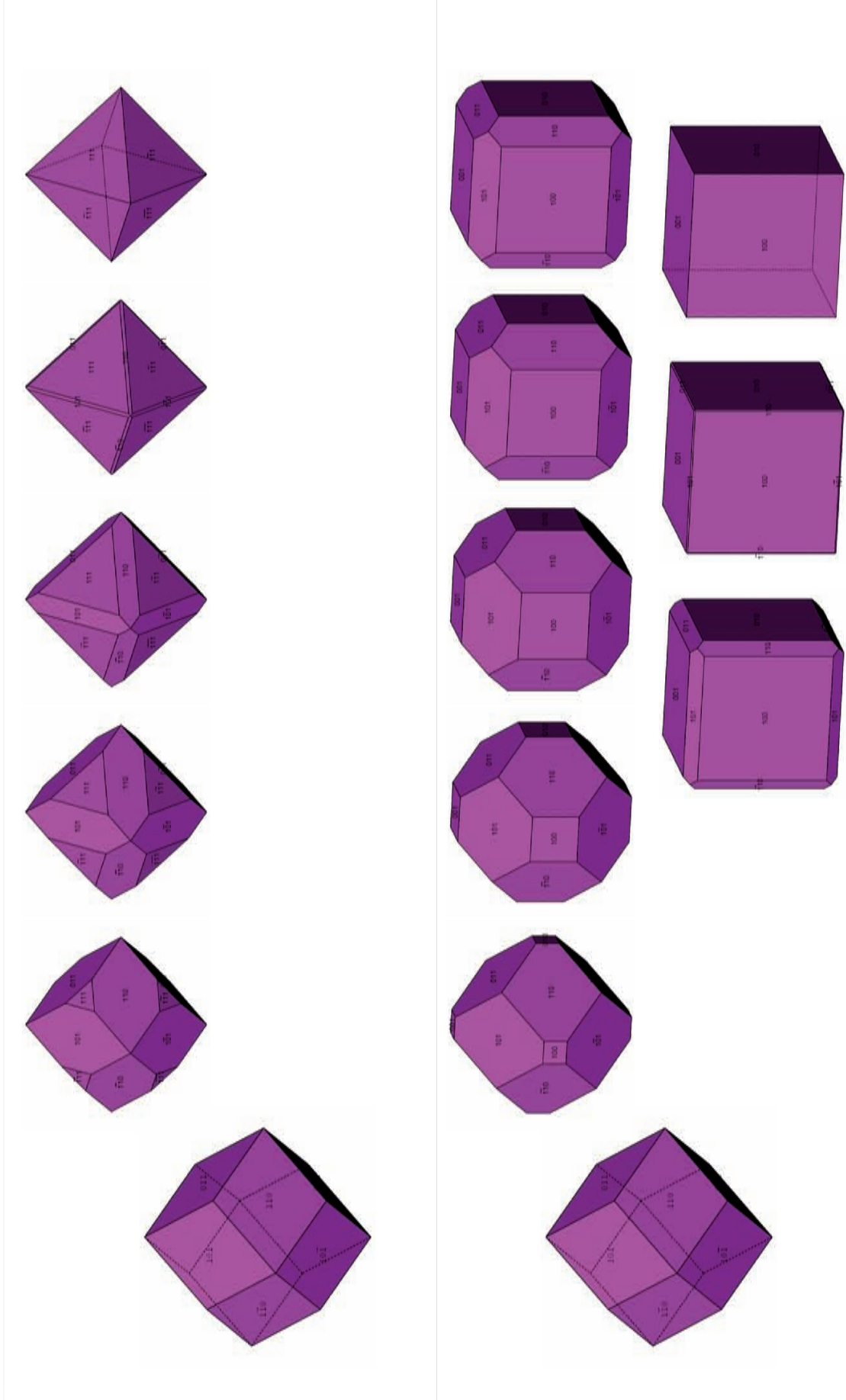
# Vom Würfel zum Oktaeder



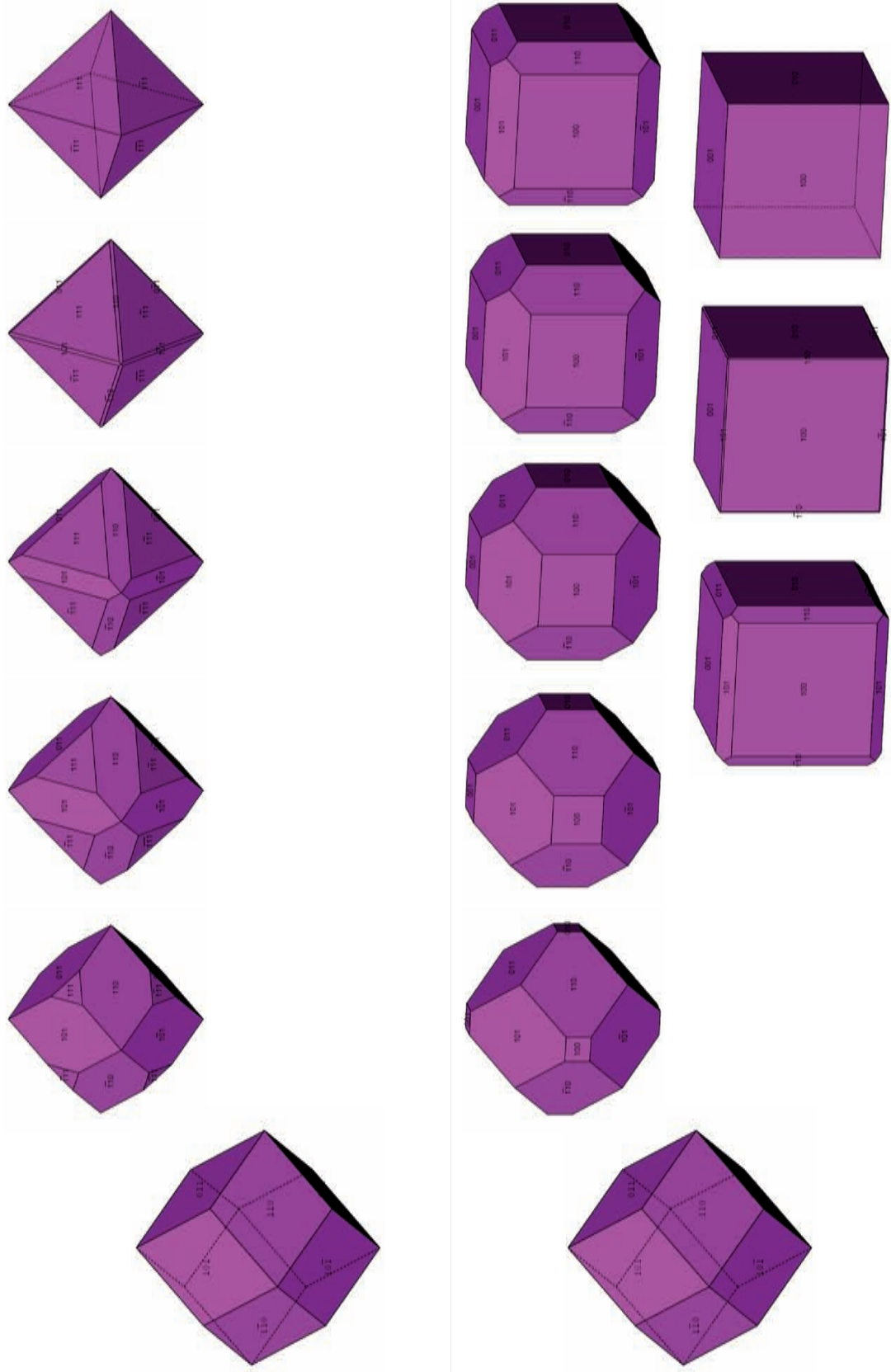
# De cubo para octaedro



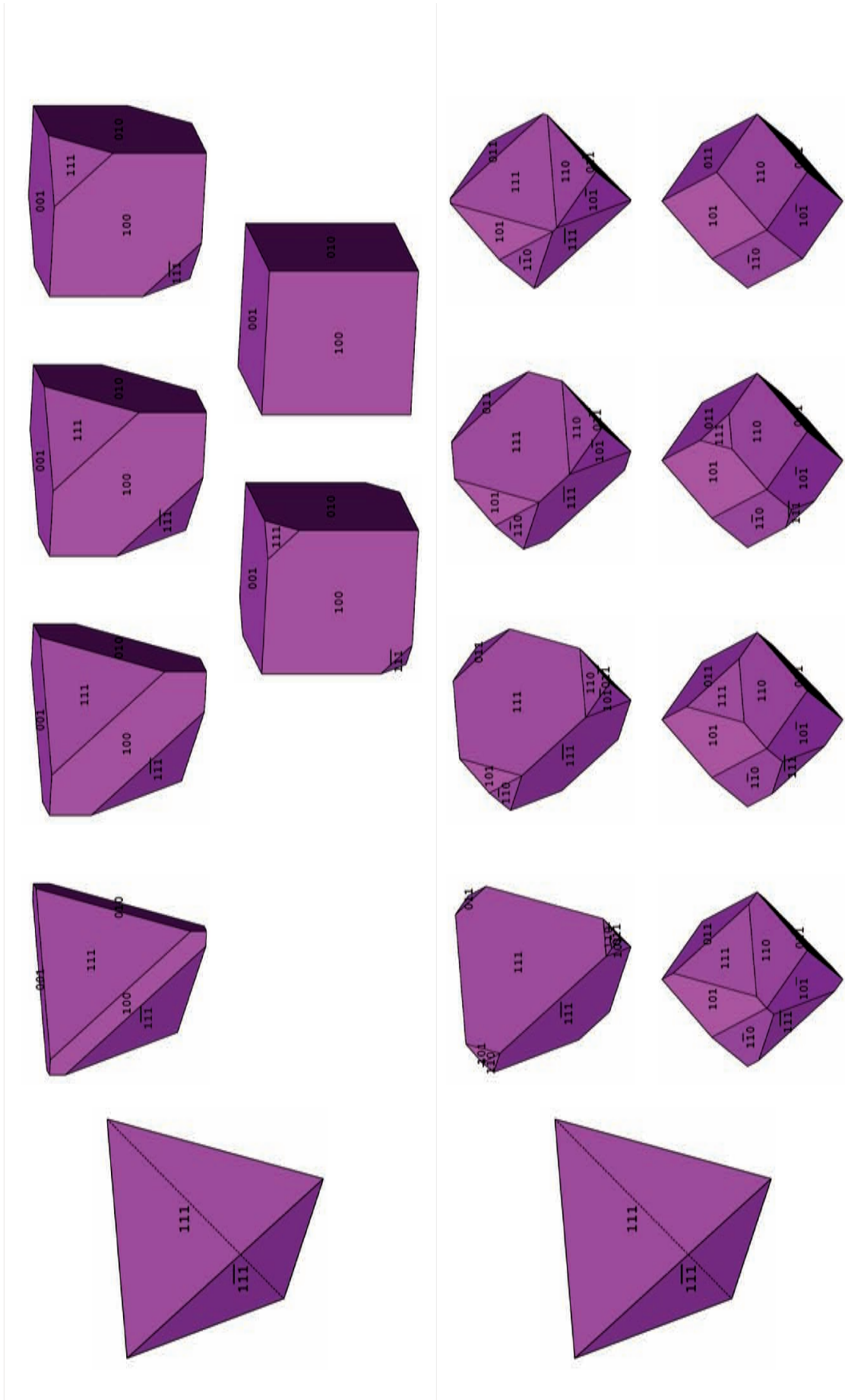
# Kombination des Rhombendodekaeder mit Oktaeder und Würfel



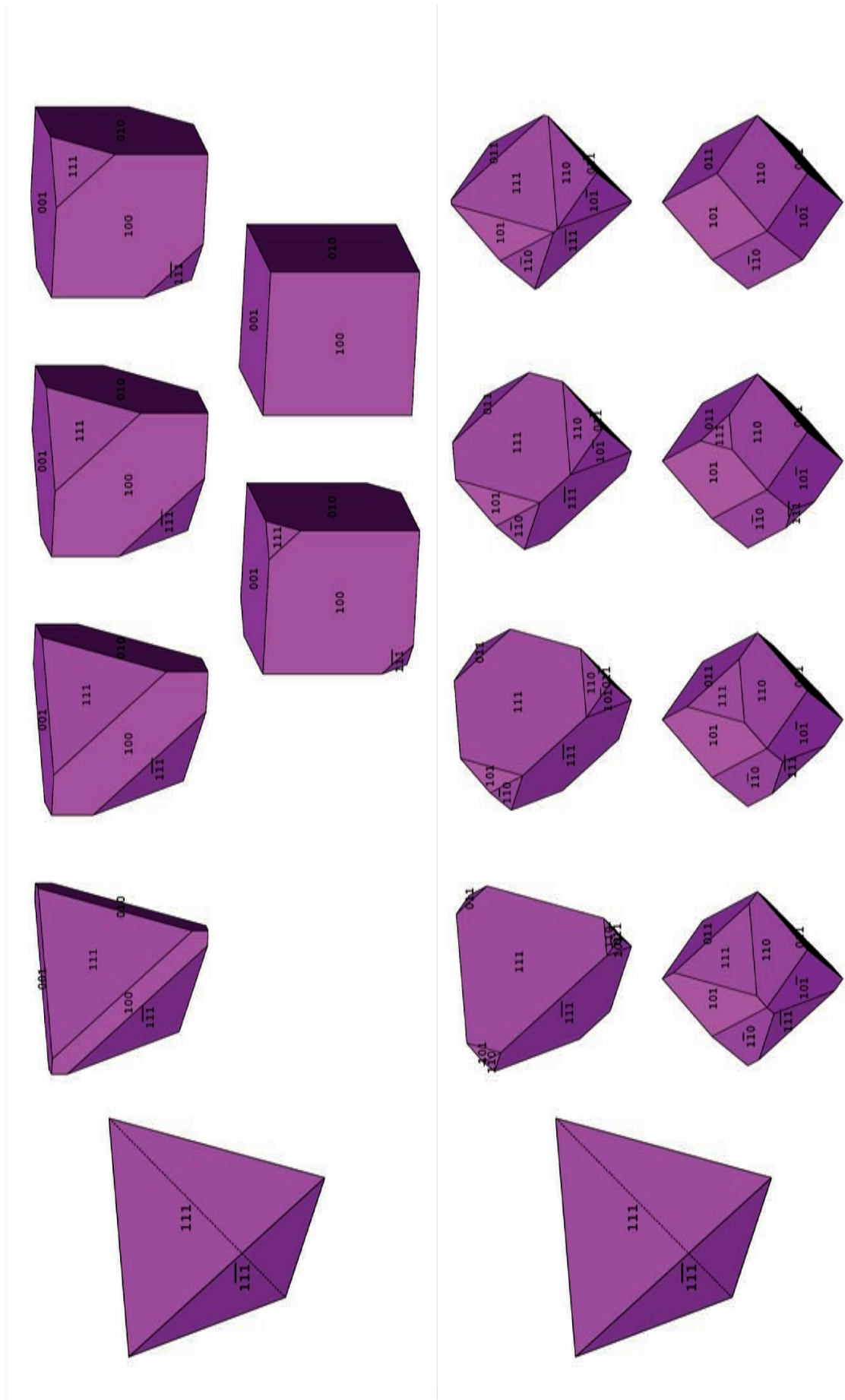
# Combinação de rombododecaedro com octaedro e cubo



