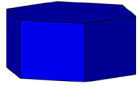
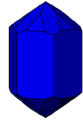


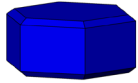
Apatit:



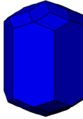
Hexagonales Prisma
Basispinakoid



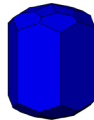
Hexagonales Prisma
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.



Hexagonales Prisma
Basispinakoid
Hex. Dipyramide I. St.

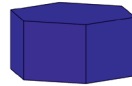


Hexagonales Prisma
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.



Hexagonales Prisma
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.

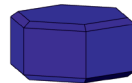
Apatite:



Hexagonal prism
Basispinacoid



Hexagonal prism
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.



Hexagonal prism
Basispinacoid
Hex. dipyramid I. pos.



Hexagonal prism
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.



Hexagonal prism
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.

HERBERT PÖLLMANN & CHIMEDNOROV OTGONBAYAR

**SYMMETRIE UND DIE 32 PUNKTGRUPPEN IN DEUTSCHER UND
ENGLISCHER SPRACHE**

**SYMMETRY AND 32 POINTGROUPS IN GERMAN AND
ENGLISH LANGUAGE**



HALLE (SAALE) 2017

HALLESCHES JAHRBUCH FÜR GEOWISSENSCHAFTEN

Herausgeber

Institut für Geowissenschaften und Geographie
der Martin - Luther Universität Halle-Wittenberg

G. BORG M. FRÜHAUF C. FÜRST
C. GLÄSSER H. HEINISCH
H. PÖLLMANN P. WYCISK

Schriftleitung

D. MERTMANN T. DEGEN S. STÖBER

BEIHEFT 39

Halle (Saale) 2017

Institut für Geowissenschaften und Geographie
der Martin - Luther Universität Halle-Wittenberg



Anschrift von Herausgebern und Schriftleitung:

Martin-Luther-Universität Halle-Wittenberg
Institut für Geowissenschaften und Geographie
Von Seckendorff - Platz 3/4
D-06120 Halle (Saale)

e-mail: hjg@geo.uni-halle.de

Schriftleitung:

D. Mertmann T. Degen S. Stöber

P-ISSN: 2193-1313, E-ISSN: 2196-3622

© 2017 im Selbstverlag des Instituts für Geowissenschaften und Geographie
der Martin-Luther-Universität Halle-Wittenberg
Alle Rechte vorbehalten

Vorwort

Im vorliegenden zusammenfassenden Buch wird versucht, die Grundlagen der Kristallographie und insbesondere der Symmetriellehre in deutscher und englischer Sprache darzustellen. Die Zusammenfassung der Symmetriellehre basierend auf den 32 Punktgruppen wird erweitert durch Anwendungsbeispiele der makroskopischen Form von wichtigen Mineralen sowie Hinweisen zu den Raumgruppen.

Die Zusammenarbeit zwischen deutschen und brasilianischen Wissenschaftlern über lange Jahre basiert hierbei nicht nur auf der Kooperation in wissenschaftlichen Projekten sondern wurde darüber hinaus auch durch einen regen Wissenschaftler- und Studentenaustausch weiter intensiviert. Die Unterstützung des Deutschen Akademischen Austauschdienstes (DAAD) auf deutscher Seite, bzw. Capes/Cnpq auf brasilianischer Seite soll an dieser Stelle besonders hervorgehoben werden.

Die Idee, Sprachbarrieren durch gemeinsame Grundlagenpublikationen abzubauen, ergab sich aus den Schwierigkeiten von Austauschstudenten, Vorlesungen und Übungen in einer fremden Sprache und zudem einer fremden Fachsprache zu verstehen. Oft werden fehlende Kenntnisse der deutschen oder portugiesischen Sprache dann über Englisch ausgeglichen.

Das Buch ist vor allem dahingehend ausgerichtet, zu helfen, erste Sprachbarrieren zu überwinden und entsprechendes Grundlagenwissen anschaulich darzustellen. Durch die Komplexität des Gesamtgebietes kann hier natürlich nur ein kleiner Ausschnitt aufgezeigt werden, der trotzdem Basisdaten in beiden Sprachen darstellt und als Einführung verwendet werden kann.

Hier soll bewusst der Versuch unternommen werden, die für den Austausch zwischen Brasilien und Deutschland notwendigen Sprachen in den Vordergrund zu stellen. Es ist geplant, diese Zusammenstellung in den kommenden Jahren auszuweiten, zu vervollständigen sowie auch eine entsprechende Zusammenstellung in englischer Sprache zur Verfügung zu stellen.

Besonderer Dank gebührt Frau Prof. Dr. Dorothee Mertmann für die geduldige und hilfreiche Umsetzung des Manuskriptes in die vorliegende Endform.

Herbert Pöllmann

Halle, Oktober 2017

Preface

The present book tries to give some introductory basic knowledge of crystallography, especially for the training of symmetry in German and English language. The concentration of training is mostly due to the 32 point groups and adjacent knowledge on application examples of macroscopic forms of important minerals including some hints on space groups.

The collaboration between German and other international groups of scientist makes it necessary to have some tools available to discuss relevant crystallographic details in different languages. This is also based on exchange of students and scientists in different fields. The first booklet in this field was published in German and Portuguese language.

The idea to overcome basic barriers due to language knowledge should be overcome by using different translations in different languages.

Therefore especially an English translation of this basic book is highly necessary.

The book focuses mainly on helping to overcome barriers in different languages and show the basic knowledge in crystallography clearly.

The complexity of this documentation makes it necessary to show only a small part, but containing despite all details in 2 languages. It can be used as an introductory work.

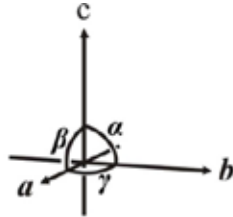
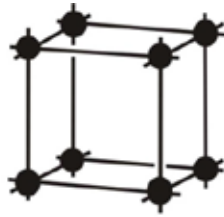
The trial, to consciously include different languages for this topic should help to minimize language barriers and to focus on the scientific work. This bilingual book will be offered also in other languages now or in the future. Also some extensions to other important topics are planned for the future.

Special thanks are due to Prof. Dr. Dorothee Mertmann for her patience and helpful transformation of the manuscript into the present form of the book.

Herbert Pöllmann

Halle, Oktober 2017

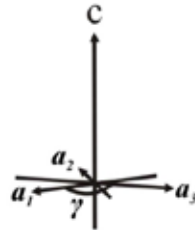
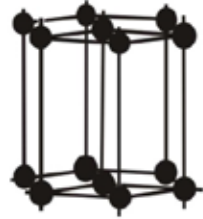
Die 7 Kristallsysteme



Kubisch:

$$a_0 = b_0 = c_0$$

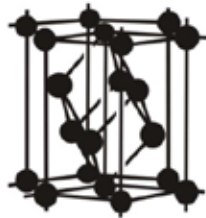
$$\alpha = \beta = \gamma = 90^\circ$$



Hexagonal:

$$a_1 = a_2 = a_3 \neq c_0$$

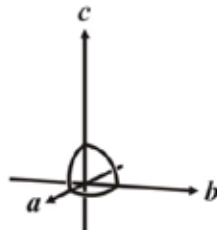
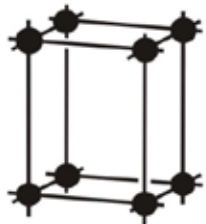
$$\alpha = \beta = 90^\circ; \gamma = 120^\circ$$



Rhomboedrisch:

$$a_1 = a_2 = a_3$$

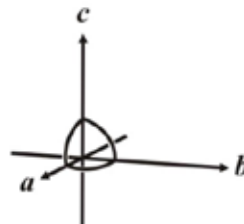
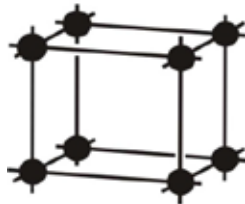
$$\alpha_1 = \alpha_2 = \alpha_3 \neq 90^\circ$$



Tetragonal:

$$a_0 = b_0 \neq c_0$$

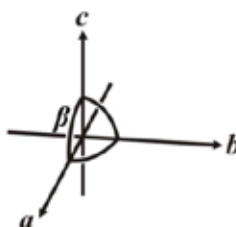
$$\alpha = \beta = \gamma = 90^\circ$$



Orthorhombisch:

$$a_0 \neq b_0 \neq c_0$$

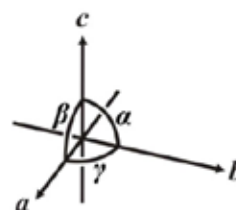
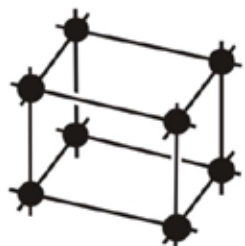
$$\alpha = \beta = \gamma = 90^\circ$$



Monoklin:

$$a_0 \neq b_0 \neq c_0$$

$$\alpha = \gamma = 90^\circ; \beta \neq 90^\circ$$

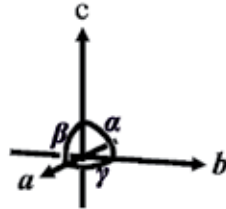
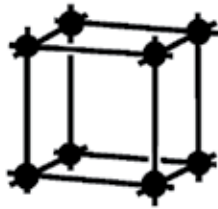


Triklin:

$$a_0 \neq b_0 \neq c_0$$

$$\alpha \neq \beta \neq \gamma \neq 90^\circ$$

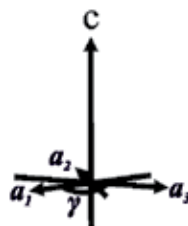
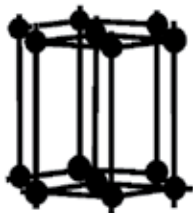
The 7 crystal systems



Cubic:

$$a_0 = b_0 = c_0$$

$$\alpha = \beta = \gamma = 90^\circ$$



Hexagonal:

$$a_1 = a_2 = a_3 \neq c_0$$

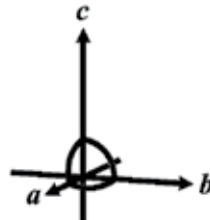
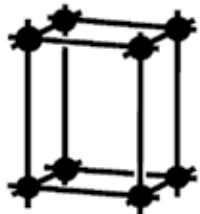
$$\alpha = \beta = 90^\circ; \gamma = 120^\circ$$



Rhombohedral:

$$a_1 = a_2 = a_3$$

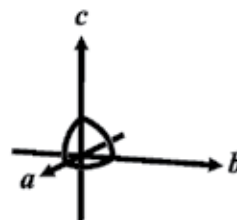
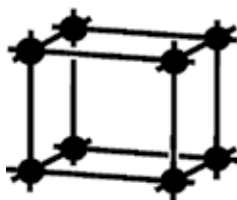
$$\alpha_1 = \alpha_2 = \alpha_3 \neq 90^\circ$$



Tetragonal:

$$a_0 = b_0 \neq c_0$$

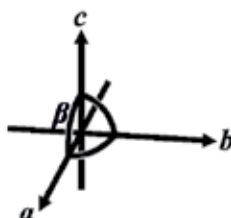
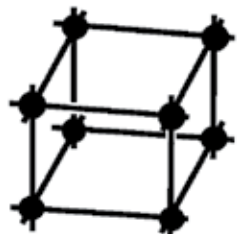
$$\alpha = \beta = \gamma = 90^\circ$$



Orthorhombic:

$$a_0 \neq b_0 \neq c_0$$

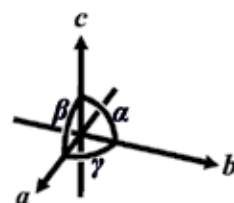
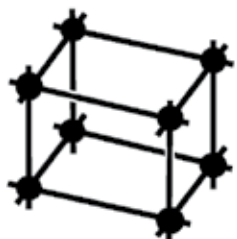
$$\alpha = \beta = \gamma = 90^\circ$$



Monoclinic:

$$a_0 \neq b_0 \neq c_0$$

$$\alpha = \gamma = 90^\circ; \beta \neq 90^\circ$$



Triclinic:

$$a_0 \neq b_0 \neq c_0$$

$$\alpha \neq \beta \neq \gamma \neq 90^\circ$$

Häufigkeit von Mineralen

5091 Minerale sind derzeit bekannt (www.mindat.org, August 2017). Nach einer alten Aufstellung sind:

Kubisch	5 Crystal classes	346	9,9%
Hexagonal	7 Crystal classes	326	9,3%
Trigonal-Rhomboedrisch	5 Crystal classes	295	8,4%
Tetragonal	7 Crystal classes	278	7,9%
Orthorhombisch	3 Crystal classes	774	22%
Monoklin	3 Crystal classes	1129	32,2%
Triklin	2 Crystal classes	332	9,5%
Amorph		30	0,8%

aus „Minerale: Bestimmen nach äußeren Kennzeichen / HOCHLEITNER; PHILIPSBORN; WEINER“

Wirkung der verschiedenen Drehachsen

Name	Symbol	Wirkung
Inversion	* / i	Inversion am Zentrum
Zweizählige Drehachse	• / 2	Drehungen um 180°
Spiegelebene / inverse zweizählige D.	m = $\overline{2}$	Spiegelung an einer Ebene
Dreizählige Drehachse	▲ / 3	Drehungen um 120°
Inverse dreizählige D.	△ / $\overline{3}$	Drehungen um 120° und Inversion
Vierzählige Drehachse	◆ / 4	Drehungen um 90°
Inverse vierzählige D.	◇ / $\overline{4}$	Drehungen um 90° und Inversion
Sechszählige Drehachse	● / 6	Drehungen um 60°
Inverse sechszählige D.	⊕ / $\overline{6}$	Drehungen um 60° und Inversion

Occurrence frequency of minerals

5091 minerals are known to date (www.mindat.org, august 2017).

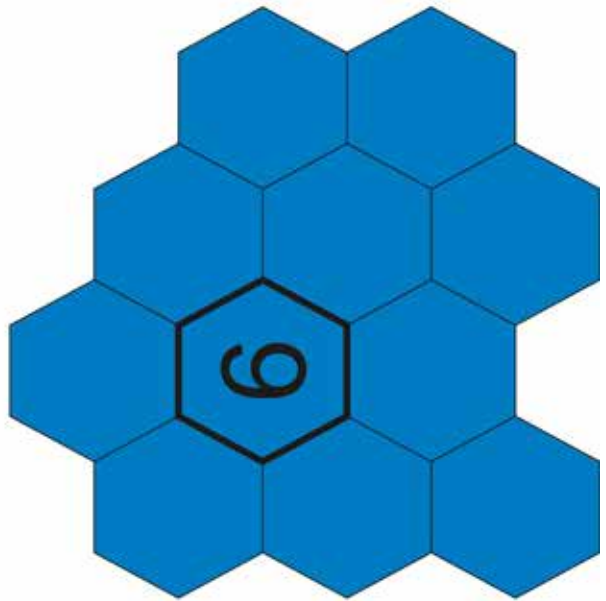
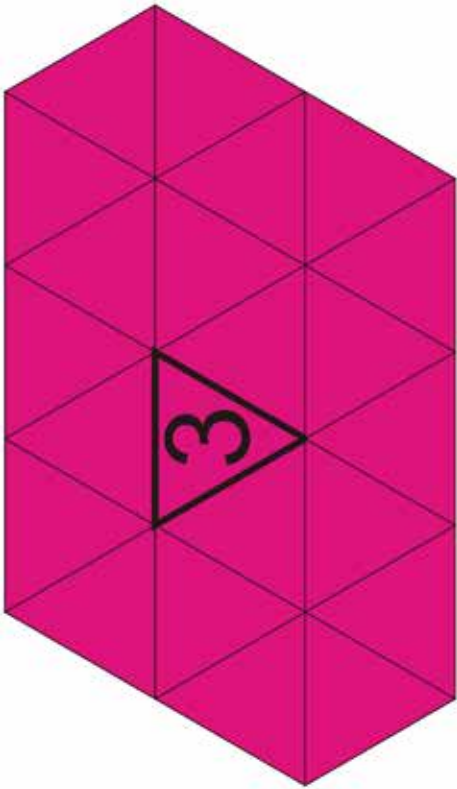
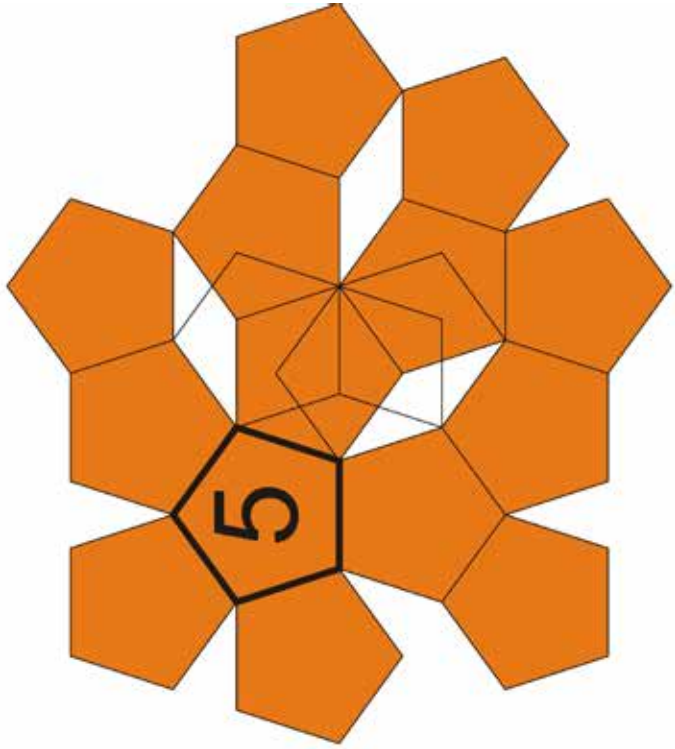
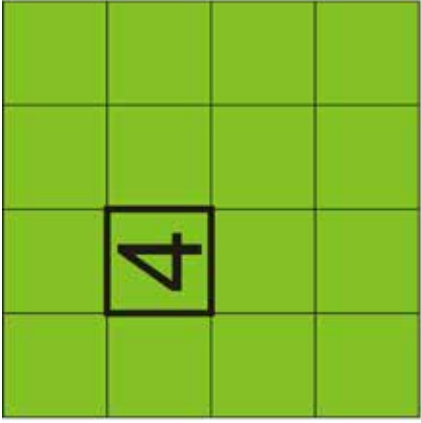
Following crystal systems are based on older data sets:

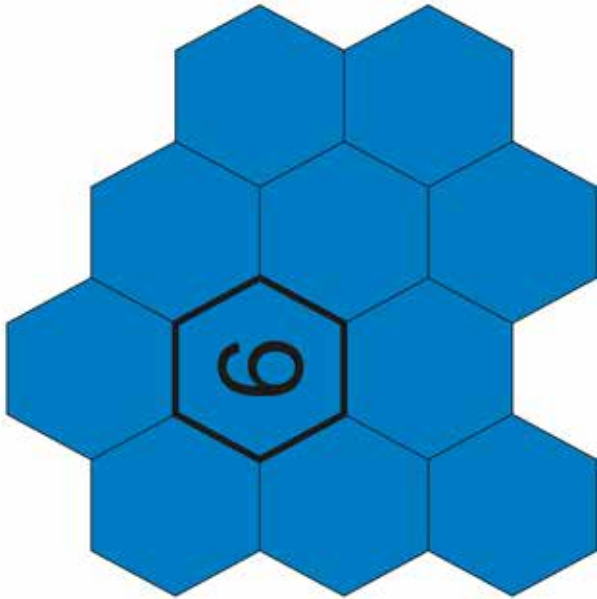
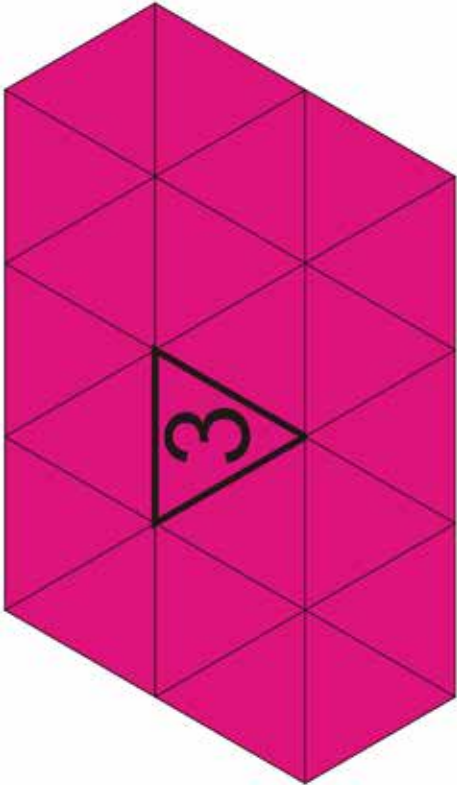
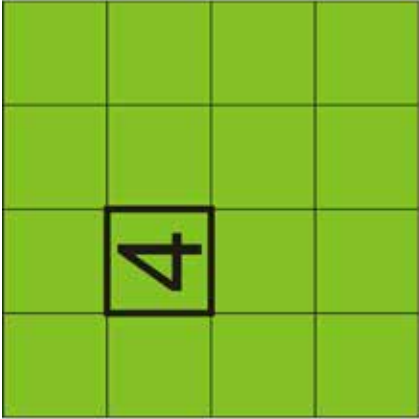
Cubic	5 Crystal classes	346	9,9%
Hexagonal	7 Crystal classes	326	9,3%
trigonal - rhombohedral	5 Crystal classes	295	8,4%
tetragonal	7 Crystal classes	278	7,9%
orthorhombic	3 Crystal classes	774	22%
monoclinic	3 Crystal classes	1129	32,2%
triclinic	2 Crystal classes	332	9,5%
amorphous		30	0,8%

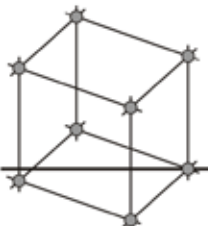
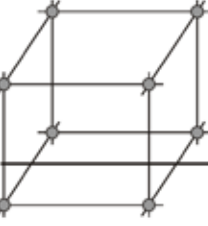

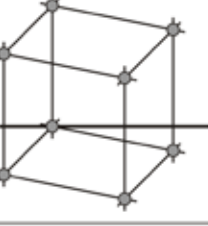
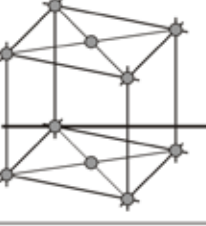
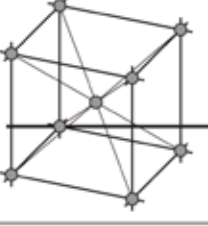

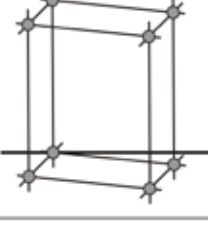
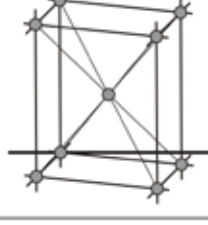
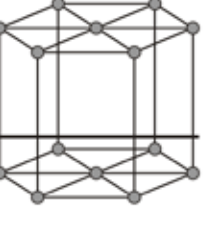
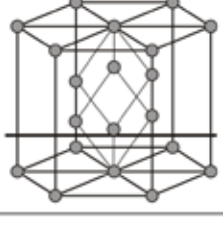
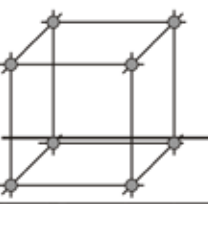
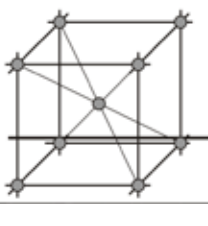
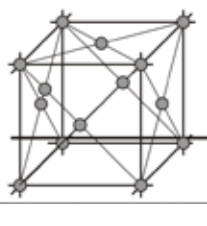
from „Minerale: Bestimmen nach äußeren Kennzeichen / HOCHLEITNER; PHILIPSBORN; WEINER“

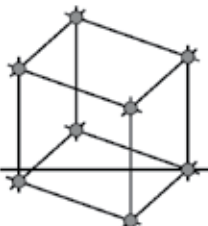
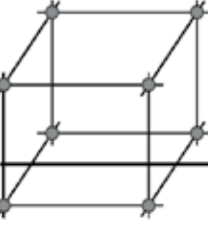

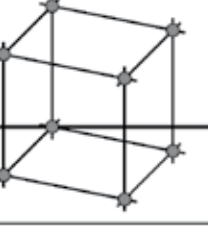
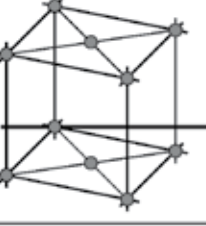
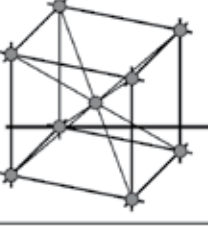

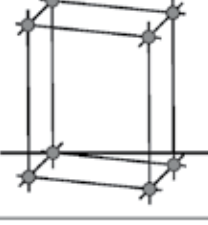
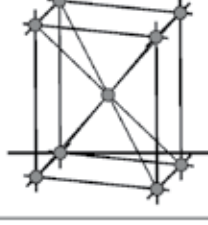
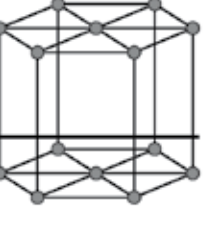
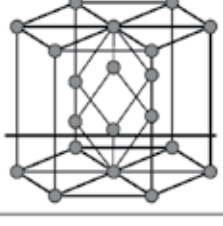
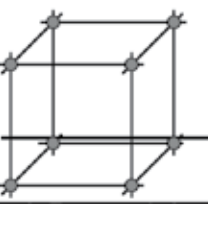
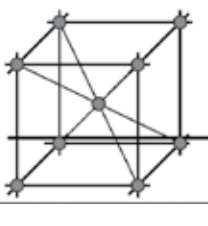
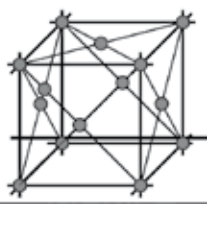
Influence of different axes

Name	Symbol	Influence
Inversion	* / i	Inversion at the center
Two-fold rotation axis	\bullet / 2	Rotation of 180°
Mirror plane / inverse two-fold rotation axis	$m = \bar{2}$	A reflection at a plane
Three-fold rotation axis	\blacktriangle / 3	Rotation of 120°
Inverse three-fold rotation axis	\triangle / $\bar{3}$	Rotation of 120° and Inversion
Four-fold rotation axis	\blacklozenge / 4	Rotation of 90°
Inverse four-fold rotation axis	\lozenge / $\bar{4}$	Rotation of 90° and Inversion
Six-fold rotation axis	\bullet / 6	Rotation of 60°
Inverse six-fold rotation axis	\blacklozenge / $\bar{6}$	Rotation of 60° and Inversion





	P	C	I	F
triklin				
monoklin			identisch mit C-Gitter	identisch mit C-Gitter
ortho- rhombisch				
tetragonal		identisch mit P-Gitter		identisch mit I-Gitter
trigonal				
hexagonal				
kubisch		unmöglich		

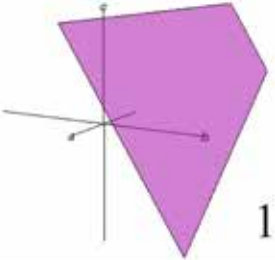
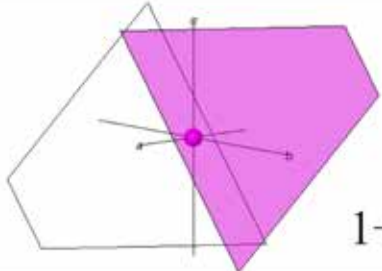
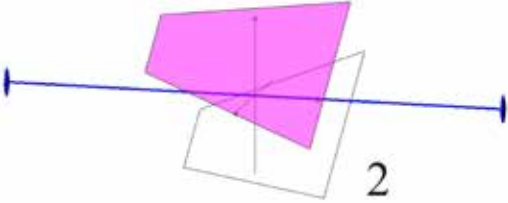
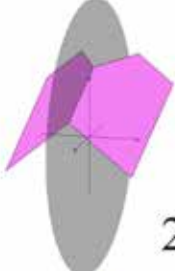
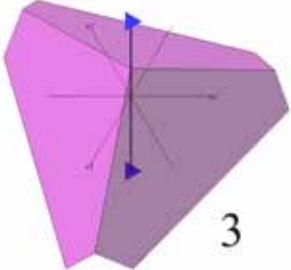
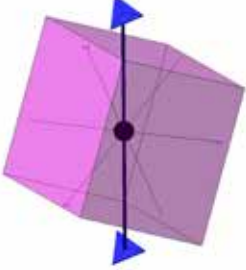
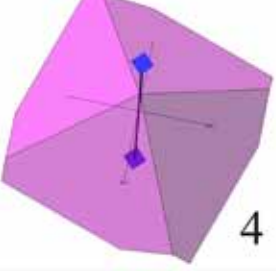

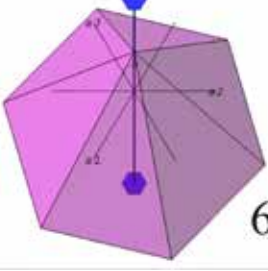
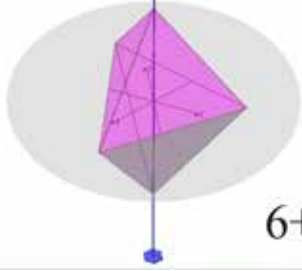
	P	C	I	F
triclinic				
monoclinic			identical with C-lattice	identical with C-lattice
orthorhombic				
tetragonal		identical with P-lattice		identical with I-lattice
trigonal				
hexagonal				
cubic		impossible		

Die Symmetrieelemente

(normale Drehachsen, Drehachsen kombiniert mit Inversion)

kombiniert mit
Inversionszentrum

normal

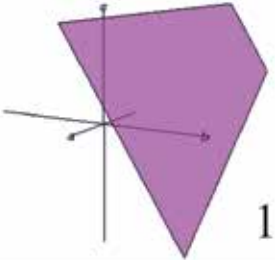
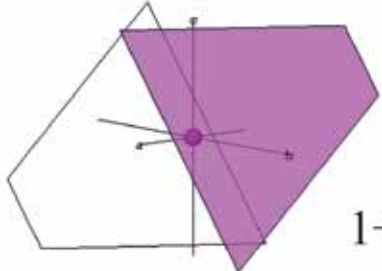
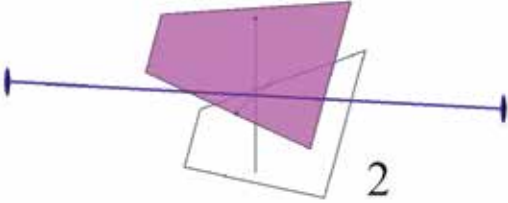
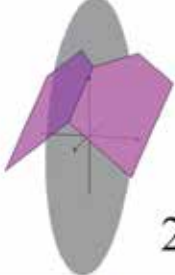
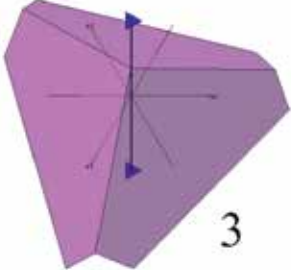
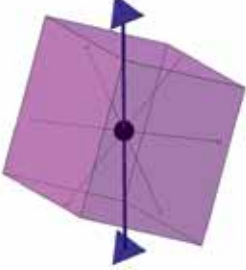
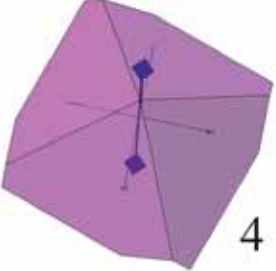

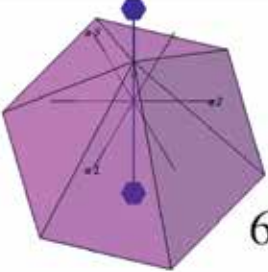
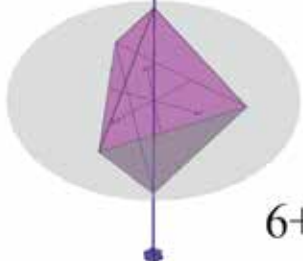
normal		kombiniert mit Inversionszentrum
 <p>1</p>	1	 <p>1+i</p>
 <p>2</p>	2	 <p>2+i = m</p>
 <p>3</p>	3	 <p>3+i</p>
 <p>4</p>	4	 <p>4+i</p>
 <p>6</p>	6	 <p>6+i</p>

Symmetry elements

(normal rotation axes, rotation axes combined with inversion)

combined with
inversion center

normal

normal		combined with inversion center
 <p>1</p>	<p>1</p>	 <p>1+i</p>
 <p>2</p>	<p>2</p>	 <p>2+i = m</p>
 <p>3</p>	<p>3</p>	 <p>3+i</p>
 <p>4</p>	<p>4</p>	 <p>4+i</p>
 <p>6</p>	<p>6</p>	 <p>6+i</p>