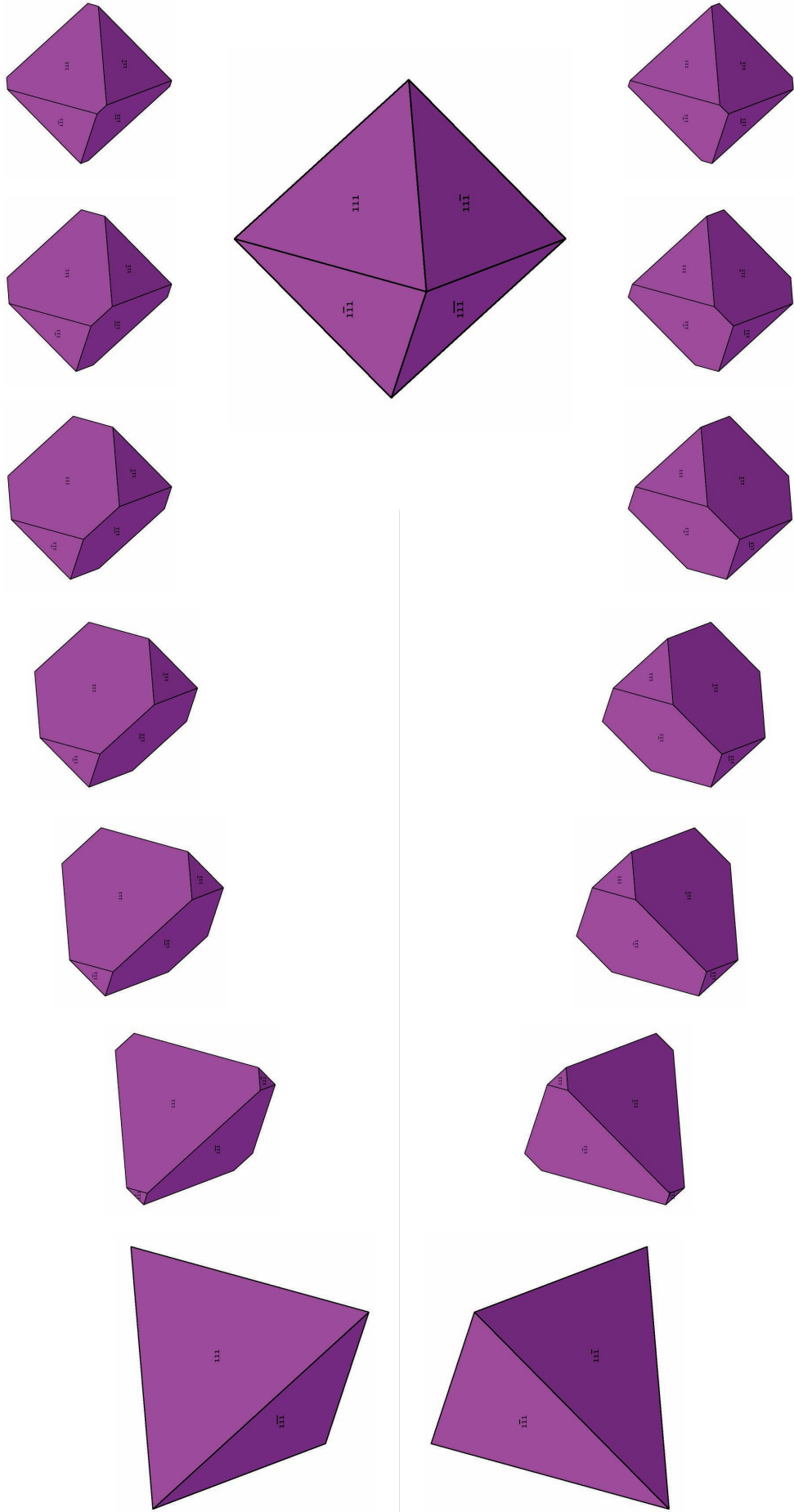
# Kombination des Tetraeder mit Würfel und Rhombendodekaeder



# Combinações de tetraedros com cubos e rombododecaedro

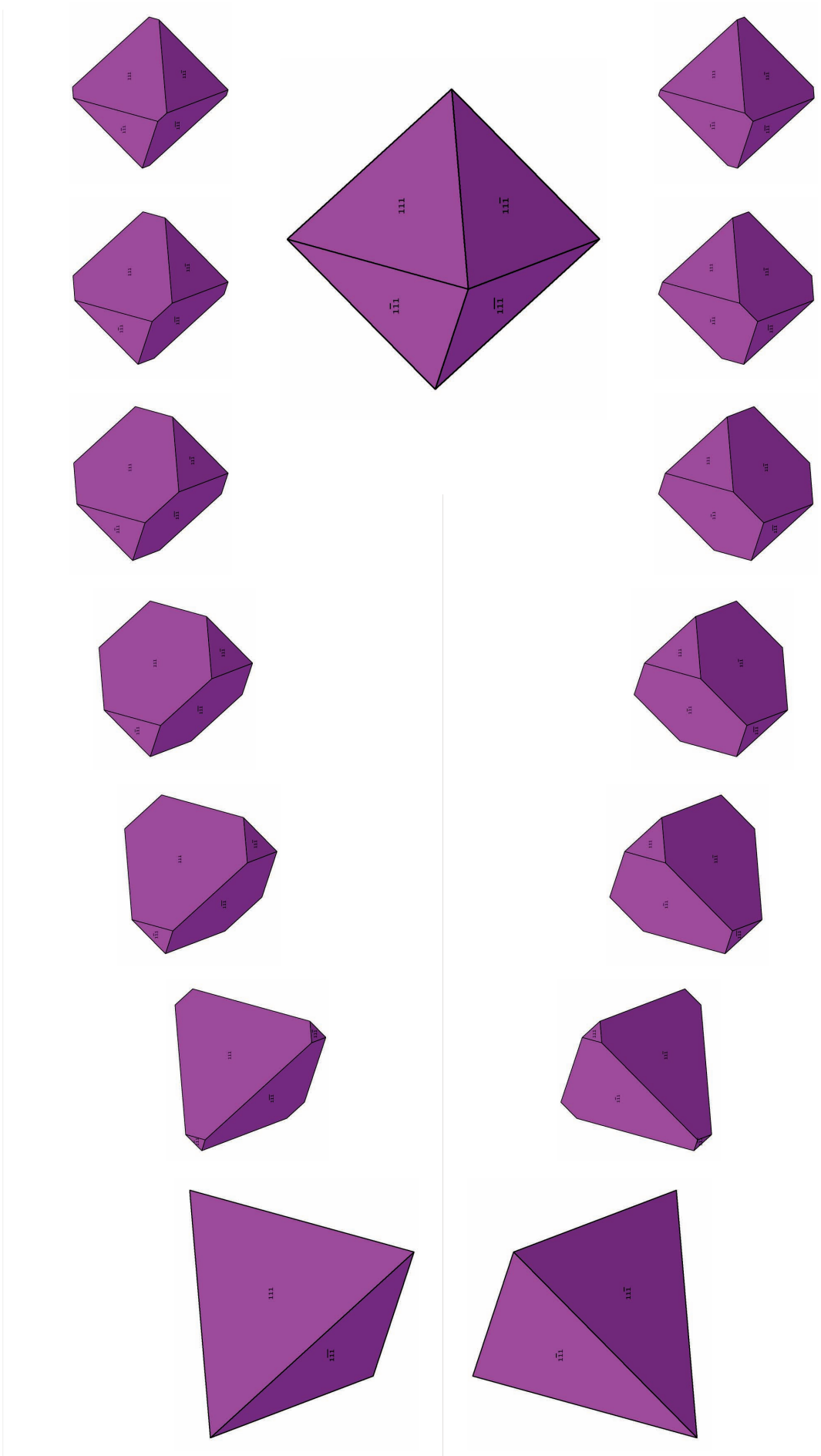


# Kombination von positivem und negativem Tetraeder

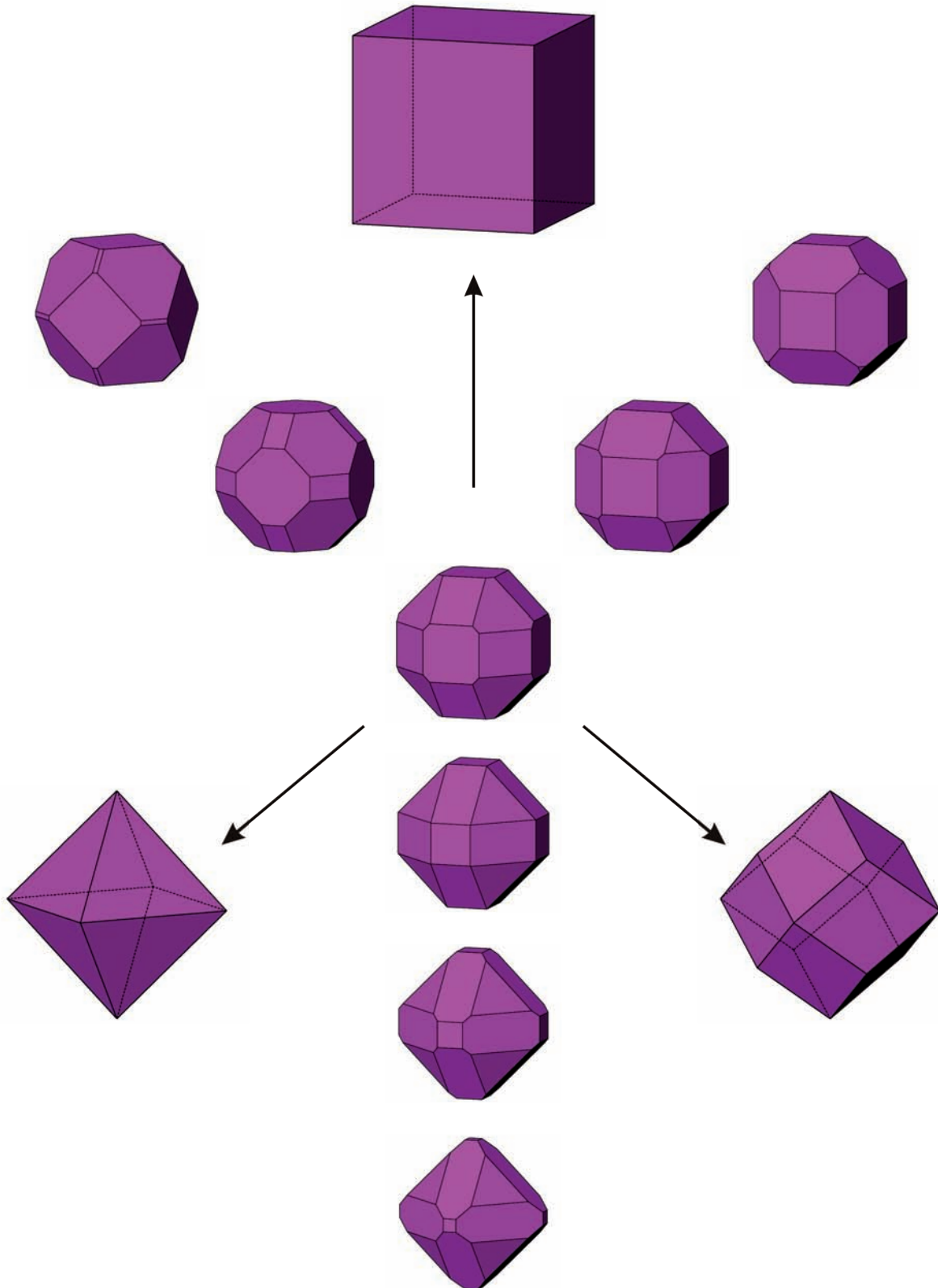




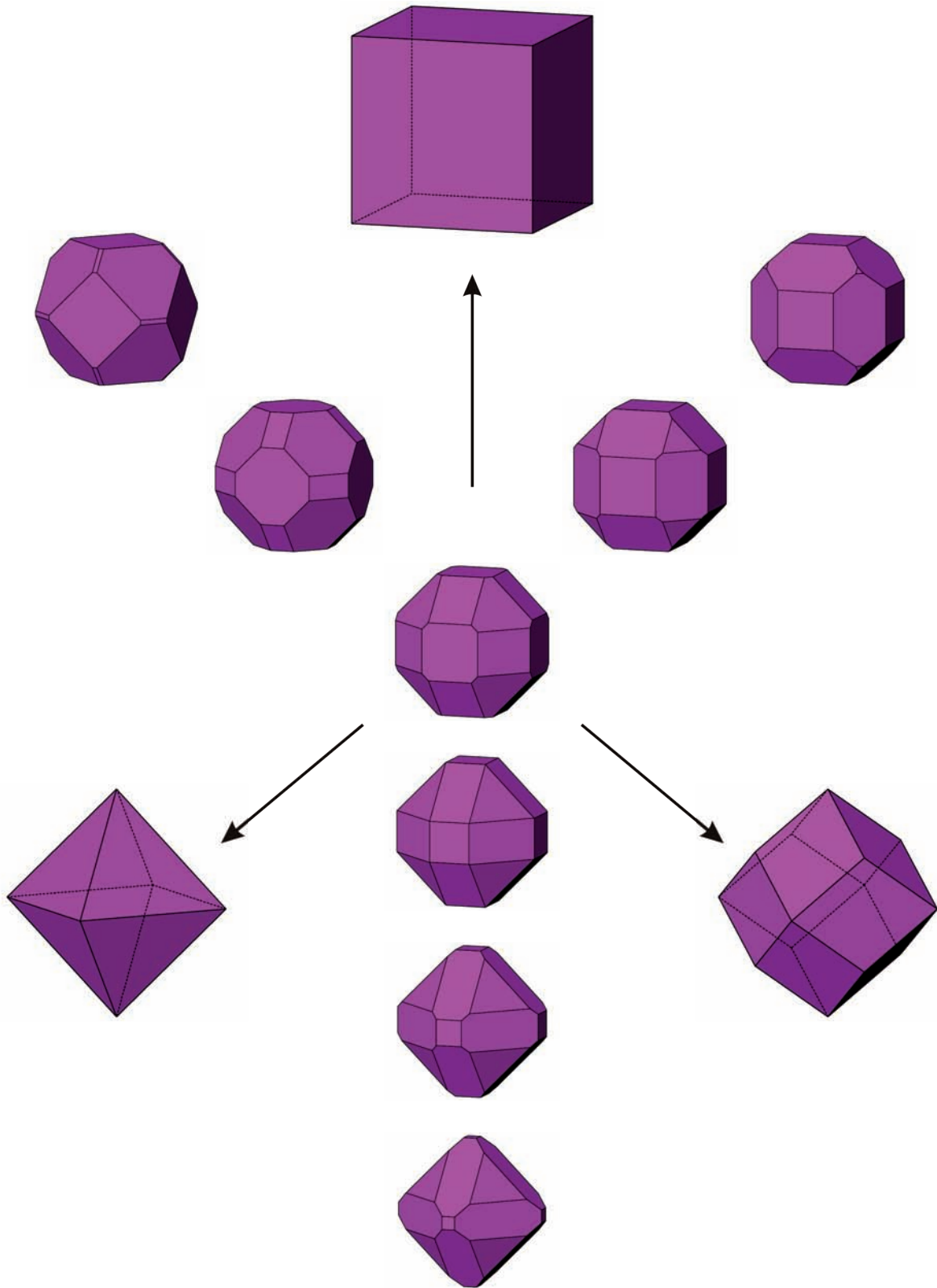
# Combinações de tetraedros positivos e negativos



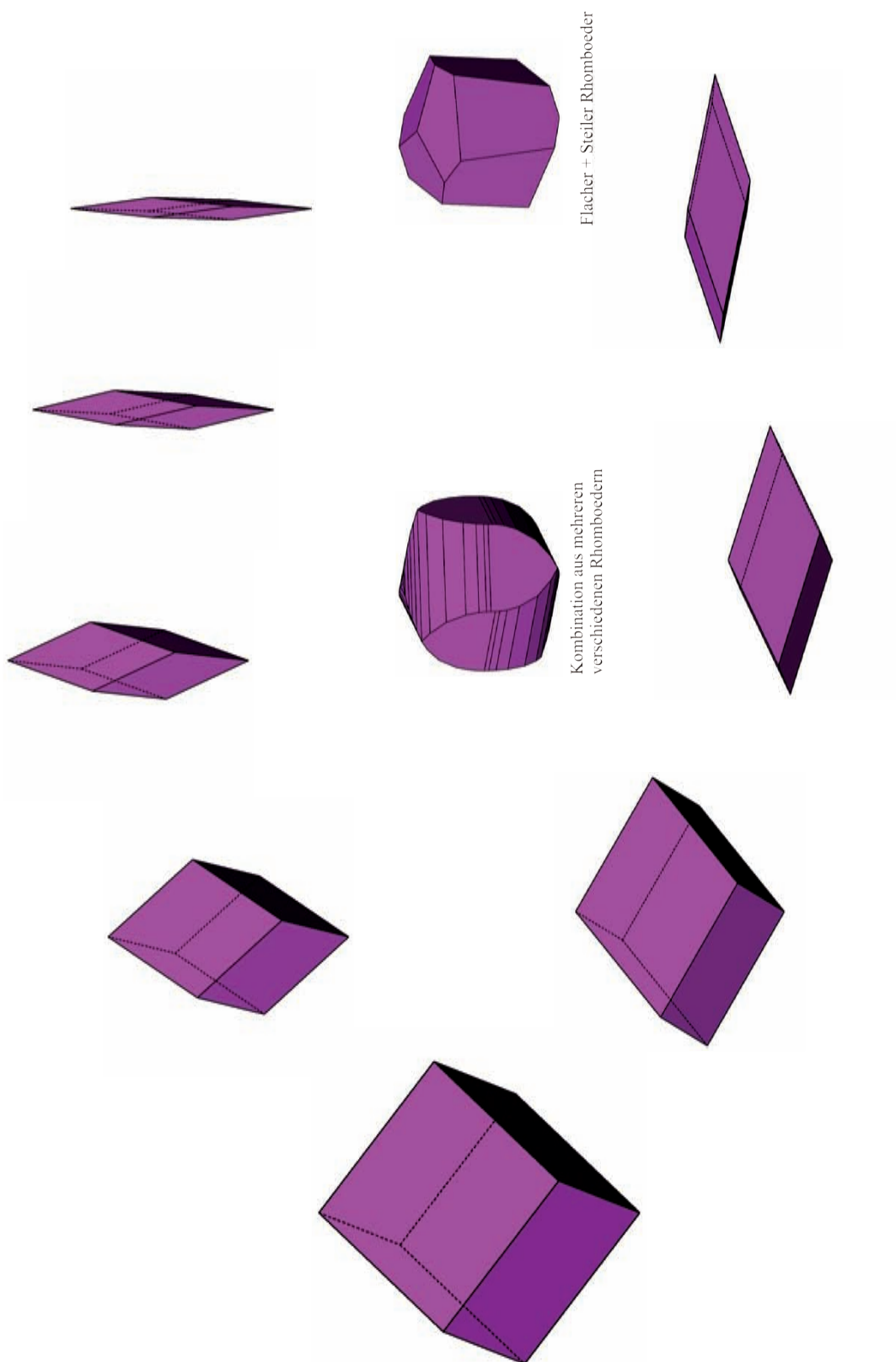
## Kombinationen aus Würfel, Oktaeder und Rhombendodekaeder



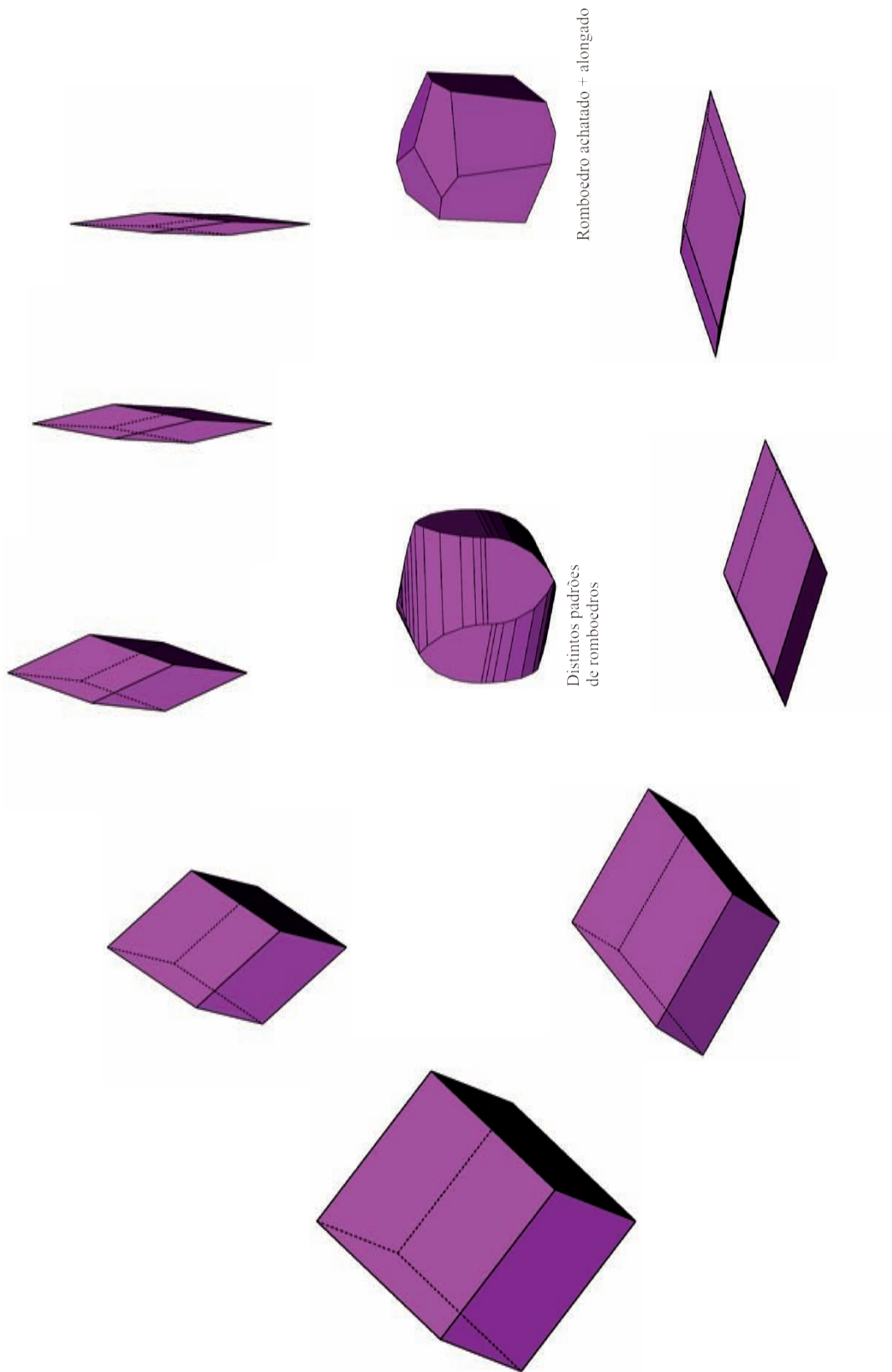
### Combinações de cubo, octaedro e rombododecaedro



## Verschieden steile und flache Rhomboeder

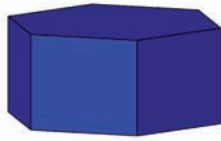


# Combinações de distintos estilos de romboedros achatados e alongados



## **Einige ausgewählte Formen und Kombinationen nach Mineralbeispielen**

**Alguns exemplos selecionados  
de combinações de formas  
entre os minerais**

Apatit:

Hexagonales Prisma  
Basispinakoid



Hexagonales Prisma  
Hex. Dipyramide I. St.  
Hex. Dipyramide II. St.



Hexagonales Prisma  
Basispinakoid  
Hex. Dipyramide I. St.



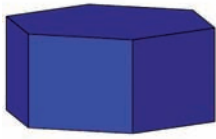
Hexagonales Prisma  
Hex. Dipyramide I. St.  
Hex. Dipyramide II. St.



Hexagonales Prisma  
Hex. Dipyramide I. St.  
Hex. Dipyramide II. St.

Chemische Zusammensetzung	: $\text{Ca}_5[(\text{PO}_4)_3(\text{F,Cl,OH})]$
Kristallsystem	: Hexagonal, 6/m
Elementarzelle	: $a_0=9.38 \text{ \AA}$ , $c_0=6.89 \text{ \AA}$
Raumgruppe	: $P 6_3/m$
Ausbildung	: Derb, eingewachsen, körnig, kryptokristallin, krustenartig, xx säulig, dicktafelig, nadelig

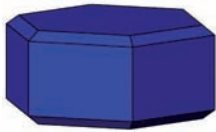


Apatita:

Prisma hexagonal  
Base pinacoidal



Prisma hexagonal  
Dipirâmide hexagonal I. pos.  
Dipirâmide II. Posição



Prisma hexagonal  
Base pinacoidal  
Dipirâmide dihexagonal  
I. pos.

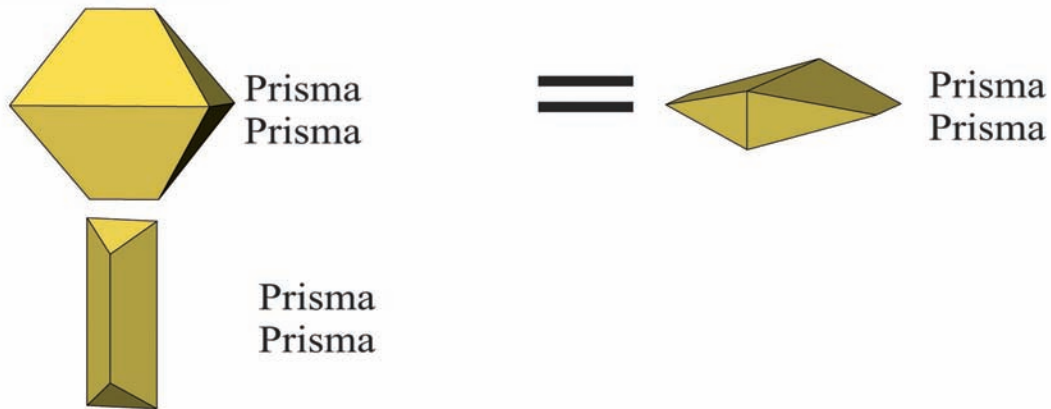


Prisma hexagonal  
Dipirâmide hexagonal I. pos.  
Dipirâmide II. Posição

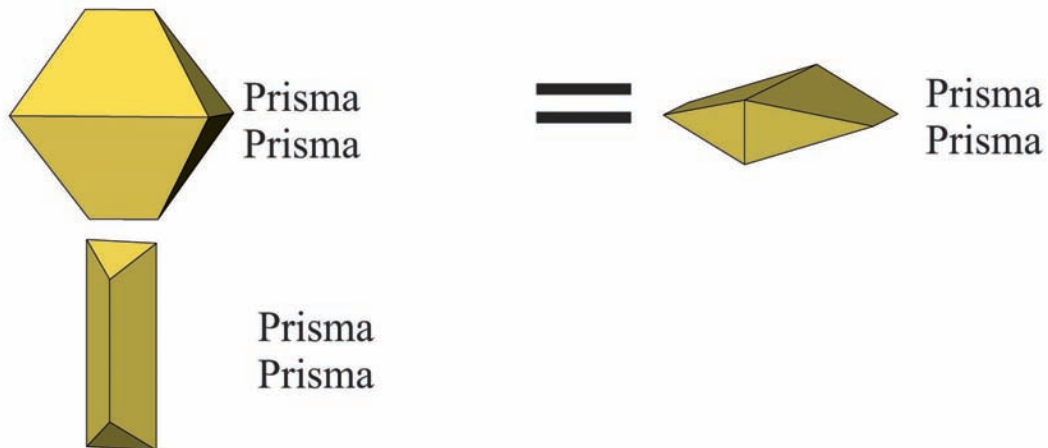


Prisma hexagonal  
Dipirâmide hexagonal I. pos.  
Dipirâmide II. Posição

Fórmula química	: $\text{Ca}_5[(\text{PO}_4)_3(\text{F}, \text{Cl}, \text{OH})]$
Sistema cristalino	: Hexagonal, 6/m
Célula unitária	: $a_0=9.38 \text{ \AA}$ , $c_0=6.89 \text{ \AA}$
Grupo espacial	: P 6 <sub>3</sub> /m
Características	: maciço, granular, criptocristalino, com intercrescimento, xx colunar, tabular, acicular

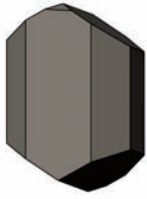
Arsenopyrit:**„Arsenkies“**

Chemische Zusammensetzung	: FeAsS
Kristallsystem	: Monoklin, pseudorhombisch
Elementarzelle	: $a_0 = 5.74 \text{ \AA}$ , $b_0 = 5.68 \text{ \AA}$ , $c_0 = 5.79 \text{ \AA}$ , $\beta = 112.17^\circ$
Raumgruppe	: $P 2_1/c$
Ausbildung	: idiomorphe Kristalle, körnig

Arsenopirita:

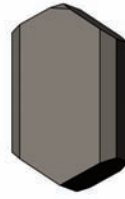
Fórmula química  
 Sistema cristalino  
 Célula unitária  
 Grupo espacial  
 Características

: FeAsS  
 : Monoclínico, pseudorômbico  
 :  $a_0 = 5.74 \text{ \AA}$ ,  $b_0 = 5.68 \text{ \AA}$ ,  $c_0 = 5.79 \text{ \AA}$ ,  $\beta = 112.17^\circ$   
 :  $P 2_1/c$   
 : cristais idiomorfos, granular

Augit:

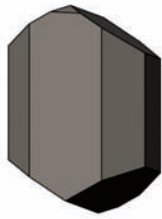
Pinakoid  
Prisma  
Pinakoid  
Pinakoid  
Prisma

=



Pinakoid  
Prisma  
Pinakoid  
Pinakoid  
Prisma

Chemische Zusammensetzung	: (Ca, Mg, Fe, Al, Ti) [(Si, Al) <sub>2</sub> O <sub>6</sub> ]
Kristallsystem	: Monoklin
Elementarzelle	: $a_0 = 9.69 \text{ \AA}$ , $b_0 = 8.84 \text{ \AA}$ , $c_0 = 5.28 \text{ \AA}$ , $\beta = 106.3^\circ$
Raumgruppe	: C 2/c
Ausbildung	: Kurzprismatische bis tafelige xx, achteckiger Querschnitt, oft zonar aufgebaut

Augita:

Pinacóide  
Prisma  
Pinacóide  
Pinacóide  
Prisma

=



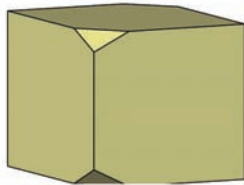
Pinacóide  
Prisma  
Pinacóide  
Pinacóide  
Prisma

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

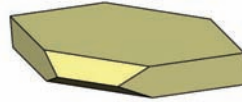
: (Ca, Mg, Fe, Al, Ti) [(Si, Al)<sub>2</sub>O<sub>6</sub>]  
: Monoclínico  
:  $a_0 = 9.69 \text{ \AA}$ ,  $b_0 = 8.84 \text{ \AA}$ ,  $c_0 = 5.28 \text{ \AA}$ ,  $\beta = 106.3^\circ$   
: C 2/c  
: de pequenos prismas até tabular xx; seções octógonas, geralmente zonado.