Die Symmetrieelemente

Symmetrie	Symbole nach	Symbol				
		*	●	▲	◆	●
(polare Drehachse)	Schönflies	C_1	C_2	C_3	C_4	C_6
	Hermann-Mauguin	1	2	3	4	6
Kombination von Drehachsen und horizontalen Symmetrieebenen	Schönflies		C_{2h}	C_{3h}	C_{4h}	C_{6h}
	Hermann-Mauguin		2/m	$3/m (\bar{6})$	4/m	6/m
Kombination von Drehachsen und vertikalen Symmetrieebenen	Schönflies	C_s	C_{2v}	C_{3v}	C_{4v}	C_{6v}
	Hermann-Mauguin	m	mm2	3m	4mm	6mm
Kombination von Drehachsen, horizontalen und vertikalen Symmetrieebenen	Schönflies		D_{2h}	D_{3h}	D_{4h}	D_{6h}
	Hermann-Mauguin		mmm	$\bar{6}2m$	4/mmm	6/mmm
Kombination mit zweizähligen Drehachsen	Schönflies		D_2	D_3	D_4	D_6
	Hermann-Mauguin		222	32	422	622
Symmetriezentrum und Inversionsachsen	Schönflies	C_i		C_{3i}	S_4	
	Hermann-Mauguin	T	$\bar{2} (m)$	$\bar{3}$	$\bar{4}$	$6 (3/m)$
				$\bar{3}m$	$\bar{4}2m$	$\bar{6}2m$

Elements of symmetry

Symmetry	Symbols according to	Symbol				
		*	●	▲	◆	◆
(polar rotation axis)	Schoenflies	C_1	C_2	C_3	C_4	C_6
	Hermann-Mauguin	1	2	3	4	6
Combination of rotary axes and horizontal symmetry planes	Schoenflies		C_{2h}	C_{3h}	C_{4h}	C_{6h}
	Hermann-Mauguin		$2/m$	$3/m (\bar{6})$	$4/m$	$6/m$
Combination of rotary axes and vertical symmetry planes	Schoenflies	C_s	C_{2v}	C_{3v}	C_{4v}	C_{6v}
	Hermann-Mauguin	m	$mm2$	$3m$	$4mm$	$6mm$
Combination of rotary axes, horizontal and vertical symmetry planes	Schoenflies		D_{2h}	D_{3h}	D_{4h}	D_{6h}
	Hermann-Mauguin		mmm	$\bar{6}2m$	$4/mmm$	$6/mmm$
Combination with two-fold rotary axes	Schoenflies		D_2	D_3	D_4	D_6
	Hermann-Mauguin		222	32	422	622
Symmetry center and inversion axes	Schoenflies	C_i		C_{3i}	S_4	
	Hermann-Mauguin	$\bar{1}$	$\bar{2} (m)$	3	4	$6 (3/m)$
				$\bar{3}m$	$\bar{4}2m$	$\bar{6}2m$

Die 32 Punktgruppen

Kristallsystem	Triklin	Monoklin	Orthorhombisch	Trigonal	Hexagonal	Tetragonal	Kubisch
Holoedrie	$\bar{1}$	$\frac{2}{m}$	$\frac{2}{m} \frac{2}{m} \frac{2}{m}$	$\frac{2}{3} \frac{2}{m}$	$\frac{6}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{2}{m} \frac{2}{m}$
Blickrichtung	- keine-	[010]	[100] [010] [001]	[001] [100]	[001] [100] [110]	[001] [100] [110]	[001] [111] [110]
Punktgruppen	1 $\bar{1}$	2 m $\frac{2}{m}$	2 2 2 m m 2 $\frac{2}{m} \frac{2}{m} \frac{2}{m}$	3 3 2 3 m $\bar{3}$ $\frac{2}{3} \frac{2}{m}$	6 6 2 2 6 m m $\bar{6}$ $\bar{6} 2 m$ $\bar{6} m 2$ $\frac{6}{m}$ $\frac{6}{m} \frac{2}{m} \frac{2}{m}$	4 4 2 2 4 m m $\bar{4}$ $\bar{4} 2 m$ $\bar{4} m 2$ $\frac{4}{m}$ $\frac{4}{m} \frac{2}{m} \frac{2}{m}$	2 3 4 3 2 $\frac{2}{m} \frac{2}{m} \frac{3}{m}$ $\bar{4} 3 m$ $\frac{4}{m} \frac{2}{m} \frac{2}{m}$

32 Crystallographic point groups

Crystal systems	Triclinic	Monoclinic	Orthorhombic	Trigonal	Hexagonal	Tetragonal	Cubic
Holoedry	$\bar{1}$	$\frac{2}{m}$	$\frac{2}{m} \frac{2}{m} \frac{2}{m}$	$\frac{3}{m}$	$\frac{6}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{2}{m} \frac{2}{m}$	$\frac{4}{m} \frac{3}{m} \frac{2}{m}$
Viewing direction	- keine-	[010]	[100] [010] [001]	[001] [100]	[001] [100] [110]	[001] [100] [110]	[001] [111] [110]
Point groups	1 $\bar{1}$	2 m $\frac{2}{m}$	2 2 2 m m 2 $\frac{2}{m} \frac{2}{m} \frac{2}{m}$	3 3 2 3 m $\bar{3}$ $\frac{3}{2} \frac{2}{m}$	6 6 2 2 6 m m $\bar{6}$ $\bar{6} 2 m$ $\bar{6} m 2$ $\frac{6}{m}$ $\frac{6}{m} \frac{2}{m} \frac{2}{m}$	4 4 2 2 4 m m $\bar{4}$ $\bar{4} 2 m$ $\bar{4} m 2$ $\frac{4}{m}$ $\frac{4}{m} \frac{2}{m} \frac{2}{m}$	2 3 4 3 2 $\frac{2}{m} \frac{3}{m}$ $\bar{4} 3 m$ $\frac{4}{m} \frac{3}{m} \frac{2}{m}$

Die 32 Kristallklassen

Anzahl der Klassen	Bezeichnung nach Herman-Mauguin		Bezeichnung der Kristallklasse
	komplett	abgekürzt	
Triklines System			
2 Klassen	1	1	Triklin Pedial
	-1	-1	Triklin Pinakoidal
Monoklines System			
3 Klassen	$2/m$	$2/m$	Monoklin Prismatisch
	m	m	Monoklin Domatisch
	2	2	Monoklin Sphenoid
Orthorhombisches System			
3 Klassen	$2/m2/m/m$	mmm	Orthorhombisch Dipyramidal
	$mm2$	$mm2$	Orthorhombisch Pyramidal
	222	222	Orthorhombisch Disphenoid
Tetragonales System			
7 Klassen	$4/m2/m2/m$	$4/mmm$	Ditetragonal Dipyramidal
	$4mm$	$4mm$	Ditetragonal Pyramidal
	422	422	Tetragonal Trapezoedrisch
	$4/m$	$4/m$	Tetragonal Dipyramidal
	4	4	Tetragonal Pyramidal
	$4-2 m$	$4-2 m$	Tetragonal Skalenoedrisch
	-4	-4	Tetragonal Disphenoidisch
Hexagonales System:			
7 Klassen	$6/m2/m2/m$	$6/mmm$	Dihexagonal Dipyramidal
	$6mm$	$6mm$	Dihexagonal Pyramidal
	622	622	Hexagonal Trapezoedrisch
	$6/m$	$6/m$	Hexagonal Dipyramidal
	6	6	Hexagonal Pyramidal
	$-6m2$	$-6m2$	Ditrigonal Dipyramidal
	-6	-6	Trigonal Dipyramidal
Trigonales System:			
5 Klassen	$-3 2m$	$-3m$	Ditrigonal Skalenoedrisch
	$3m$	$3m$	Ditrigonal Pyramidal
	32	32	Trigonal Trapezoedrisch
	3	3	Trigonal Pyramidal
	-3	-3	Trigonal Rhomboedrisch
Kubisches System			
5 Klassen	$4/m-3 2/m$	$m3m$	Hexakisoktaedrisch
	432	432	Pentagonikositetraedrisch
	$2/m -3$	$m3$	Disdodekaedrisch
	$-4 3m$	$-4 3m$	Hexakistetraedrisch
	23	23	Tetraedrisch Pentagonododekaedrisch

The 32 crystal classes

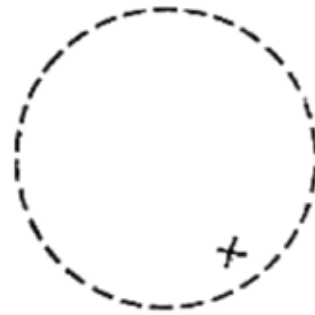
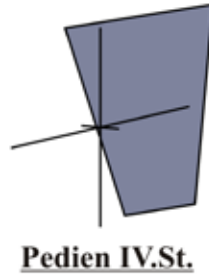
Amount of Classes	Symbol according Hermann-Mauguin		Crystal class
	complete	Abrev.	
Triclinic System			
2 Classes	1	1	Triclinic Pedial
	-1	-1	Triclinic Pinakoidal
Monoclinic System			
3 Classes	$2/m$	$2/m$	Monoclinic Prismatic
	m	m	Monoclinic Domatic
	2	2	Monoclinic Sphenoid
Orthorhombic System			
3 Classes	$2/m2/m/m$	mmm	Orthorhombic Dipyramidal
	$mm2$	$mm2$	Orthorhombic Pyramidal
	222	222	Orthorhombic Disphenoid
Tetragonal System			
7 Classes	$4/m2/m2/m$	$4/mmm$	Ditetragonal Dipyramidal
	$4mm$	$4mm$	Ditetragonal Pyramidal
	422	422	Tetragonal Trapezoedric
	$4/m$	$4/m$	Tetragonal Dipyramidal
	4	4	Tetragonal Pyramidal
	$4-2 m$	$4-2 m$	Tetragonal Scenoedric
	-4	-4	Tetragonal Disphenoidic
Hexagonal System:			
7 Classes	$6/m2/m2/m$	$6/mmm$	Dihexagonal Dipyramidal
	$6mm$	$6mm$	Dihexagonal Pyramidal
	622	622	Hexagonal Trapezoedric
	$6/m$	$6/m$	Hexagonal Dipyramidal
	6	6	Hexagonal Pyramidal
	$-6m2$	$-6m2$	Ditrigonal Dipyramidal
	-6	-6	Trigonal Dipyramidal
Trigonal System:			
5 Classes	$-3 2m$	$-3m$	Ditrigonal Scenoedric
	$3m$	$3m$	Ditrigonal Pyramidal
	32	32	Trigonal Trapezoedric
	3	3	Trigonal Pyramidal
	-3	-3	Trigonal Rhomboedric
Cubic System			
5 Classes	$4/m-3 2/m$	$m3m$	Hexakisoktaedric
	432	432	Pentagonikositetraedric
	$2/m -3$	$m3$	Disdodekaedric
	$-4 3m$	$-4 3m$	Hexakistetraedric
	23	23	Tetraedric Pentagondodekaedric

Triklines Kristallsystem

Trikline Hemiedrie
Triklin-pediale Klasse

Symbol: 1 oder C_1

Allgemeine Form:
{hkl}



{0kl}	{h0l}	{hk0}
<p><u>Pedien I.St.</u></p>	<p><u>Pedien II.St.</u></p>	<p><u>Pedien III.St.</u></p>
{100}	{010}	{001}
<p><u>I. Pedion</u></p>	<p><u>II. Pedion</u></p>	<p><u>III. Pedion</u></p>

Mineral-Bsp.:

Sinnerite $Cu_6As_4S_9$

Hartite $C_{20}H_{34}$

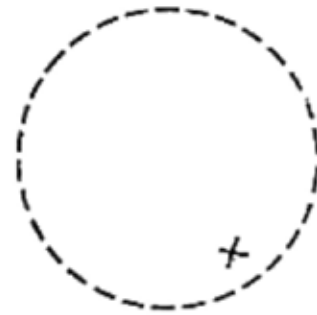
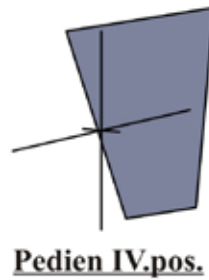
Nekoite $Ca_3[Si_6O_{15}] \cdot 7H_2O$

Triclinic crystal system

Triclinic hemiedry
Triclinic pedial class

Symbol: 1 or C_1

General form:
{hkl}



{0kl}	{h0l}	{hk0}
<u>Pedien I.pos.</u>	<u>Pedien II.pos.</u>	<u>Pedien III.pos.</u>
{100}	{010}	{001}
<u>I. Pedion</u>	<u>II. Pedion</u>	<u>III. Pedion</u>

mineral examples:

Sinnerite $Cu_6As_4S_9$

Hartite $C_{20}H_{34}$

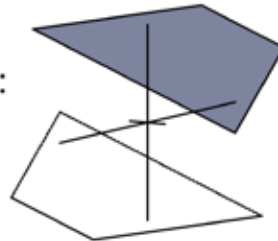
Nekoite $Ca_3[Si_6O_{15}] \cdot 7H_2O$

Triklines Kristallsystem

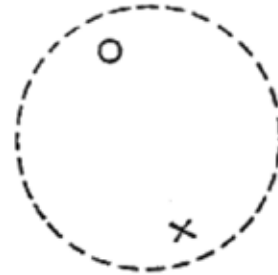
Triklone Holoedrie
Triklin-pinakoidale Klasse

Symbol: $\bar{1}$ oder C_i

Allgemeine Form:
{hkl}



Pinakoide IV.St.



{0kl}	{h0l}	{hk0}
<p><u>Pinakoide I.St.</u></p>	<p><u>Pinakoide II.St.</u></p>	<p><u>Pinakoide III.St.</u></p>
{100}	{010}	{001}
<p><u>I. Pinakoid</u></p>	<p><u>II. Pinakoid</u></p>	<p><u>III. Pinakoid</u></p>

Mineral-Bsp.:

Albit $\text{Na}[\text{AlSi}_3\text{O}_8]$

Kyanit/Disthen $\text{Al}_2[\text{O}/\text{SiO}_4]$

Wollastonit $\text{Ca}[\text{SiO}_3]$

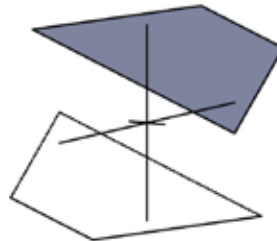
Axinit $\text{Ca}_2(\text{Fe},\text{Mn})\text{Al}_2[\text{BO}_3\text{OH}/\text{Si}_4\text{O}_{12}]$

Triclinic crystal system

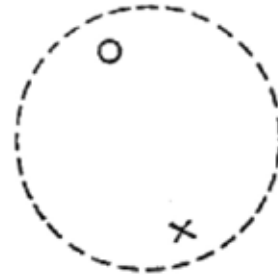
Triclinic holoedry
Triclinic pinacoidal class


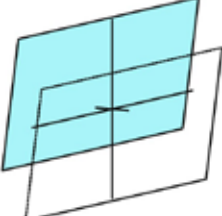
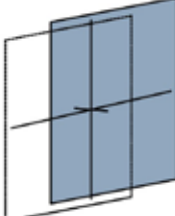
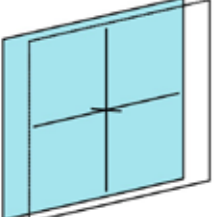


Symbol: $\bar{1}$ or C_i

General form:
{hkl}



Pinacoid IV.pos.



{0kl}	{h0l}	{hk0}
 <p data-bbox="300 1144 459 1173"><u>Pinacoid I.St.</u></p>	 <p data-bbox="715 1144 874 1173"><u>Pinacoid II.St.</u></p>	 <p data-bbox="1120 1144 1295 1173"><u>Pinacoid III.St.</u></p>
{100}	{010}	{001}
 <p data-bbox="316 1552 450 1581"><u>I. Pinacoid</u></p>	 <p data-bbox="730 1552 865 1581"><u>II. Pinacoid</u></p>	 <p data-bbox="1136 1552 1279 1581"><u>III. Pinacoid</u></p>

mineral examples:

Albite $Na[AlSi_3O_8]$

Kyanite/Disthene $Al_2[O/SiO_4]$

Wollastonite $Ca[SiO_3]$

Axinite $Ca_2(Fe,Mn)Al_2[BO_3OH/Si_4O_{12}]$

Monoklines Kristallsystem

Monokline Hemimorphie

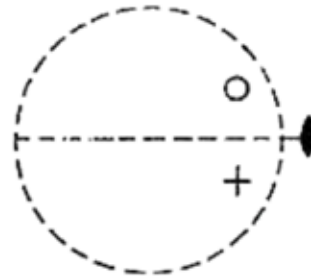
Symbol: 2 oder C_2

Monoklin-sphenoidische Klasse

Allgemeine Form:
 $\{hkl\}$



Sphenoide IV.St.



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<p><u>Sphenoide I.St.</u></p>	<p><u>Pinakoide II.St.</u></p>	<p><u>Sphenoide III.St.</u></p>
$\{100\}$	$\{010\}$	$\{001\}$
<p><u>I. Pinakoid</u></p>	<p><u>II. Pedion</u></p>	<p><u>III. Pinakoid</u></p>

Mineral-Bsp.:

(Rohr-/Kandis-)Zucker $C_{12}H_{22}O_{11}$

Weinsäure $C_4H_6O_6$

Latiumite $(Ca,K)_8(Al,Mg,Fe)[(Si,Al)_{10}O_{25}/(SO_4)]$

Monoclinic crystal system

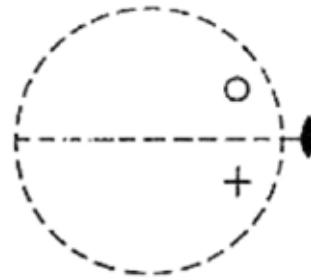
Monoclinic hemimorphy
 Monoclinic sphenoidal class

Symbol: 2 or C_2

General form:
 $\{hkl\}$



Sphenoid IV.pos.



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<p><u>Sphenoid I.pos.</u></p>	<p><u>Pinacoid II.pos.</u></p>	<p><u>Sphenoid III.pos.</u></p>
$\{100\}$	$\{010\}$	$\{001\}$
<p><u>I. Pinacoid</u></p>	<p><u>II. Pedion</u></p>	<p><u>III. Pinacoid</u></p>

mineral examples:

(cane/candy)sugar $C_{12}H_{22}O_{11}$

Tartaric acid $C_4H_6O_6$

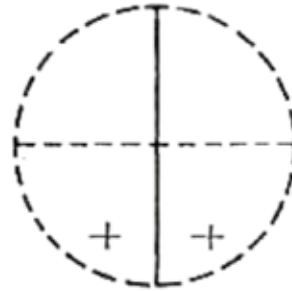
Latiumite $(Ca,K)_8(Al,Mg,Fe)[(Si,Al)_{10}O_{25}/(SO_4)]$

Monoklines Kristallsystem

Monokline Hemiedrie II. Art
Monoklin-domatische Klasse

Symbol: m oder C_s

Allgemeine Form:
 $\{hkl\}$



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<p><u>Domen I.St.</u></p>	<p><u>Pediën II.St.</u></p>	<p><u>Domen III.St.</u></p>
$\{100\}$	$\{010\}$	$\{001\}$
<p><u>I. Pedion</u></p>	<p><u>II. Pinakoid</u></p>	<p><u>III. Pedion</u></p>

Mineral-Bsp.:

Posnjakite $Cu_4[(OH)_6/SO_4] \cdot H_2O$

Hilgardit $Ca_2[Cl/B_5O_8(OH)_2]$

Skolezit $Ca[Al_2Si_3O_{10}] \cdot 3H_2O$

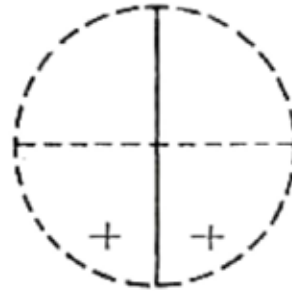
Monoclinic crystal system

Monoclinic hemiedry II. type Symbol: m or C_s
 Monoclinic domatic class

General form:
 $\{hkl\}$



Domen IV.pos.



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<p><u>Domen I.pos.</u></p>	<p><u>Pedien II.pos.</u></p>	<p><u>Domen III.pos.</u></p>
$\{100\}$	$\{010\}$	$\{001\}$
<p><u>I. Pedion</u></p>	<p><u>II. Pinacoid</u></p>	<p><u>III. Pedion</u></p>

mineral examples:

Posnjakite $Cu_4[(OH)_6/SO_4] \cdot H_2O$

Hilgardite $Ca_2[Cl/B_5O_8(OH)_2]$

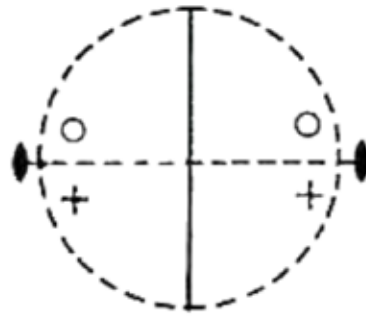
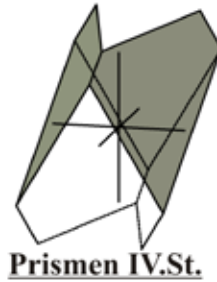
Scolecite $Ca[Al_2Si_3O_{10}] \cdot 3H_2O$

Monoklines Kristallsystem

Monokline Holoedrie
Monoklin-prismatische Klasse

Symbol: $2/m$ oder C_{2h}

Allgemeine Form:
{hkl}



{0kl}	{h0l}	{hk0}
<p>Prismen I.St.</p>	<p>Pinakoide II.St.</p>	<p>Prismen III.St.</p>
{100}	{010}	{001}
<p>I. Pinakoid</p>	<p>II. Pinakoid</p>	<p>III. Pinakoid</p>

Mineral-Bsp.:

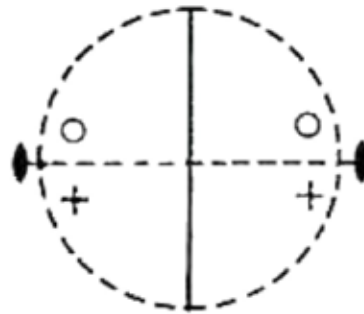
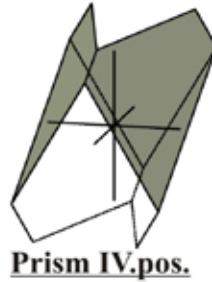
- häufigste Kristallklasse
- siehe Zusatzseite

Monoclinic crystal system

Monoclinic Holoedry
Monoclinic prismatic class

Symbol: $2/m$ or C_{2h}

General form:
{hkl}



{0kl}	{h0l}	{hk0}
<p>Prism I.pos.</p>	<p>Pinacoid II.pos.</p>	<p>Prism III.pos.</p>
{100}	{010}	{001}
<p>I. Pinacoid</p>	<p>II. Pinacoid</p>	<p>III. Pinacoid</p>

mineral examples:

- most common crystal class
- see additional page

Mineral-Bsp. für 2/m: (häufigste Kristallklasse)Gips $\text{Ca}[\text{SO}_4] \cdot 2\text{H}_2\text{O}$ Diopsid $\text{Ca}(\text{Mg}, \text{Fe})[\text{Si}_2\text{O}_6]$ Orthoklas $\text{K}[\text{AlSi}_3\text{O}_8]$ Soda $\text{Na}_2[\text{CO}_3] \cdot 10\text{H}_2\text{O}$ Arsenopyrit FeAsS Azurit $\text{Cu}_3[(\text{CO}_3)_2/(\text{OH})_2]$ Biotit $\text{K}(\text{Mg}, \text{Fe}^{2+}, \text{Mn}^{2+})_3[(\text{OH}, \text{F})_2/(\text{Al}, \text{Fe}^{3+}, \text{Ti}^{3+})\text{Si}_3\text{O}_{10}]$ Borax $\text{Na}_2[\text{B}_4\text{O}_5/(\text{OH})_4] \cdot 8\text{H}_2\text{O}$ Chalkosin Cu_2S Epidot $\text{Ca}_2(\text{Fe}, \text{Al})_3[(\text{Si}_2\text{O}_7)/(\text{SiO}_4)_3/(\text{OH})_2]$ Hornblende $\text{Ca}_2(\text{Mg}, \text{Fe}, \text{Al})_5[(\text{Al}, \text{Si})_8\text{O}_{22}/(\text{OH})_2]$ Kryolith Na_3AlF_6 Malachit $\text{Cu}_2[(\text{CO}_3)/(\text{OH})_2]$ Muskovit $\text{KAl}_2[\text{Si}_3\text{AlO}_{10}/(\text{OH}, \text{F})_2]$ Phlogopit $\text{KMg}_3[\text{Si}_3\text{AlO}_{10}/(\text{F}, \text{OH})_2]$ Realgar AsS Sanidin $(\text{K}, \text{Na})[\text{Al}_3\text{SiO}_8]$ Staurolith $(\text{Fe}^{2+}, \text{Mg}, \text{Zn})_2\text{Al}_9[(\text{Si}, \text{Al})_4\text{O}_{22}/(\text{OH})_2]$ Titanit $\text{CaTi}[\text{O}/\text{SiO}_4]$

mineral examples for 2/m: (most common crystal class)Gypsum $\text{Ca}[\text{SO}_4] \cdot 2\text{H}_2\text{O}$ Diopside $\text{Ca}(\text{Mg}, \text{Fe})[\text{Si}_2\text{O}_6]$ Orthoclase $\text{K}[\text{AlSi}_3\text{O}_8]$ Soda $\text{Na}_2[\text{CO}_3] \cdot 10\text{H}_2\text{O}$ Arsenopyrite FeAsS Azurite $\text{Cu}_3[(\text{CO}_3)_2/(\text{OH})_2]$ Biotite $\text{K}(\text{Mg}, \text{Fe}^{2+}, \text{Mn}^{2+})_3[(\text{OH}, \text{F})_2/(\text{Al}, \text{Fe}^{3+}, \text{Ti}^{3+})\text{Si}_3\text{O}_{10}]$ Borax $\text{Na}_2[\text{B}_4\text{O}_5/(\text{OH})_4] \cdot 8\text{H}_2\text{O}$ Chalcocite Cu_2S Epidote $\text{Ca}_2(\text{Fe}, \text{Al})_3[(\text{Si}_2\text{O}_7)/(\text{SiO}_4)_3/(\text{OH})_2]$ Hornblende $\text{Ca}_2(\text{Mg}, \text{Fe}, \text{Al})_5 [(\text{Al}, \text{Si})_8\text{O}_{22}/(\text{OH})_2]$ Cryolite Na_3AlF_6 Malachite $\text{Cu}_2[(\text{CO}_3)/(\text{OH})_2]$ Muscovite $\text{KAl}_2[\text{Si}_3\text{AlO}_{10}/(\text{OH}, \text{F})_2]$ Phlogopite $\text{KMg}_3[\text{Si}_3\text{AlO}_{10}/(\text{F}, \text{OH})_2]$ Realgar AsS Sanidine $(\text{K}, \text{Na})[\text{Al}_3\text{SiO}_8]$ Staurolite $(\text{Fe}^{2+}, \text{Mg}, \text{Zn})_2\text{Al}_9[(\text{Si}, \text{Al})_4\text{O}_{22}/(\text{OH})_2]$ Titanite $\text{CaTi}[\text{O}/\text{SiO}_4]$

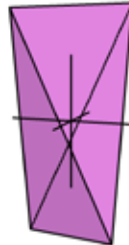
Orthorhombisches Kristallsystem

Rhombische Hemiedrie

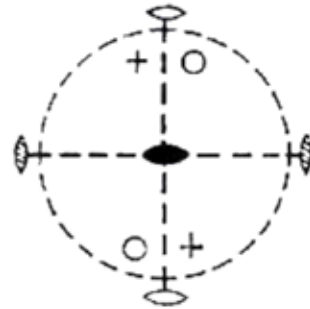
Symbol: 222 oder D_2

Rhombisch-disphenoidische Klasse

Allgemeine Form:
 $\{hkl\}$



Rhombische Disphenoide



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<p><u>Prismen I.St.</u></p>	<p><u>Prismen II.St.</u></p>	<p><u>Prismen III.St.</u></p>
$\{100\}$	$\{010\}$	$\{001\}$
<p><u>I. Pinakoid</u></p>	<p><u>II. Pinakoid</u></p>	<p><u>III. Pinakoid</u></p>

Mineral-Bsp.:

Epsomit (Bittersalz) $MgSO_4 \cdot 7H_2O$

Zinkvitriol/Goslarit $ZnSO_4 \cdot 7H_2O$

Austinit $CaZn[(AsO_4)/(OH)]$

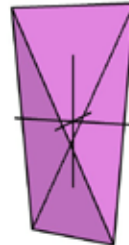
Arsenoclasite $Mn_5[(AsO_4)_2/(OH)_4]$

Orthorhombic crystal system

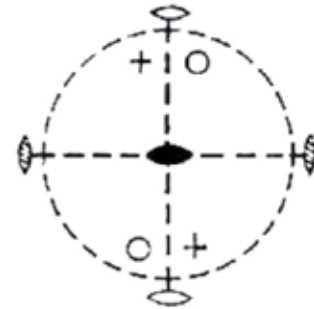
Rhombic hemiedry
Rhombic-disphenoidal class

Symbol: 222 or D_2

General form:
 $\{hkl\}$



Rhombic disphenoids



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<p>Prism I.pos.</p>	<p>Prism II.pos.</p>	<p>Prism III.pos.</p>
$\{100\}$	$\{010\}$	$\{001\}$
<p>I. Pinacoid</p>	<p>II. Pinacoid</p>	<p>III. Pinacoid</p>

mineral examples:

Epsomite (Epsom salt) $MgSO_4 \cdot 7H_2O$

Zinc vitriol/ Goslarite $ZnSO_4 \cdot 7H_2O$

Austinite $CaZn[(AsO_4)/(OH)]$

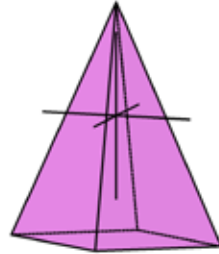
Arsenoclasite $Mn_5[(AsO_4)_2/(OH)_4]$

Orthorhombisches Kristallsystem

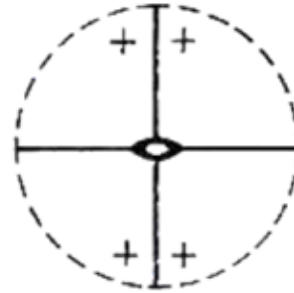
Rhombische Hemimorphie
Rhombisch-pyramidale Klasse

Symbol: $mm2$ oder C_{2v}
(mm)

Allgemeine Form:
{hkl}



Rhombische Pyramiden



{0kl}	{h0l}	{hk0}
<p data-bbox="311 1142 454 1176"><u>Domen I.St.</u></p>	<p data-bbox="726 1142 869 1176"><u>Domen II.St.</u></p>	<p data-bbox="1125 1142 1300 1176"><u>Prismen III.St.</u></p>
{100}	{010}	{001}
<p data-bbox="311 1556 454 1590"><u>I. Pinakoid</u></p>	<p data-bbox="726 1556 869 1590"><u>II. Pinakoid</u></p>	<p data-bbox="1149 1556 1276 1590"><u>III. Pedien</u></p>

Mineral-Bsp.:

Hemimorphit/Kieselzinkerz $Zn_4[(OH)_2/Si_2O_7]*H_2O$

Struvit $MgNH_4[PO_4]*6H_2O$

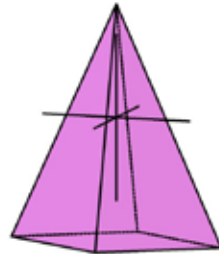
Prehnit $Ca_2Al_2[Si_3O_{10}/(OH)_2]$

Orthorhombic crystal system

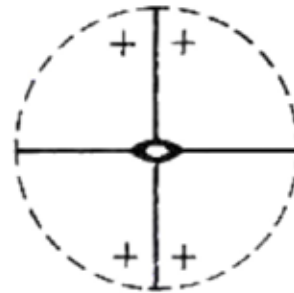
Rhombic Hemimorphy
Rhombic-pyramidal class

Symbol: $mm2$ or C_{2v}
(mm)

General form:
{hkl}



Rhombic pyramids



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
$\{100\}$	$\{010\}$	$\{001\}$

mineral examples:

Hemimorphite/Kieselgalmey $Zn_4[(OH)_2/Si_2O_7]*H_2O$

Struvite $MgNH_4[PO_4]*6H_2O$

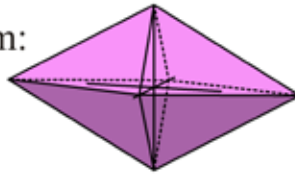
Prehnite $Ca_2Al_2[Si_3O_{10}/(OH)_2]$

Orthorhombisches Kristallsystem

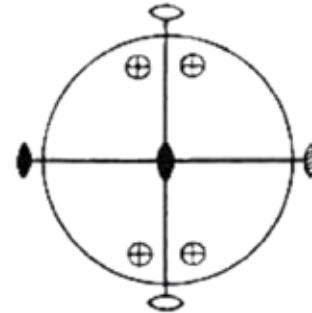
Rhombische Holoedrie
Rhombisch-dipyramidale Klasse

Symbol: $2/m\ 2/m\ 2/m$
(mmm) oder D_{2h}

Allgemeine Form:
 $\{hkl\}$



Rhombische Dipyramiden



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<u>Prismen I.St.</u>	<u>Prismen II.St.</u>	<u>Prismen III.St.</u>
$\{100\}$	$\{010\}$	$\{001\}$
<u>I. Pinakoid</u>	<u>II. Pinakoid</u>	<u>III. Pinakoid</u>

Mineral-Bsp.:

Schwefel S

Aragonit $\text{Ca}[\text{CO}_3]$

Anhydrit $\text{Ca}[\text{SO}_4]$

Baryt $\text{Ba}[\text{SO}_4]$

Topas $\text{Al}_2[\text{F}_2/\text{SiO}_4]$

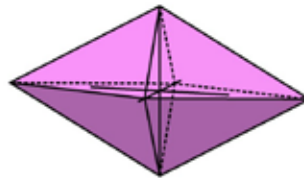
Olivin $(\text{Mg,Fe})_2[\text{SiO}_4]$

Orthorhombic crystal system

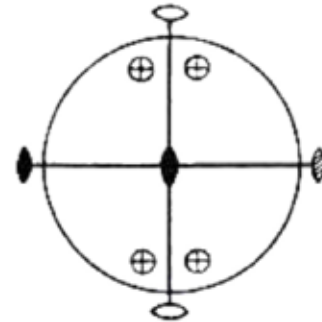
Rhombic Holoedry
Rhombic-bipyramidal class

Symbol: $2/m\ 2/m\ 2/m$
(mmm) or D_{2h}

General form:
 $\{hkl\}$



Rhombic bipyramid



$\{0kl\}$	$\{h0l\}$	$\{hk0\}$
<u>Prism I.pos.</u>	<u>Prism II.pos.</u>	<u>Prism III.pos.</u>
$\{100\}$	$\{010\}$	$\{001\}$
<u>I. Pinacoid</u>	<u>II. Pinacoid</u>	<u>III. Pinacoid</u>

mineral examples:

Sulphur S

Baryte $Ba[SO_4]$

Aragonite $Ca[CO_3]$

Topaz $Al_2[F_2/SiO_4]$

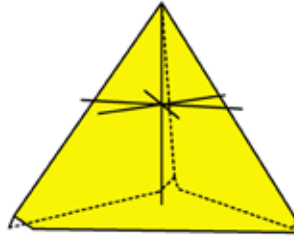
Anhydrite $Ca[SO_4]$

Olivine $(Mg,Fe)_2[SiO_4]$

Trigonales Kristallsystem

Hemimorphie d. trig. Tetartoedrie Symbol: 3 oder C_3
 Trigonal-pyramidale Klasse

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Trigonale Pyramiden III. St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Trig. Pyramiden I.St.</u>	<u>Trig. Pyramiden II.St.</u>	<u>Trig. Prismen III.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Trig. Prismen I.St.</u>	<u>Trig. Prismen II.St.</u>	<u>Basispedien</u>

Mineral-Bsp.:

Susannite $Pb_4[(SO_4)/(CO_3)_2/(OH)_2]$

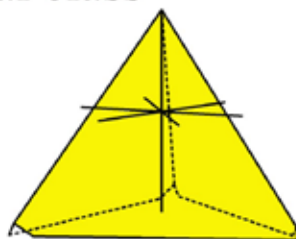
Carlinit TlS_2

Bleigermanat(Tieftemp.Modifikation) $Pb_5Ge_3O_{11}$

Trigonal crystal system

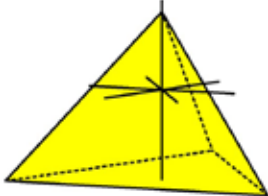
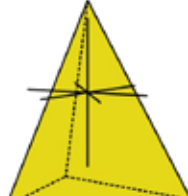
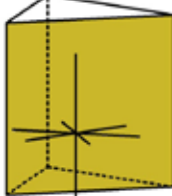
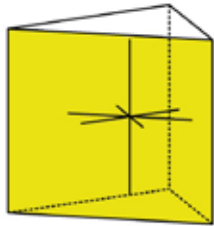
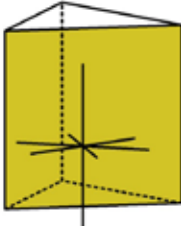
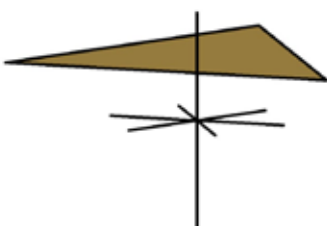
Hemimorphy of trig. Tetartoedry Symbol: 3 or C_3
 Trigonal pyramidal class

General form:
 $\{hkl\} / \{hkil\}$



Trigonal pyramid III. pos.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
Trig. Pyramid I.pos.	Trig. Pyramid II.pos.	Trig. Prism III.pos.
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
Trig. Prism I.pos.	Trig. Prism II.pos.	Basispedia

mineral examples:

Susannite $Pb_4[(SO_4)/(CO_3)_2/(OH)_2]$

Carlinite TlS_2

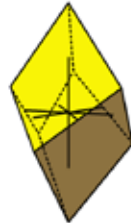
Lead germanate (low-temperature modification) $Pb_5Ge_3O_{11}$

Trigonales Kristallsystem

Hexagonal-trapezoedrische
Tetartoedrie
Trigonal-trapezoedrische Klasse

Symbol: 32 oder D_3

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Trig. Trapezoeder



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Rhomboeder I.St.</u>	<u>Trig. Dipyramiden II.St.</u>	<u>Ditrig. Prismen II.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Hex. Prisma I.St.</u>	<u>Trig. Prismen II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Zinnober/Cinnabarit HgS

Tiefquarz SiO_2

Selen Se

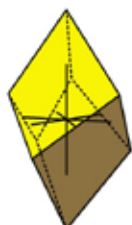
Tellur Te

Trigonal crystal system

Hexagonal trapezohedral tetartoeidry
 Trigonal trapezohedral Class


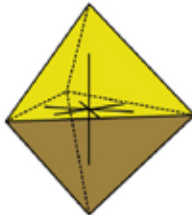
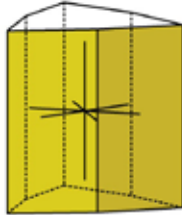
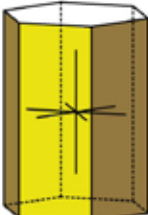
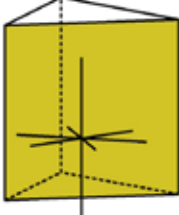
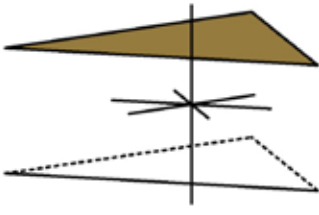
Symbol: 32 or D_3

General form:
 $\{hkl\} / \{hkil\}$



Trig. Trapezohedron



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Rhombohedron I.pos.</u>	<u>Trig. Bipyramid II.pos.</u>	<u>Ditrig. Prism II.pos.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex. Prism I.pos.</u>	<u>Trig. Prism II.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Cinnabar HgS

Quartz SiO_2

Selenium Se

Tellurium Te

Trigonales Kristallsystem

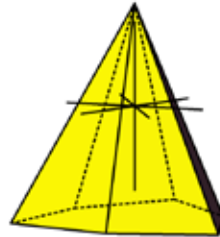
Hemimorphie d. rhomboedrischen

Hemiedrie

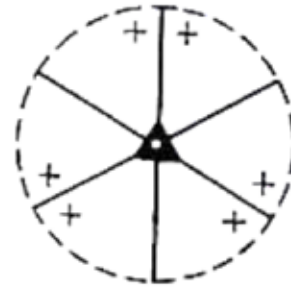
Ditrigonal-pyramidale Klasse

Symbol: $3m$ oder C_{3v}

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Ditrig. Pyramiden I.St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hh1\} / \{hh2\bar{h}l\}$	$\{hk0\} / \{hki0\}$
<u>Trig. Pyramiden I.St.</u>	<u>Hex. Pyramiden II.St.</u>	<u>Ditrig. Prismen II.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Trig. Prismen I.St.</u>	<u>Hex. Prisma II.St.</u>	<u>Basispedien</u>

Mineral-Bsp.:

Turmalin $(X)(Y_3)(Z_6)[Si_6O_{18}/(BO_3)_3/(V_3)(W)]$

Proustit Ag_3AsS_3

Lithiumniobat $LiNbO_3$

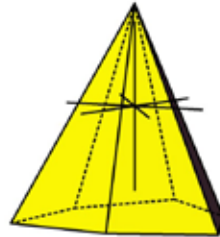
Trigonal pyramidal class

Hemimorphy of rhomboedric
hemiedry

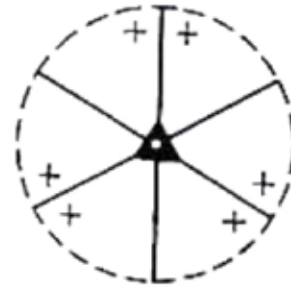
Symbol: $3m$ or C_{3v}

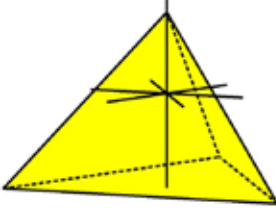
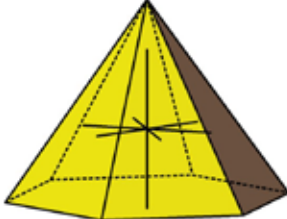
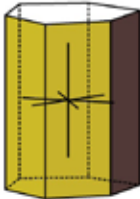
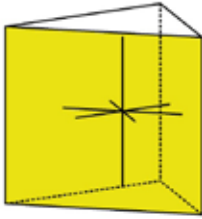
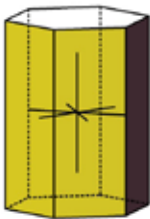
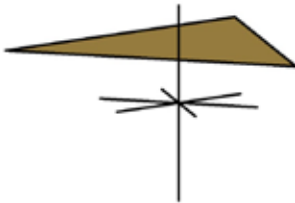
Ditrigonal-pyramidal class

General form:
 $\{hkl\} / \{hkil\}$



Ditrig. Pyramid I.pos.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hh1\} / \{hh2\bar{h}l\}$	$\{hk0\} / \{hki0\}$
		
<u>Trig. Pyramid I.pos.</u>	<u>Hex. Pyramid II.pos.</u>	<u>Ditrig. Prism II.pos.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Trig. Prism I.pos.</u>	<u>Hex. Prisma II.pos.</u>	<u>Basispedia</u>

mineral examples:

Tourmaline $(X)(Y_3)(Z_6)[Si_6O_{18}/(BO_3)_3/(V_3)(W)]$

Proustite Ag_3AsS_3

Lithium niobate $LiNbO_3$

Trigonales Kristallsystem

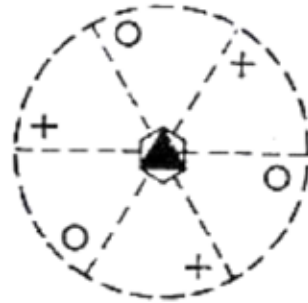
Hexagonal-rhomboedrische
Tetartoedrie
Rhomboedrische Klasse




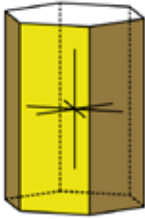
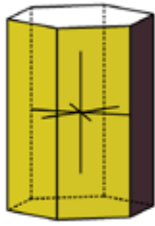
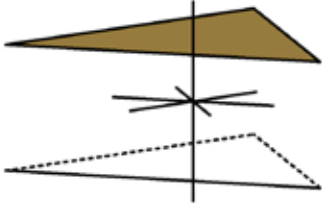
Symbol: $\bar{3}$ oder C_{3i}

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Rhomboeder III.St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
Rhomboeder I.St.	Rhomboeder II.St.	Hex. Prismen II.St.
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
Hex. Prisma I.St.	Hex. Prisma II.St.	Basispinakoid

Mineral-Bsp.:

Dolomit $\text{CaMg}[(\text{CO}_3)_2]$

Diopas $\text{Cu}[\text{Si}_6\text{O}_{18}]$

Phenakit Be_2SiO_4

Trigonal crystal system

Hexagonal-rhomboedral
tetartoedry
Rhombohedral class




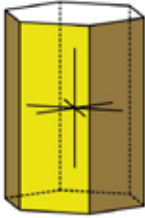
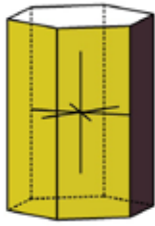
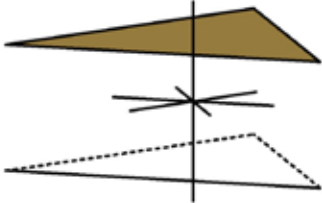
Symbol: $\bar{3}$ or C_{3i}

General form:
 $\{hkl\} / \{hkil\}$



Rhombohedron III.pos.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
Rhombohedron I.pos.	Rhombohedron II.pos.	Hex. Prism II.pos.
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
Hex. Prism I.pos.	Hex. Prism II.pos.	Basispinacoid

mineral examples:

Dolomite $CaMg[(CO_3)_2]$

Diopside $Cu[Si_6O_{18}]$

Phenakite Be_2SiO_4

Trigonales Kristallsystem

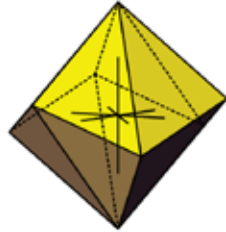
Hexagonal-rhomboedrische

Hemiedrie

Ditrigonal-skalenoedrische Klasse



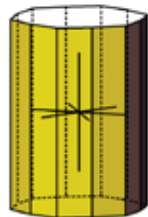
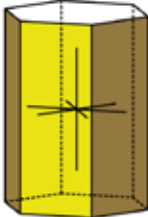
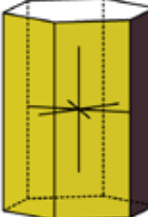
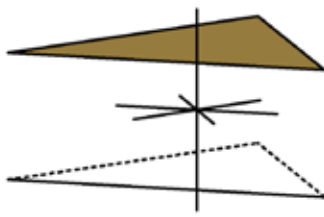
Symbol: $\bar{3} 2/m$ oder D_{3d}
($\bar{3}m$)

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Ditrig. Skalenoeder



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Rhomboeder I.St.</u>	<u>Hex. Dipyramiden II.St.</u>	<u>Dihex. Prismen</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex. Prisma I.St.</u>	<u>Hex. Prisma II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Calcit CaCO_3

Korund Al_2O_3

Hämatit Fe_2O_3

Siderit FeCO_3

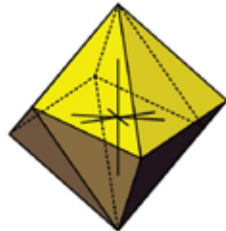
Rhodochrosit MnCO_3

Trigonal crystal system

Hexagonal-rhombohedral
hemiedry
Ditrigonal-scalenohedral class

Symbol: $\bar{3} 2/m$ or D_{3d}
($\bar{3}m$)

General form:
 $\{hkl\} / \{hkil\}$



Ditrig. Scalenohedron



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Rhombohedron I.pos.</u>	<u>Hex. Dipyramid II.pos.</u>	<u>Dihex. Prism</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Hex. Prism I.pos.</u>	<u>Hex. Prisma II.pos.</u>	<u>Basispinacoid</u>

mineral examples:

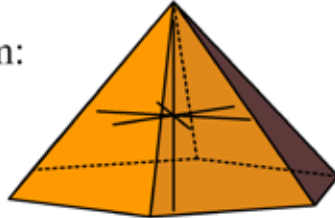
Calcite CaCO_3 Siderite FeCO_3
 Corundum Al_2O_3 Rhodochrosite MnCO_3
 Hematite Fe_2O_3

Hexagonales Kristallsystem

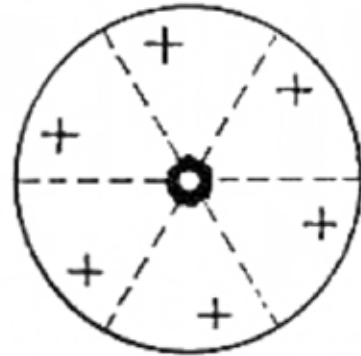
Hexagonale Tetartoedrie
Hexagonal-pyramidale Klasse

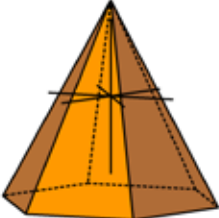

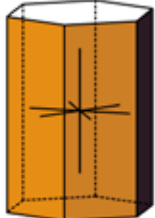
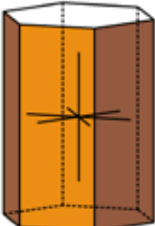
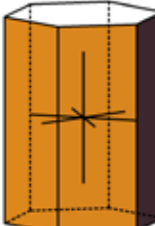
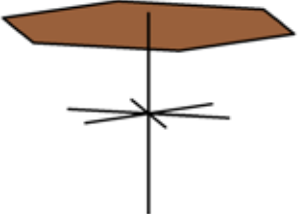
Symbol: 6 oder C_6

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Hexagonale Pyramiden III. St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Hex.Pyramiden I.St.</u>	<u>Hex. Pyramiden II.St.</u>	<u>Hex.Prismen III.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex.Prisma I.St.</u>	<u>Hex.Prisma II.St.</u>	<u>Basispedien</u>

Mineral-Bsp.:

Lithiumiodat α - LiIO_3

Nephelin $(\text{Na},\text{K})[\text{AlSiO}_4]$

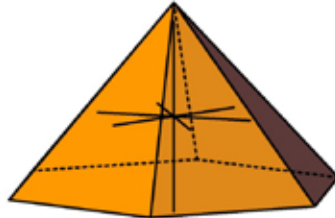
Thaumasit $\text{Ca}_3[\text{SO}_4/\text{CO}_3/\text{Si}(\text{OH})_6] \cdot 12\text{H}_2\text{O}$

Hexagonal crystal system

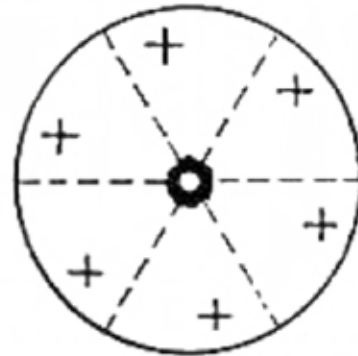
Hexagonal tetartoeidry
Hexagonal-pyramidal class

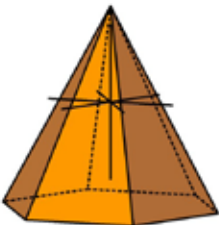
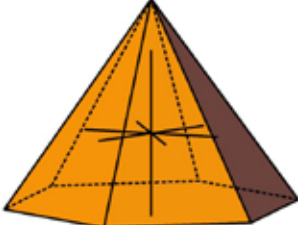

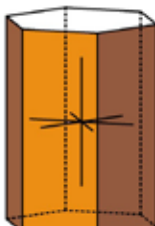

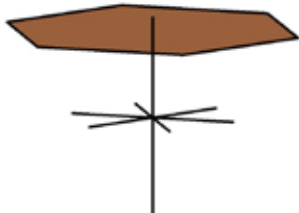
Symbol: 6 or C_6

General form:
 $\{hkl\} / \{hkil\}$



Hexagonale Pyramiden III. St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Hex.Pyramid I.pos.</u>	<u>Hex. Pyramid II.pos.</u>	<u>Hex.Prism III.pos.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex.Prism I.pos.</u>	<u>Hex.Prism II.pos.</u>	<u>Basispedia</u>

mineral examples:

Lithium iodate α - LiIO_3

Nepheline $(\text{Na},\text{K})[\text{AlSiO}_4]$

Thaumasite $\text{Ca}_3[\text{SO}_4/\text{CO}_3/\text{Si}(\text{OH})_6] \cdot 12\text{H}_2\text{O}$

Hexagonales Kristallsystem

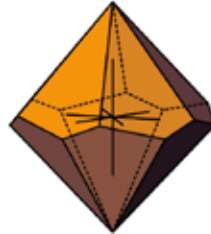
Hexagonal-trapezoedrische

Hemiedrie

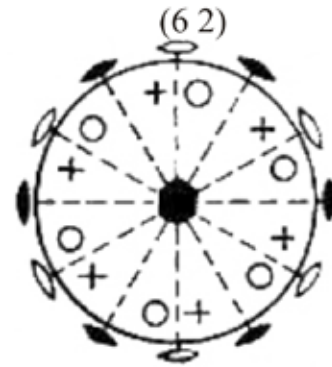
Hexagonal-trapezoedrische Klasse

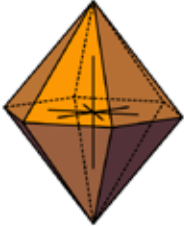
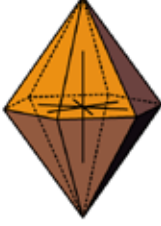
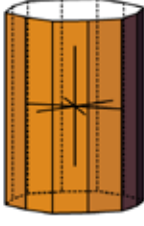
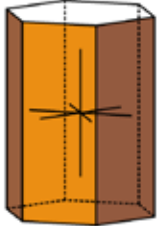
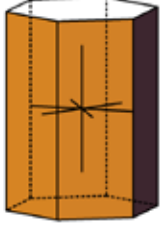
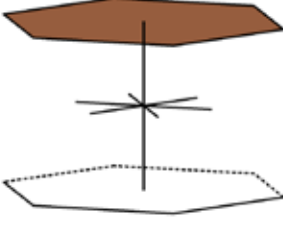
Symbol: 622 oder D_6

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Hexagonaler Trapezoeder



$\{h0l\} / \{h0\bar{l}\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Hex.Dipyramiden I.St.</u>	<u>Hex. Dipyramiden II.St.</u>	<u>Dihex. Prismen</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex.Prisma I.St.</u>	<u>Hex. Prisma II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Hochquarz SiO_2

Virgilite $\text{LiAlSi}_2\text{O}_6$

Rhabdophan $(\text{Ce},\text{La})\text{PO}_4 \cdot \text{H}_2\text{O}$

Hexagonal crystal system

Hexagonal- trapezohedral
hemiedry

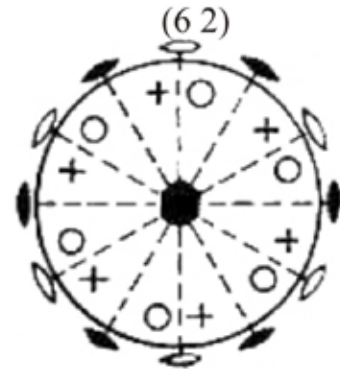
Symbol: 622 or D_6

Hexagonal-trapezohedral Class

General form:
 $\{hkl\} / \{hkil\}$



Hexagonal Trapezohedron



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Hex.Dipyramid I.pos.</u>	<u>Hex. Dipyramid II.pos.</u>	<u>Dihex. Prism</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Hex.Prism I.St.</u>	<u>Hex. Prism II.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Quartz-beta SiO_2

Virgilite $\text{LiAlSi}_2\text{O}_6$

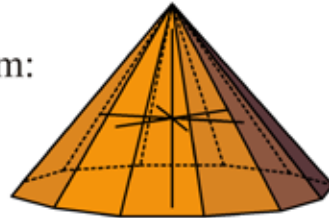
Rhabdophane $(\text{Ce},\text{La})\text{PO}_4 \cdot \text{H}_2\text{O}$

Hexagonales Kristallsystem

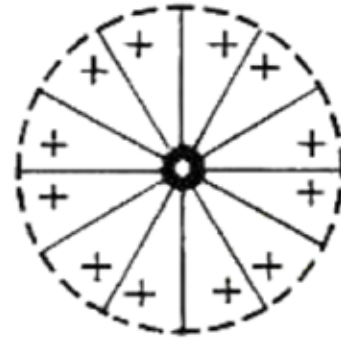
Hexagonale Hemimorphie
Dihexagonal-pyramidale Klasse

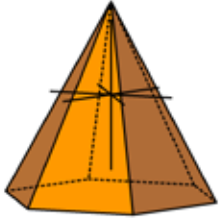
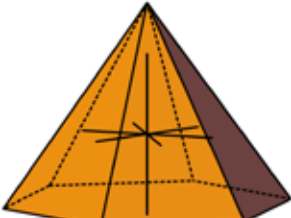
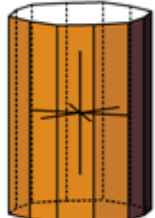
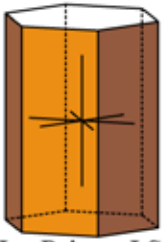
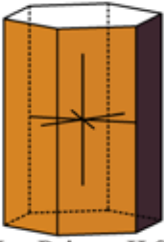
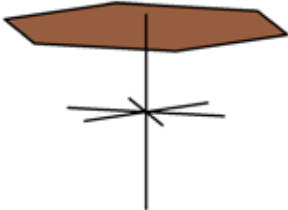
Symbol: $6mm$ oder C_{6v}

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Dihexagonale Pyramide



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Hex.Pyramiden I.St.</u>	<u>Hex. Pyramiden II.St.</u>	<u>Dihex.Prismen</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex.Prisma I.St.</u>	<u>Hex.Prisma II.St.</u>	<u>Basispedien</u>

Mineral-Bsp.:

Wurzit ZnS

Zinkit ZnO

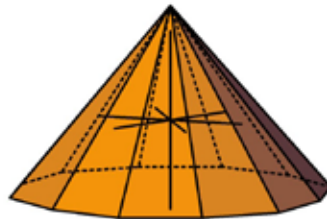
Greenockit CdS

Hexagonal crystal system

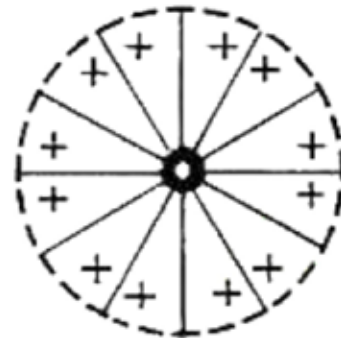
Hexagonal Hemimorphy
Dihexagonal-pyramidal class

Symbol: $6mm$ or C_{6v}

General form:
 $\{hkl\} / \{hkil\}$



Dihexagonal Pyramid



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
Hex.Pyramid I.pos.	Hex. Pyramid II.pos.	Dihex.Prism
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
Hex.Prism I.pos.	Hex.Prism II.pos.	Basispedia

mineral examples:

Wurzite ZnS

Zincite ZnO

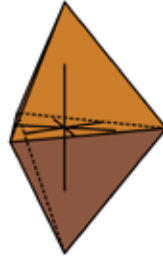
Greenockite CdS

Hexagonales Kristallsystem

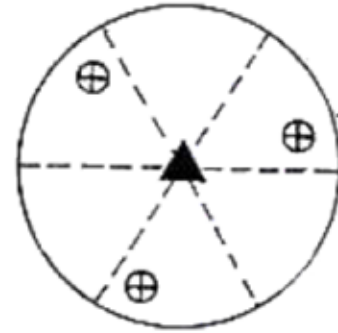
Trigonale Tetartoedrie
Trigonal-dipyramidale Klasse

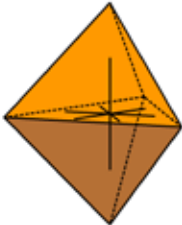
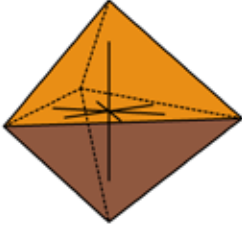
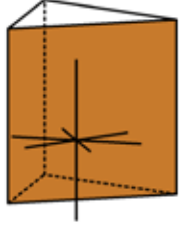
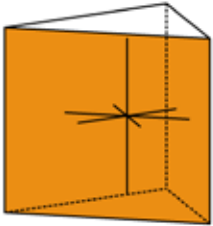
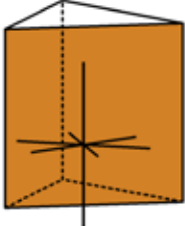
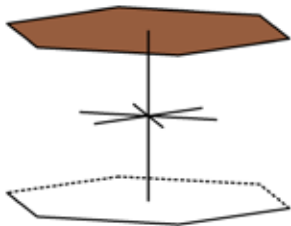
Symbol: $\bar{6}$ oder C_{3h}

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Trig. Dipyramiden III.St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Trig. Dipyramiden I.St.</u>	<u>Trig. Dipyramiden II.St.</u>	<u>Trig. Prismen III.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Trig. Prismen I.St.</u>	<u>Trig. Prismen II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Bleigermanat (Hochtemp. Modifikation) $Pb_5[Ge_3O_{11}]$

Penfieldit $Pb_2[Cl_3/OH]$

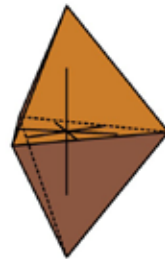
Ganomalit $Pb_9Ca_5Mn[(Si_2O_7)_3/(SiO_4)_3]$

Hexagonal crystal system

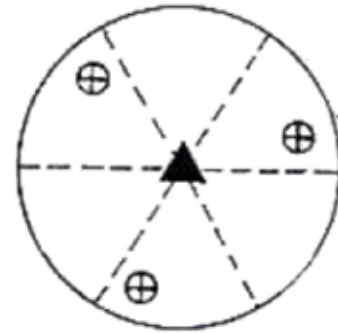
Trigonal tetartoedry
Trigonal-dipyramidal classe

Symbol: $\bar{6}$ or C_{3h}

General form:
 $\{hkl\} / \{hkil\}$



Trig. Dipyramid III.pos.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Trig. Dipyramid I.pos.</u>	<u>Trig. Dipyramid II.pos.</u>	<u>Trig.Prism III.pos.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Trig.Prism I.pos.</u>	<u>Trig.Prism II.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Lead germanate (Hightemp. Modification) $Pb_5[Ge_3O_{11}]$

Penfieldite $Pb_2[Cl_3/OH]$

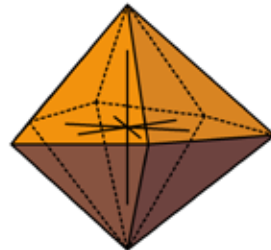
Ganomalite $Pb_9Ca_5Mn[(Si_2O_7)_3/(SiO_4)_3]$

Hexagonales Kristallsystem

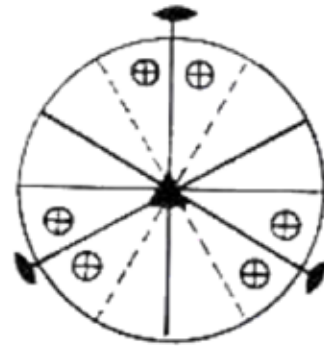
Trigonale Hemiedrie
Ditrigonal-dipyramidale Klasse

Symbol: $\bar{6}2m$ oder D_{3h}

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Ditrig. Dipyramiden



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Trig. Dipyramiden I.St.</u>	<u>Hex. Dipyramiden II.St.</u>	<u>Ditrig.Prismen III.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Trig.Prismen I.St.</u>	<u>Hex.Prismen II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Benitoit $BaTi[Si_3O_9]$

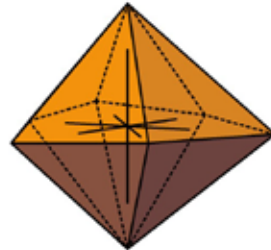
Belkovit $Ba_3(Nb,Ti)_6(Si_2O_7)_2O_{12}$

Hexagonal crystal system

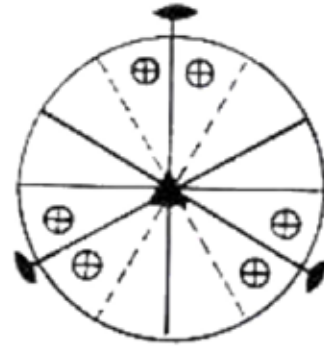
Trigonal Hemiedry
Ditrigonal-dipyramidal class

Symbol: $\bar{6}2m$ or D_{3h}

General form:
 $\{hkl\} / \{hkil\}$



Ditrig. Dipyramid



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
Trig. Dipyramid I.pos.	Hex. Dipyramid II.pos.	Ditrig.Prism III.pos.
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
Trig.Prism I.pos.	Hex.Prism II.pos.	Basispinacoid

mineral examples:

Benitoite $BaTi[Si_3O_9]$

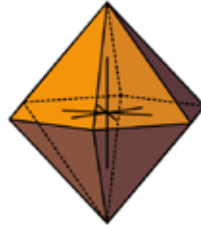
Belkovite $Ba_3(Nb,Ti)_6(Si_2O_7)_2O_{12}$

Hexagonales Kristallsystem

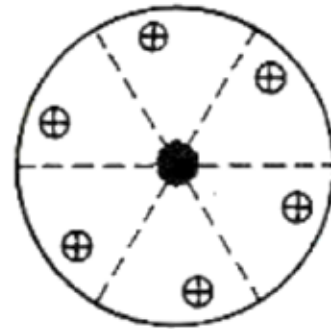
Hexagonal-pyramidale
Hemiedrie
Hexagonal-dipyramidale Klasse

Symbol: $6/m$ oder C_{6v}

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Hexagonale Dipyramide III.St.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Hex.Dipyramiden I.St.</u>	<u>Hex. Dipyramiden II.St.</u>	<u>Hex.Prismen III.St.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Hex.Prisma I.St.</u>	<u>Hex.Prisma II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Apatit $(Ca, Ba, Pb, Sr, etc.)_5[(PO_4, CO_3)_3/(F, Cl, OH)]$

Pyromorphit $Pb_5[(PO_4)_3/Cl]$

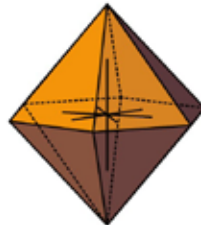
Vanadinit $Pb_5[(VO_4)_3/Cl]$

Hexagonal crystal system

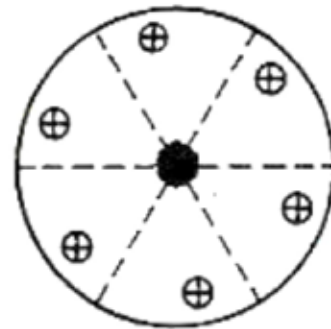
Hexagonal-pyramidal Hemiedry
Hexagonal-dipyramidal class

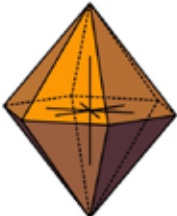


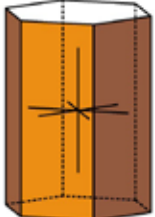
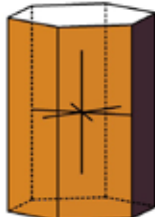
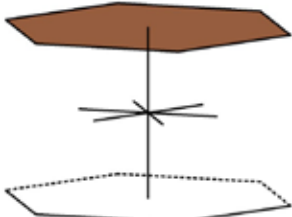
Symbol: $6/m$ or C_{6v}

General form:
 $\{hkl\} / \{hkil\}$



Hexagonal Dipyramid III.pos.