Baryt:

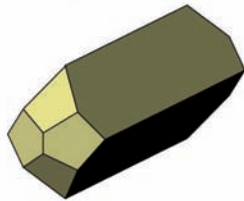
Basispinakoid  
Prisma



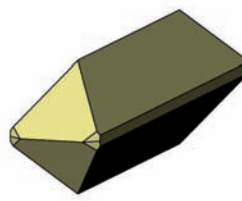
Basispinakoid  
Prisma  
Prisma



Basispinakoid  
Prisma  
Prisma



Prisma  
Prisma  
Prisma



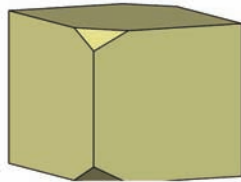
Prisma  
Prisma  
Pinakoid  
Dipyramide

Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

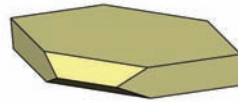
: BaSO<sub>4</sub>  
: Rhombisch, 2/m 2/m 2/m  
: a<sub>0</sub> = 8.88Å, b<sub>0</sub> = 5.45Å, c<sub>0</sub> = 7.15Å  
: P b n m  
: Dicht, feinkristallin, grobspätig, blättrig, xx meist tafelig oder meißelförmig

Barita:

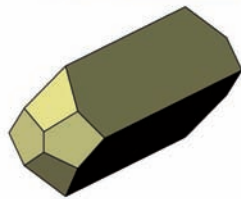
Pinacóide basal  
Prisma



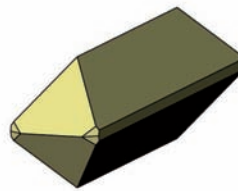
Pinacóide basal  
Prisma  
Prisma



Pinacóide basal  
Prisma  
Prisma



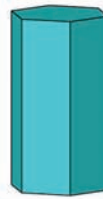
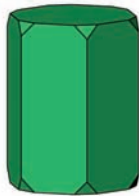
Prisma  
Prisma  
Prisma



Prisma  
Prisma  
Pinacóide  
Dipirâmide

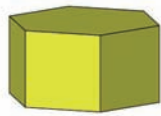
Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

: BaSO<sub>4</sub>  
: Rômbico, 2/m 2/m 2/m  
: a<sub>0</sub>= 8.88Å, b<sub>0</sub>= 5.45Å, c<sub>0</sub>= 7.15Å  
: P b n m  
: denso, fino, prismático, laminar, xx principalmente tabular ou miceliforme (bastonete)

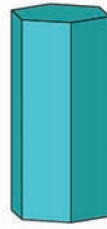
Beryll:Hexagonales Prisma  
BasispinakoidHexagonales Prisma  
BasispinakoidHexagonales Prisma  
Basispinakoid  
Hex. Dipyramide II. St.Hexagonales Prisma  
Basispinakoid  
Hex. Dipyramide I. St.  
Hex. Dipyramide II. St.

Chemische Zusammensetzung	: $\text{Be}_3\text{Al}_2[\text{Si}_6\text{O}_{18}]$
Kristallsystem	: hexagonal
Elementarzelle	: $a_0 = 9.22\text{\AA}$ , $c_0 = 9.19\text{\AA}$
Raumgruppe	: $P\ 6/m\ m\ c$
Ausbildung	: säulig, stengelig, dominierend Prisma (1 0 – 1 0) und Pinakoid (0 0 0 1)

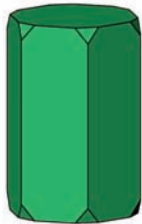


Berilo:

Prisma hexagonal  
Pinacóide basal



Prisma hexagonal  
Pinacóide basal



Prisma hexagonal  
Pinacóide basal  
Dipirâmide hexagonal  
II. Posição

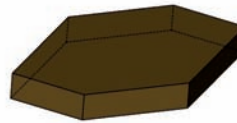


Prisma hexagonal  
Pinacóide basal  
Dipriâmide hexagonal I. pos.  
Dipirâmide hexagonal II.pos.

Fórmula química	: $\text{Be}_3\text{Al}_2[\text{Si}_6\text{O}_{18}]$
Sistema cristalino	: hexagonal
Célula unitária	: $a_0 = 9.22\text{Å}$ , $c_0 = 9.19\text{Å}$
Grupo espacial	: $P\ 6/m\ m\ c$
Características	: colunar, fibroso, predominantemente prismático (1 0 - 1 0) e pinacoidal (0001)

Biotit:

Basispinakoid  
Prisma  
Pinakoid



Basispinakoid  
Prisma  
Pinakoid



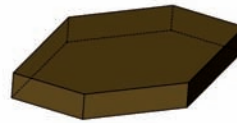
Basispinakoid  
Prisma  
Pinakoid  
Prisma

Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

:  $K(Mg, Fe)_3[(OH)_2/AlSi_3O_{10}]$   
: Monoklin, trioktaedrisches Dreischichtsilikat  
:  $a_0 = 5.35 \text{ \AA}$ ,  $b_0 = 9.26 \text{ \AA}$ ,  $c_0 = 10.23 \text{ \AA}$ ,  $\beta = 100.3^\circ$   
: C 2/m  
: Tafelig, plattig, schuppig

*Biotita:*

Pinacóide basal  
Prisma  
Pinacóide

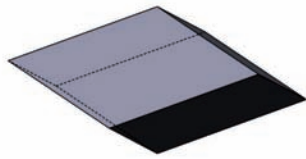


Pinacóide basal  
Prisma  
Pinacóide



Pinacóide basal  
Prisma Pinacóide  
Prisma

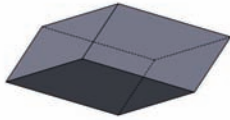
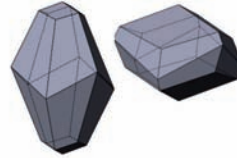
Fórmula química	: $K(Mg, Fe)_3[(OH)_2/AlSi_3O_{10}]$
Sistema cristalino	: Monoclínico, três folhas de silicatos tri-octaédricos
Célula unitária	: $a_0 = 5.35 \text{ \AA}$ , $b_0 = 9.26 \text{ \AA}$ , $c_0 = 10.23 \text{ \AA}$ , $\beta = 100.3^\circ$
Grupo espacial	: C 2/m
Características	: tabular, micáceo, liso, laminar

Calcit:

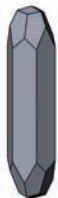
positiver Rhomboeder



Skalenoeder

negativer  
RhomboederSkalenoeder  
(+) Rhomboeder

negativer Rhomboeder

Prisma  
(-) Rhomboederpos. Rhomboeder  
Prisma  
SkalenoederPrisma  
(+) Rhomboeder

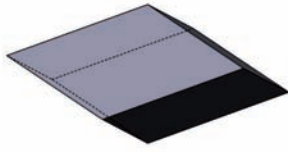
Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

:  $\text{CaCO}_3$   
: Trigonal,  $-3m$   
:  $a_0 = 4.99$ ,  $c_0 = 17.06$

: R-3c

: Gesteinsbildend, dicht, oolithisch, derb  
grobspätig, xx sehr verschieden (formenreiches  
Mineral!), z. B. Rhomboeder und Skalenoeder,  
säulige xx u.a.

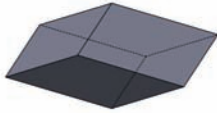
Calcita:



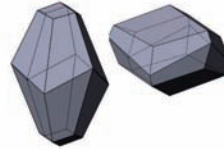
Romboedro Positivo



Escalenoedro



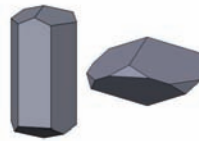
Romboedro negativo achatado



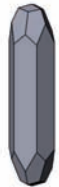
Escalenoedro  
Romboedro positivo



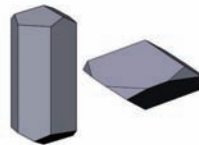
Romboedro negativo alongado



Prisma  
Romboedro negativo



Romboedro positivo  
Prisma  
Escalenoedro

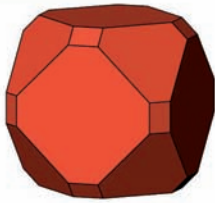


Prisma  
Romboedro positivo

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

: CaCO<sub>3</sub>  
: Trigonal, -3m  
: a<sub>0</sub>= 4.99, c<sub>0</sub>= 17.06

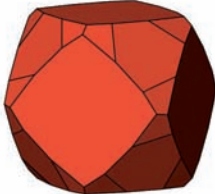
: R-3c  
: formador de rocha, denso, oolítico, maciço, colunar, xx  
muitas variações (mineral com vários hábitos possíveis!),  
p.ex.: romboédrico e escalenoédrico, xx e outros

Cuprit:

Hexaeder  
Oktaeder  
Rhombendodekaeder

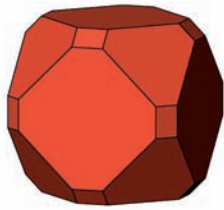


Hexaeder  
Oktaeder  
Rhombendodekaeder

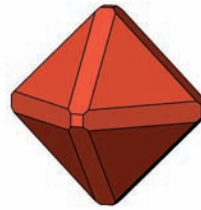


Hexaeder  
Oktaeder  
Pentagonikositetraeder

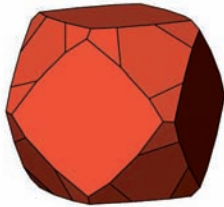
Chemische Zusammensetzung	: Cu <sub>2</sub> O
Kristallsystem	: Kubisch
Elementarzelle	: a <sub>0</sub> =4.27 Å
Raumgruppe	: Pn3m
Ausbildung	: Derb, eingesprengt, Würfel, Oktaeder, nadelige x

Cuprita:

Hexaedro (cubo) =  
 Octaedro =  
 Rombododecaedro



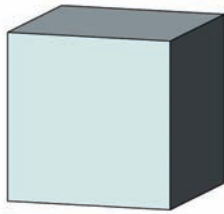
Hexaedro (cubo)  
 Octaedro  
 Rombododecaedro



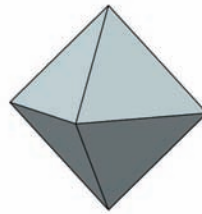
Hexaedro (cubo)  
 Octaedro  
 Pentagonicositetraedro

Fórmula química  
 Sistema cristalino  
 Célula unitária  
 Grupo espacial  
 Características

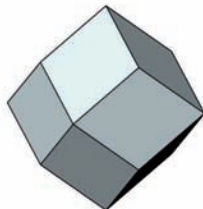
: Cu<sub>2</sub>O  
 : Isométrico  
 : a<sub>0</sub>=4.27 Å  
 : Pn3m  
 : maciço, disseminado, cúbico, octaédrico, acicular x

Diamant:

Hexaeder



Oktaeder



Rhombendodekaeder



Tetrakis hexaeder

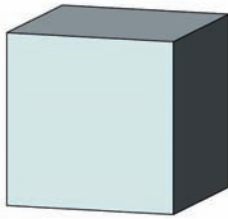


Hexakisoktaeder

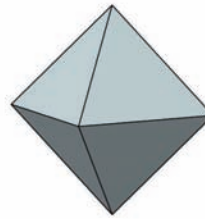
Oktaeder  
Hexakisoktaeder

Chemische Zusammensetzung	: C
Kristallsystem	: Kubisch
Elementarzelle	: $a_0 = 3.559 \text{ \AA}$
Raumgruppe	: F d3m
Ausbildung	: Oktaeder, Rhombendodekaeder, Würfel, xx meist abgerundet

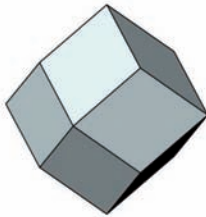


Diamante:

Hexaedro (cubo)



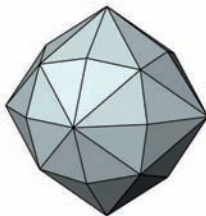
Octaedro



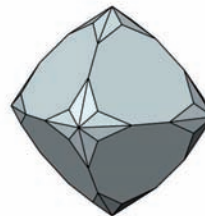
Rombododecaedro



Tetrahexaedro

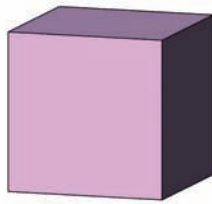


Hexaoctaedro

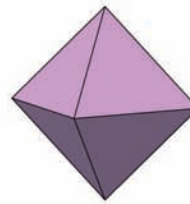
Octaedro  
Hexaoctaedro

Fórmula química  
 Sistema cristalino  
 Célula unitária  
 Grupo espacial  
 Características

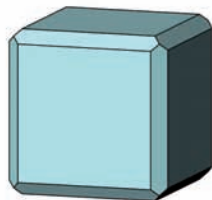
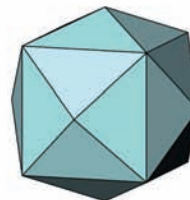
: C  
 : Isométrico  
 :  $a_0 = 3.559 \text{ \AA}$   
 : F d3m  
 : octaédrico, romboédrico, cubo, xx por vezes arredondado

Fluorit:

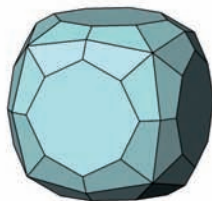
Hexaeder



Oktaeder

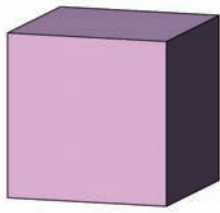
Hexaeder  
OktaederOktaeder  
RhombendodekaederHexaeder  
Rhombendodekaeder

Tetrakis hexaeder

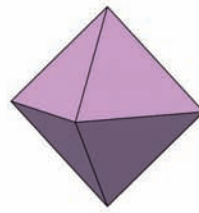
Hexaeder  
Hexakisoktaeder

Chemische Zusammensetzung  
 Kristallsystem  
 Elementarzelle  
 Raumgruppe  
 Ausbildung

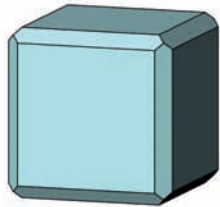
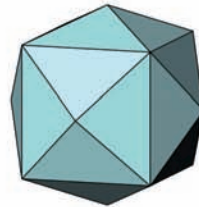
:  $\text{CaF}_2$   
 : Kubisch  
 :  $a_0 = 5.46 \text{ \AA}$   
 :  $F m\bar{3}m$   
 : Derb, grobspätig, Würfel, Oktaeder,  
 Rhombendodekaeder, oft grobkristallin,

Fluorita:

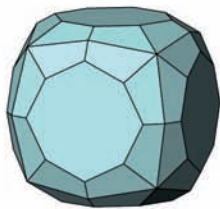
Hexaedro (cubo)



Octaedro

Hexaedro  
OctaedroOctaedro  
RombododecaedroHexaedro  
Rombododecaedro

Tetrahexaedro

Hexaedro  
Hexaoctaedro

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

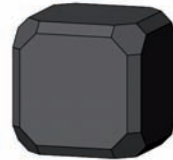
:  $\text{CaF}_2$   
: Isométrico  
:  $a_0 = 5.46 \text{ \AA}$   
:  $F m\bar{3}m$   
: maciço, prismático, cúbico, octaédrico,  
rombododecaédrico; geralmente desenvolve cristais

Galenit:

Oktaeder



Würfel

div.  
Kub-OktaederWürfel  
Oktaeder  
Rhombendodekaeder

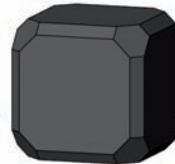
Chemische Zusammensetzung	: PbS
Kristallsystem	: Kubisch, NaCl-Gitter
Elementarzelle	: $a_0 = 5.94 \text{ \AA}$
Raumgruppe	: F m 3 m
Ausbildung	: Derby, eingesprengt, Würfel, Oktaeder, Rhombendodekaeder, oft Kombinationen

Galena:

Octaedro

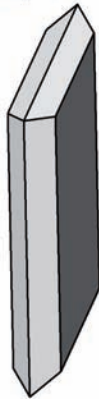


Cubo (hexaedro)

Distintas combinações  
de hexaedro e octaedroCubo (hexaedro)  
Octaedro  
Rombododecaedro

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

: PbS  
: Isométrico, reticulados de NaCl  
:  $a_0 = 5.94 \text{ \AA}$   
: F m 3 m  
: maciço, disseminado, cúbico, octaédrico, dodecaédrico  
rômbico, geralmente associado a outros minerais

Gips:

Prisma  
Prisma  
Pinakoid



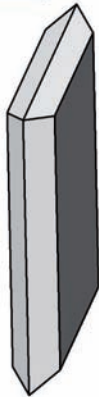
Prisma  
Prisma  
Prisma  
Prisma  
Pinakoid



Prisma  
Basispinakoid  
Pinakoid

Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

:  $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$   
: Monoklin,  $2/m$   
:  $a_0 = 5.68\text{\AA}$ ,  $b_0 = 15.18\text{\AA}$ ,  $c_0 = 6.89\text{\AA}$ ,  $\beta = 113.83$   
:  $A 2/a$   
: Dicht, feinkörnig, faserig, xx nadelig, prismatisch,  
tafelig, oft Schwalbenschwanzwillinge