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Hex.Dipyramid I.pos.</u>	<u>Hex. Dipyramid II.pos.</u>	<u>Hex.Prism III.pos.</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex.Prism I.pos.</u>	<u>Hex.Prism II.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Apatite $(Ca, Ba, Pb, Sr, etc.)_5[(PO_4, CO_3)_3/(F, Cl, OH)]$

Pyromorphite $Pb_5[(PO_4)_3/Cl]$

Vanadinite $Pb_5[(VO_4)_3/Cl]$

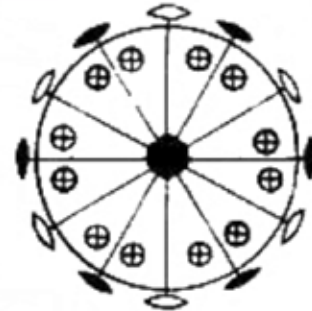
Hexagonales Kristallsystem

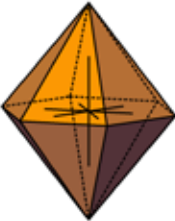

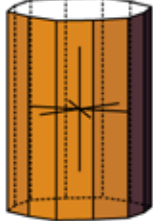
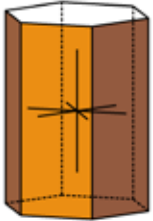
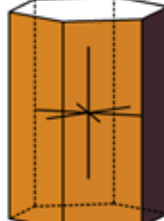
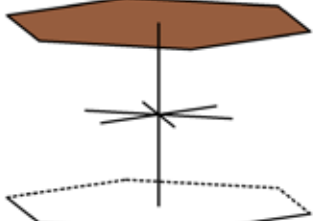
Hexagonale Holoedrie Symbol: $6/mmm$ oder D_{6h}
 Dihexagonal-dipyramidale Klasse ($6/m\ 2/m\ 2/m$)

Allgemeine Form:
 $\{hkl\} / \{hkil\}$



Dihexagonale Dipyramiden



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
		
<u>Hex.Dipyramiden I.St.</u>	<u>Hex.Dipyramiden II.St.</u>	<u>Dihex.Prismen</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
		
<u>Hex.Prisma I.St.</u>	<u>Hex.Prisma II.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Beryll $Be_3Al_2[Si_6O_{18}]$ Covellin CuS

Graphit C Eis H_2O

Hochtridymit SiO_2

Hexagonal crystal system

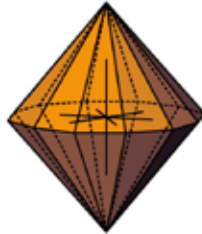
Hexagonal holoedry

Symbol: $6/mmm$ or D_{6h}

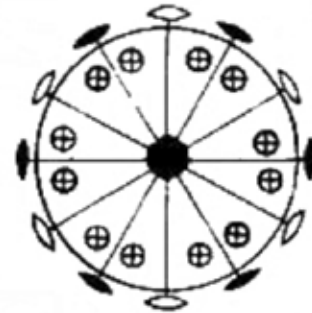
Dihexagonal-dipyramidal class

($6/m\ 2/m\ 2/m$)

General form:
 $\{hkl\} / \{hkil\}$



Dihexagonal Dipyramid



$\{h0l\} / \{h0\bar{h}l\}$	$\{hhl\} / \{hh\bar{2}hl\}$	$\{hk0\} / \{hki0\}$
<u>Hex.Dipyramid I.pos.</u>	<u>Hex.Dipyramid II.pos.</u>	<u>Dihex.Prism</u>
$\{100\} / \{10\bar{1}0\}$	$\{110\} / \{11\bar{2}0\}$	$\{001\} / \{0001\}$
<u>Hex.Prism I.pos.</u>	<u>Hex.Prism II.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Beryl $Be_3Al_2[Si_6O_{18}]$

Covellite CuS

Graphite C

Ice H_2O

High tridymite SiO_2

Tetragonales Kristallsystem

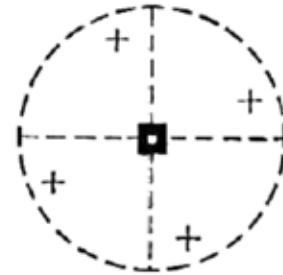
Hemimorphie d. tetragonal-
pyramidalen Hemiedrie
Tetragonal-pyramidale Klasse



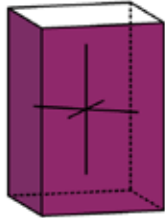
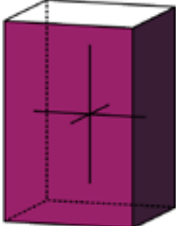
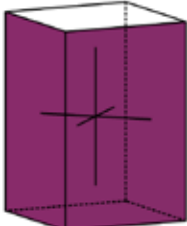
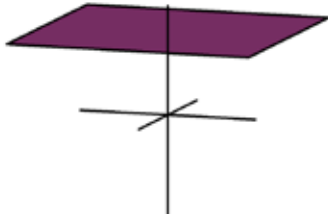
Symbol: 4 oder C_4

Allgemeine Form:
 $\{hkl\}$



Tetrag. Pyramiden III.St.



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag.Pyramiden II.St.</u>	<u>Tetrag.Pyramiden I.St.</u>	<u>Tetrag. Prismen III.St.</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag.Prisma II.St.</u>	<u>Tetrag.Prisma I.St.</u>	<u>Basispedien</u>

Mineral-Bsp.:

Piypite $K_2Cu_2[O/(SO_4)_2]$

Pinnoite $Mg[B_2O_4]*3H_2O$

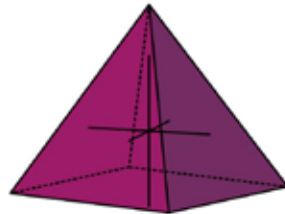
Percleveite-(Ce) $(Ce,La,Nd)_2[Si_2O_7]$

Tetragonal crystal system

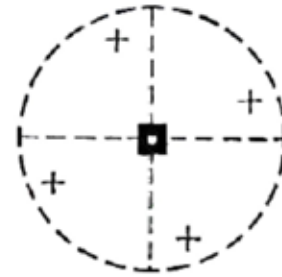
Hemimorphy of tetragonal -
pyramidal hemiedry
Tetragonal-pyramidal class



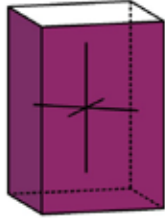
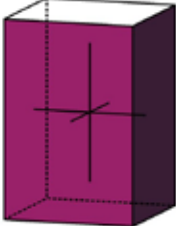
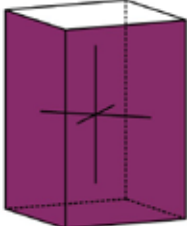
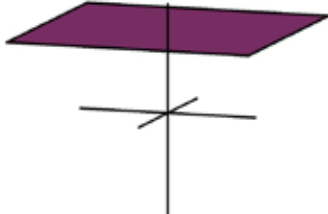
Symbol: 4 or C_4

General form:
 $\{hkl\}$



Tetrag. Pyramid III.pos.



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag.Pyramid II.pos.</u>	<u>Tetrag.Pyramid I.pos.</u>	<u>Tetrag. Prism III.pos.</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag.Prism II.pos.</u>	<u>Tetrag.Prism I.pos.</u>	<u>Basispedia</u>

mineral examples:

Piypite $K_2Cu_2[O/(SO_4)_2]$

Pinnoite $Mg[B_2O_4]*3H_2O$

Percleveite-(Ce) $(Ce,La,Nd)_2[Si_2O_7]$

Tetragonales Kristallsystem

Tetragonal-trapezoedrische
Hemiedrie

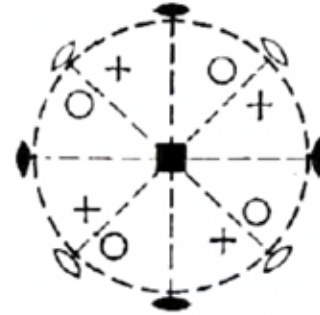
Symbol: 422 oder D_4

Tetragonal-trapezoedrische Klasse

Allgemeine Form:
 $\{hkl\}$



Tetrag. Trapezoeder



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
<u>Tetrag. Dipyramiden II. St.</u>	<u>Tetrag. Dipyramiden I. St.</u>	<u>Ditetrag. Prismen</u>
$\{100\}$	$\{110\}$	$\{001\}$
<u>Tetrag. Prisma II. St.</u>	<u>Tetrag. Prisma I. St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Retgersit $Ni[SO_4] \cdot 6H_2O$

Maucherit $Ni_{11}As_8$

Cristobalit SiO_2

Tetragonal crystal system

Tetragonal-trapezohedral hemiedry

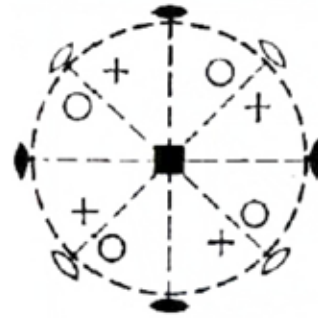
Tetragonal-trapezohedral class




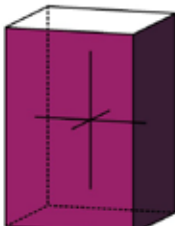
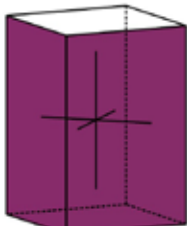
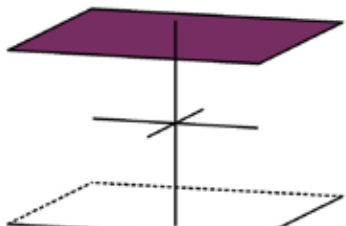
Symbol: 422 or D_4

General form:
 $\{hkl\}$



Tetrag. Trapezohedron



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag. Dipyramid II.pos.</u>	<u>Tetrag. Dipyramid I.pos.</u>	<u>Ditetrag. Prism</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag. Prism II.pos.</u>	<u>Tetrag. Prism I.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Retgersite $\text{Ni}[\text{SO}_4] \cdot 6\text{H}_2\text{O}$

Maucherite $\text{Ni}_{11}\text{As}_8$

Cristobalite SiO_2

Tetragonales Kristallsystem

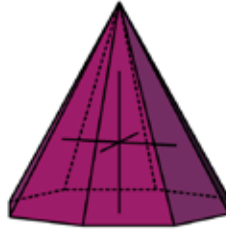
Hemimorphie d. tetragonalen

Holoedrie

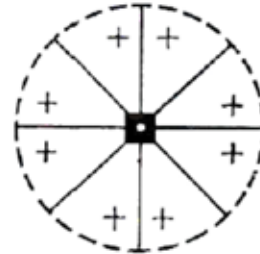
Symbol: $4mm$ oder C_{4v}



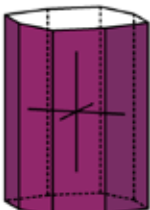

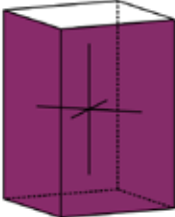
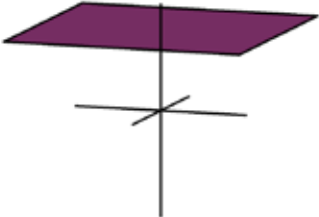
Ditetragonal-pyramidale Klasse

Allgemeine Form:
 $\{hkl\}$



Ditetrag. Pyramiden



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag. Pyramiden II. St.</u>	<u>Tetrag. Pyramiden I. St.</u>	<u>Ditetrag. Prismen</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag. Prisma II. St.</u>	<u>Tetrag. Prisma I. St.</u>	<u>Basispedien</u>

Mineral-Bsp.:

Diaboleit $Pb_2Cu[Cl_2/(OH)_4]$

Macedonite $PbTiO_3$

Nielsenite $PdCu_3$

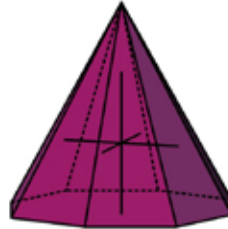
Tetragonal crystal system

Hemimorphy of tetragonal
holoedry

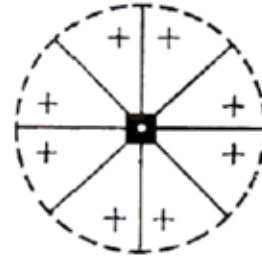
Symbol: $4mm$ or C_{4v}


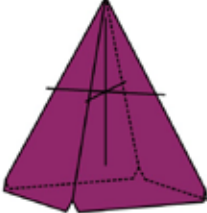
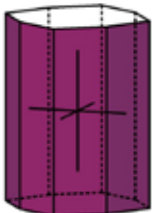

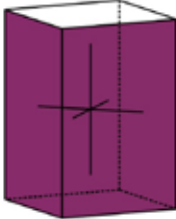
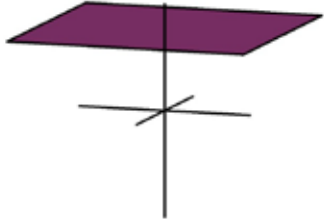
Ditetragonal-pyramidal class

General form:
 $\{hkl\}$



Ditetrag. Pyramid



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag.Pyramid II.pos.</u>	<u>Tetrag.Pyramid I.pos.</u>	<u>Ditetrag. Prism</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag.Prism II.pos.</u>	<u>Tetrag.Prism I.pos.</u>	<u>Basispedia</u>

mineral examples:

Diaboleite $Pb_2Cu[Cl_2/(OH)_4]$

Macedonite $PbTiO_3$

Nielsenite $PdCu_3$

Tetragonales Kristallsystem

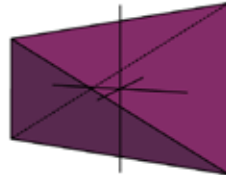
Tetragonal-sphenoidische

Tetartoedrie

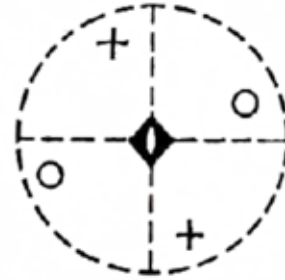
Tetragonal-disphenoidische Klasse

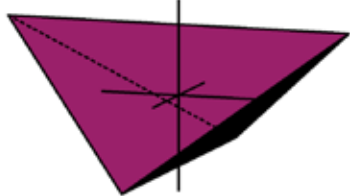
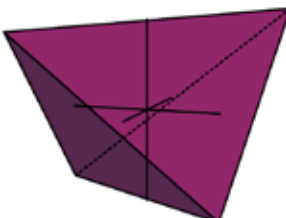
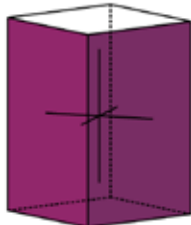
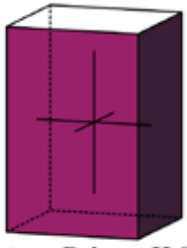
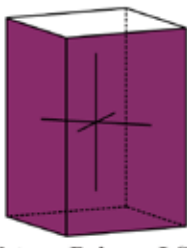
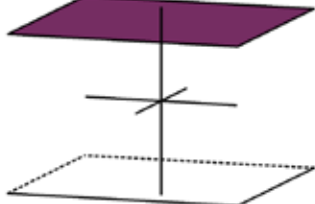
Symbol: $\bar{4}$ oder S_4

Allgemeine Form:
 $\{hkl\}$



Tetrag.Disphenoide III.St.



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag.Disphenoide II.St.</u>	<u>Tetrag.Disphenoide I.St.</u>	<u>Tetrag.Prismen III.St.</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag.Prisma II.St.</u>	<u>Tetrag.Prisma I.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Cahnit $\text{Ca}_2[\text{AsO}_4/\text{B}(\text{OH})_4]$

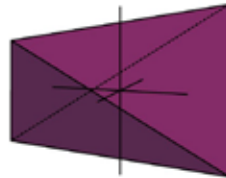
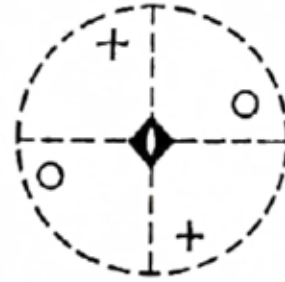
Tugtupit $\text{Na}_8[(\text{AlBeSi}_4\text{O}_{12})_2/\text{Cl}_2]$

Schreibersit $(\text{Fe},\text{Ni})_3\text{P}$

Tetragonal crystal system

Tetragonal-sphenoidal tetartoeidry

Tetragonal-disphenoidal class

Symbol: $\bar{4}$ or S_4 General form:
 $\{hkl\}$ Tetrag.Disphenoid III.pos.

$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
<u>Tetrag.Disphenoid II.pos.</u>	<u>Tetrag.Disphenoid I.pos.</u>	<u>Tetrag.Prism III.pos.</u>
$\{100\}$	$\{110\}$	$\{001\}$
<u>Tetrag.Prism II.pos.</u>	<u>Tetrag.Prism I.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Cahnite $\text{Ca}_2[\text{AsO}_4/\text{B}(\text{OH})_4]$ Tugtupite $\text{Na}_8[(\text{AlBeSi}_4\text{O}_{12})_2/\text{Cl}_2]$ Schreibersite $(\text{Fe},\text{Ni})_3\text{P}$

Tetragonales Kristallsystem

Tetragonal-sphenoidische
Hemiedrie

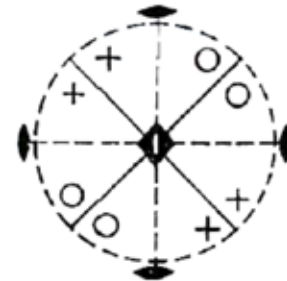
Symbol: $\bar{4}2m$ oder D_{2d}

Tetragonal-skalenoedrische Klasse

Allgemeine Form:
{hkl}



Tetrag. Skalenoeder



{h0l}	{hhl}	{hk0}
<u>Tetrag. Dipyramiden II. St.</u>	<u>Tetrag. Disphenoide I. St.</u>	<u>Ditetrag. Prismen</u>
{100}	{110}	{001}
<u>Tetrag. Prisma II. St.</u>	<u>Tetrag. Prisma I. St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Chalkopyrit CuFeS_2

Mooihoekite $\text{Cu}_9\text{Fe}_9\text{S}_{16}$

Stannit $\text{Cu}_2\text{FeSnS}_4$

Tetragonal crystal system

Tetragonal-sphenoidal hemiedry

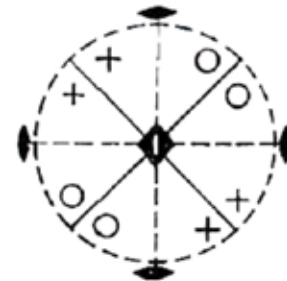
Tetragonal-scalenohedral class

Symbol: $\bar{4}2m$ or D_{2d}

General form:
 $\{hkl\}$



Tetrag. Scalenohedron



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
<u>Tetrag. Dipyramid II. pos.</u>	<u>Tetrag. Disphenoid I. pos.</u>	<u>Ditetrag. Prism</u>
$\{100\}$	$\{110\}$	$\{001\}$
<u>Tetrag. Prism II. pos.</u>	<u>Tetrag. Prism I. pos.</u>	<u>Basispinacoid</u>

mineral examples:

Chalkopyrite CuFeS_2

Mooihoekite $\text{Cu}_9\text{Fe}_9\text{S}_{16}$

Stannite $\text{Cu}_2\text{FeSnS}_4$

Tetragonales Kristallsystem

Tetragonal-dipyramidale

Hemiedrie

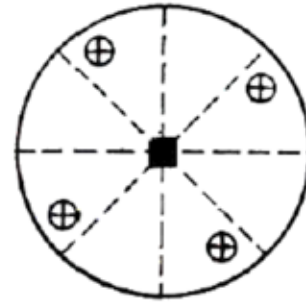
Symbol: $4/m$ oder C_{4h}



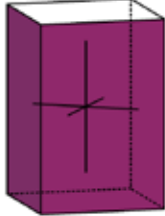
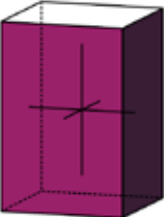
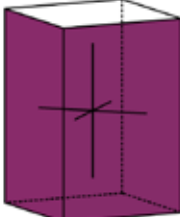
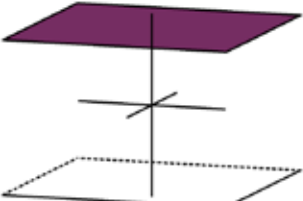
Tetragonal-dipyramidale Klasse

Allgemeine Form:
 $\{hkl\}$



Tetrag.Dipyramiden III.St.



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Tetrag.Dipyramiden II.St.</u>	<u>Tetrag.Dipyramiden I.St.</u>	<u>Tetrag.Prismen III.St.</u>
$\{100\}$	$\{110\}$	$\{001\}$
		
<u>Tetrag.Prisma II.St.</u>	<u>Tetrag.Prisma I.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Scheelit $Ca[WO_4]$

Vesuvian $Ca_{19}(Mg,Fe,Ti)_4Al_9[(OH,F)_{10}/(SiO_4)_{10}/(Si_2O_7)_4]$

Fergusonit $YNbO_4$

Tetragonal crystal system

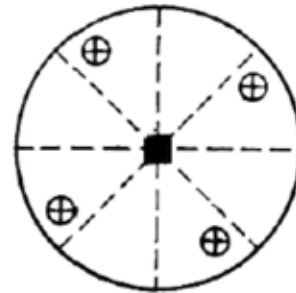
Tetragonal-dipyramidal hemiedry
 Tetragonal-dipyramidal class

Symbol: $4/m$ or C_{4h}

General form:
 $\{hkl\}$



Tetrag.Dipyramid III.pos.



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
<u>Tetrag.Dipyramid II.pos.</u>	<u>Tetrag.Dipyramid I.pos.</u>	<u>Tetrag.Prism III.pos.</u>

$\{100\}$	$\{110\}$	$\{001\}$
<u>Tetrag.Prism II.pos.</u>	<u>Tetrag.Prism I.pos.</u>	<u>Basispinacoid</u>

mineral examples:

Scheelite $Ca[WO_4]$

Vesuvianite $Ca_{19}(Mg,Fe,Ti)_4Al_9[(OH,F)_{10}/(SiO_4)_{10}/(Si_2O_7)_4]$

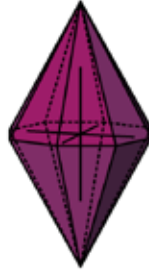
Fergusonite $YNbO_4$

Tetragonales Kristallsystem

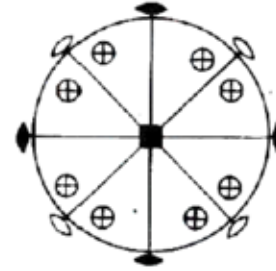
Tetragonale Holoedrie
Ditetragonal-dipyramidale
Klasse

Symbol: $4/m \ 2/m \ 2/m$
($4/m \ m \ m$) oder D_{4h}

Allgemeine Form:
 $\{hkl\}$



Ditetrag. Dipyramiden



$\{h0l\}$	$\{hhl\}$	$\{hk0\}$
<u>Tetrag.Dipyramiden II.St.</u>	<u>Tetrag.Dipyramiden I.St.</u>	<u>Ditetrag.Prismen</u>
$\{100\}$	$\{110\}$	$\{001\}$
<u>Tetrag.Prisma II.St.</u>	<u>Tetrag.Prisma I.St.</u>	<u>Basispinakoid</u>

Mineral-Bsp.:

Rutil TiO_2

Anatas TiO_2

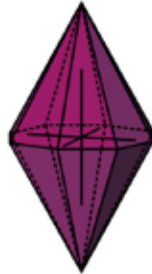
Zirkon $Zr[SiO_4]$

Tetragonal crystal system

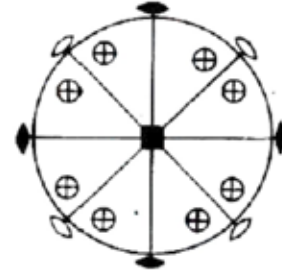
Tetragonal holoedry
Ditragonal-dipyramidal
class




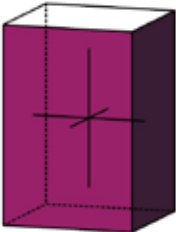
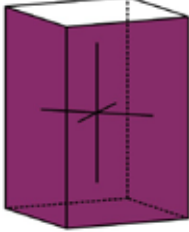
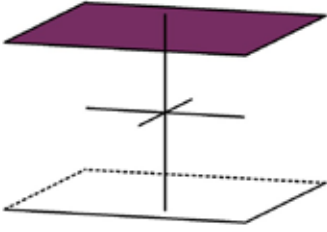
Symbol: $4/m \ 2/m \ 2/m$
($4/m \ m \ m$) or D_{4h}

General form:
{hkl}



Ditetrag. Dipyramid



{h0l}	{hhl}	{hk0}
 <p data-bbox="236 1144 528 1173"><u>Tetrag. Dipyramid II.pos.</u></p>	 <p data-bbox="660 1144 938 1173"><u>Tetrag. Dipyramid I.pos.</u></p>	 <p data-bbox="1129 1144 1299 1173"><u>Ditetrag. Prism</u></p>
{100}	{110}	{001}
 <p data-bbox="264 1559 497 1588"><u>Tetrag. Prism II.pos.</u></p>	 <p data-bbox="687 1559 908 1588"><u>Tetrag. Prism I.pos.</u></p>	 <p data-bbox="1134 1559 1294 1588"><u>Basispinacoid</u></p>

mineral examples:

Rutile TiO_2

Anatase TiO_2

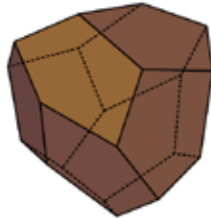
Zircon $Zr[SiO_4]$

Kubisches Kristallsystem

Kubische Tetartoedrie
Tetraedisch-pentagon-
doekaedrische Klasse

Symbol: 23 oder T

Allgemeine Form:
 $\{hkl\}$



Tetraedr.Pentagondodekaeder



$\{hll\}$	$\{hhl\}$	$\{hk0\}$
<u>Pyramidentetraeder</u>	<u>Deltoiddodekaeder</u>	<u>Pentagondodekaeder</u>
$\{111\}$	$\{110\}$	$\{100\}$
<u>Tetraeder</u>	<u>Rhombendodekaeder</u>	<u>Würfel/ Hexaeder</u>

Mineral-Bsp.:

Langbeinit $K_2Mg_2[(SO_4)_3]$

Ullmannit NiSbS

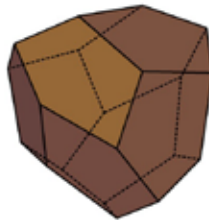
Gersdorffit NiAsS

Cubic crystal system

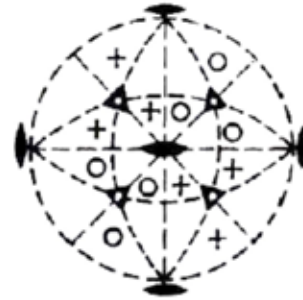
Cubic Tetartohedry
Tetrahedral-pentagon-
dodecaedric class

Symbol: 23 or T

General form:
 $\{hkl\}$



Tetrahedral Pentagondodecaeder



$\{hll\}$	$\{hhl\}$	$\{hk0\}$
<u>Pyramidal tetrahedron</u>	<u>Deltoidal dodecahedron</u>	<u>Pentagonal dodecaeder</u>
$\{111\}$	$\{110\}$	$\{100\}$
<u>Tetrahedron</u>	<u>Rhombic dodecahedron</u>	<u>Cube/Hexahedron</u>

mineral examples:

Langbeinite $K_2Mg_2[(SO_4)_3]$

Ullmannite NiSbS

Gersdorffite NiAsS

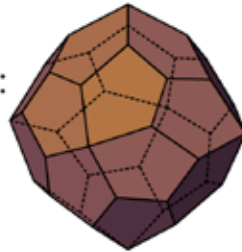
Kubisches Kristallsystem

Gyroedrische Hemiedrie

Symbol: 432 oder O

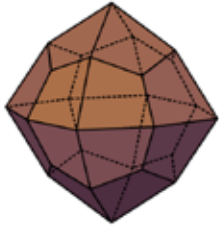
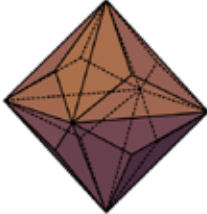

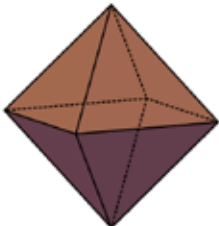
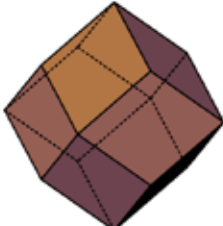
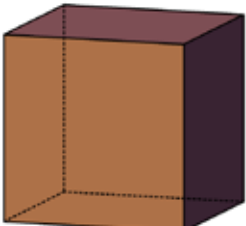
Pentagonikositetraedrische Klasse

Allgemeine Form:
 $\{hkl\}$



Pentagonikositetraeder



$\{hll\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Deltoidikositetraeder</u>	<u>Pyramidenoktaeder</u>	<u>Pyramidenwürfel</u>
$\{111\}$	$\{110\}$	$\{100\}$
		
<u>Oktaeder</u>	<u>Rhombendodekaeder</u>	<u>Würfel/ Hexaeder</u>

Mineral-Bsp.:

Maghemit $\text{Gamma-Fe}_2\text{O}_3$

Choloalit $\text{CuPb}[(\text{TeO}_3)_2]$

Petzit Ag_3AuTe_2

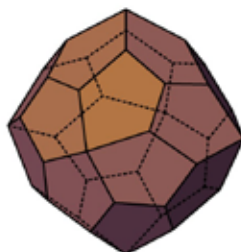
Cubic crystal system

Gyroedric hemiedry

Symbol: 432 or O

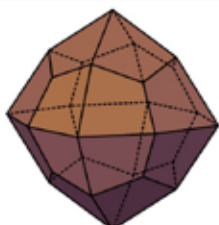
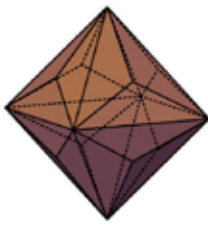

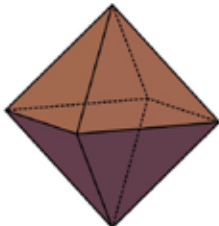
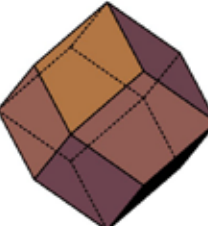
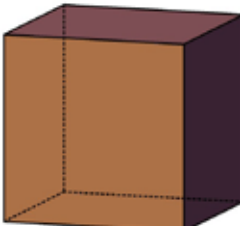
Pentagonal icositetrahedral class

General form:
 $\{hkl\}$



Pentagonal icositetrahedron



$\{hll\}$	$\{hhl\}$	$\{hk0\}$
		
<u>Deltoidal icositetrahedron</u>	<u>Pyramidal octahedron</u>	<u>Pyramidal cube</u>
$\{111\}$	$\{110\}$	$\{100\}$
		
<u>Octahedron</u>	<u>Rhombic dodecahedron</u>	<u>Cube/ Hexahedron</u>

mineral examples:

Maghemite $\text{Gamma-Fe}_2\text{O}_3$

Choloalite $\text{CuPb}[(\text{TeO}_3)_2]$

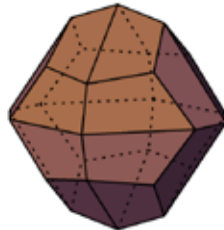
Petzite Ag_3AuTe_2

Kubisches Kristallsystem

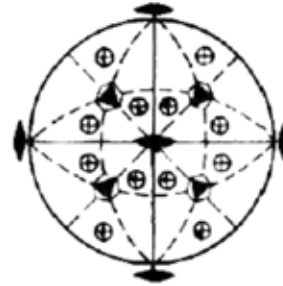
Parallelflächige Hemiedrie
Disdodekaedrische Klasse