Gipso:

Prisma  
Prisma  
Pinacóide



Prisma  
Prisma  
Prisma  
Prisma  
Pinacóide



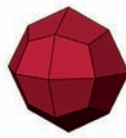
Prisma  
Picanóide basal  
Pinacóide

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

:  $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$   
: Monoclínico,  $2/m$   
:  $a_0 = 5.68\text{Å}$ ,  $b_0 = 15.18\text{Å}$ ,  $c_0 = 6.89\text{Å}$ ,  $\beta = 113.83$   
:  $A 2/a$   
: denso, fino, fibroso, xx acicular, prismático, roseta,  
tabular; frequentemente com geminação do tipo ponta  
de flecha

Granat:

Rhombendodekaeder



Deltoidikositetraeder

Rhombendodekaeder  
DeltoidikositetraederRhombendodekaeder  
Deltoidikositetraeder  
Tetrakisheptaeder  
Hexakisoktaeder  
Trisoktaeder

Chemische Zusammensetzung :  $X_3Y_2[SiO_4]_3$   
 X: Mg, Fe<sup>2+</sup>, Mn<sup>2+</sup>, Ca (8er Koord.)  
 Y: Al, Fe<sup>3+</sup>, Cr<sup>3+</sup>, V<sup>3+</sup> (6er Koord.)  
 Für (SiO<sub>4</sub>) z. T. auch (AlO<sub>4</sub>) oder (OH)

Kristallsystem : Kubisch  
 Ausbildung : Rhombendodekaeder (110), Deltoidikositetraeder (211)

**„Pyralspite“**

Pyrop  $Mg_3Al_2[SiO_4]_3$

Almandin:  $Fe_3Al_2[SiO_4]_3$

Spessartin:  $Mn_3Al_2[SiO_4]_3$

**„Ugrandite“**

Uwarowit:  $Ca_3Cr_2[SiO_4]_3$

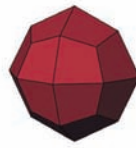
Grossular:  $Ca_3Al_2[SiO_4]_3$

Andradit:  $Ca_3Fe_2[SiO_4]_3$



Granada:

Rombododecaedro



Icositetraedro deltóide

Rombododecaedro  
Icositetraedro deltóideRombododecaedro  
Icositetraedro deltóide  
Tetrahexaedro  
Hexaoctaedro  
Trioctaedro

Fórmula química

:  $X_3Y_2[SiO_4]_3$ X: Mg, Fe<sup>2+</sup>, Mn<sup>2+</sup>, Ca (8° coord.)Y: Al, Fe<sup>3+</sup>, Cr<sup>3+</sup>, V<sup>3+</sup> (6° coord.)Para (SiO<sub>4</sub>) por vezes também (AlO<sub>4</sub>) ou (OH)

Sistema cristalino

: Isométrico (cúbico)

Características

: rombododecaédrico (110), icositetraedro deltoidal (211)

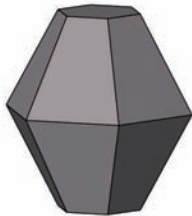
**Piralspita**Piropero  $Mg_3Al_2[SiO_4]_3$ Almandina:  $Fe_3Al_2[SiO_4]_3$ Espessartita:  $Mn_3Al_2[SiO_4]_3$ **Ugrandita**Uvarovita:  $Ca_3Cr_2[SiO_4]_3$ Grossulária:  $Ca_3Al_2[SiO_4]_3$ Andradita:  $Ca_3Fe_2[SiO_4]_3$

Hämatit:

Rhomboeder  
Basispinakoid



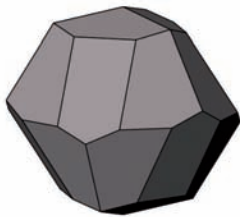
Rhomboeder  
Basispinakoid



Dipyramide  
Basispinakoid



Rhomboeder  
Dipyramide



Rhomboeder  
Dipyramide  
Rhomboeder

Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

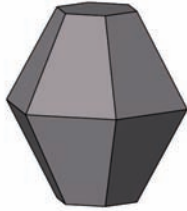
:  $\text{Fe}_2\text{O}_3$   
: Trigonal  
:  $a_0 = 5.03\text{\AA}$ ,  $c_0 = 13.74\text{\AA}$   
: R -3c  
: Erdig, schuppig, radialstrahlig, knollig, nierig, xx  
tafelig, plattig

Hematita:

Romboedro  
Pinacóide basal



Romboedro  
Pinacóide basal



Dipirâmide  
Pinaóide basal



Romboedro  
Dipirâmide



Romboedro  
Dipirâmide  
Romboedro

“diamante-negro”

Fórmula química  
Sistema cristalino  
Célula unitáriaunitária  
Grupo espacial  
Características

: Fe<sub>2</sub>O<sub>3</sub>  
: Trigonal  
: a<sub>0</sub>= 5.03Å, c<sub>0</sub>=13.74Å  
: R -3c  
: terroso, radial (dendrítico), nodular, reniforme, xx tabular, laminar

Korund:

Prisma  
Basispinakoid



Prisma  
Basispinakoid  
Dipyramide



Prisma  
Basispinakoid  
Dipyramide  
Dipyramide  
Rhomboeder

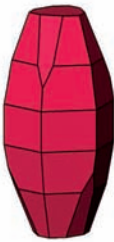
Chemische Zusammensetzung	: $\text{Al}_2\text{O}_3$
Kristallsystem	: Trigonal
Elementarzelle	: $a_0=4.75 \text{ \AA}$ , $c_0=12.98 \text{ \AA}$
Raumgruppe	: R-3c
Ausbildung	: Eingesprengt, säulige bis tonnenförmige xx, meist mit Flächenstreifung

*Corindum:*

Prisma  
Pinacóide basal



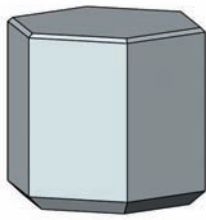
Prisma  
Pinacóide basal  
Dipirâmide



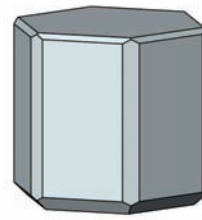
Prisma  
Pinacóide basal  
Dipirâmide  
Dipirâmide  
Romboedro

Fórmula química  
Sistema cristalino  
Célula unitáriaunitária  
Grupo espacial  
Características

:  $\text{Al}_2\text{O}_3$   
: Trigonal  
:  $a_0 = 4.75 \text{ \AA}$ ,  $c_0 = 12.98 \text{ \AA}$   
: R-3c  
: disseminado, colunar até oval, geralmente com estrias

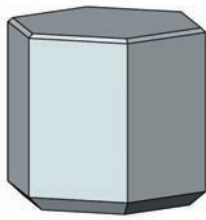
Nephelin:

Prisma  
 Basispedion  
 Basispedion  
 Pyramide  
 Pyramide

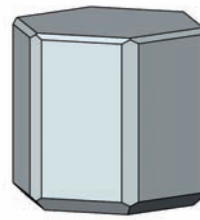


Prisma  
 Basispedion  
 Basispedion  
 Pyramide  
 Pyramide  
 Prisma

Chemische Zusammensetzung	: $\text{KNa}_3[\text{AlSiO}_4]_4$
Kristallsystem	: Hexagonal
Elementarzelle	: $a_0 = 9.99 \text{ \AA}$ , $c_0 = 8.37 \text{ \AA}$
Raumgruppe	: $P 6_3$
Ausbildung	: Derb, selten säulige xx

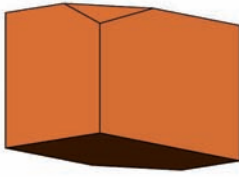
Nefelina:

Prisma  
Péδιο basal  
Péδιο basal  
Pirâmide  
Pirâmide

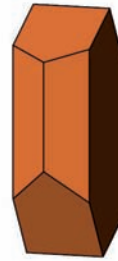


Prisma  
Péδιο basal  
Péδιο basal  
Pirâmide  
Pirâmide  
Prisma

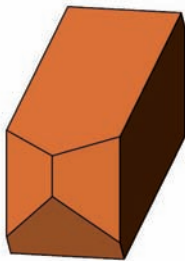
Fórmula química	: $\text{KNa}_3[\text{AlSiO}_4]_4$
Sistema cristalino	: Hexagonal
Célula unitária	: $a_0 = 9.99 \text{ \AA}$ , $c_0 = 8.37 \text{ \AA}$
Grupo espacial	: $P 6_3$
Características	: maciço, raramente colunar xx

Orthoklas:

Prisma  
Pinakoid  
Pinakoid



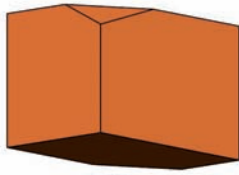
Prisma  
Basispinakoid  
Pinakoid  
Pinakoid



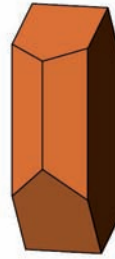
Prisma  
Pinakoid  
Basispinakoid  
Pinakoid

Chemische Zusammensetzung	: $K[AlSi_3O_8]$ „Kalifeldspat“
Kristallsystem	: Monoklin
Elementarzelle	: $a_0 = 8.62 \text{ \AA}$ , $b_0 = 12.99 \text{ \AA}$ , $c_0 = 7.19 \text{ \AA}$ , $\beta = 116.02^\circ$
Raumgruppe	: $C 2/m$
Ausbildung	: Gesteinsbildend, xx tafelig, prismatisch, häufig

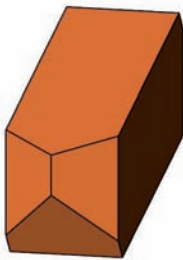


Ortoclásio:

Prisma  
Pinacóide  
Pinacóide

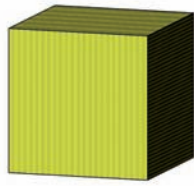


Prisma  
Pinacóide basal  
Pinacóide  
Pinacóide

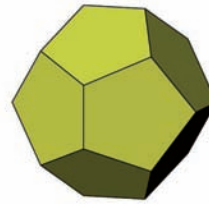


Prisma  
Pinacóide  
Pinacóide basal  
Pinacóide

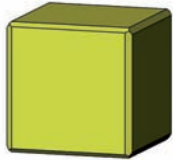
Fórmula química	: $K[AlSi_3O_8]$ k-Feldspato (Feldspato potássico)
Sistema cristalino	: Monoclínico
Célula unitária	: $a_0 = 8.62 \text{ \AA}$ , $b_0 = 12.99 \text{ \AA}$ , $c_0 = 7.19 \text{ \AA}$ , $\beta = 116.02^\circ$
Grupo espacial	: C 2/m
Características	: formador de rochas, xx tabular, prismático, frequentemente com geminações (Karlsbar, Manebach, Baveno)

Pyrit:

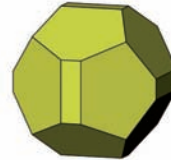
Würfel mit  
Flächenstreifung



Pentagondodekaeder



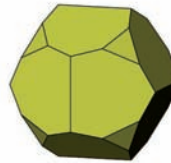
Pentagondodekaeder  
Würfel



Pentagondodekaeder  
Würfel



Pentagondodekaeder  
Oktaeder  
= "Pseudo-Ikosaeder"



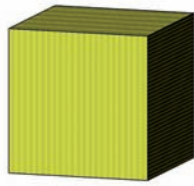
Pentagondodekaeder  
Oktaeder

Chemische Zusammensetzung  
Kristallsystem

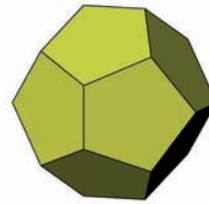
: FeS<sub>2</sub>  
: Kubisch: Struktur: ähnlich NaCl, S<sub>2</sub>-Hanteln liegen  
parallel (1 1 1)

Elementarzelle  
Raumgruppe  
Ausbildung

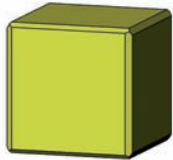
: a<sub>0</sub> = 5.42 Å  
: P a 3  
: Derb, eingesprengt, Würfel (oft mit Flächenstreifung),  
Oktaeder, Pentagondodekaeder

Pirita:

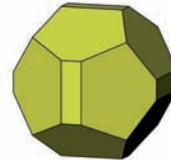
Cubo (hexaedro) com  
estrias nas faces



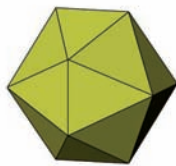
Pentagondodecaedro



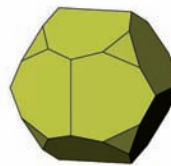
Pentagondodecaedro  
Cubo (hexaedro)



Pentagondodecaedro  
Cubo (hexaedro)



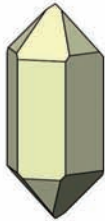
Pentagondodecaedro  
Octaedro  
= Pseudo-isosaedro



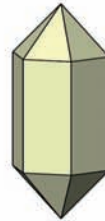
Pentagondodecaedro  
Octaedro

## “Ouro de Tolo“

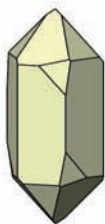
Fórmula química	: FeS <sub>2</sub>
Sistema cristalino	: Isométrico (estrutura cúbica similar NaCl, com ligações S <sub>2</sub> paralelas (1 1 1))
Célula unitária	: a <sub>0</sub> = 5.42 Å
Grupo espacial	: P a 3
Características	: maciço, disseminado, cúbico (geralmente estriado), octaédrico, pentadodecaédrico

Quarz:

Hexagonales Prisma  
pos. Rhomboeder  
neg. Rhomboeder



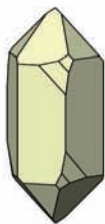
Hexagonales Prisma  
Hex. Dipyramide  
Hochquarz!  
(andere Symetrie)



Hexagonales Prisma  
pos. Rhomboeder  
neg. Rhomboeder  
rechter Trapezoeder



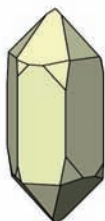
Hexagonales Prisma  
pos. Rhomboeder  
neg. Rhomboeder  
linker Trapezoeder



Hexagonales Prisma  
pos. Rhomboeder  
neg. Rhomboeder  
rechter Trapezoeder  
rechte Dipyramide



Hexagonales Prisma  
pos. Rhomboeder  
neg. Rhomboeder  
linker Trapezoeder  
linke Dipyramide



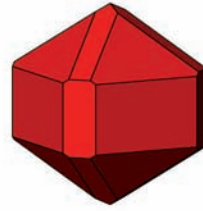
Hexagonales Prisma  
pos. Rhomboeder  
neg. Rhomboeder  
“linker”  
“rechter” Trapezoeder

Chemische Zusammensetzung	: SiO <sub>2</sub>
Kristallsystem	: Trigonal
Elementarzelle	: a <sub>0</sub> = 4.91 Å, c <sub>0</sub> = 5.40 Å
Raumgruppe	: P 3 <sub>1</sub> 2 1, P 3 <sub>2</sub> 2 1
Ausbildung	: derb, schöne Kristalle, z.T. wasserklar

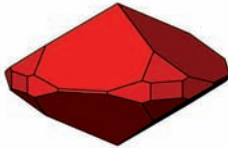


Rutil:

Prisma  
Prisma  
Dipyramide



Prisma  
Prisma  
Dipyramide  
Dipyramide

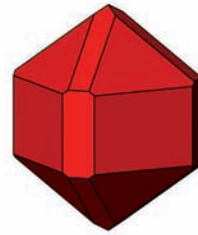


Prisma  
Prisma  
Dipyramide  
Dipyramide  
ditetragonales Prisma

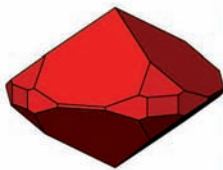
Chemische Zusammensetzung	: $\text{TiO}_2$
Kristallsystem	: Tetragonal
Elementarzelle	: $a_0=4.594\text{\AA}$ , $c_0=2.958\text{\AA}$
Raumgruppe	: $P 4/ m n m$
Ausbildung	: Derb, nadelige, stengelig, häufig Verzwilligung