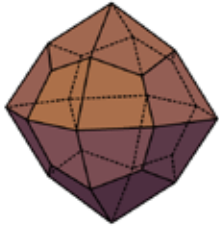
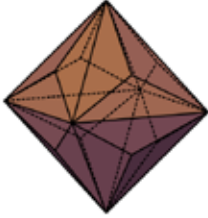
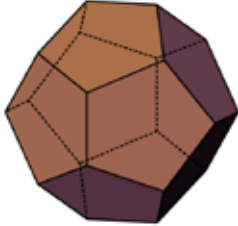
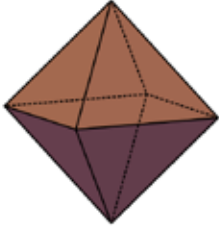
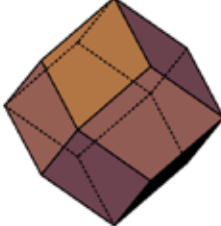
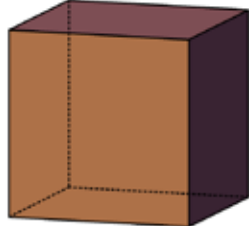
Symbol: $2/m\bar{3}$ oder T_h
($m\bar{3}$)

Allgemeine Form:
{hkl}



Disdodekaeder



{hll}	{hhl}	{hk0}
		
<u>Deltoidikositetraeder</u>	<u>Pyramidenoktaeder</u>	<u>Pentagondodekaeder</u>
{111}	{110}	{100}
		
<u>Oktaeder</u>	<u>Rhombendodekaeder</u>	<u>Würfel/Hexaeder</u>

Mineral-Bsp.:

Pyrit FeS_2

Cobaltin $CoAsS$

Alaune, z.B. $KAl[SO_4]_2 \cdot 12H_2O$

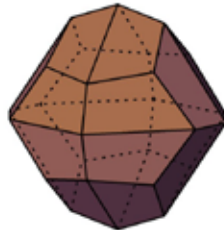
Bixbyit $(Mn,Fe)_2O_3$

Cubic crystal system

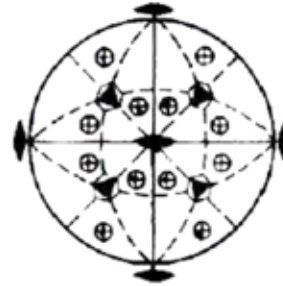
Parallel Hemiedry
Disdodecaedric class

Symbol: $2/m\bar{3}$ or T_h
($m\bar{3}$)

General form:
 $\{hkl\}$



Disdodecahedron



$\{hll\}$	$\{hhl\}$	$\{hk0\}$
<u>Deltoidal icositetrahedron</u>	<u>Pyramidal octahedron</u>	<u>Pentagonal dodecahedron</u>
$\{111\}$	$\{110\}$	$\{100\}$
<u>Octahedron</u>	<u>Rhombic dodecahedron</u>	<u>Cube/Hexahedron</u>

mineral examples:

Pyrite FeS_2

Cobaltite $CoAsS$

Alum, z.B. $KAl[SO_4]_2 \cdot 12H_2O$

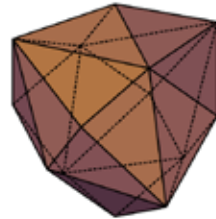
Bixbyite $(Mn,Fe)_2O_3$

Kubisches Kristallsystem

Geneigtflächige Hemiedrie
Hexakistetraedrische Klasse

Symbol: $\bar{4}3m$ oder T_d

Allgemeine Form:
{hkl}



Hexakistetraeder



{hll}	{hhl}	{hk0}
<u>Pyramidentetraeder</u>	<u>Deltoiddodekaeder</u>	<u>Pyramidenwürfel</u>
{111}	{110}	{100}
<u>Tetraeder</u>	<u>Rhombendodekaeder</u>	<u>Würfel/ Hexaeder</u>

Mineral-Bsp.:

Sphalerit/Zinkblende ZnS

Tetraedrit $(\text{Cu,Fe})_{12}[\text{Sb}_4\text{S}_{13}]$

Sodalith $\text{Na}_8[(\text{Cl})_2(\text{AlSiO}_4)_6]$

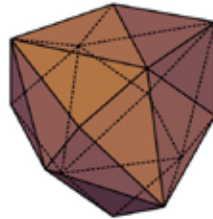
Mayenit $\text{Ca}_{12}\text{Al}_{14}\text{O}_{33}$

Cubic crystal system

Inclined Hemiedry
Hexacistetrahedral class

Symbol: $\bar{4}3m$ or T_d

General form:
{hkl}



Hexacistetrahedron



{hll}	{hhl}	{hk0}
<u>Pyramidal tetrahedron</u>	<u>Deltoidal dodecahedron</u>	<u>Pyramidal cube</u>
{111}	{110}	{100}
<u>Tetrahedron</u>	<u>Rhombic dodecahedron</u>	<u>Cube/Hexahedron</u>

mineral examples:

Sphalerite/Zinc blende ZnS

Tetrahedrite $(Cu,Fe)_{12}[Sb_4S_{13}]$

Sodalite $Na_8[(Cl)_2(AlSiO_4)_6]$

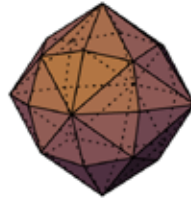
Mayenite $Ca_{12}Al_{14}O_{33}$

Kubisches Kristallsystem

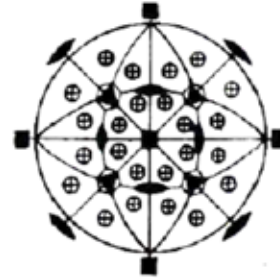
Kubische Holoedrie
Hexakisoktaedrische Klasse

Symbol: $4/m \bar{3} 2/m$ oder O_h
($m\bar{3}m$)

Allgemeine Form:
{hkl}



Hexakisoktaeder



{hll}	{hhl}	{hk0}
<u>Deltoidikositetraeder</u>	<u>Pyramidenoktaeder</u>	<u>Pyramidenwürfel</u>
{111}	{110}	{100}
<u>Oktaeder</u>	<u>Rhombendodekaeder</u>	<u>Würfel/Hexaeder</u>

Mineral-Bsp.:

Elemente: Au, Ag, Cu, Pt,
Pb, Fe, W, Si, C(Diamant)
Halt/Steinsalz NaCl

Galenit/Bleiglanz PbS
Fluorit/Flußspat CaF_2
Spinelle, z.B. $MgAl_2O_4$
Granate $Me^{II}_3Me^{III}_2[SiO_4]_3$

Cubic crystal system

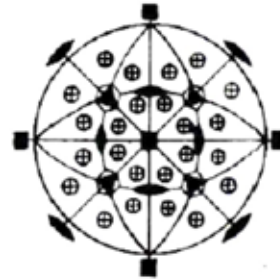
Cubic holoedry
Disdyakis dodecahedral class

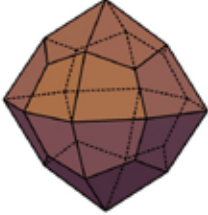
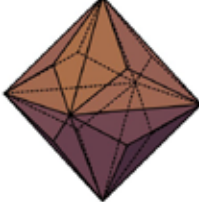
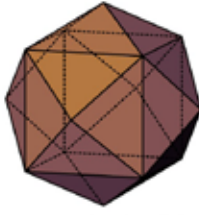
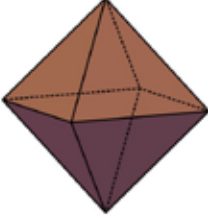
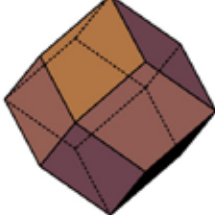
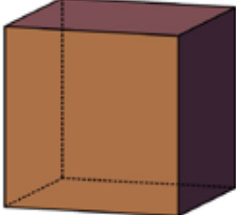
Symbol: $4/m \bar{3} 2/m$ or O_h
($m\bar{3}m$)

General form:
{hkl}



Disdyakis dodecahedron



{hll}	{hhl}	{hk0}
 <p><u>Deltoidal icositetrahedron</u></p>	 <p><u>Pyramidal octahedron</u></p>	 <p><u>Pyramidal cubic</u></p>
{111}	{110}	{100}
 <p><u>Octahedron</u></p>	 <p><u>Rhombic dodecahedron</u></p>	 <p><u>Cube/Hexahedron</u></p>

mineral examples:

Elements: Au, Ag, Cu, Pt,
Pb, Fe, W, Si, C(Diamond)
Halite/ Rock salt NaCl

Galena /Lead glance PbS
Fluorite/Flusse CaF_2
Spinel, z.B. $MgAl_2O_4$
Garnet $Me^{II}_3Me^{III}_2[SiO_4]_3$

Beispiele für korrele Kristallformen

Formenausbildung

Positiv - Negativ

Rechts - Links

Oben - Unten

Vorn - Hinten

Examples for correlate crystal forms

Crystal forms

Positive - Negative

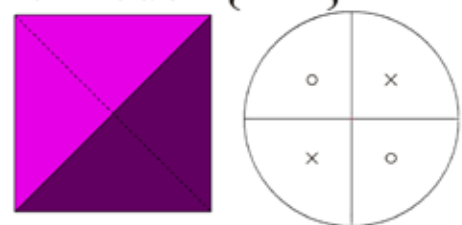
Right - Left

Up - Down

Front - Above

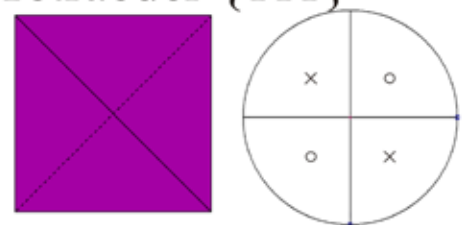
Kubisches Kristallsystem

Tetraeder $\{\bar{111}\}$



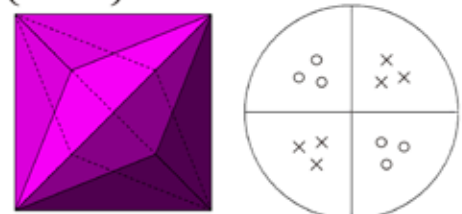
negativ

Tetraeder $\{111\}$



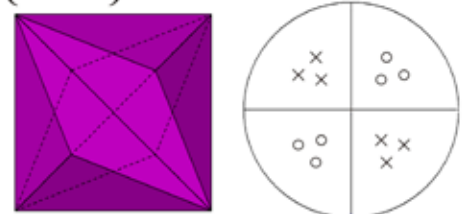
positiv

Pyramidentetraeder $\{223\}$



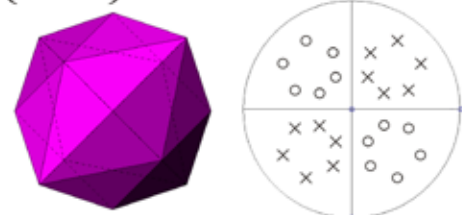
positiv

Pyramidentetraeder $\{\bar{2}23\}$



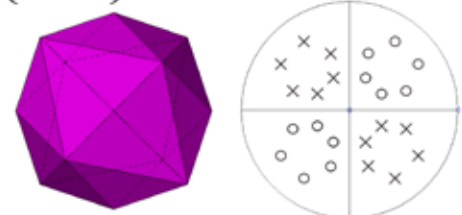
negativ

Hexakistetraeder $\{123\}$



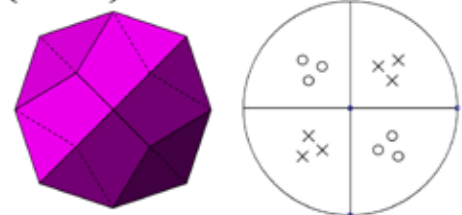
positiv

Hexakistetraeder $\{\bar{1}23\}$



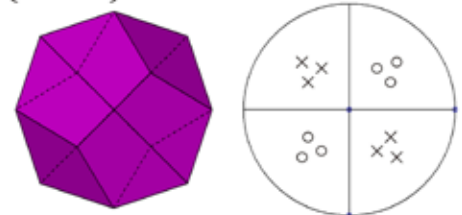
negativ

Deltoiddodekaeder $\{332\}$



positiv

Deltoiddodekaeder $\{\bar{3}32\}$

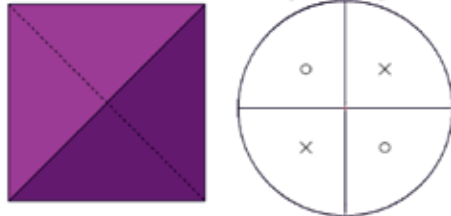


negativ

Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

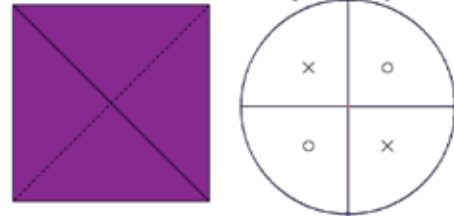
Cubic crystal system

Tetrahedron $\{\bar{1}\bar{1}\bar{1}\}$



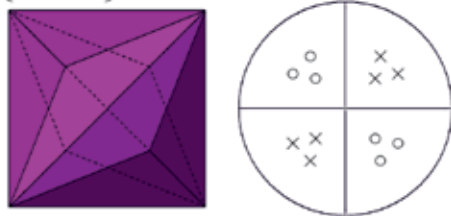
negative

Tetrahedron $\{111\}$



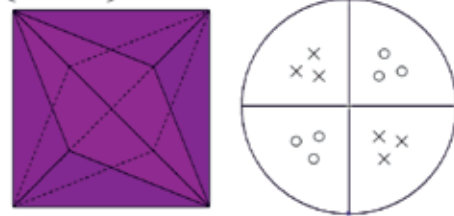
positive

Pyramidal tetrahedron $\{223\}$



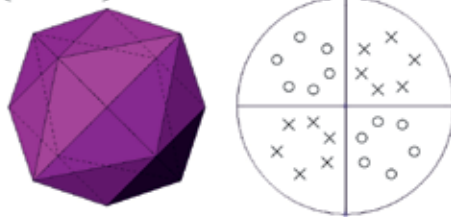
positive

Pyramidal tetrahedron $\{\bar{2}\bar{2}\bar{3}\}$



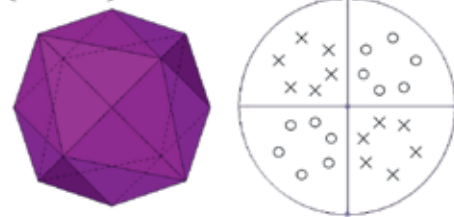
negative

Hexacistetrahedron $\{123\}$



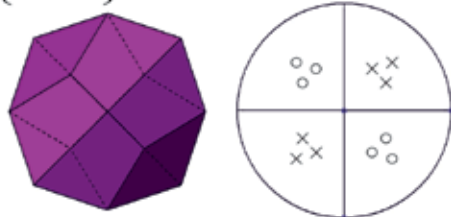
positive

Hexacistetrahedron $\{\bar{1}\bar{2}\bar{3}\}$



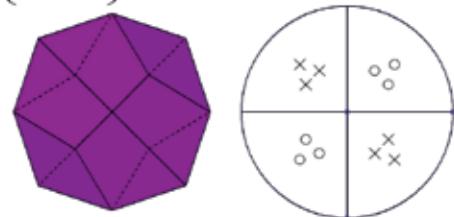
negative

Deltoid dodecahedron $\{332\}$



positive

Deltoid dodecahedron $\{\bar{3}\bar{3}\bar{2}\}$

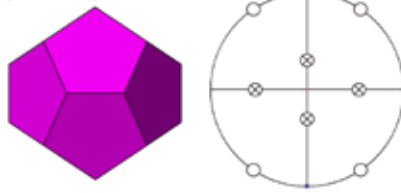


negative

These stereographic projections do not contain any symmetry elements; they are merely reference lines

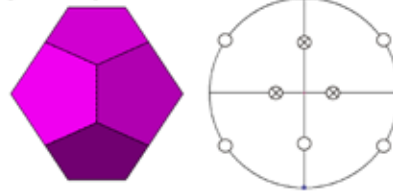
Kubisches Kristallsystem

Pentagondodekaeder
{230}



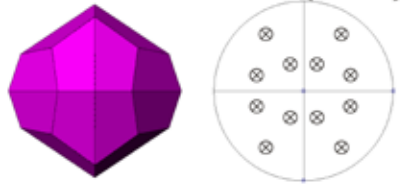
rechts
(positiv)

Pentagondodekaeder
{320}



links
(negativ)

Disdodekaeder {213}



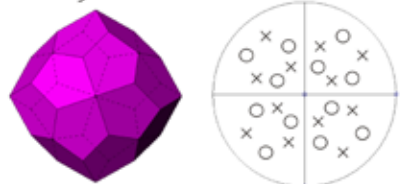
links
(positiv)

Disdodekaeder {123}



rechts
(negativ)

Pentagonikositetraeder
{123} rechts

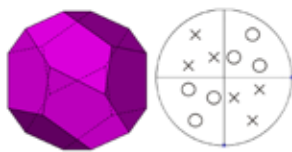


Pentagonikositetraeder
{213} links



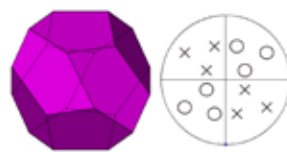
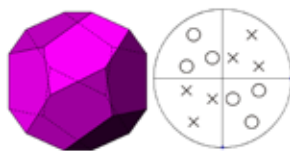
Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

Tetraedrischer Pentagondodekaeder



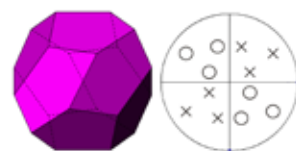
{123} positiv
rechts

{ $\bar{1}23$ } negativ
rechts



{213} positiv
links

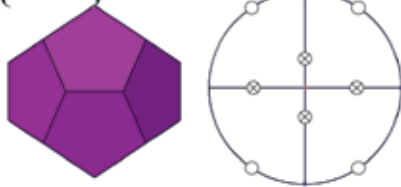
{ $\bar{2}13$ } negativ
links



Cubic crystal system

Pentagonal dodecahedron

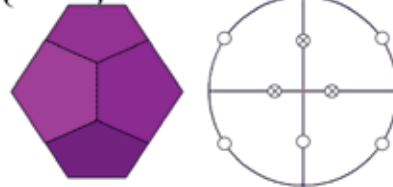
{230}



right
(positive)

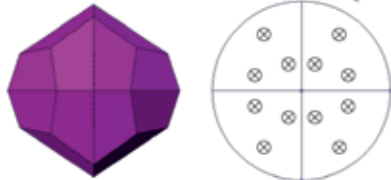
Pentagonal dodecahedron

{320}



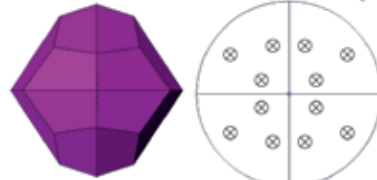
left
(negative)

Disdodecahedron {213}



left
(positive)

Disdodecahedron {123}

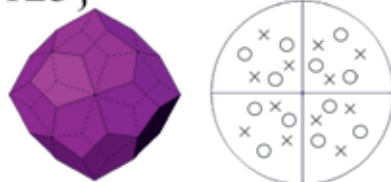


right
(negative)

Pentagonal icositetrahedron

{123}

right



Pentagonal icositetrahedron

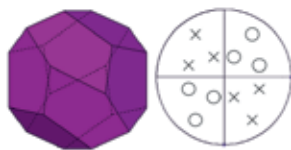
{213}

left



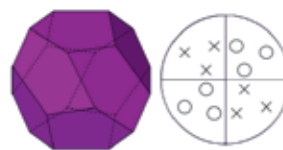
These stereographic projections do not contain any symmetry elements; they are merely reference lines.

Tetrahedral pentagonal dodecahedron



{123} positive
right

{ $\bar{1}23$ } negative
right



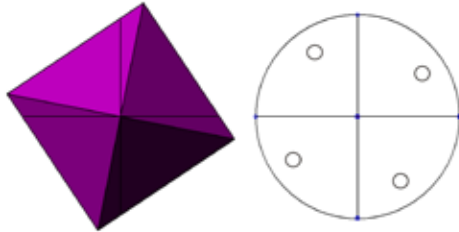
{213} positive
left

{ $\bar{2}13$ } negative
left



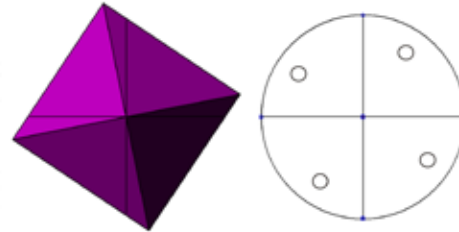
Tetragonales Kristallsystem

Pyramide $\{321\}$



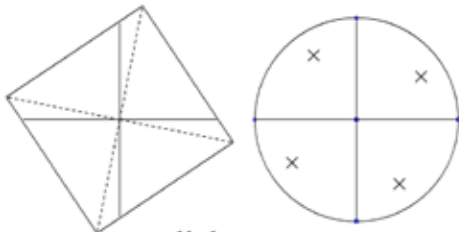
links
oben

Pyramide $\{231\}$



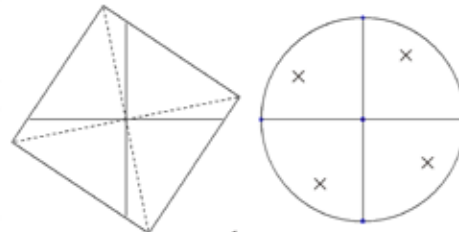
rechts
oben

Pyramide $\{32\bar{1}\}$



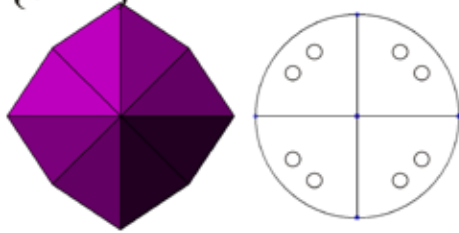
links
unten

Pyramide $\{23\bar{1}\}$



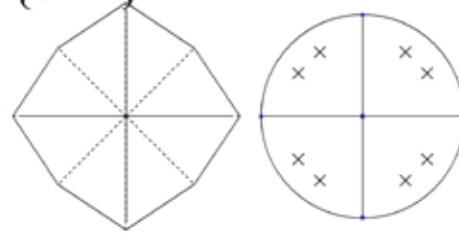
rechts
unten

Ditetrag. Pyramide
 $\{321\}$



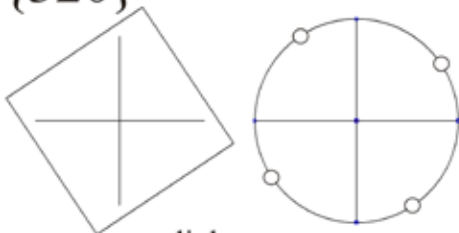
unten

Ditetrag. Pyramide
 $\{32\bar{1}\}$



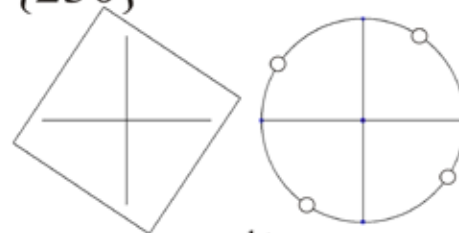
oben

Prisma III.Stellung
 $\{320\}$



links

Prisma III.Stellung
 $\{230\}$

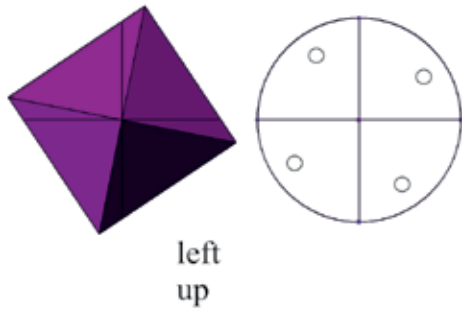


rechts

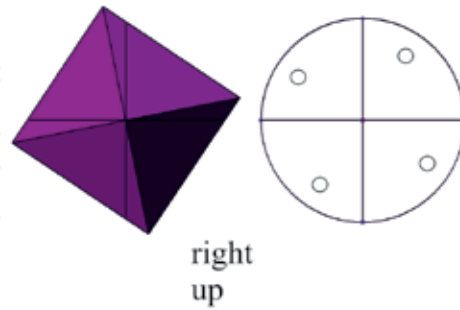
Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

Tetragonal crystal system

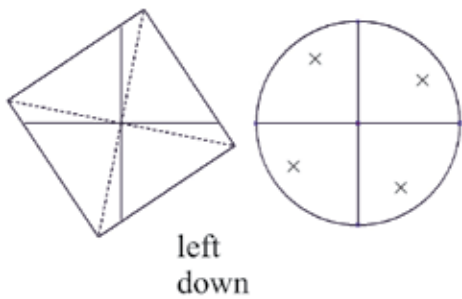
Pyramid {321}



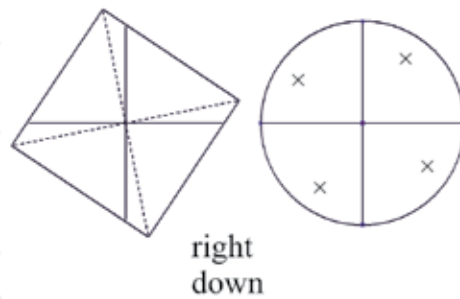
Pyramid {231}



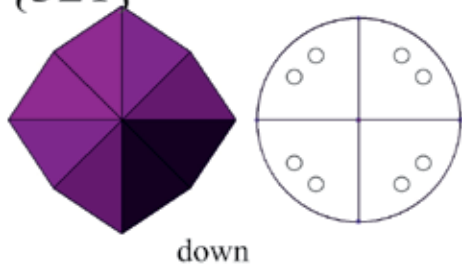
Pyramid {321̄}



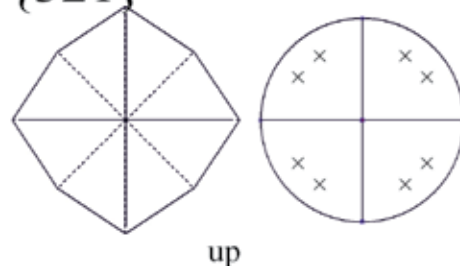
Pyramid {231̄}



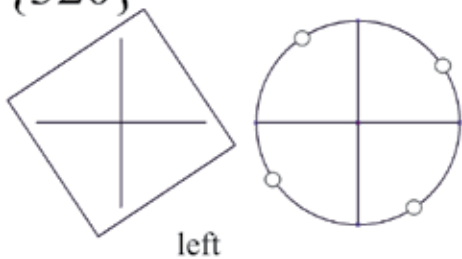
Ditetrag. Pyramid
{321}



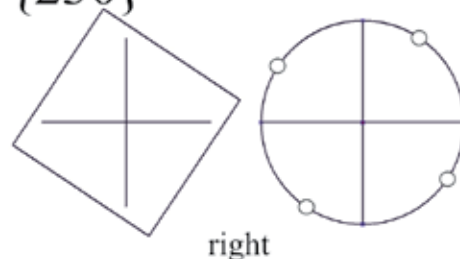
Ditetrag. Pyramid
{321̄}



Prism III.Position
{320}



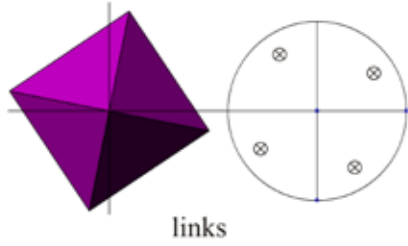
Prism III.Position
{230}



These stereographic projections do not contain any symmetry elements: they are merely reference lines

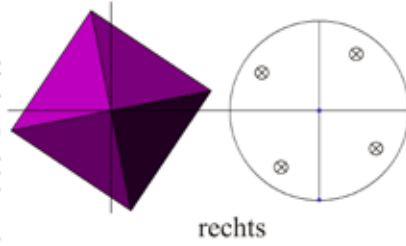
Tetragonales Kristallsystem

Dipyramide {321}



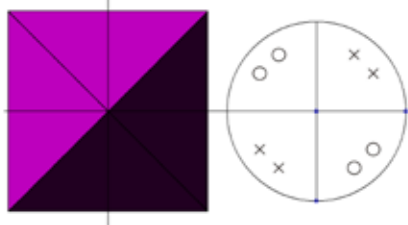
links

Dipyramide {231}



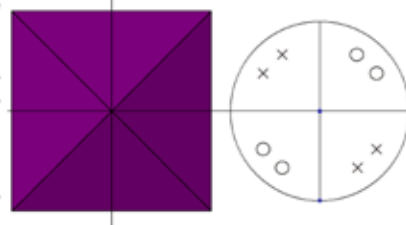
rechts

Skalenoeder {321}



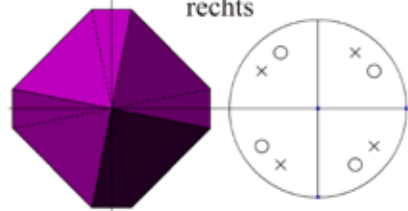
positiv

Skalenoeder {3̄21}



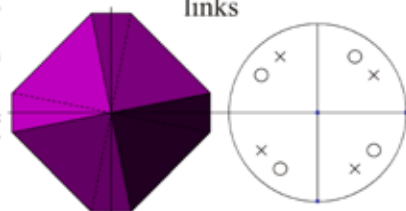
negativ

Trapezoeder {321}



rechts

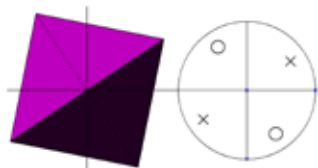
Trapezoeder {231}



links

Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

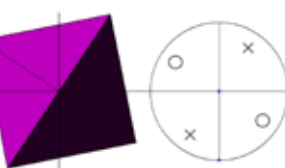
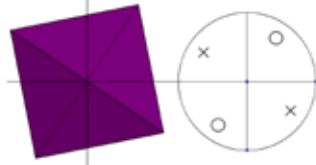
Tetragonales Disphenoid



links
positiv

{321}

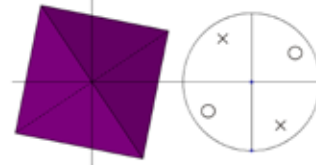
links
negativ
{3̄21}



rechts
positiv

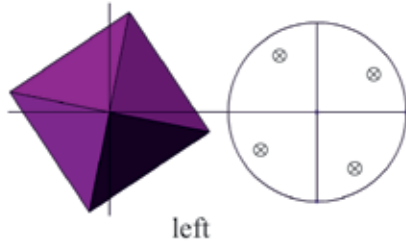
{231}

rechts
negativ
{2̄31}



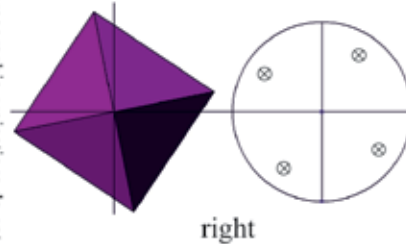
Tetragonal crystal system

Dipyramid {321}



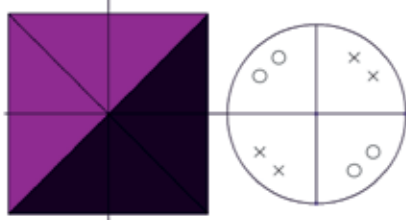
left

Dipyramid {231}



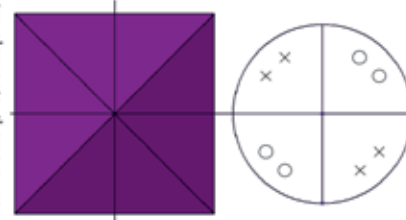
right

Scalenohedron {321}



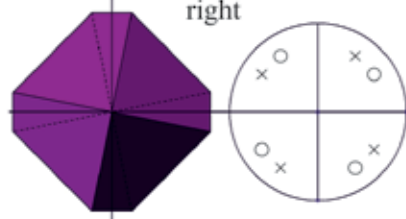
positive

Scalenohedron {3̄21}



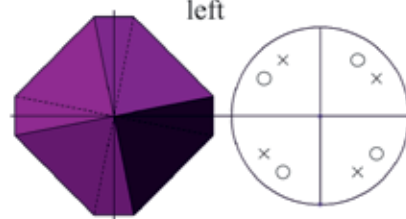
negative

Trapezohedron {321}



right

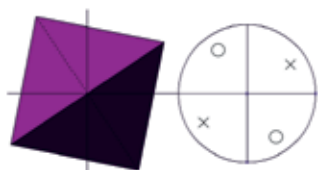
Trapezohedron {231}



left

These stereographic projections do not contain any symmetry elements; they are merely reference lines