Rutilo:

Prisma  
Prisma  
Dipirâmide

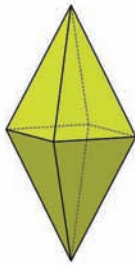


Prisma  
Prisma  
Dipirâmide  
Dipirâmide

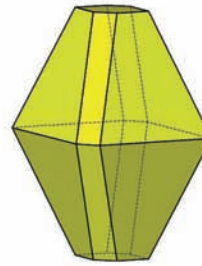
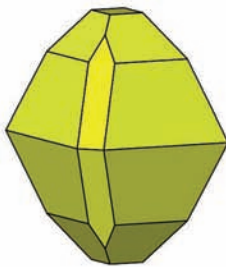


Prisma  
Prisma  
Dipirâmide  
Dipirâmide  
Prisma ditetragonal

Fórmula química	: $\text{TiO}_2$
Sistema cristalino	: Tetragonal
Célula unitária	: $a_0=4.594\text{Å}$ , $c_0=2.958\text{Å}$
Grupo espacial	: $P 4/ m n m$
Características	: granular, acicular, fibroso, geralmente geminado (ângulo $120^\circ$ ), com agregados reticulados

Schwefel:

Dipyramide

Dipyramide  
Basispinakoid  
PrismaDipyramide  
Dipyramide  
Basispinakoid  
Prisma**Kopfbild**  
Dipyramide  
Dipyramide  
Basispinakoid  
Prisma

Chemische Zusammensetzung: S

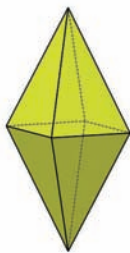
Kristallsystem : orthorhombisch

Elementarzelle :  $a_0 = 10.45 \text{ \AA}$ ,  $b_0 = 12.85 \text{ \AA}$ ,  $c_0 = 24.46 \text{ \AA}$ 

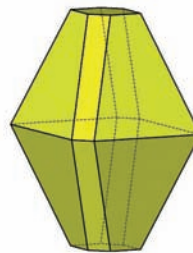
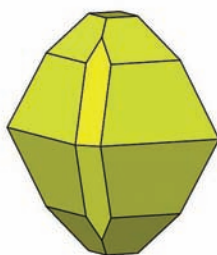
Raumgruppe : F ddd

Besonderheiten : bis  $95^\circ$  orthorhombisch, ab  $95^\circ$  monoklin,  
Schmelzpunkt  $119,2^\circ \text{ C}$ Ringförmige  $S_8$ -Moleküle, zwischen den Molekülen  
nur Van-der-Waalsche BindungAusbildung : Derbe, dichte, erdige Aggregate, pyramidale und  
sphenoidische xx

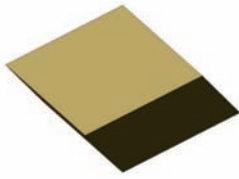


Enxofre:

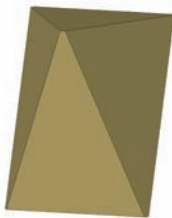
Dipirâmide

Dipirâmide  
Pinacóide basal  
PrismaDipirâmide  
Dipirâmide  
Pinacóide basal  
Prisma**Imagem de cima**  
Dipirâmide  
Dipirâmide  
Pinacóide basal  
Prisma

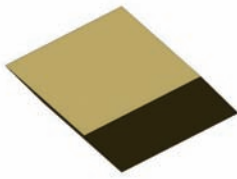
Fórmula química	: S
Sistema cristalino	: ortorrômbico
Célula unitária	: $a_0 = 10.45 \text{ \AA}$ , $b_0 = 12.85 \text{ \AA}$ , $c_0 = 24.46 \text{ \AA}$
Grupo espacial	: F ddd
Particularidades	: até 95°C ortorrômbico, a partir de 95°C monoclinico, ponto de fusão 119,2 °C Moléculas S8 anelares; entre as moléculas apenas ligações do tipo Van-der-Waal
Características	: maciço, denso, agregados terrosos, piramidal e esfenóide xx

Siderit:

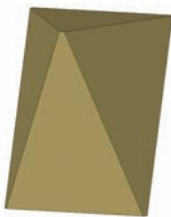
Rhomboeder

Rhomboeder  
SkalenoederRhomboeder  
Skalenoeder  
Skalenoeder  
RhomboederRhomboeder  
BasispinakoidRhomboeder  
Basispinakoid  
Rhomboeder  
Skalenoeder

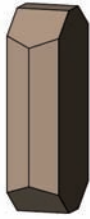
Chemische Zusammensetzung	: $\text{FeCO}_3$
Kristallsystem	: Trigonal, $-3m$
Elementarzelle	: $a_0 = 4.72$ , $c_0 = 15.46$
Raumgruppe	: $R -3c$
Ausbildung	: Feinkörnig bis grobspätig, rhomboedrische xx

Siderita:

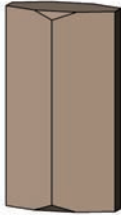
Romboedro

Romboedro  
EscalenoedroRomboedro  
Escalenoedro  
Escalenoedro  
RomboedroRomboedro  
Pinacóide basalRomboedro  
Pinacóide basal  
Romboedro  
Escalenoedro

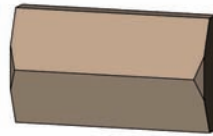
Fórmula química	: FeCO <sub>3</sub>
Sistema cristalino	: Trigonal, -3m
Célula unitáriaunitária	: a <sub>0</sub> = 4.72, c <sub>0</sub> = 15.46
Grupo espacial	: R -3c
Características	: fino até colunar, romboédrico xx

Staurolith:

Prisma  
Basispinakoid  
Pinakoid  
Prisma



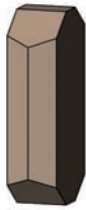
Prisma  
Basispinakoid  
Pinakoid  
Prisma



Prisma  
Basispinakoid  
Pinakoid  
Prisma

Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

:  $(\text{Fe,Mg})_2\text{Al}_6(\text{Si,Al})_4\text{O}_{20}(\text{O,OH})_2$   
: Monoklin  
:  $a_0 = 7.863 \text{ \AA}$ ,  $b_0 = 16.61 \text{ \AA}$ ,  $c_0 = 5.65 \text{ \AA}$ ,  $\beta = 90-93^\circ$   
: C 2/m  
: Gesteinsbildend, xx tafelig, prismatisch, häufig

Estaurolita:

Prisma  
Pinacóide basal  
Pinacóide  
Prisma



Prisma  
Pinacóide basal  
Pinacóide  
Prisma



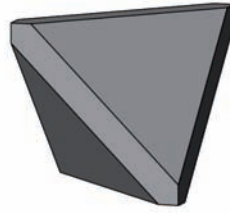
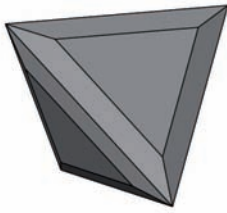
Prisma  
Pinacóide basal  
Pinacóide  
Prisma

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

:  $(\text{Fe,Mg})_2\text{Al}_9(\text{Si,Al})_4\text{O}_{20}(\text{O,OH})_2$   
: Monoclínico  
:  $a_0 = 7.863 \text{ \AA}$ ,  $b_0 = 16.61 \text{ \AA}$ ,  $c_0 = 5.65 \text{ \AA}$ ,  $\beta = 90-93^\circ$   
: C 2/m  
: formador de rocha, xx tabular, prismática, mineral frequente

Tetraedrit:

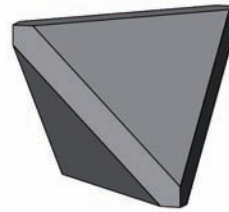
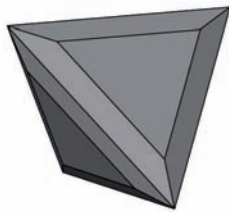
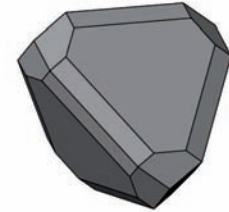
Tetraeder

Tetraeder  
HexaederTetraeder  
TristetraederTetraeder  
Tristetraeder  
Rhomböeder

Chemische Zusammensetzung	: $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$ , z. T. mit Ag-, Hg, Zn-Gehalten
Kristallsystem	: kubisch
Elementarzelle	: $a_0 = 10.36 \text{ \AA}$
Raumgruppe	: I -4 3 m
Ausbildung	: derb, tetraedr. Ausbildung, mit Calcit, Quarz verwachsen

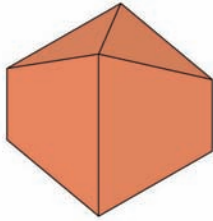
*Tetraedrita:*

Tetraedro

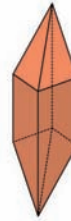
Tetraedro  
HexaedroTetraedro  
TristetrahedroTetraedro  
Tristetrahedro  
Romboedro

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

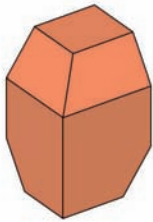
:  $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$ , por vezes com contribuições de Ag, Hg, Zn  
: Isométrico (cúbico)  
:  $a_0 = 10.36 \text{ \AA}$   
: I-43m  
: maciço; formação tetraédrica alternada com calcita e quartzo

*Titanit:*

Prisma  
Prisma



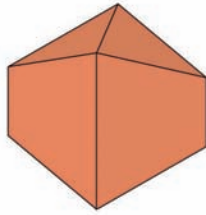
Prisma  
Prisma



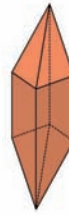
Prisma  
Basispinakoid  
Prisma

Chemische Zusammensetzung	: CaTi [O / SiO <sub>4</sub> ]
Kristallsystem	: Monoklin
Elementarzelle	: a <sub>0</sub> =7.06Å, b <sub>0</sub> =8,71 Å, c <sub>0</sub> =6.56Å, β =113.8 °
Raumgruppe	: P2 <sub>1</sub> /a
Ausbildung	: Eingesprengter, xx oft Briefkuvertförmig, stengelig, tafelig

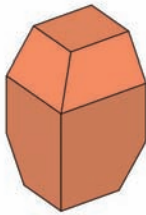


*Titanita:*

Prisma  
Prisma

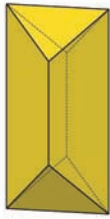


Prisma  
Prisma

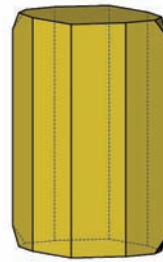


Prisma  
Pinacóide basal  
Prisma

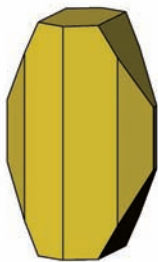
Fórmula química	: CaTi [O / SiO <sub>4</sub> ]
Sistema cristalino	: Monoclínico
Célula unitária	: a <sub>0</sub> =7.06Å, b <sub>0</sub> =8,71 Å, c <sub>0</sub> =6.56Å, β =113.8 °
Grupo espacial	: P2 <sub>1</sub> /a
Características	: disseminado, xx geralmente losangular e com terminações em cunha, compacto, tabular

Topas:

Hexagonales Prisma  
Basispinakoid



Prisma  
Prisma  
Basispinakoid  
Prisma



Prisma  
Prisma  
Basispinakoid  
Prisma  
Pinakoid

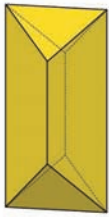


Prisma  
Prisma  
Basispinakoid  
Prisma  
Pinakoid

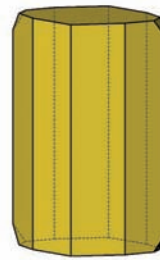
Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

: CaTi [O / SiO<sub>4</sub>]  
: Monoklin  
: a<sub>0</sub>=7.06Å, b<sub>0</sub>=8,71 Å, c<sub>0</sub>=6.56Å, β =113.8 °  
: P2<sub>1</sub>/a  
: Eingesprengter, xx oft Briefkuvertförmig, stengelig,  
tafelig

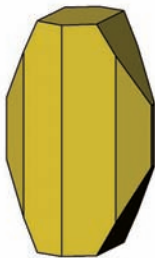
Topázio:



Prisma  
Pinacóide basal



Prisma  
Prisma  
Pinacóide basal  
Prisma



Prisma  
Prisma  
Pinacóide basal =  
Prisma  
Pinacóide



Prisma  
Prisma  
Pinacóide basal  
Prisma  
Pinacóide

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

:  $\text{Al}_2[\text{F}_2/\text{SiO}_4]$   
: ortorrômbico  
:  $a_0 = 4.65\text{Å}$ ,  $b_0 = 8.8\text{Å}$ ,  $c_0 = 8.4\text{Å}$   
:  $\text{Pbnm}$   
: granular, fibroso, prismático, miceliforme (bastonete)

Turmalin

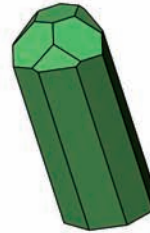
versch. Köpfe:



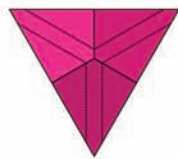
Basispedion  
 Trigonales Prisma  
 Hexagonales Prisma



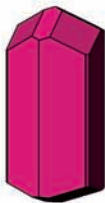
Pyramide  
 Pyramide  
 Trigonales Prisma  
 Hexagonales Prisma



Pyramide  
 Pyramide  
 Trigonales Prisma  
 Hexagonales Prisma



Pyramide  
 Trigonales Prisma  
 Pyramide  
 Pyramide



Pyramide  
 Pyramide  
 Trigonales Prisma  
 Pyramide  
 Pyramide



Pyramide  
 Trigonales Prisma  
 Hexagonales Prisma  
 Pyramide  
 Basispedion

# Turmalina

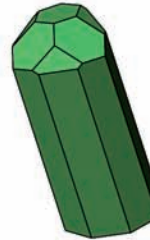
*Terminações Diferentes:*



Pédio basal  
Prisma trigonal  
Prisma hexagonal



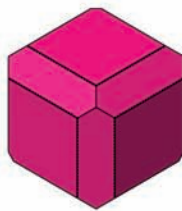
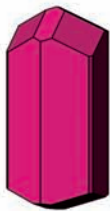
Pirâmide  
Pirâmide  
Prisma trigonal  
Prisma hexagonal



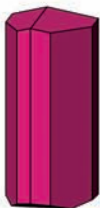
Pirâmide  
Pirâmide  
Prisma trigonal  
Prisma hexagonal



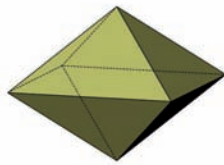
Pirâmide  
Prisma trigonal  
Pirâmide  
Pirâmide



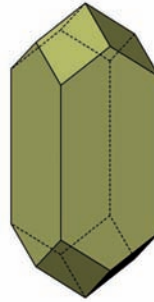
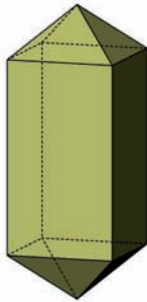
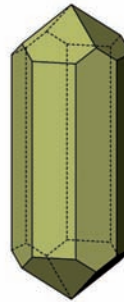
Pirâmide  
Pirâmide  
Prisma trigonal  
Pirâmide  
Pirâmide



Pirâmide  
Prisma trigonal  
Prisma hexagonal  
Pirâmide  
Péδιο basal

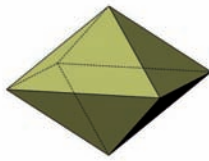
Zirkon:

Dipyramide

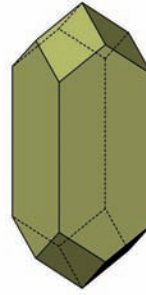
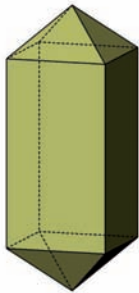
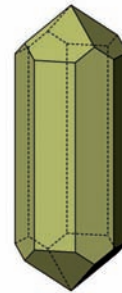
Dipyramide  
und Prisma  
I. StellungDipyramide  
und Prisma  
II. StellungDipyramide  
Prisma I. St.  
Prisma II. St.

Chemische Zusammensetzung  
Kristallsystem  
Elementarzelle  
Raumgruppe  
Ausbildung

:  $\text{Zr}[\text{SiO}_4]$ , enthält diadoch Hf, SEE  
: Tetragonal  
:  $a_0 = 6.60 \text{ \AA}$ ,  $c_0 = 5.98 \text{ \AA}$   
:  $I 4_1/a m d$   
: Körner, häufig idiomorphe xx, kurzsäulig,  
prismatisch

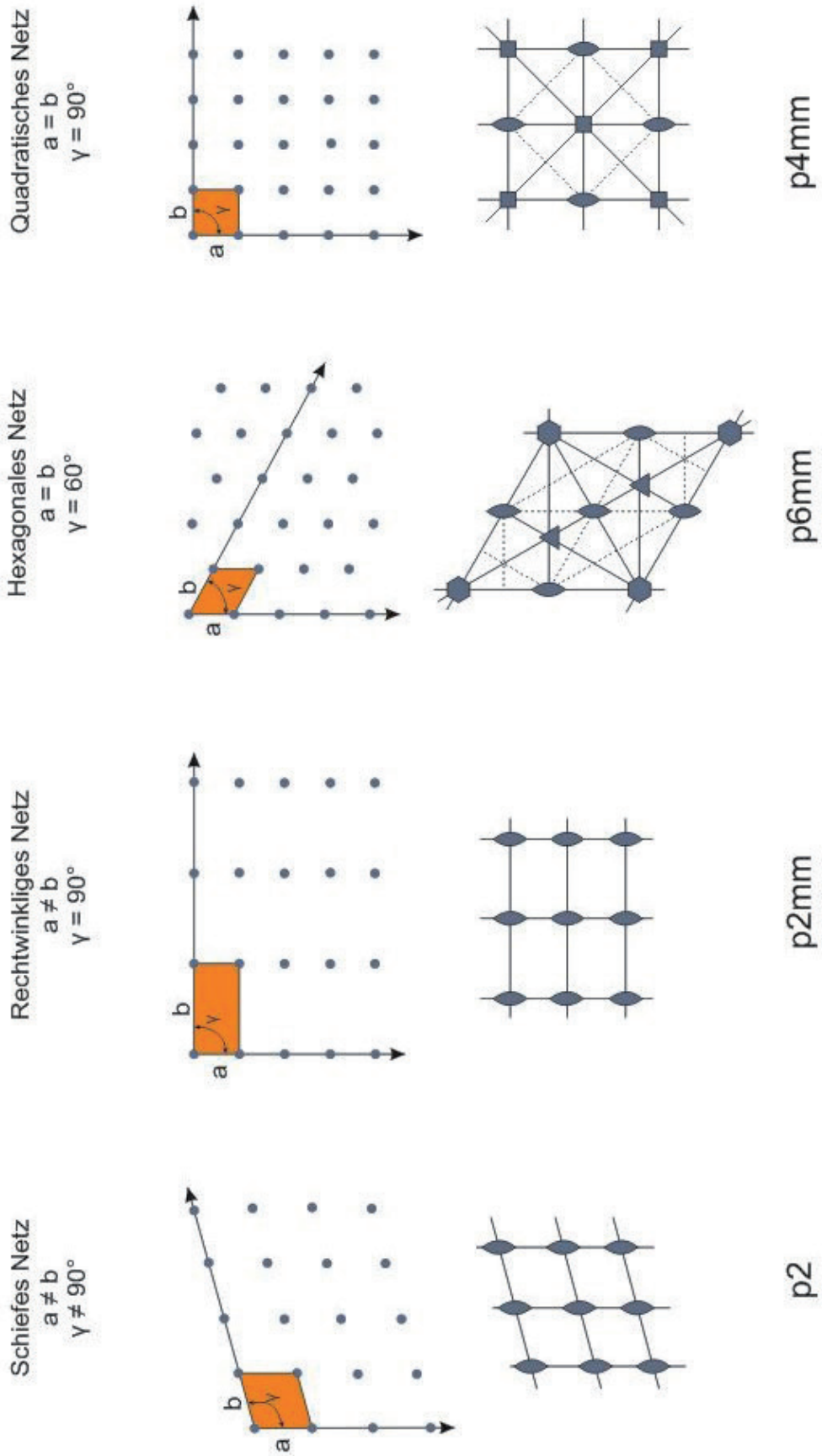
Zircão:

Dipirâmide

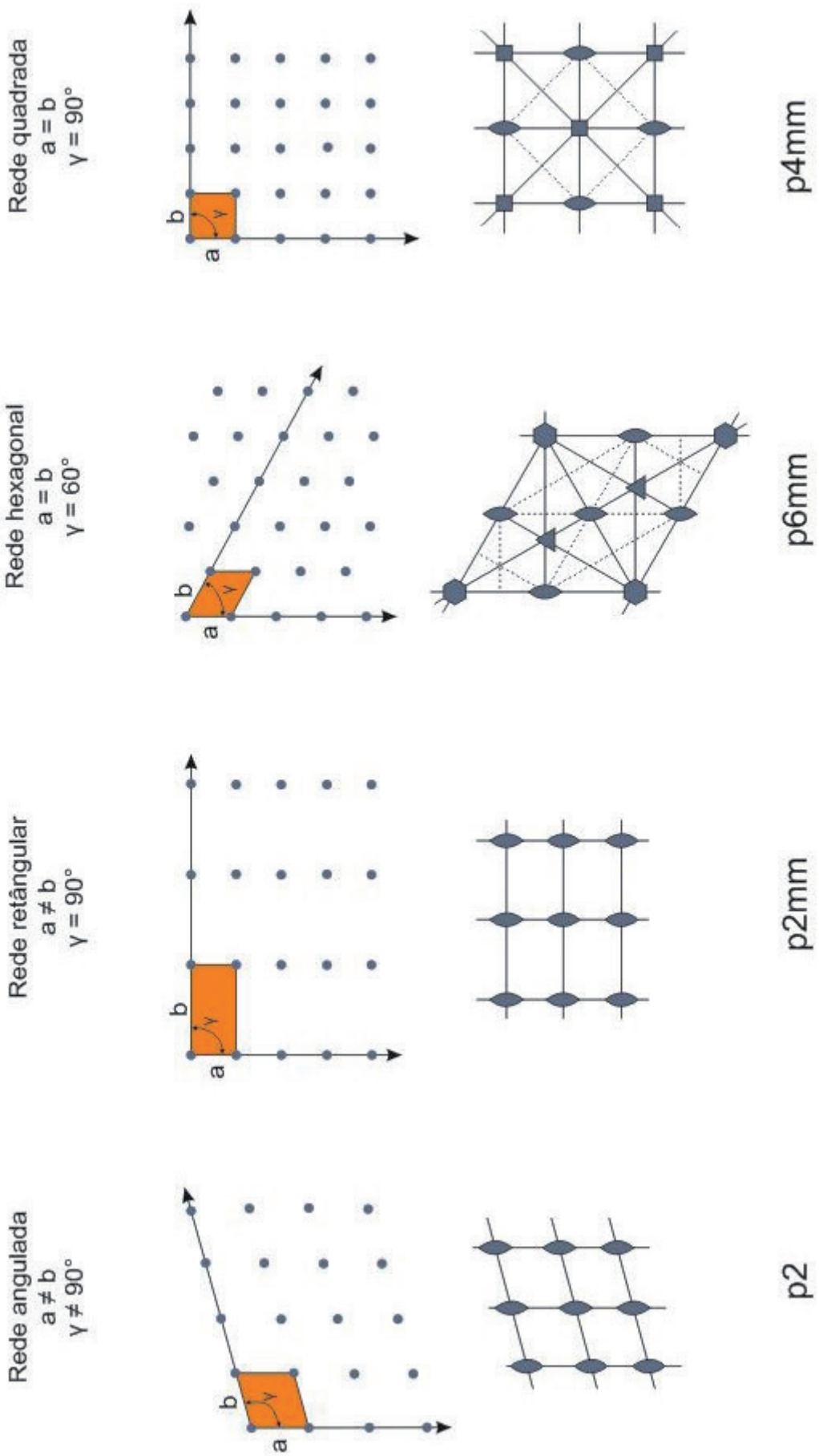
Dipirâmide  
e Prisma Posição IDipirâmide  
Prisma Posição IIDipirâmide  
Prisma posição I  
Prisma posição II

Fórmula química  
Sistema cristalino  
Célula unitária  
Grupo espacial  
Características

:  $Zr[SiO_4]$ , contém Hf, ETR  
: Tetragonal  
:  $a_0 = 6.60 \text{ \AA}$ ,  $c_0 = 5.98 \text{ \AA}$   
:  $I 4_1/a m d$   
: granular, geralmente idiomorfo xx, colunar, prismático

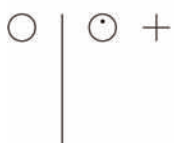






## Symbolik der Symmetrieebenen

Symmetrieoperation	Symbol	Symbol zur Zeichenebene		Bemerkung
		senkrecht	parallel	
Spiegelung	m			falls die Spiegelebene über der Zeichenebene liegt, wird die Höhe in Bruchteilen der Gitterkonstante angegeben $\frac{1}{4}$ = Spiegelebene liegt um $\frac{1}{4}$ über der Zeichenebene in der Elementarzelle
Gleit- spiegelung, achsial	a, b			Gleit- spiegelung um $\frac{\bar{a}}{2} \parallel$ a-Achse $\frac{\bar{b}}{2} \parallel$ b-Achse $\frac{\bar{c}}{2} \parallel$ c-Achse falls die Spiegelebene über d. Zeichenebene liegt, wird die Höhe in Bruchteilen der Gitterkonstante angegeben
	c		keines	
Gleit- spiegelung, diagonal	n			$\frac{\bar{t}}{2} (\bar{a}+\bar{b})/2 \parallel (001)$ $\frac{\bar{t}}{2} (\bar{a}+\bar{c})/2 \parallel (010)$ $\frac{\bar{t}}{2} (\bar{b}+\bar{c})/2 \parallel (100)$ im tetragonalen und kubischen Fall $\frac{\bar{t}}{2} (\bar{a}+\bar{b}+\bar{c})/2 \parallel (111)$
Diamant- gleit- spiegelung	d			$\frac{\bar{t}}{4} (\bar{a}+\bar{b})/4$ $\frac{\bar{t}}{4} (\bar{a}+\bar{c})/4$ $\frac{\bar{t}}{4} (\bar{b}+\bar{c})/4$ im tetragonalen und kubischen Fall $\frac{\bar{t}}{4} (\bar{a}+\bar{b}+\bar{c})/4$



Wirkung einer Spiegelebene  $\perp$  zur Projektionsebene:  
Original- und Bildpunkt werden durch ein Komma voneinander unterschieden. Die Lage der Projektionsebene wird durch + (oberhalb) oder - (unterhalb) beschrieben.

## Símbolos dos planos de simetria

Operação de simetria	símbolo	símbolo para o plano do desenho		comentar
		vertical	paralelo	
Reflexão	m			caso o plano de simetria encontra-se acima do plano do desenho, sua altura é indicada em partes da constante do retículo. 1/4 significa que o plano da simetria encontra-se 1/4 acima do plano do desenho dentro da cela elementar.
Plano de reflexão translativo (planos de deslizamento), axial	a, b			Plano de reflexão translativo por $\bar{a}/2 \parallel$ eixo a $\bar{b}/2 \parallel$ eixo b $c/2 \parallel$ eixo c caso o plano de reflexão encontra-se (é situado) acima do plano de desenho, seu montante é indicado em parte da constante do retículo.
	c		nenhum	
Plano de reflexão translativo (planos de deslizamento), diagonal	n			$\bar{t}(\bar{a}+\bar{b})/2 \parallel (001)$ $\bar{t}(\bar{a}+\bar{c})/2 \parallel (010)$ $\bar{t}(\bar{b}+\bar{c})/2 \parallel (100)$ no caso tetragonal e cúbico $\bar{t}(\bar{a}+\bar{b}+\bar{c})/2 \parallel (111)$
Plano de reflexão translativo (planos de deslizamento), diamante	d			$\bar{t}(\bar{a}+\bar{b})/4$ $\bar{t}(\bar{a}+\bar{c})/4$ $\bar{t}(\bar{b}+\bar{c})/4$ no caso tetragonal e cúbico $\bar{t}(\bar{a}+\bar{b}+\bar{c})/4$






















Efeito de um plano de reflexão  $\perp$  ao plano de projeção: conto original e de demonstração serão separados por vírgula. A posição do plano da projeção estão sendo indicada por + (acima) ou - (abaixo).

## Symbolik der Symmetrieachsen

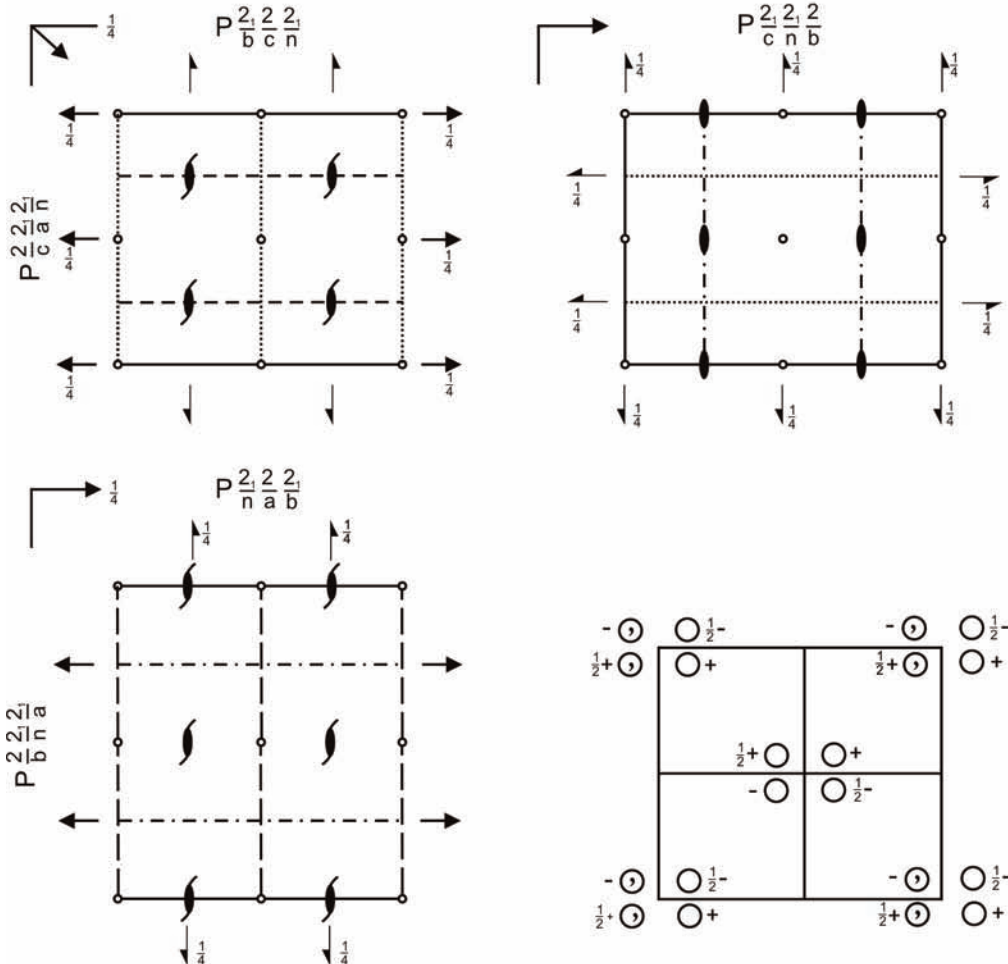
Symmetrieachse	Symbol	graphisches Symbol	Translation    Symmetrieachse bei rechtshändiger Symmetrieoperation
Symmetrieachse	1		keine
Inversionszentrum	$\bar{1}$		keine
2-zählige Drehung	2		keine
2-zählige Schraubung	$2_1$		1/2
3-zählige Drehung	3		keine
3-zählige Schraubung	$3_1$		1/3
	$3_2$		2/3
3-zählige Inversionsdrehachse	$\bar{3}$		keine
4-zählige Drehung	4		keine
4-zählige Schraubung	$4_1$		1/4
	$4_2$		1/2
	$4_3$		3/4
4-zählige Inversionsdrehachse	$\bar{4}$		keine
6-zählige Drehung	6		keine
6-zählige Schraubung	$6_1$		1/6
	$6_2$		2/6
	$6_3$		3/6
	$6_4$		4/6
	$6_5$		5/6
6-zählige Inversionsdrehachse	$\bar{6}$		keine