Tetragonal Disphenoid

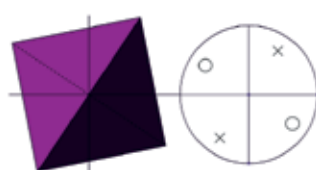
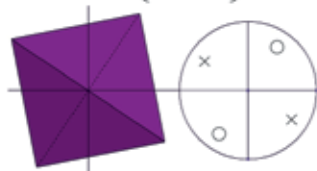


left
positive

{321}

left
negative

{3̄21}



right
positive

{231}

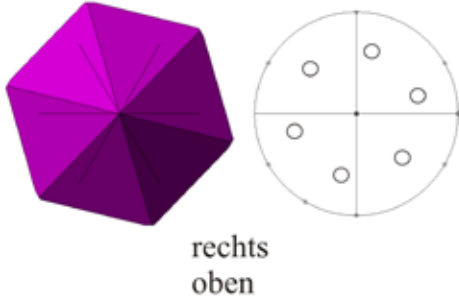
right
negative

{2̄31}

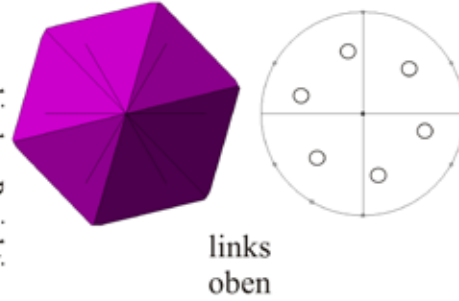


Hexagonales Kristallsystem

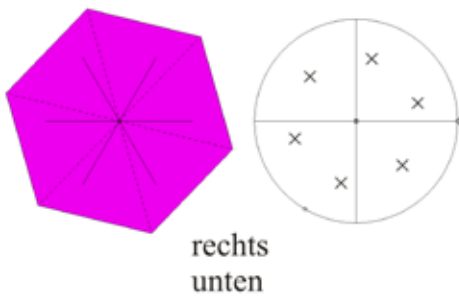
Pyramide $\{132\}$



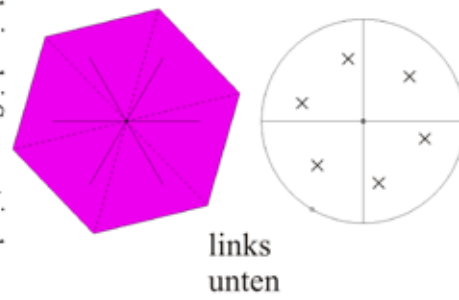
Pyramide $\{312\}$



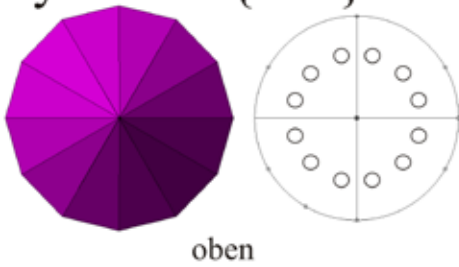
Pyramide $\{13\bar{2}\}$



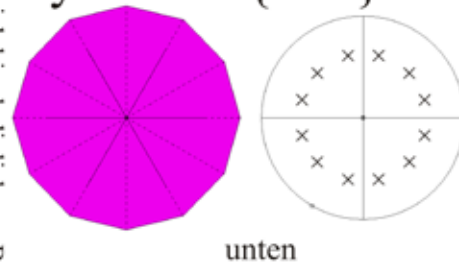
Pyramide $\{31\bar{2}\}$



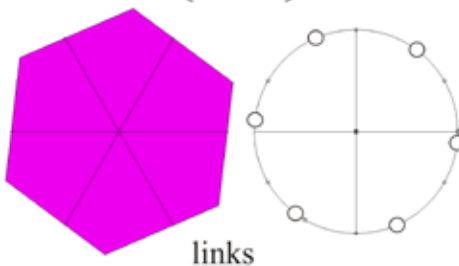
Dihexagonale
Pyramide $\{312\}$



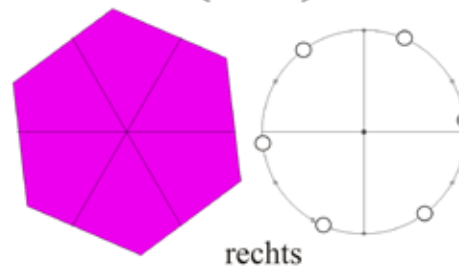
Dihexagonale
Pyramide $\{31\bar{2}\}$



Prisma $\{320\}$



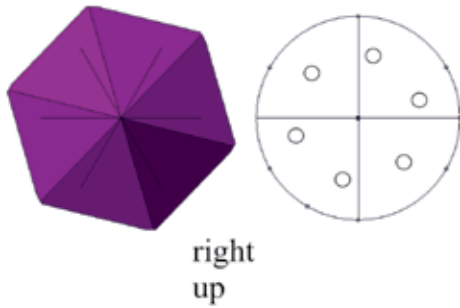
Prisma $\{230\}$



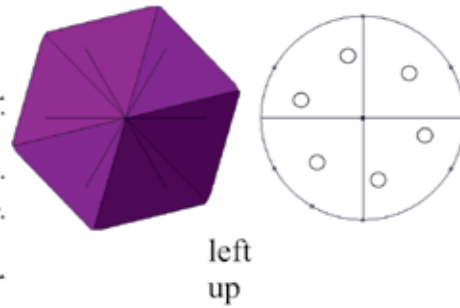
Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

Hexagonal crystal system

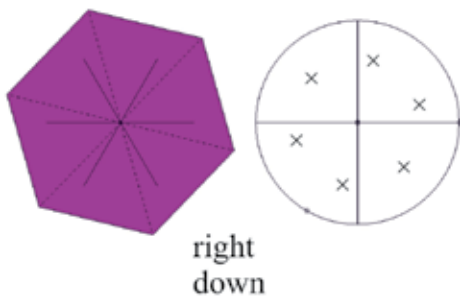
Pyramid {132}



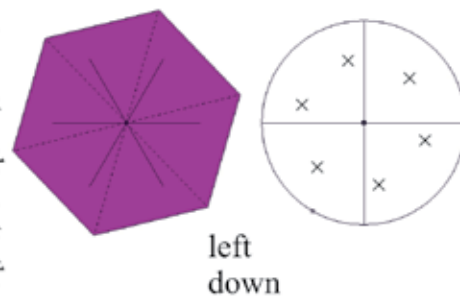
Pyramid {312}



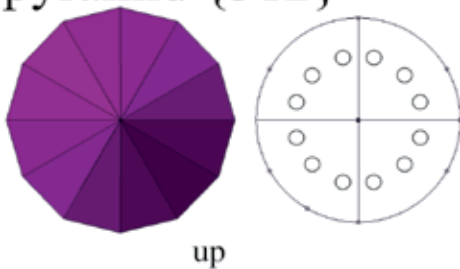
Pyramid {13 $\bar{2}$ }



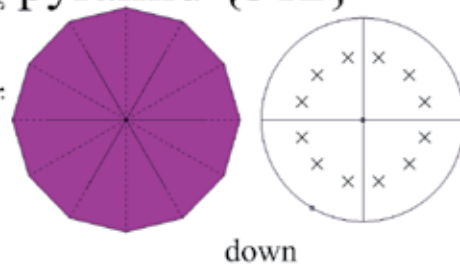
Pyramid {31 $\bar{2}$ }



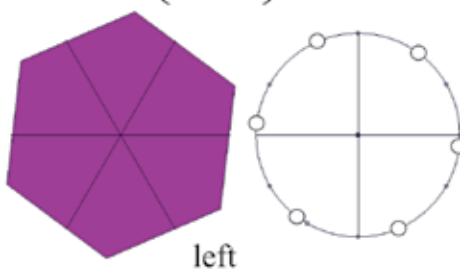
Dihexagonal pyramid {312}



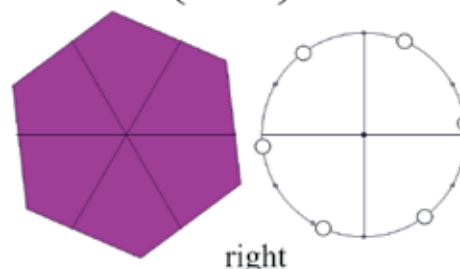
Dihexagonal pyramid {31 $\bar{2}$ }



Prism {320}



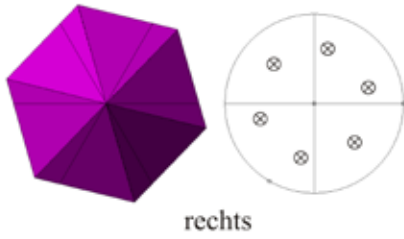
Prism {230}



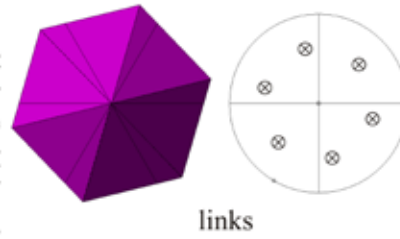
These stereographic projections do not contain any symmetry elements; they are merely reference lines

Hexagonales Kristallsystem

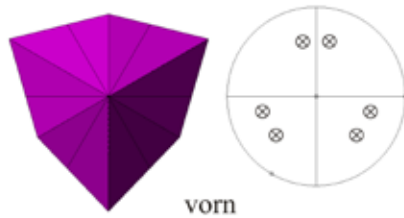
Dipyramide {132}



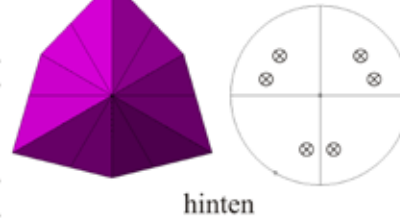
Dipyramide {312}



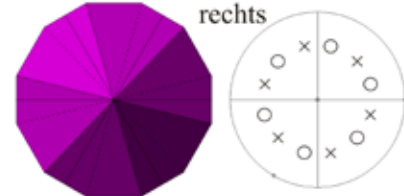
Ditrigonale Dipyramide {132}



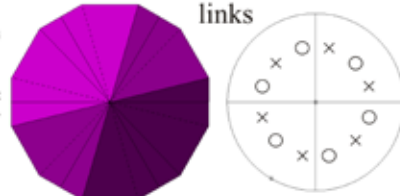
Ditrigonale Dipyramide {312}



Trapezoeder {132}



Trapezoeder {312}



Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

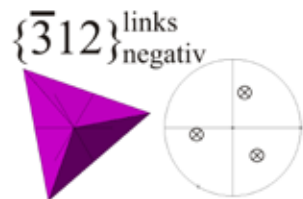
Trigonale Dipyramide



$\{1\bar{3}2\}$ rechts negativ



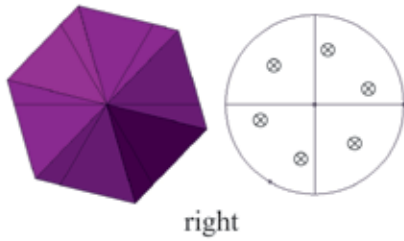
$\{312\}$ links positiv



$\{\bar{3}12\}$ links negativ

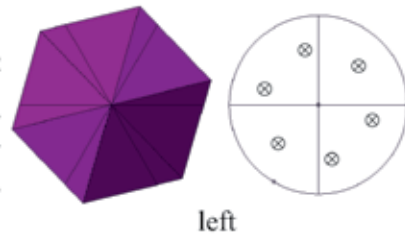
Hexagonal crystal system

Dipyramid {132}



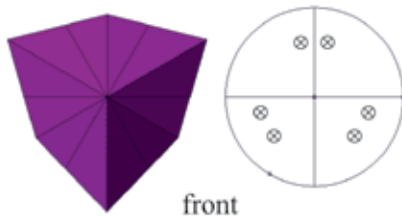
right

Dipyramid {312}



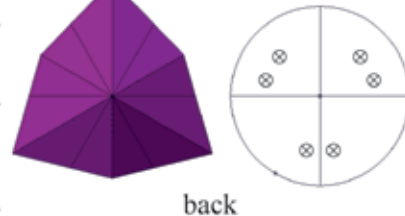
left

Ditrigonal dipyramid {132}



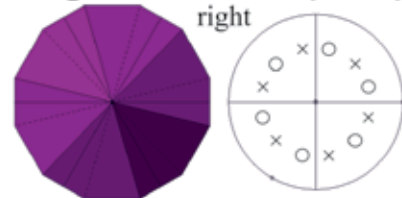
front

Ditrigonal dipyramid {312}



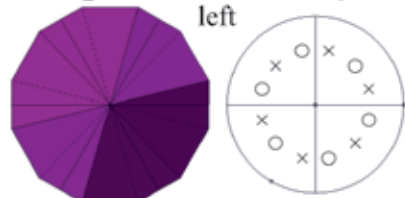
back

Trapezohedron {132}



right

Trapezohedron {312}



left

These stereographic projections do not contain any symmetry elements; they are merely reference lines

Trigonal Dipyramid



{132} right positive



$\bar{1}32$ right negative



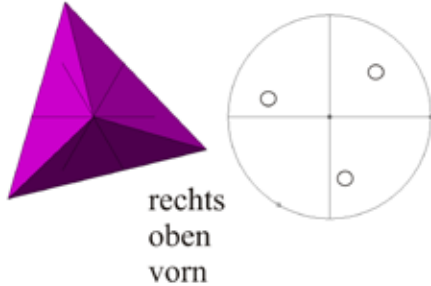
{312} left positive



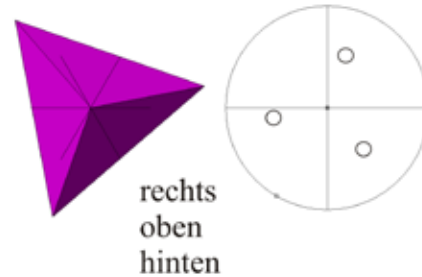
$\bar{3}12$ left negative

Trigonales Kristallsystem

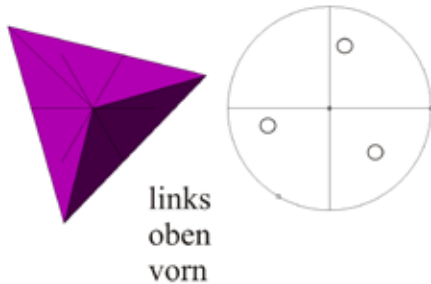
Pyramide $\{312\}$



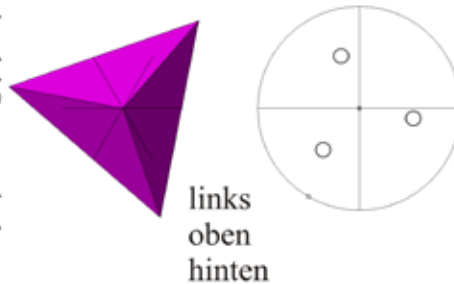
Pyramide $\{\bar{3}12\}$



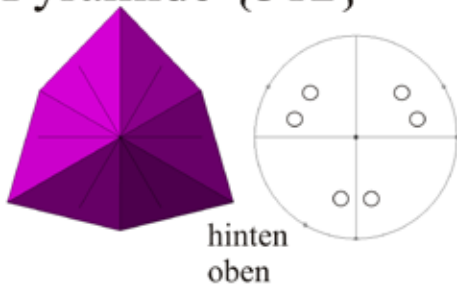
Pyramide $\{132\}$



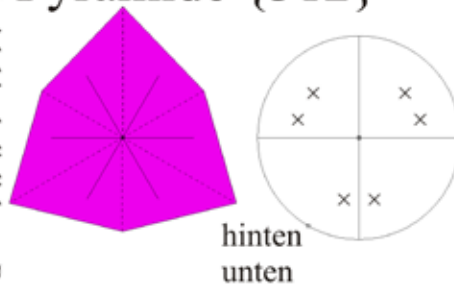
Pyramide $\{\bar{1}32\}$



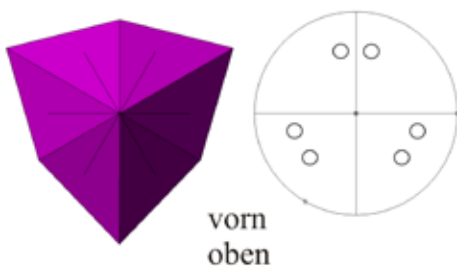
Ditrigonale
Pyramide $\{312\}$



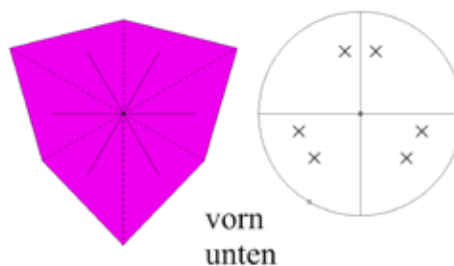
Ditrigonale
Pyramide $\{31\bar{2}\}$



Ditrigonale
Pyramide $\{132\}$



Ditrigonale
Pyramide $\{13\bar{2}\}$

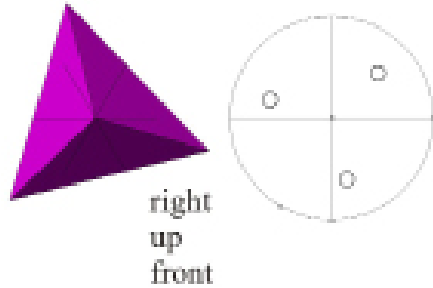


Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

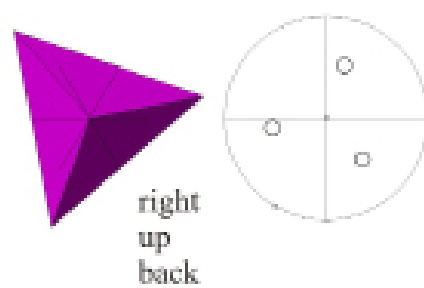
analog
dazu
"unten"

Trigonal crystal system

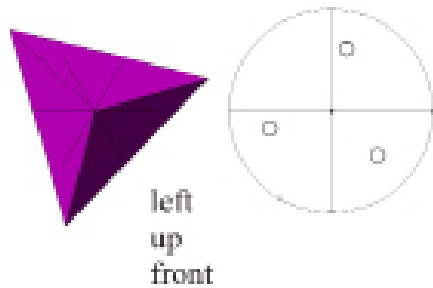
Pyramid {312}



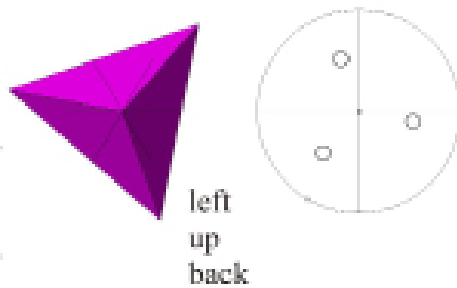
Pyramid { $\bar{3}12$ }



Pyramid {132}

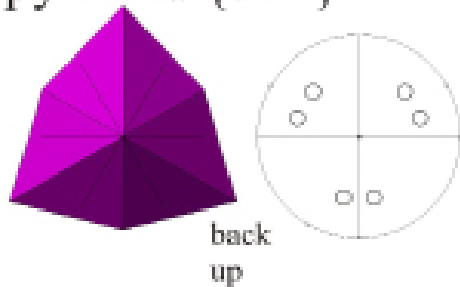


Pyramid { $\bar{1}32$ }

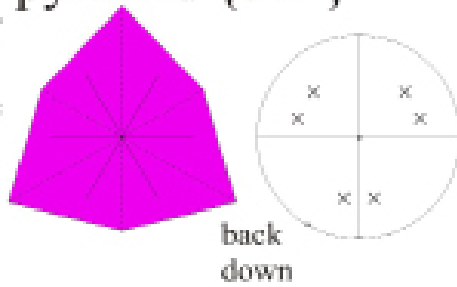


analog
to
"down"

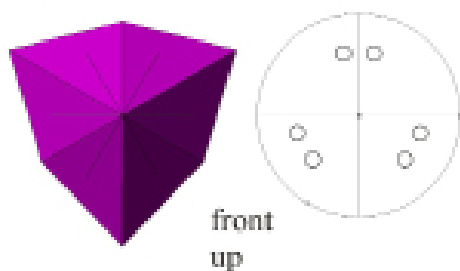
Ditrigonal
pyramid {312}



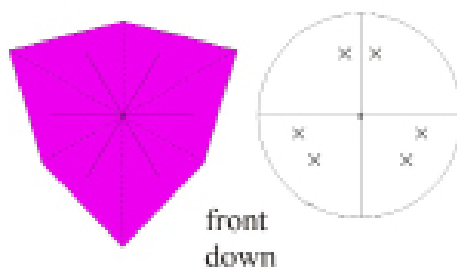
Ditrigonal
pyramid { $31\bar{2}$ }



Ditrigonal
pyramid {132}



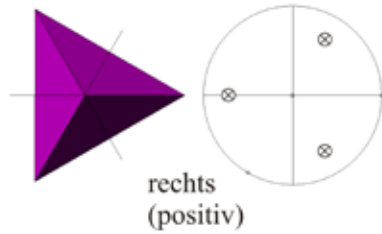
Ditrigonal
pyramid { $13\bar{2}$ }



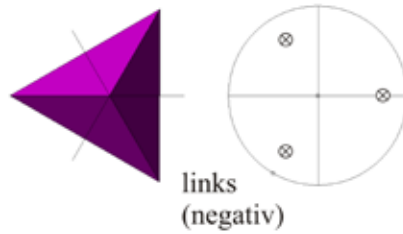
These stereographic projections do not contain any symmetry elements; they are merely reference lines

Trigonales Kristallsystem

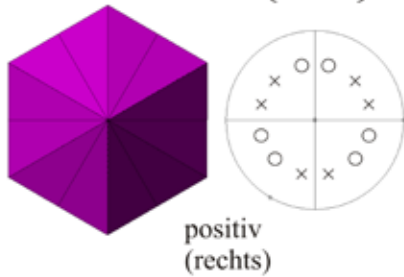
Dipyramide $\{332\}$



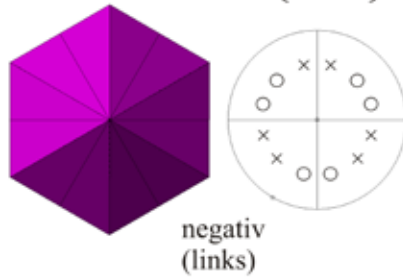
Dipyramide $\{\bar{3}32\}$



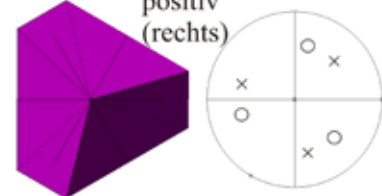
Ditrigonaler Skalenoeder $\{132\}$



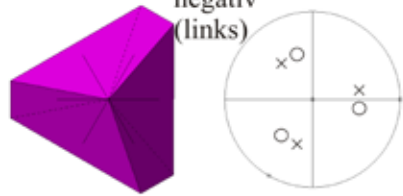
Ditrigonaler Skalenoeder $\{312\}$



Trapezoeder $\{132\}$
positiv
(rechts)



Trapezoeder $\{\bar{1}32\}$
negativ
(links)



Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente, es handelt sich hier lediglich um Bezugslinien

Trigonaler Rhomboeder



$\{\bar{1}32\}$ rechts
negativ

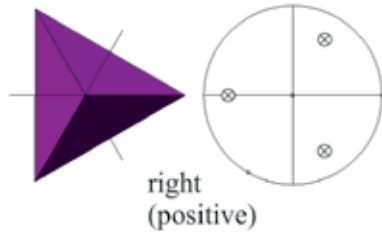


$\{\bar{3}12\}$ links
negativ

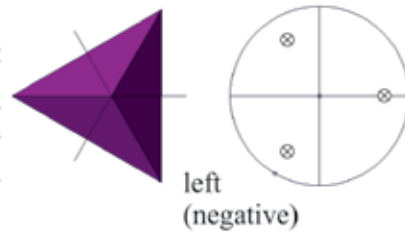


Trigonal crystal system

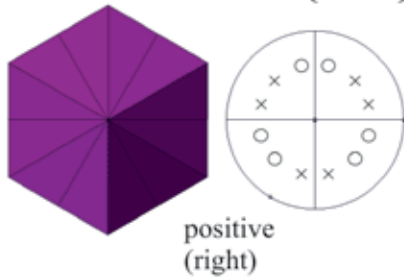
Dipyramid $\{332\}$



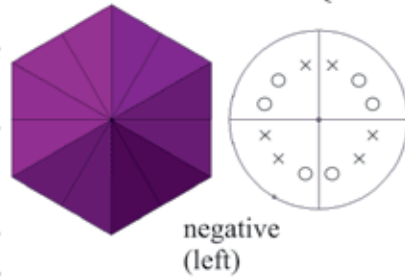
Dipyramid $\{\bar{3}\bar{3}2\}$



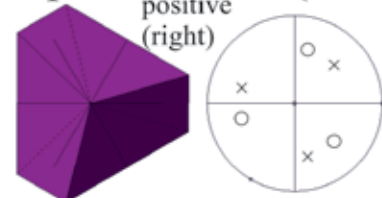
Ditrigonal scalenohedron $\{132\}$



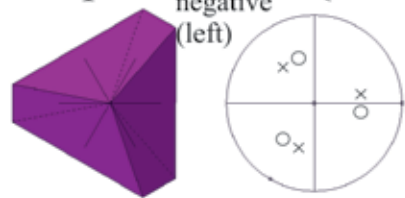
Ditrigonal scalenohedron $\{312\}$



Trapezohedron $\{132\}$



Trapezohedron $\{\bar{1}\bar{3}2\}$



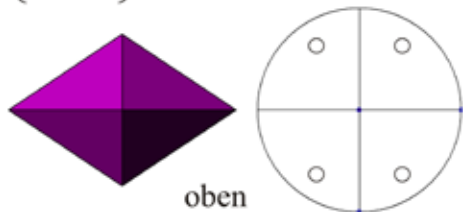
These stereographic projections do not contain any symmetry elements; they are merely reference lines

Trigonal rhombohedron

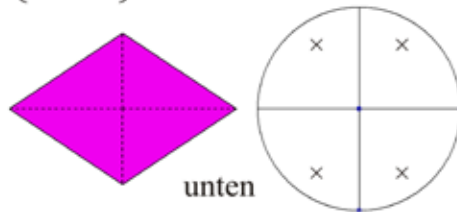


Orthorhombisches Kristallsystem

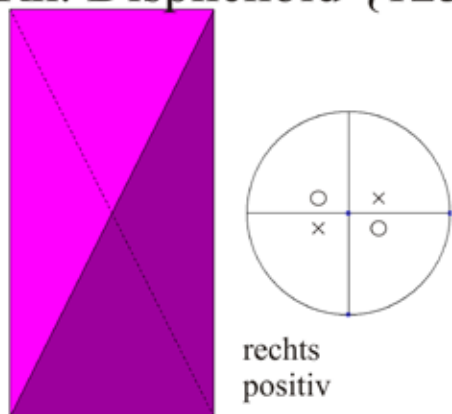
Rhombische Pyramide
{321}



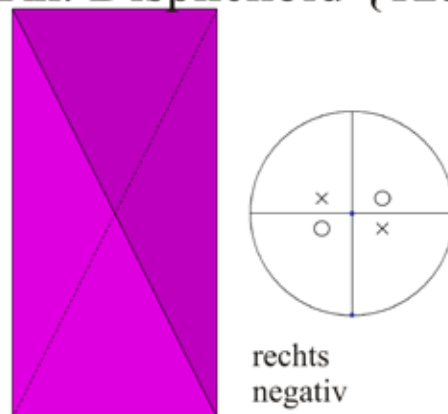
Rhombische Pyramide
{321}



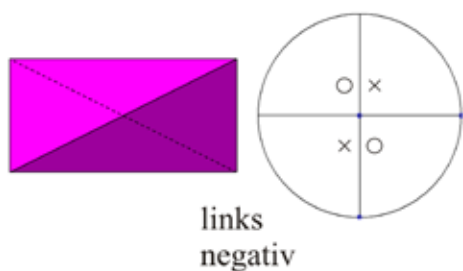
Rh. Disphenoid {123}



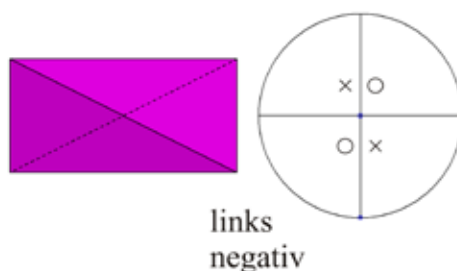
Rh. Disphenoid $\{\bar{1}23\}$



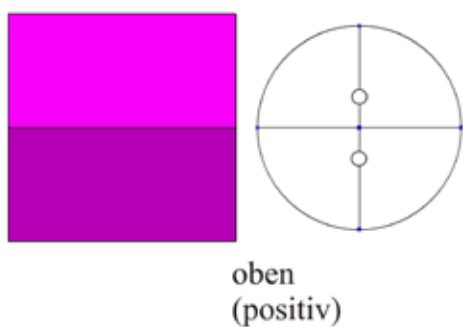
Rh. Disphenoid {213}



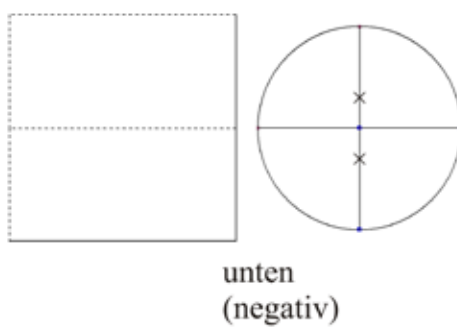
Rh. Disphenoid $\{\bar{2}13\}$



Doma II.Stellung
{203}



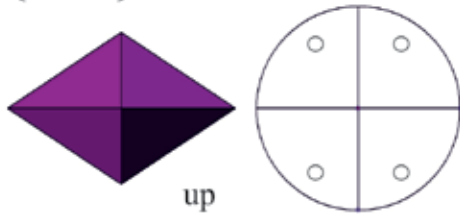
Doma II.Stellung
 $\{\bar{2}03\}$



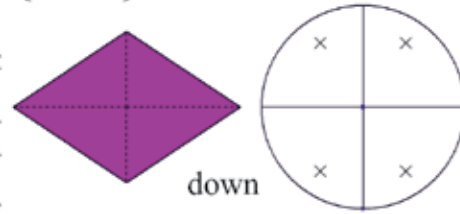
Diese stereographischen Projektionen beinhalten keinerlei Symmetrieelemente; es handelt sich hier lediglich um Bezugslinien

Orthorhombic crystal system

Rhombic pyramid
{321}

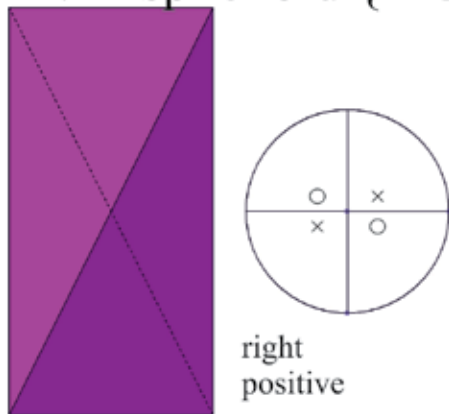


Rhombic Pyramid
{321}

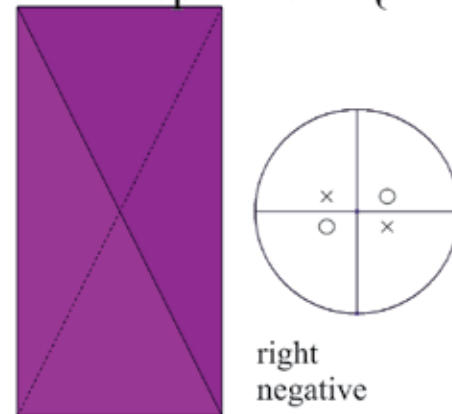


These stereographic projections do not contain any symmetry elements; they are merely reference lines

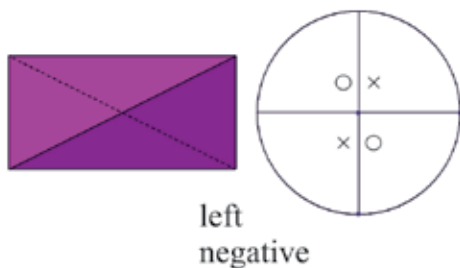
Rh. Disphenoid {123}



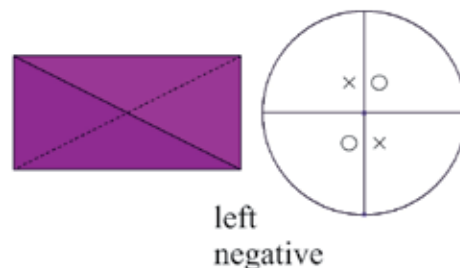
Rh. Disphenoid {1̄23}



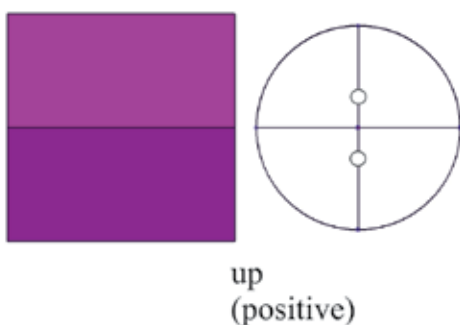
Rh. Disphenoid {213}



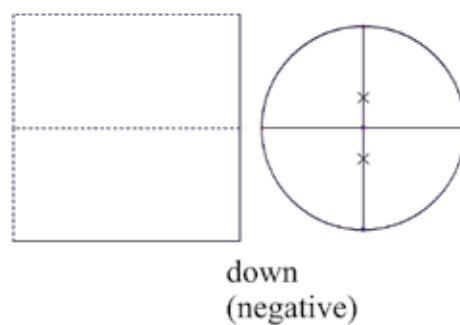
Rh. Disphenoid {2̄13}



Doma II. position
{203}



Doma II. position
{2̄03}



System	Klasse	Symmetrieelement	polare Richtungen im Kristall	nichtpolare Richtungen im Kristall
triklin	1	-	* alle	keine
	2	$1\vec{0}$	* b	alle \perp b
monoklin	m	$1m$ ($m \equiv \vec{2}$)	a, c und alle Zonen die in (010) liegen	b
	mm2	$1\vec{0} + 1m + 1m$	* c	alle \perp c
tetragonal	4	$1\vec{0}$	* c	alle \perp c
	4mm	$1\vec{0} + 2m + 2m$	* c	alle \perp c
trigonal	3	$\vec{3}$	* c und alle Zonen \perp c	keine
	32	$1\# + 3\vec{0}$	a_1, a_2, a_3 und alle Zonen \perp c	alle \perp a_1, a_2, a_3
hexagonal	3m	$1\vec{3} + 3m$	* c	a_1, a_2, a_3
	6	$1\vec{0}$	* c	alle \perp c
hexagonal	$\vec{6}$	$1\vec{0} \equiv \# + m$	* a_1, a_2, a_3 und alle Zonen \perp c	c
	6mm	$1\vec{0} + 3m + 3m$	* c	alle \perp c
kubisch	$\vec{6}2m$	$1\vec{0} + 3\vec{0} + 3m$	alle \bullet	a_1, a_2, a_3 und alle Zonen \perp auf \bullet die dazwischenliegen
	23	$3\bullet + 4\vec{3}$	a_1, a_2, a_3	alle \perp a_1, a_2, a_3
	$\vec{4}3m$	$3\vec{0} + 4\vec{3} + 6m$	a_1, a_2, a_3	alle \perp a_1, a_2, a_3