## Símbolos dos eixos de simetria

Eixo de simetria	Símbolo	Símbolográfico	Translação    Eixo de simetria Operação de simetria no sentido direito
eixo de simetria	1		sem
centro de simetria	$\bar{1}$		sem
rotação (2)	2		sem
eixo de rotação com translação (2) (eixos parafuso)	$2_1$		1/2
rotação (3)	3		sem
eixo de rotação com translação (3) (eixos parafuso)	$3_1$		1/3
	$3_2$		2/3
centro de inversão rotativo (3)	$\bar{3}$		sem
rotação (4)	4		sem
eixo de rotação com translação (4) (eixos parafuso)	$4_1$		1/4
	$4_2$		1/2
	$4_3$		3/4
centro de inversão rotativo (4)	$\bar{4}$		sem
rotação (6)	6		sem
eixo de rotação com translação (6) (eixos parafuso)	$6_1$		1/6
	$6_2$		2/6
	$6_3$		3/6
	$6_4$		4/6
	$6_5$		5/6
centro de inversão rotativo (6)	$\bar{6}$		sem

## Beispiel einer Raumgruppe **Pbcn**

Pbcn  $D_{2h}^{14}$  mmm Orthorhombisches Kristallsystem  
 P  $2_1/b$   $2/c$   $2_1/n$  Patterson-Symmetrie Pmmm



Ursprung bei  $\bar{1}$  in  $1\ c\ 1$

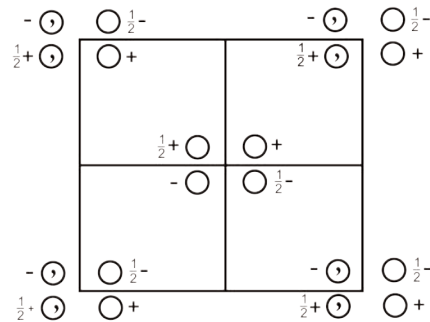
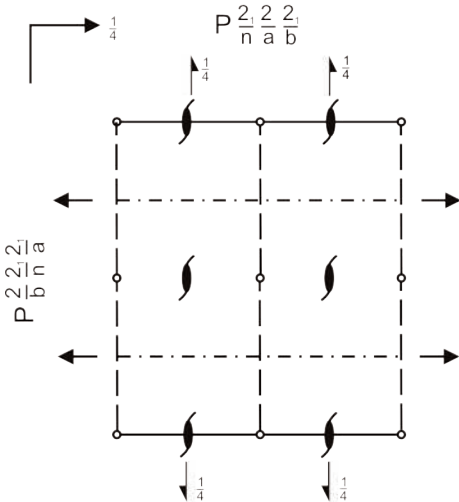
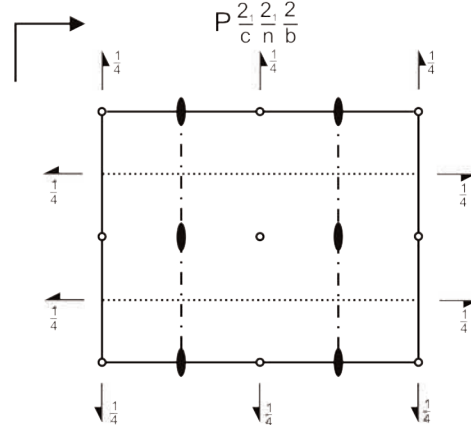
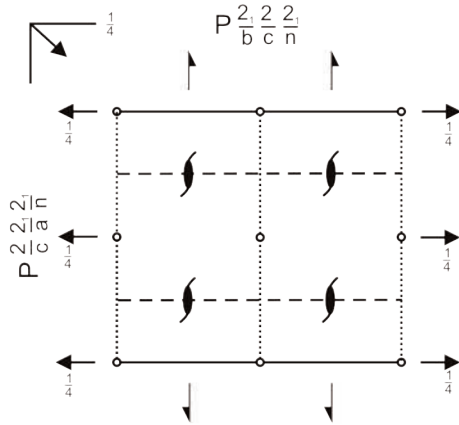
Asymmetrische Einheit  $0 \leq x \leq 1/2$   $0 \leq y \leq 1/2$   $0 \leq z \leq 1/2$

Symmetrie Operationen

- |                            |                              |                              |
|----------------------------|------------------------------|------------------------------|
| (1) 1                      | (2) $2(0,0,1/2)$ $1/4,1/4,z$ | (3) $2\ 0,y,1/4$             |
| (4) $2(1/2,0,0)$ $x,1/4,0$ | (5) $\bar{1}\ 0,0,0$         | (6) $n(1/2,1/2,0)$ $x,y,1/4$ |
| (7) $c\ x,0,z$             | (8) $b\ 1/4,y,z$             |                              |

## Exemplo grupo de espaço **Pbcn**

Pbcn  $D_{2h}^{14}$  mmm Sistema cristalino ortorrômbico  
 P  $2_1/b$   $2/c$   $2_1/n$  Simetria Patterson Pmmm



Origem:  $\overline{1}$  em  $1\ c\ 1$

Unidade assimétrica  $0 \leq x \leq \frac{1}{2}$   $0 \leq y \leq \frac{1}{2}$   $0 \leq z \leq \frac{1}{2}$

Operação de simetria

- |  |  |  |
|--|--|--|
| (1) 1                                      | (2) $2(0,0,\frac{1}{2})$ $\frac{1}{4},\frac{1}{4},z$ | (3) $2\ 0,y,\frac{1}{4}$                             |
| (4) $2(\frac{1}{2},0,0)$ $x,\frac{1}{4},0$ | (5) $\overline{1}\ 0,0,0$                            | (6) $n(\frac{1}{2},\frac{1}{2},0)$ $x,y,\frac{1}{4}$ |
| (7) $c\ x,0,z$                             | (8) $b\ \frac{1}{4},y,z$                             |  |

## Die 230 Raumgruppen

Kristallsystem	Punktgruppe	Raumgruppen			
Triklin	1 -1	P1 P1			
Monoklin	2 M 2/m	P2 Pm P2/m P2 <sub>1</sub> /c	P2 <sub>1</sub> Pc P2 <sub>1</sub> /m C2/c	C2 Cm C2/m	Cc P2/c
Orthorhombisch	222  mm2   mmm	P222 C222 <sub>1</sub> I2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub> Pmm2 Pca2 <sub>1</sub> Pna2 <sub>1</sub> Ccc2 Aba2 Iba2 Pmmm Pmma Pbam Pmnn Cmcm Cmma Immm	P222 <sub>1</sub> C222 Pmc2 <sub>1</sub> Pnc2 Pnn2 Amm2 Fmm2 Ima2 Pnnn Pnna Pvnn Pbcn Cmca Ccca Ibam	P2 <sub>1</sub> 2 <sub>1</sub> 2 F222 Pcc2 Pmn2 <sub>1</sub> Cmm2 Abm2 Fdd2 Pccm Pmna Pbcm Pbca Cmmm Fmmm Ibca	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub> I222 Pma2 Pba2 Cmc2 <sub>1</sub> Ama2 Imm2 Pban Pcca Pnmm Pnma Cccm Fddd Imma
Tetragonal	4  -4 4/m  422  4mm  -42m  4/mmm	P4 I4 P-4 P4/m I4/m P422 P4 <sub>2</sub> 22 I422 P4mm P4cc I4mm P-42m P-4m2 I-4m2 P4/mmm P4/mbm P4 <sub>2</sub> /mmc P4 <sub>2</sub> /mbc I4/mmm	P4 <sub>1</sub> I4 <sub>1</sub> I-4 P4 <sub>2</sub> m I4 <sub>1</sub> /a P4 <sub>2</sub> 2 P4 <sub>2</sub> 2 <sub>1</sub> 2 I4 <sub>2</sub> 2 P4bm P4nc I4cm P-42c P-4c2 I-4c2 P4/mcc P4/mnc P4 <sub>2</sub> /mcm P4 <sub>2</sub> /mnm I4/mcm	P4 <sub>2</sub>   P4/n  P4 <sub>1</sub> 22 P4 <sub>3</sub> 22 P4 <sub>2</sub> cm P4 <sub>2</sub> mc I4md P-4 <sub>2</sub> m P-4b2 I-42m P4/nbm P4/nmm P4 <sub>2</sub> /nbc P4 <sub>2</sub> /nmc I4 <sub>1</sub> /amd	P4 <sub>3</sub>   P4 <sub>2</sub> /n  P4 <sub>1</sub> 2 <sub>1</sub> 2 P4 <sub>3</sub> 2 <sub>1</sub> 2 P4 <sub>2</sub> nm P4 <sub>2</sub> bc I4 <sub>1</sub> cd P-4 <sub>2</sub> c P-4n2 I-42d P4/nnc P4/ncc P4 <sub>2</sub> /nmm P4 <sub>2</sub> /ncm I4 <sub>1</sub> /acd
Trigonal	3 -3 32  3m  -3m	P3 P-3 P312 P3 <sub>2</sub> 12 P3m1 R3m P-31m R-3m	P3 <sub>1</sub> R-3 P321 P3 <sub>2</sub> 21 P31m R3c P-31c R-3c	P3 <sub>2</sub>  P3 <sub>1</sub> 12 R32 P3c1 P-3m1	R3  P3 <sub>1</sub> 21  P31c  P-3c1
Hexagonal	6  -6 6/m 622  6mm -6m2 6/mmm	P6 P6 <sub>4</sub> P-6 P6/m P622 P6 <sub>4</sub> 22 P6mm P-6m2 P6/mmm	P6 <sub>1</sub> P6 <sub>3</sub>  P6 <sub>3</sub> /m P6 <sub>1</sub> 22 P6 <sub>3</sub> 22 P6cc P-6c2 P6/mcc	P6 <sub>5</sub>   P6 <sub>3</sub> 22 P6 <sub>3</sub> cm P-62m P6 <sub>3</sub> /mm	P6 <sub>2</sub>   P6 <sub>2</sub> 22 P6 <sub>3</sub> mc P-62c P6 <sub>3</sub> /mmc
Kubisch	23  m-3  432  -43m  m-3m	P23 I2 <sub>3</sub> Pm-3 Im-3 P432 I432 P-43m F-43c Pm-3m Fm-3m Im-3m	F23 Pn-3 Pa-3 P4 <sub>2</sub> 32 P4 <sub>3</sub> 32 F-43m I-43d Pn-3n Fm-3c Ia-3d	I23 Fm-3 Ia-3 F432 P4 <sub>1</sub> 32 I-43m Pm-3n Fd-3m	P2 <sub>3</sub>  Fd-3  F4 <sub>1</sub> 32 I4 <sub>1</sub> 32 P-43n Pn-3m Fd-3c



## 230 grupos espaciais

Sistemas cristalinos	Grupo de pontos ou classes cristalinas	Grupo espaciais			
Triclínico	1 -1	P1 P1			
Monoclínico	2 M 2/m	P2 Pm P2/m P2 <sub>1</sub> /c	P2 <sub>1</sub> Pc P2 <sub>1</sub> /m C2/c	C2 Cm C2/m	Cc P2/c
Ortorômbico	222  mm2    mmm	P222 C222 <sub>1</sub> I2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>  Pmm2 Pca2 <sub>1</sub> Pna2 <sub>1</sub> Ccc2 Aba2 Iba2 Pmmm Pmma Pbam Pmmn Cmcm Cmma Immm	P222 <sub>1</sub> C222  Pmc2 <sub>1</sub> Pnc2 Pnn2 Amm2 Fmm2 Ima2 Pnnn Pnna Pvvn Pbcn Cmca Ccca Ibam	P2 <sub>1</sub> 2 <sub>1</sub> 2 F222  Pcc2 Pmn2 <sub>1</sub> Cmm2 Abm2 Fdd2  Pccm Pmna Pbcm Pbca Cmmm Fmmm Ibca	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub> I222  Pma2 Pba2 Cmc2 <sub>1</sub> Ama2 Imm2  Pban Pcca Pnm Pnma Cccm Fddd Imma
Tetragonal	4 -4 4/m 422 4mm -42m 4/mmm	P4 I4 P4 P4/m I4/m P422 P4 <sub>2</sub> 22 I422 P4mm P4cc I4mm P-42m P-4m2 I-4m2 P4/mmm P4/mbm P4 <sub>2</sub> /mmc P4 <sub>2</sub> /mbc I4/mmm	P4 <sub>1</sub> I4 <sub>1</sub> I-4 P4 <sub>2</sub> m I4 <sub>1</sub> /a P4 <sub>2</sub> 2 P4 <sub>2</sub> 2 <sub>1</sub> 2 I4 <sub>1</sub> 22 P4bm P4nc I4cm P-42c P-4c2 I-4c2 P4/mcc P4/mnc P4 <sub>2</sub> /mcm P4 <sub>2</sub> /mnm I4/mcm	P4 <sub>2</sub>  P4/n P4 <sub>1</sub> 22 P4 <sub>3</sub> 22  P4 <sub>2</sub> cm P4 <sub>2</sub> mc I4 <sub>1</sub> md P-4 <sub>2</sub> m P-4b2 I-42m P4/nbm P4/nmm P4 <sub>2</sub> /nbc P4 <sub>2</sub> /nmc I4 <sub>1</sub> /amd	P4 <sub>3</sub>  P4 <sub>2</sub> /n P4 <sub>1</sub> 2 <sub>1</sub> 2 P4 <sub>3</sub> 2 <sub>1</sub> 2  P4 <sub>2</sub> n P4 <sub>2</sub> bc I4 <sub>1</sub> cd P-4 <sub>2</sub> c P-4n2 I-42d P4/nnc P4/ncc P4 <sub>2</sub> /nmm P4 <sub>2</sub> /ncm I4 <sub>1</sub> /acd
Trigonal	3 -3 32 3m -3m	P3 P-3 P312 P3 <sub>2</sub> 12 P3m1 R3m P-31m R-3m	P3 <sub>1</sub> R-3 P321 P3 <sub>2</sub> 21 P31m R3c P-31c R-3c	P3 <sub>2</sub> R32 P3c1 P-3m1	R3 P3 <sub>2</sub> 1 P31c P-3c1
Hexagonal	6 -6 6/m 622 6mm -6m2 6/mmm	P6 P6 <sub>4</sub> P-6 P6/m P622 P6 <sub>4</sub> 22 P6mm P-6m2 P6/mmm	P6 <sub>1</sub> P6 <sub>3</sub>  P6 <sub>2</sub> /m P6 <sub>1</sub> 22 P6 <sub>2</sub> 22 P6cc P-6c2 P6/mcc	P6 <sub>5</sub>  P6 <sub>5</sub> 22  P6 <sub>3</sub> cm P-62m P6 <sub>3</sub> /mm	P6 <sub>2</sub>  P6 <sub>2</sub> 22  P6 <sub>3</sub> mc P-62c P6 <sub>3</sub> /mmc
Cúbico ou Isométrico	23 m-3 432 -43m m-3m	P23 I2 <sub>1</sub> 3 Pm-3 Im-3 P432 I432 P-43m F-43c Pm-3m Fm-3m Im-3m	F23 Pn-3 Pa-3 P4 <sub>2</sub> 32 P4 <sub>3</sub> 32 F-43m I-43d Pn-3n Fm-3c Ia-3d	I23 Fm-3 Ia-3 F432 P4 <sub>1</sub> 32 I-43m Pm-3n Fd-3m	P2 <sub>1</sub> 3 Fd-3 F4 <sub>3</sub> 2 I4 <sub>1</sub> 32 P-43n Pn-3m Fd-3c

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