Weitere azentrische Klassen mit polaren Richtungen: 222, 4, 422, 42m, 622, 432

* = Klassen mit einzigartigen polaren Achsen, zu denen keine äquivalenten Achsen existieren:
ferroelektrische Stoffe

System	Class	Symmetry element	Polar directions in crystal	Not polar direction in crystal
triclinic	1	-	* all	None
monoclinic	2	$1\bar{1}\bar{0}$	* b	all \perp b
	m	$1m (m \equiv \bar{2})$	a, c and all Zones lying in (010)	b
orthorhombic	mm2	$1\bar{1}\bar{0} + 1m + 1m$	* c	all \perp c
	4	$1\bar{2}$	* c	all \perp c
tetragonal	4mm	$1\bar{2} + 2m + 2m$	* c	all \perp c
	3	$\bar{4}$	* c and all Zones \perp c	keine
trigonal	32	$1\bar{3} + 3\bar{0}$	a_1, a_2, a_3 and all Zones \perp c	all \perp a_1, a_2, a_3
	3m	$1\bar{3} + 3m$	* c	a_1, a_2, a_3
hexagonal	6	$1\bar{6}$	* c	all \perp c
	$\bar{6}$	$1\bar{6} \equiv \bar{3} + m$	* a_1, a_2, a_3 and all Zones \perp c	c
	6mm	$1\bar{6} + 3m + 3m$	* c	all \perp c
	$\bar{6}2m$	$1\bar{6} + 3\bar{0} + 3m$	all \bullet	a_1, a_2, a_3 and all Zones \perp on \bullet in between
cubic	23	$3\bar{2} + 4\bar{3}$	a_1, a_2, a_3	all \perp a_1, a_2, a_3
	$\bar{4}3m$	$3\bar{4} + 4\bar{3} + 6m$	a_1, a_2, a_3	all \perp a_1, a_2, a_3

Other acentric classes with polar directions: 222, 4, 422, 42m, 622, 432

* = Classes with unique polar Axes, with no equivalent axes : ferroelectric materials

Die Kombination von Formen

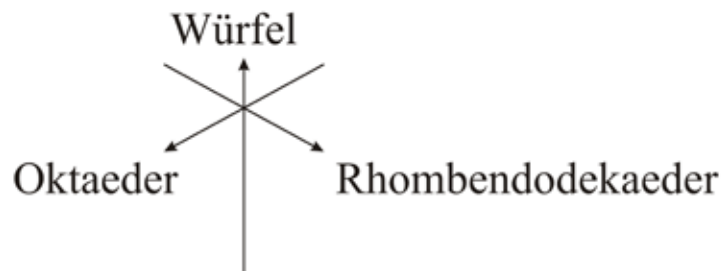
Zwei Formen: Würfel ↔ Oktaeder

Rhombendodekaeder → Würfel
Rhombendodekaeder → Oktaeder

Tetraeder → Würfel
Tetraeder → Rhombendodekaeder

positiver Tetraeder ↔ negativer Tetraeder

Drei Formen:



Kombinationen von Flächen und Steilen Rhomboedern

The combination of forms

Two forms:

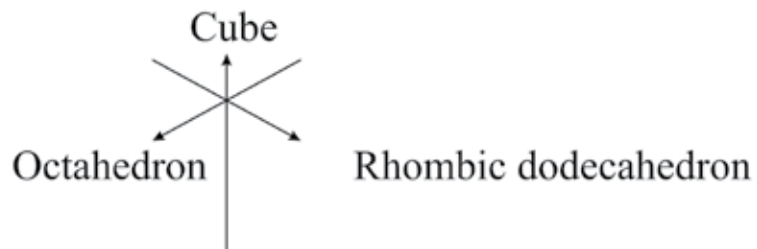
Cube ↔ Octahedron

Rhombic dodecahedron → Cube
 Rhombic dodecahedron → Octahedron

Tetrahedron → Cube
 Tetrahedron → Rhombic dodecahedron

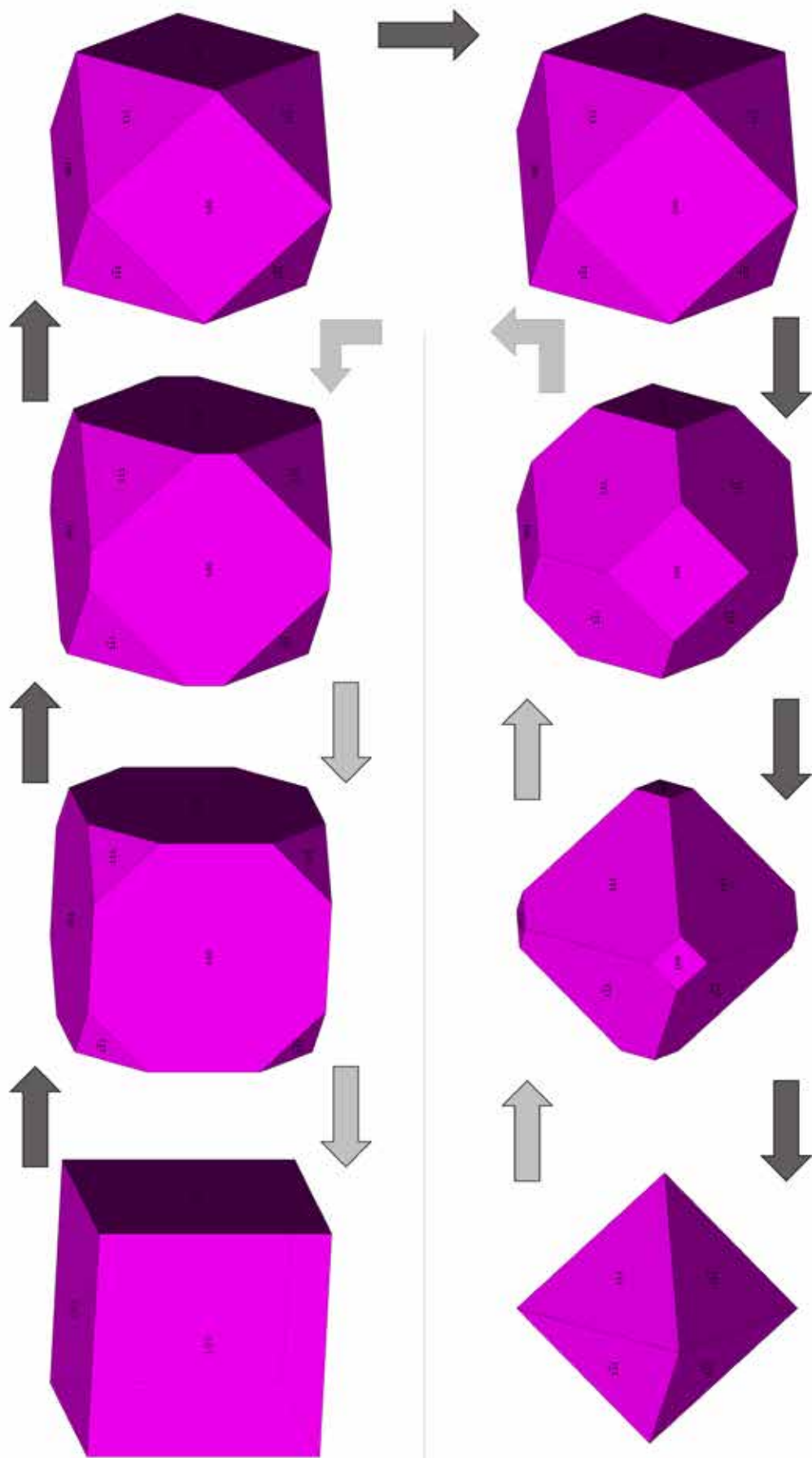
positive tetrahedron ↔ negative tetrahedron

Three forms:

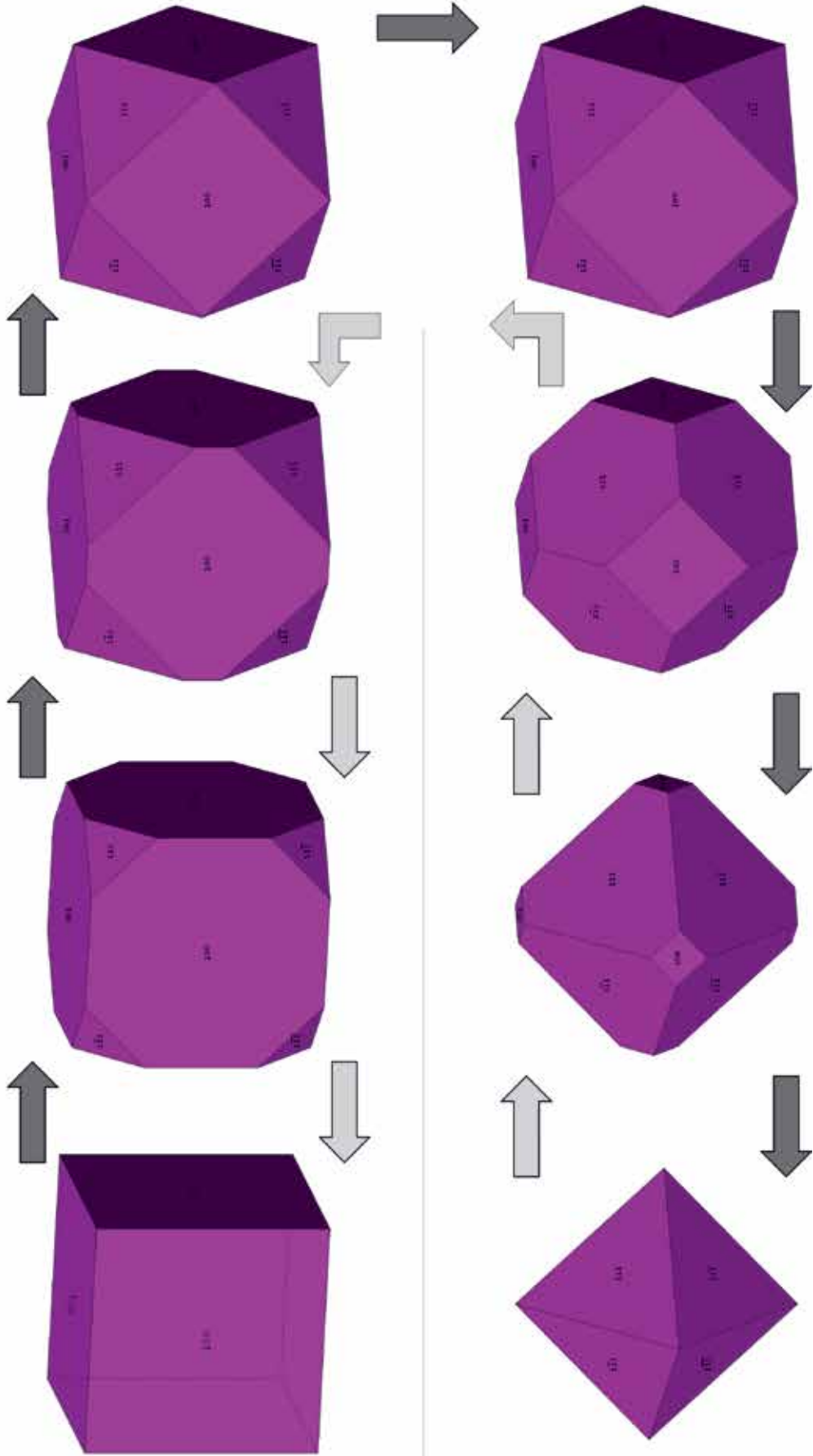


Combinations of flat and steep rhomboids

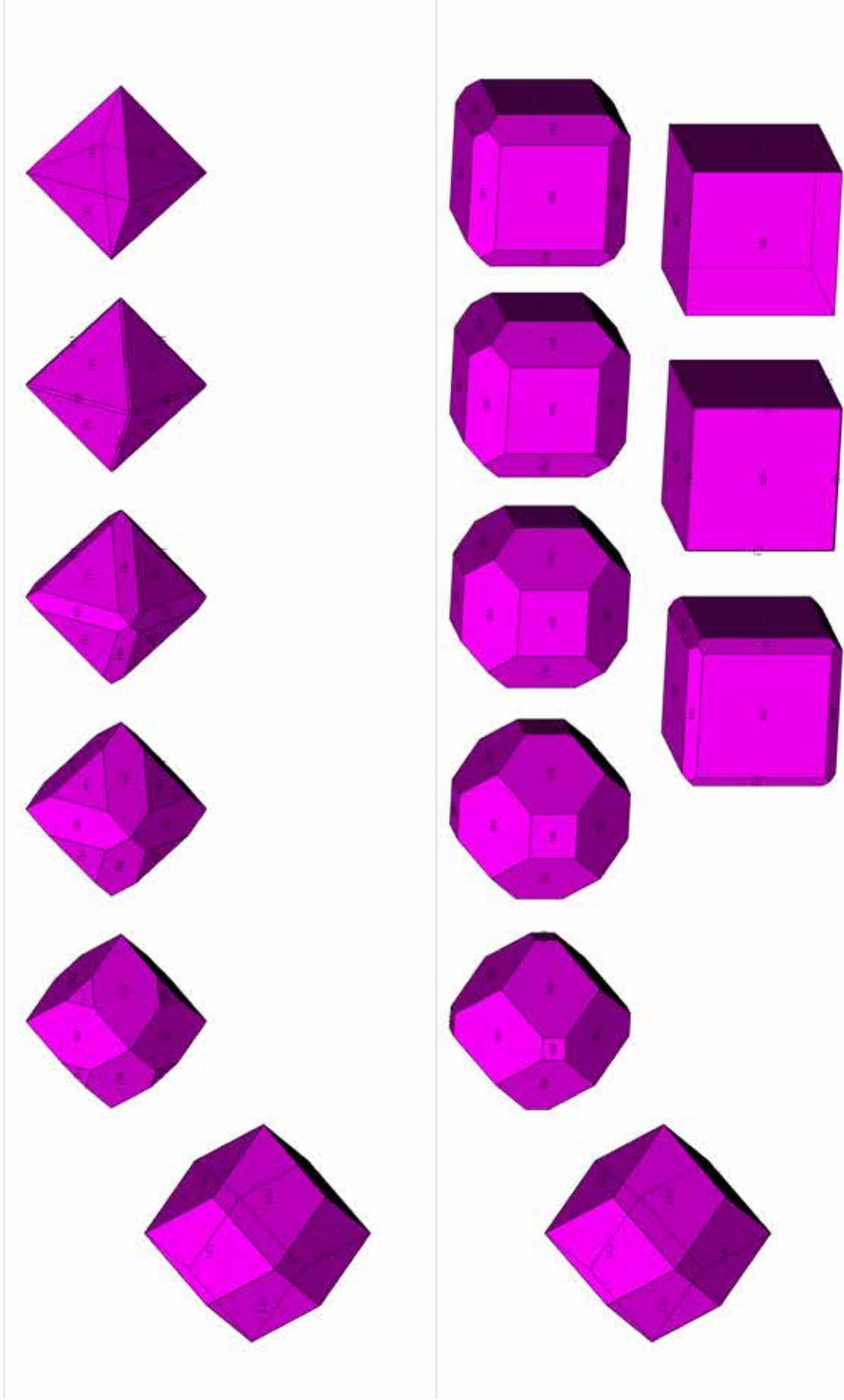
Vom Würfel zum Oktaeder



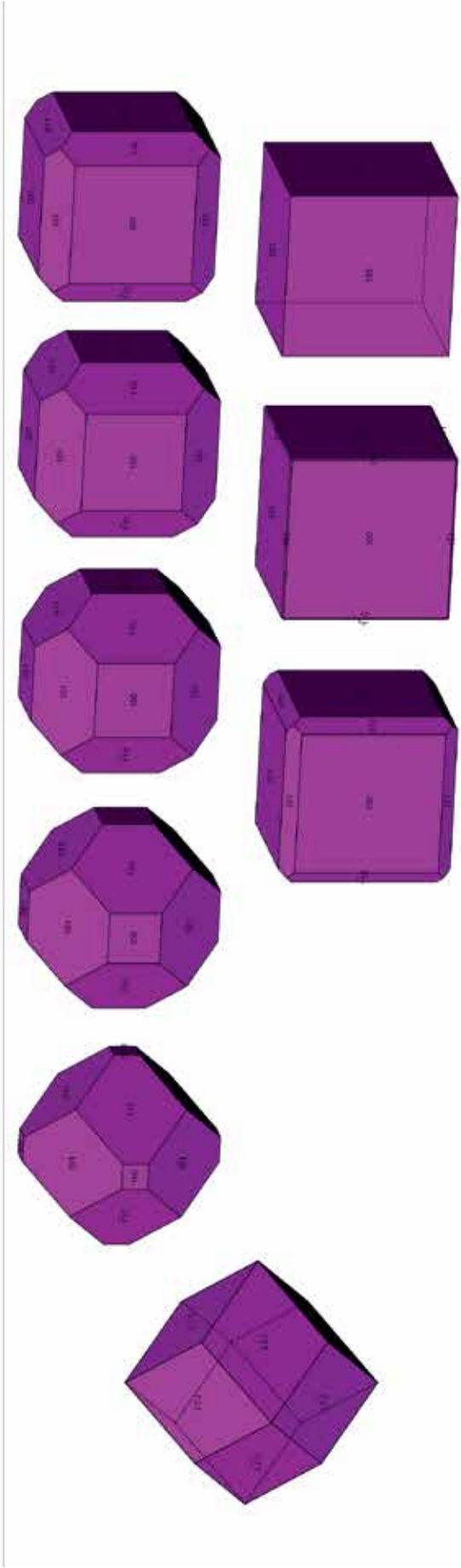
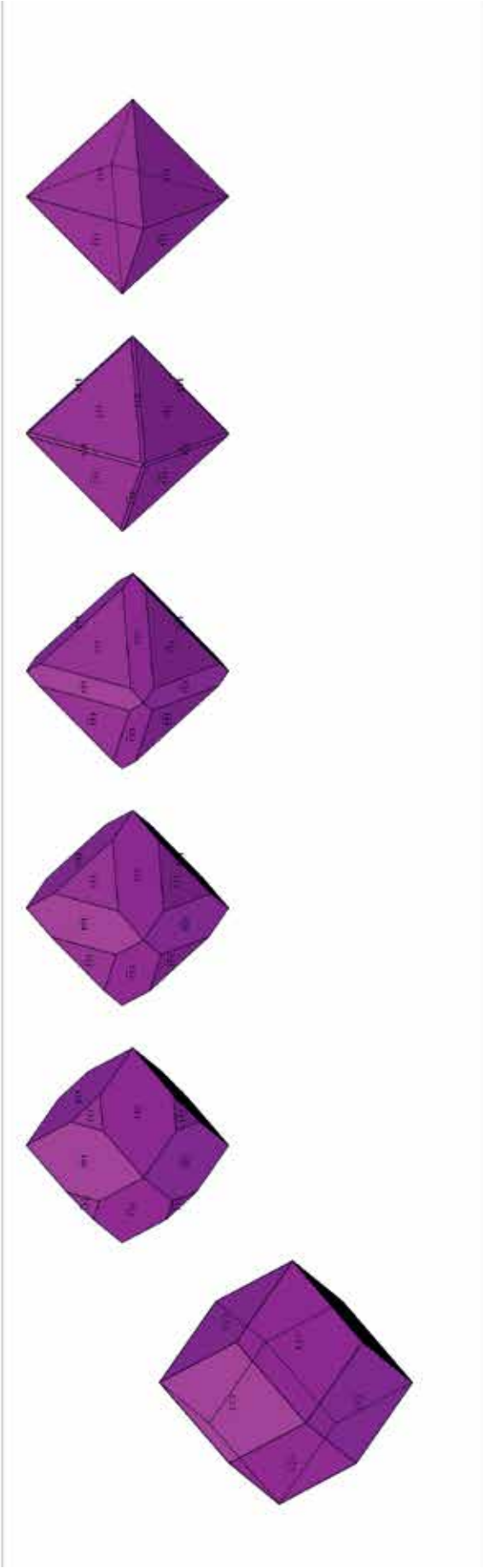
From cube to octahedron



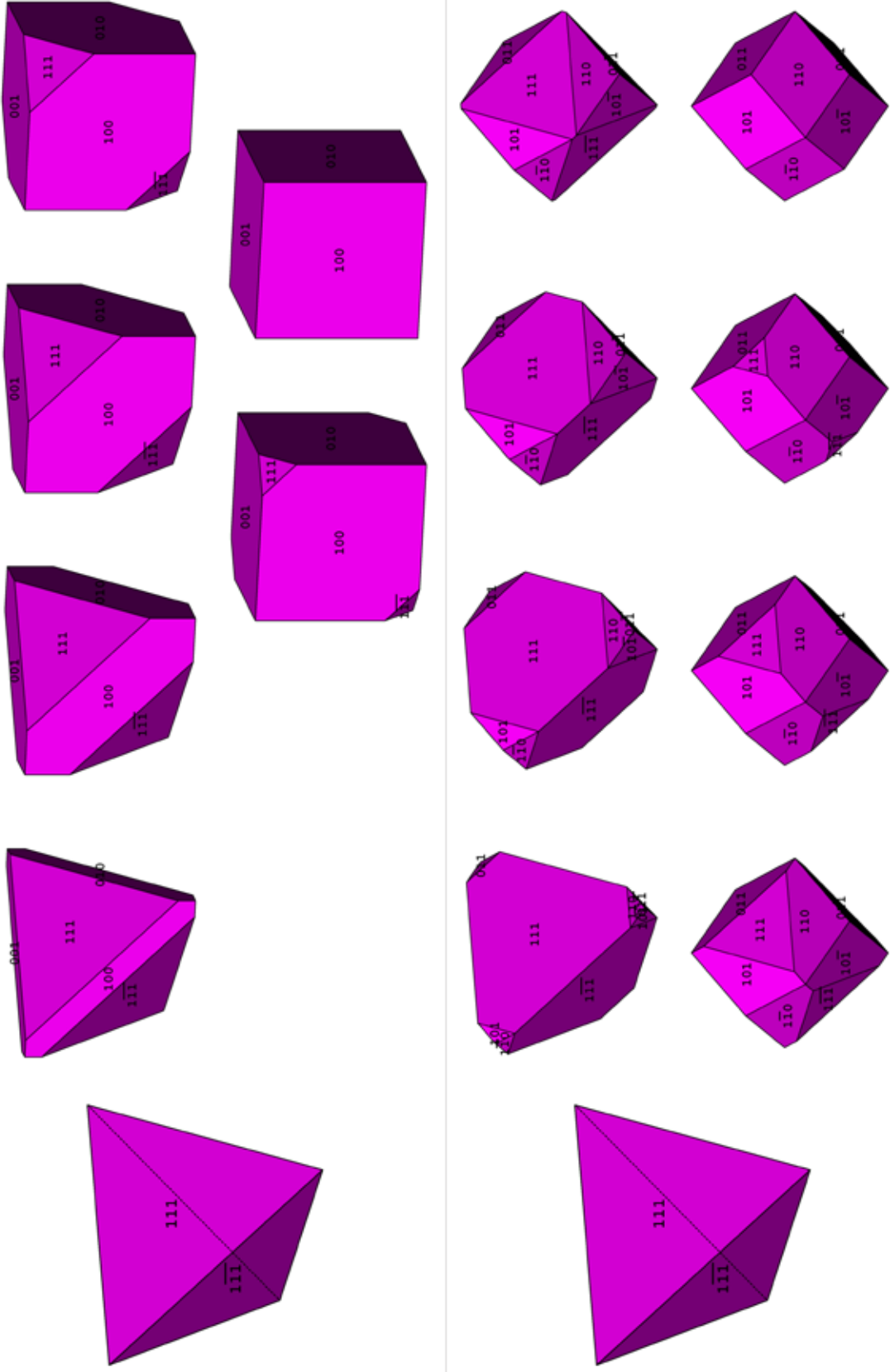
Kombination des Rhombendodekaeder mit Oktaeder und Würfel



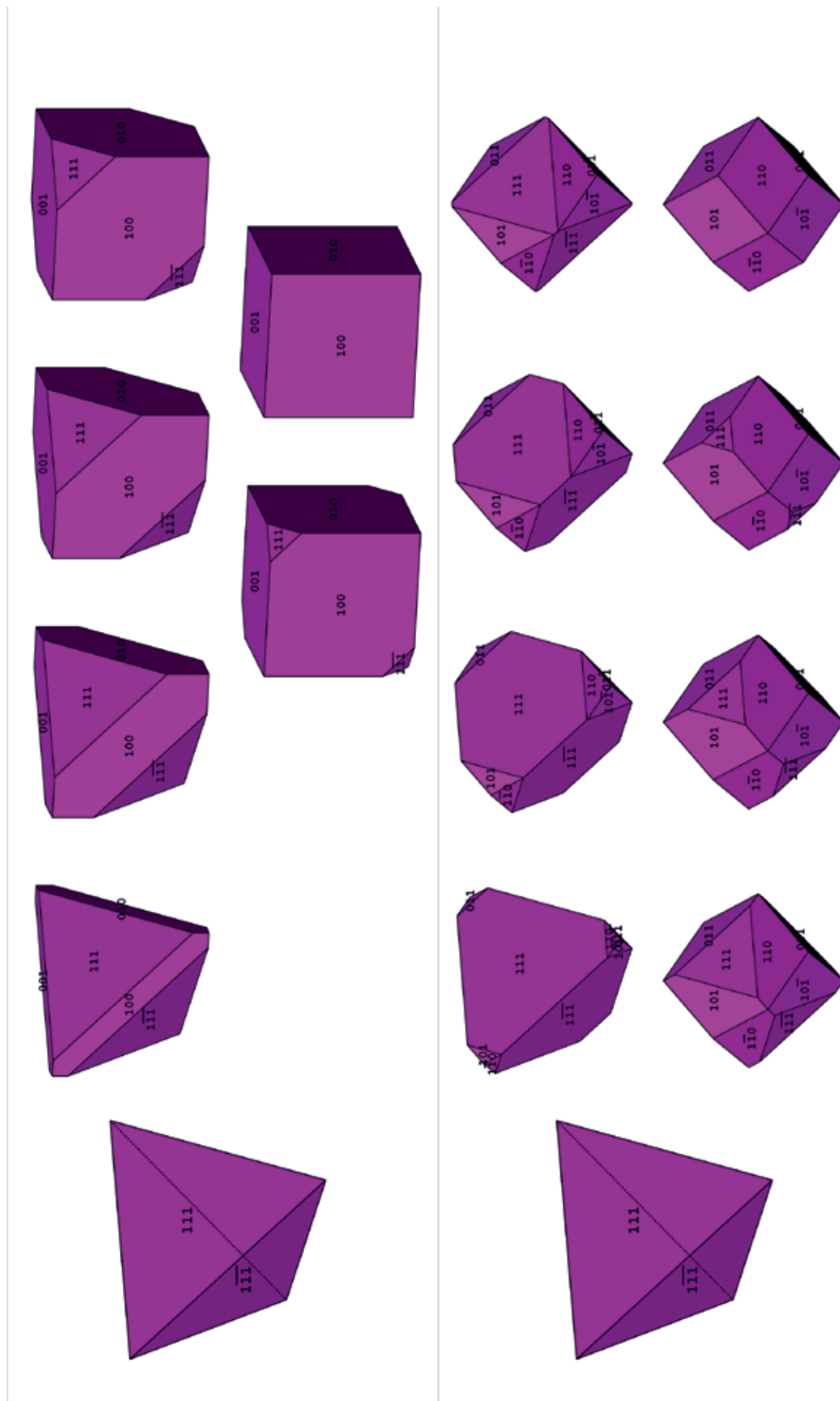
Combination of rhombic dodecahedron with octahedron and cube



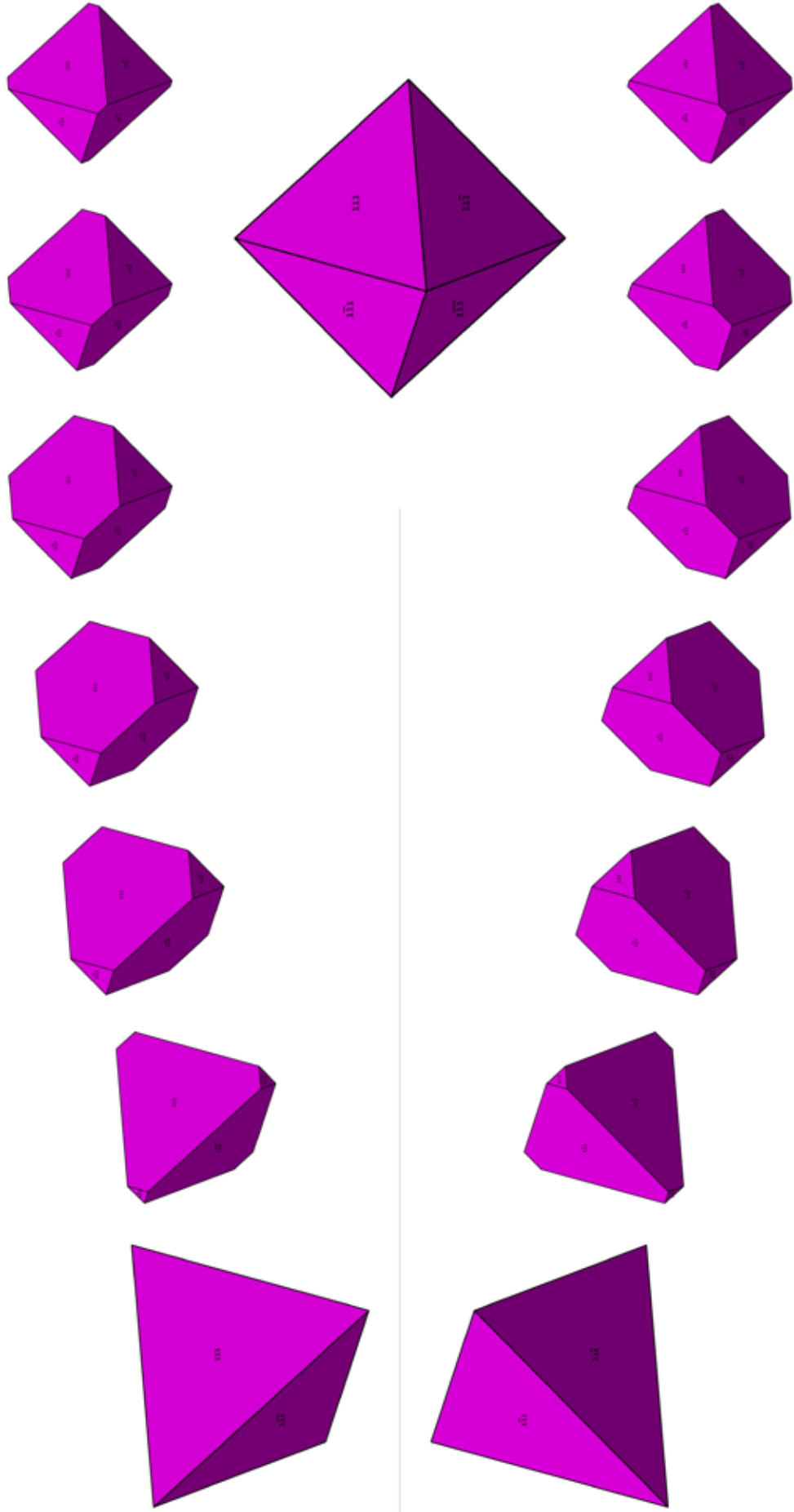
Kombination des Tetraeder mit Würfel und Rhombendodekaeder



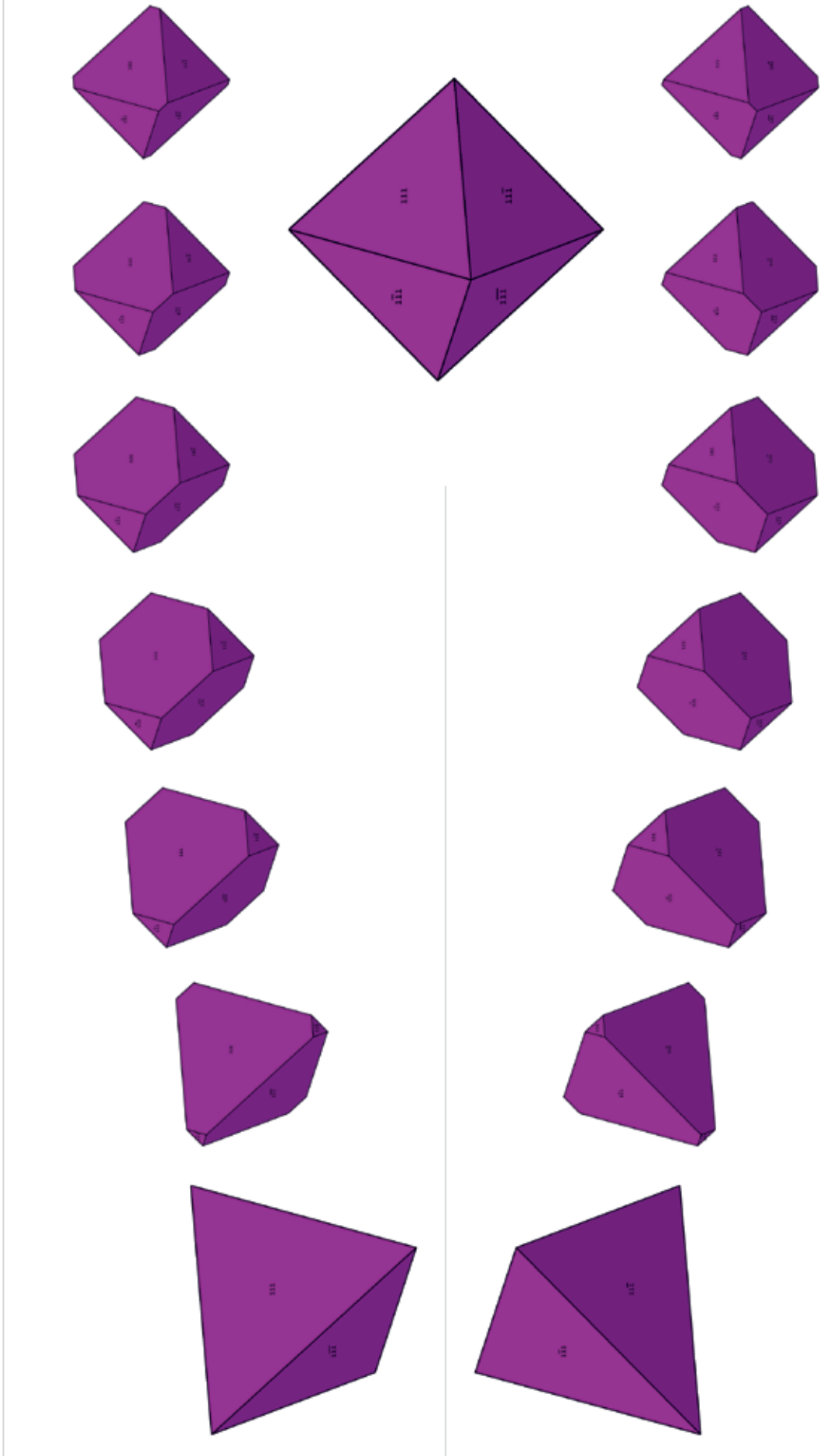
Combination of tetrahedron with cube and rhombic dodecahedron



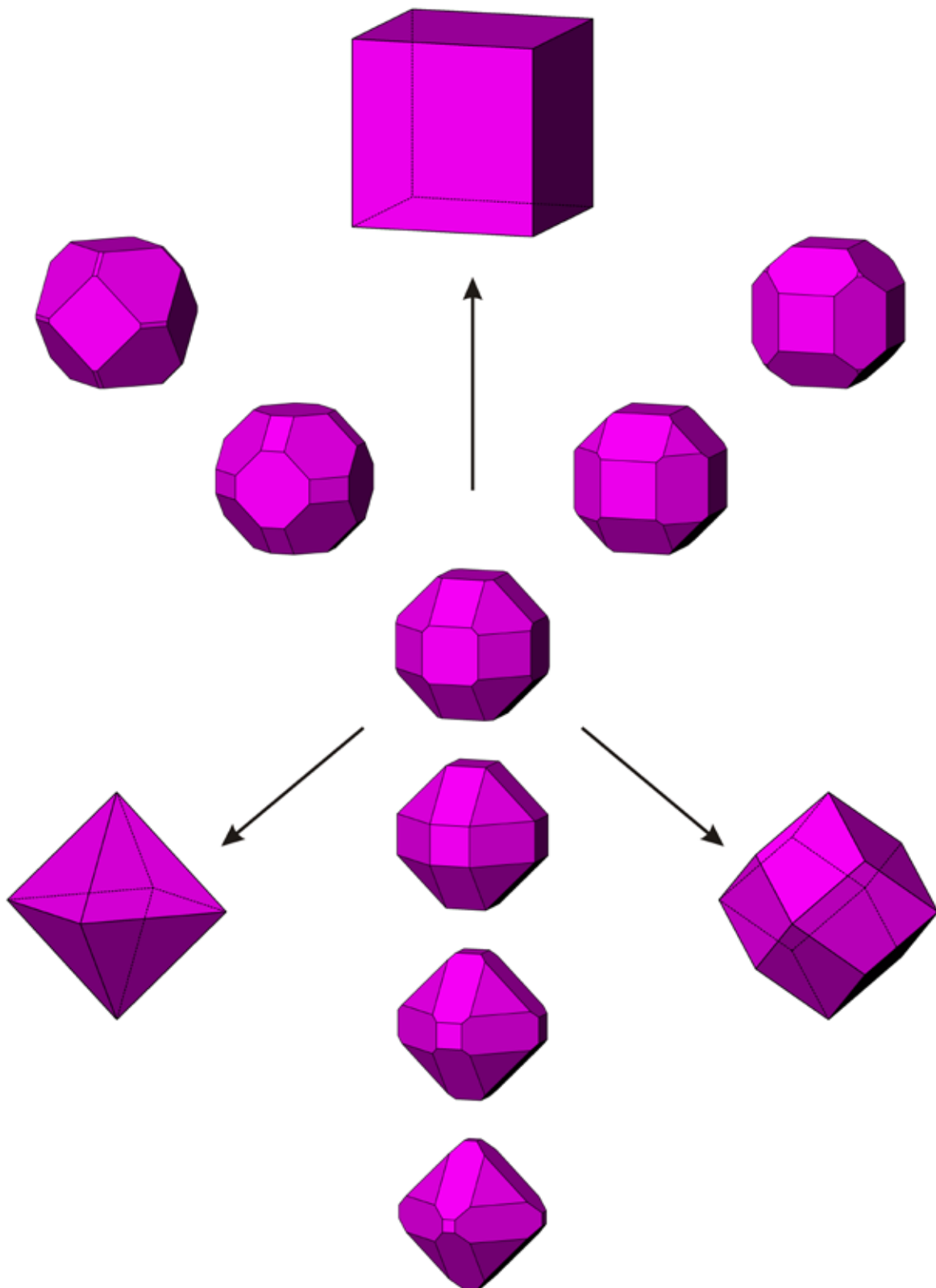
Kombination von positivem und negativem Tetraeder



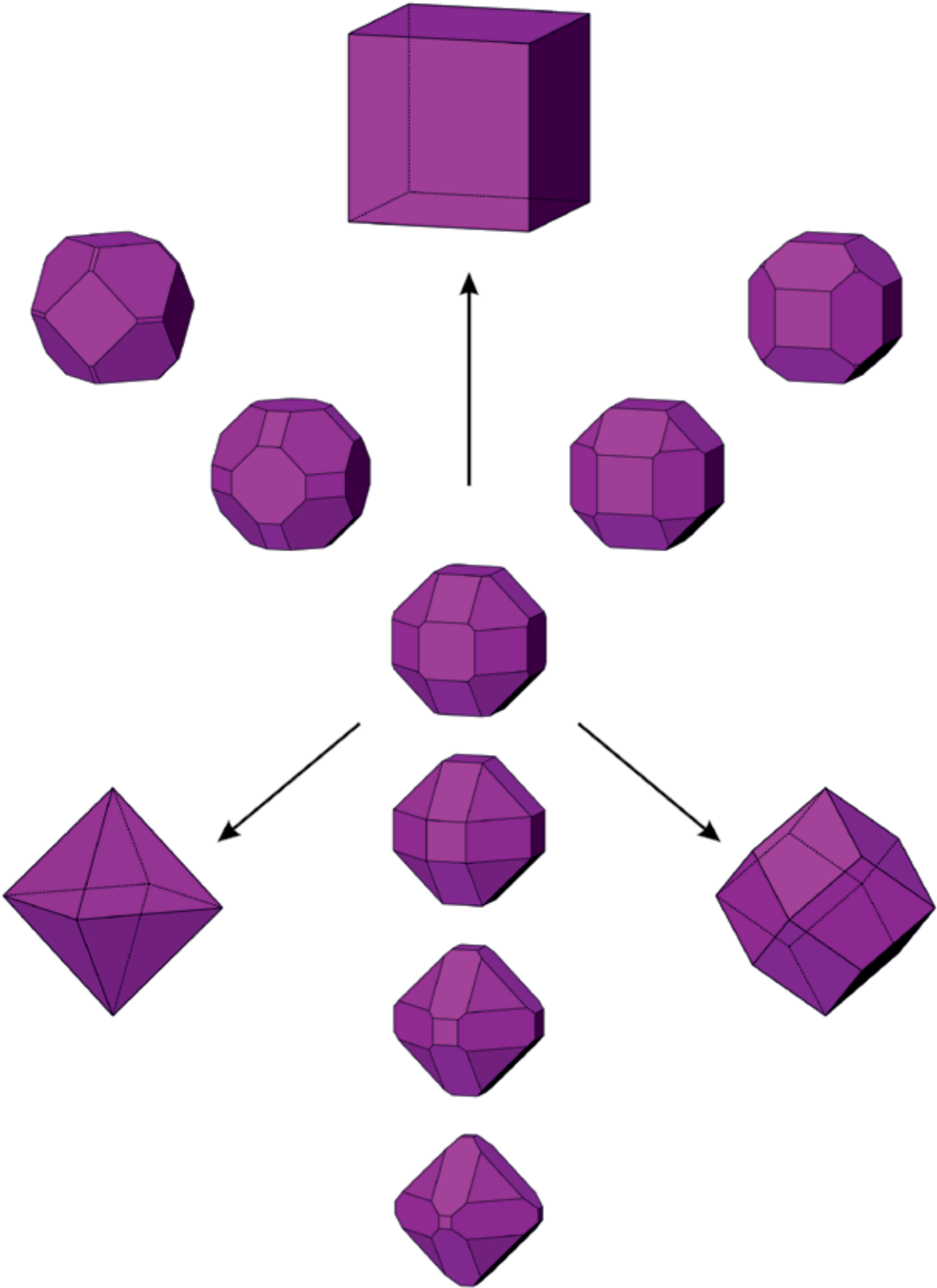
Combination of positive and negative tetrahedron



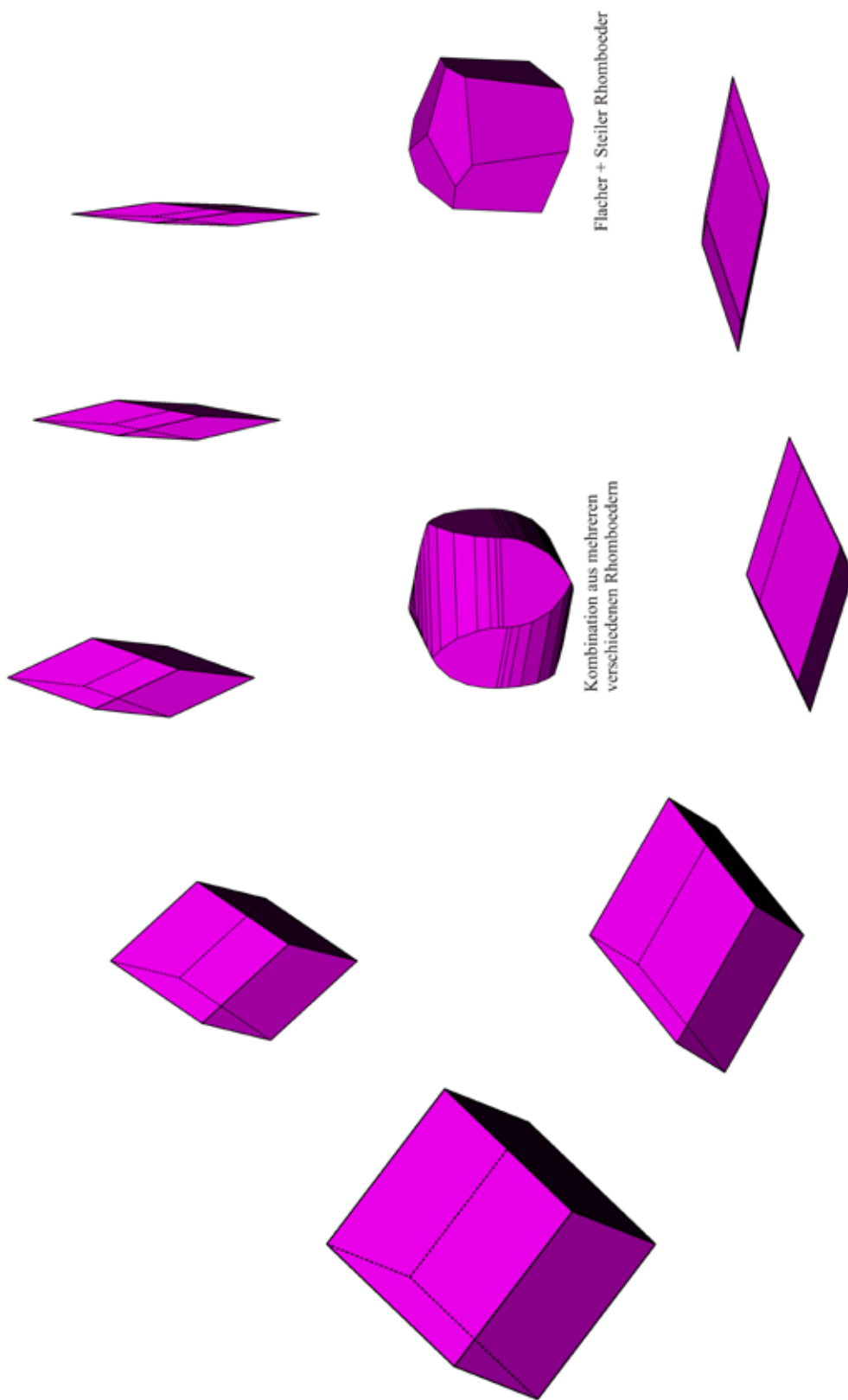
Kombinationen aus Würfel, Oktaeder und Rhombendodekaeder



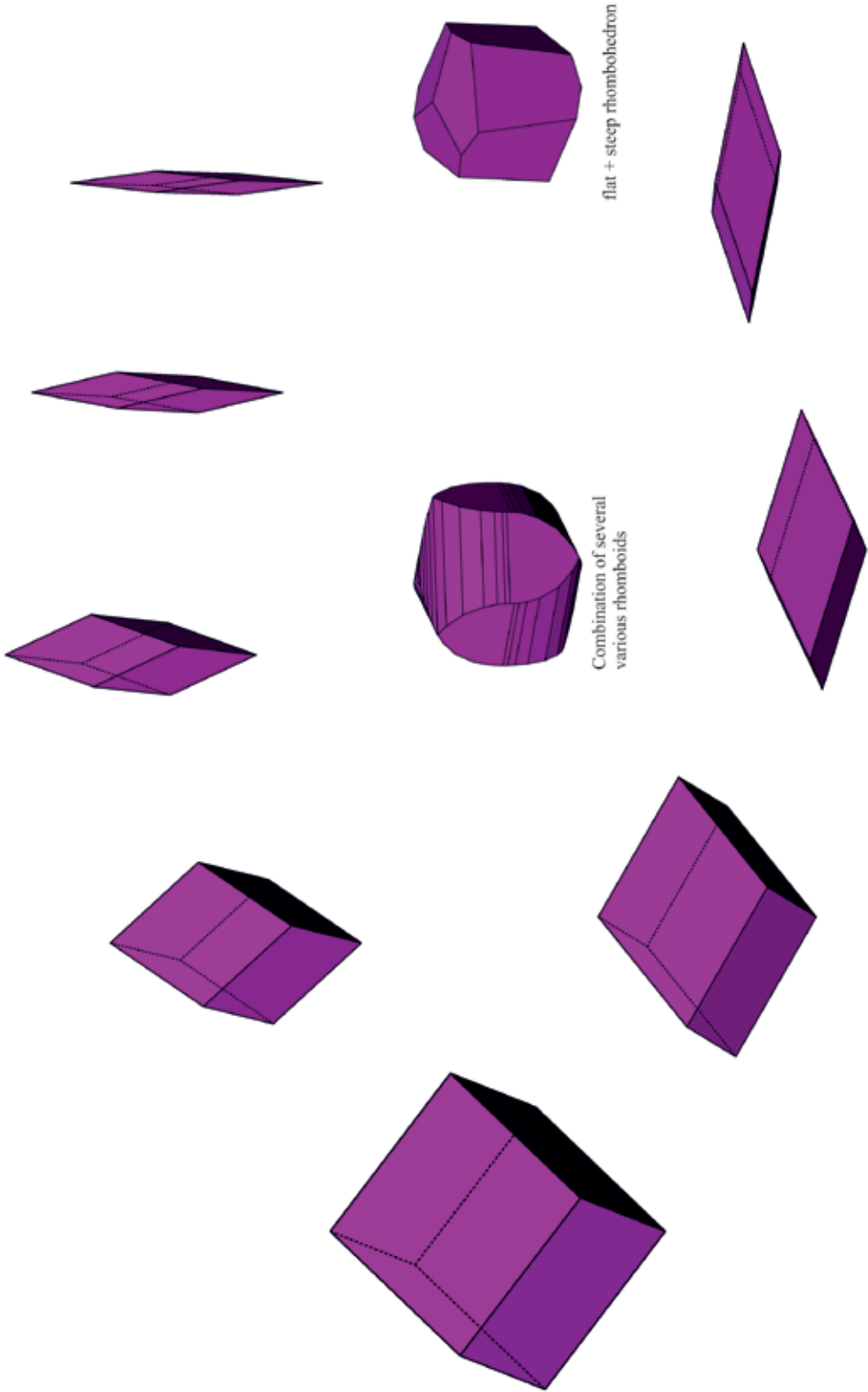
Combinations of cube, octahedron and rhombic dodecahedron



Verschieden steile und flache Rhomboeder

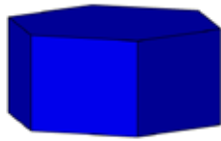


Different steep and flat rhombohedron

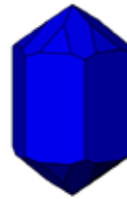


**Einige ausgewählte Formen
und Kombinationen
nach Mineralbeispielen**

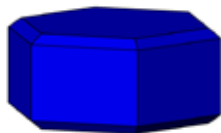
**Some selected crystal forms
and combinations
using some mineral examples**

Apatit:

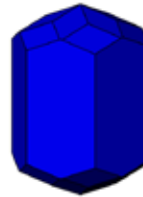
Hexagonales Prisma
Basispinakoid



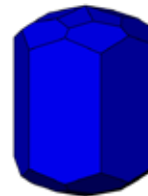
Hexagonales Prisma
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.



Hexagonales Prisma
Basispinakoid
Hex. Dipyramide I. St.

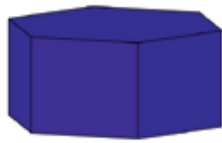


Hexagonales Prisma
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.



Hexagonales Prisma
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.

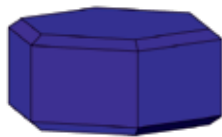
Chemische Zusammensetzung	: $\text{Ca}_5[(\text{PO}_4)_3(\text{F,Cl,OH})]$
Kristallsystem	: Hexagonal, $6/m$
Elementarzelle	: $a_0=9.38 \text{ \AA}$, $c_0=6.89 \text{ \AA}$
Raumgruppe	: $P 63/m$
Ausbildung	: Derb, eingewachsen, körnig, kryptokristallin, krustenartig, xx säulig, dicktafelig, nadelig

Apatite:

Hexagonal prism
Basispinacoid



Hexagonal prism
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.



Hexagonal prism
Basispinacoid
Hex. dipyramid I. pos.

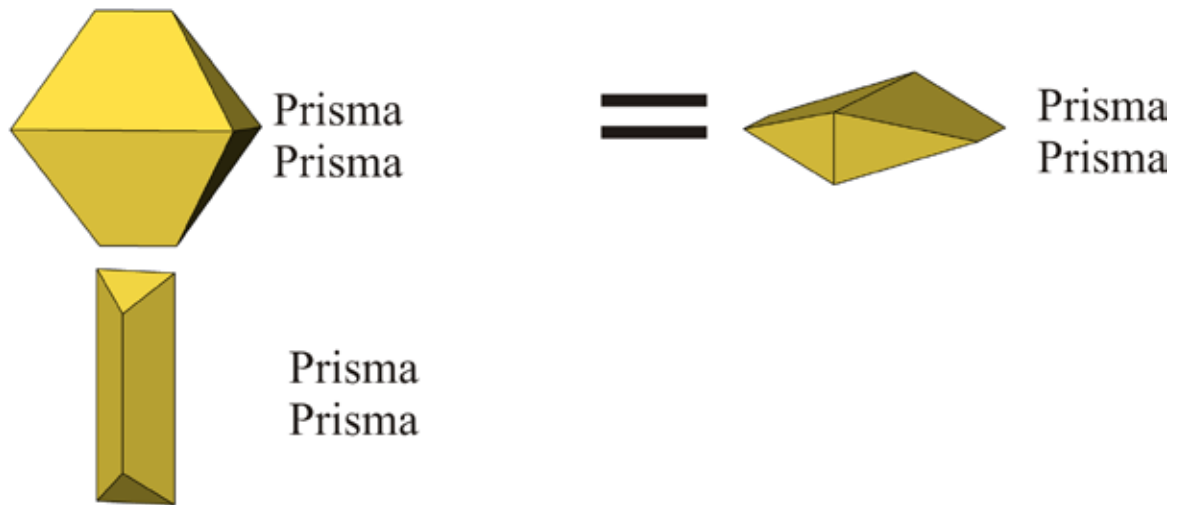


Hexagonal prism
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.



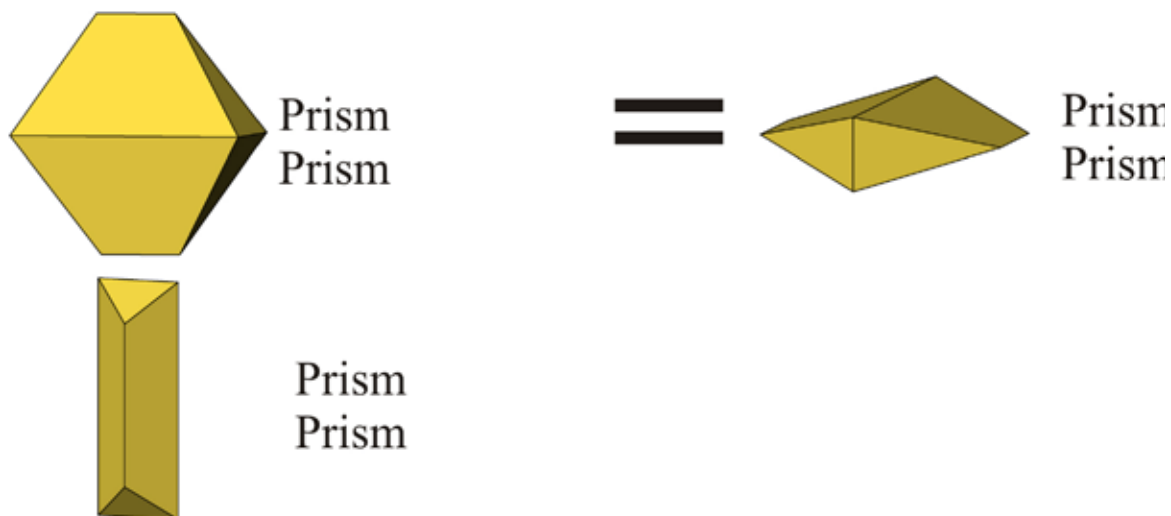
Hexagonal prism
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.

Chemical composition	: $\text{Ca}_5[(\text{PO}_4)_3(\text{F}, \text{Cl}, \text{OH})]$
Crystal system	: Hexagonal, 6/m
Elementary cell	: $a_0=9.38 \text{ \AA}$, $c_0=6.89 \text{ \AA}$
Space group	: P 63/m
Forms	: Massive, intergrown, granular, cryptocrystalline, crusts, xx columnar, tabular, needles

Arsenopyrit:

Chemische Zusammensetzung
 Kristallsystem
 Elementarzelle
 Raumgruppe
 Ausbildung

: FeAsS
 : Monoklin, pseudorhombisch
 : $a_0 = 5.74 \text{ \AA}$, $b_0 = 5.68 \text{ \AA}$, $c_0 = 5.79 \text{ \AA}$, $\beta = 112.17^\circ$
 : P 21/c
 : idiomorphe Kristalle, körnig

Arsenopyrite:

Chemical composition	: FeAsS
Crystal system	: Monoclinic, pseudorhombic
Elementary cell	: $a_0 = 5.74 \text{ \AA}$, $b_0 = 5.68 \text{ \AA}$, $c_0 = 5.79 \text{ \AA}$, $\beta = 112.17^\circ$
Space group	: P 21/c
Forms	: idiomorphic crystals, granular

Augit:

Pinakoid
Prisma
Pinakoid
Pinakoid
Prisma



Pinakoid
Prisma
Pinakoid
Pinakoid
Prisma

Chemische Zusammensetzung	: (Ca, Mg, Fe, Al, Ti) [(Si, Al) ₂ O ₆]
Kristallsystem	: Monoklin
Elementarzelle	: $a_0 = 9.69 \text{ \AA}$, $b_0 = 8.84 \text{ \AA}$, $c_0 = 5.28 \text{ \AA}$, $\beta = 106.3^\circ$
Raumgruppe	: C 2/c
Ausbildung	: Kurzprismatische bis tafelige xx, achteckige Querschnitt, oft zonar aufgebaut

Augite:

Pinacoid
Prism
Pinacoid
Pinacoid
Prism

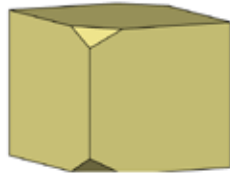


Pinacoid
Prism
Pinacoid
Pinacoid
Prism

Chemical composition	: (Ca, Mg, Fe, Al, Ti) [(Si, Al) ₂ O ₆]
Crystal system	: monoclinic
Elementary cell	: a ₀ = 9.69 Å, b ₀ = 8.84 Å, c ₀ = 5.28 Å, β = 106.3°
Space group	: C 2/c
Forms	: short prismatic to tabular xx, 8-cornered cross section, often zonal

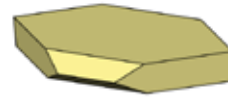
Baryt:

Basispinakoid
Prisma

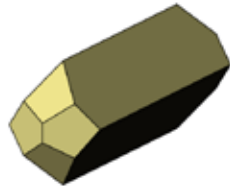


Basispinakoid
Prisma
Prisma

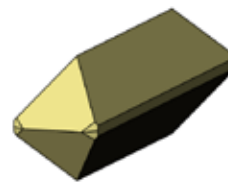
=



Basispinakoid
Prisma
Prisma



Prisma
Prisma
Prisma



Prisma
Prisma
Pinakoid
Dipyramide

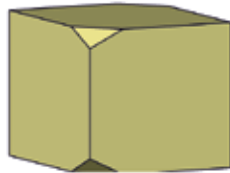
Chemische Zusammensetzung
Kristallsystem
Elementarzelle
Raumgruppe
Ausbildung

: BaSO₄
: Rhombisch, 2/m 2/m 2/m
: a₀= 8.88Å, b₀= 5.45Å, c₀= 7.15Å
: P b n m
: Dicht, feinkristallin, grobspätig, blättrig, xx meist tafelig oder
meißelförmig

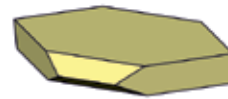
Baryte:



Basispinacoid
Prism



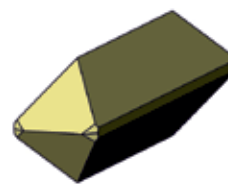
Basispinacoid
Prism
Prism



Basispinacoid
Prism
Prism



Prism
Prism
Prism



Prism
Prism
Pinacoid
Dipyrmaid

Chemical composition	: BaSO ₄
Crystal system	: Rhombic, 2/m 2/m 2/m
Elementary cell	: a ₀ = 8.88Å, b ₀ = 5.45Å, c ₀ = 7.15Å
Space group	: P b n m
Forms	: compact, fine crystalline, spary, laminated, xx mostly tabular or chisel form

Beryll:Hexagonales Prisma
BasispinakoidHexagonales Prisma
BasispinakoidHexagonales Prisma
Basispinakoid
Hex. Dipyramide II. St.Hexagonales Prisma
Basispinakoid
Hex. Dipyramide I. St.
Hex. Dipyramide II. St.

Chemische Zusammensetzung	: $\text{Be}_3\text{Al}_2 [\text{Si}_6\text{O}_{18}]$
Kristallsystem	: hexagonal
Elementarzelle	: $a_0 = 9.22\text{\AA}$, $c_0 = 9.19\text{\AA}$
Raumgruppe	: $P 6/m m c$
Ausbildung	: säulig, stengelig, dominierend Prisma (1 0 - 1 0) und Pinakoid (0 0 0 1)

Beryl:Hexagonal prism
BasispinacoidHexagonal prism
BasispinacoidHexagonal prism
Basispinacoid
Hex. dipyramid II. pos.Hexagonal prisma
Basispinacoid
Hex. dipyramid I. pos.
Hex. dipyramid II. pos.

Chemical composition	: $\text{Be}_3\text{Al}_2[\text{Si}_6\text{O}_{18}]$
Crystal system	: hexagonal
Elementary cell	: $a_0 = 9.22\text{\AA}$, $c_0 = 9.19\text{\AA}$
Space group	: $P 6/m m c$
Forms	: columnar, columnar, dominating Prism (1 0 – 1 0) and Pinakoid (0 0 0 1)

Biotit:

Basispinakoid
Prisma
Pinakoid



Basispinakoid
Prisma
Pinakoid



Basispinakoid
Prisma
Pinakoid
Prisma

Chemische Zusammensetzung	: K (Mg, Fe) ₃ [(OH) ₂ / AlSi ₃ O ₁₀]
Kristallsystem	: Monoklin, trioktaedrisches Dreischichtsilikat
Elementarzelle	: a ₀ = 5.35 Å, b ₀ = 9.26 Å, c ₀ = 10.23 Å, β = 100.3°
Raumgruppe	: C 2/m
Ausbildung	: Tafelig, plattig, schuppig

Biotite:

Basispinacoid
Prism
Pinacoid

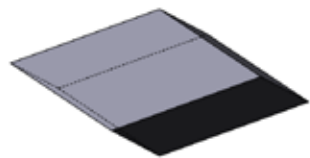


Basispinacoid
Prism
Pinacoid



Basispinacoid
Prism
Pinacoid
Prism

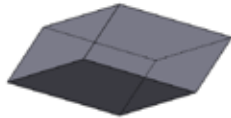
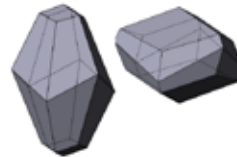
Chemical composition	: K (Mg, Fe) ₃ [(OH) ₂ / AlSi ₃ O ₁₀]
Crystal system	: monoclinic, trioktaedric 3-layered silicate
Elementary cell	: a ₀ = 5.35 Å, b ₀ = 9.26 Å, c ₀ = 10.23 Å, β = 100.3°
Space group	: C 2/m
Forms	: tabular, bladed, flaky

Calcit:

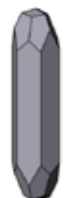
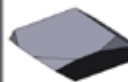
positiver Rhomboeder



Skalenoeder

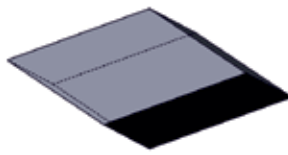
negativer
RhomboederSkalenoeder
(+) Rhomboeder

negativer Rhomboeder

Prisma
(-) Rhomboederpos. Rhomboeder
Prisma
SkalenoederPrisma
(+) Rhomboeder

Chemische Zusammensetzung	: CaCO_3
Kristallsystem	: Trigonal, $-3m$
Elementarzelle	: $a_0 = 4.99, c_0 = 17.06$
Raumgruppe	: R-3c
Ausbildung	: Gesteinsbildend, dicht, oolithisch, derb grobspätig, sehr verschieden (formenreiches Mineral!), z.B. Rhomboeder und Skalenoeder, säulige xx u.a

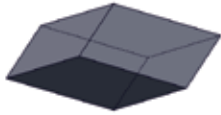
Calcite:



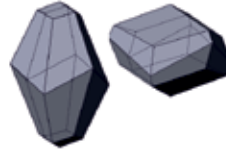
positive Rhombohedron



Scalenohedron



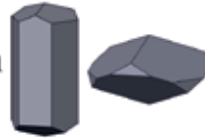
negative Rhombohedron



Skalenoeder (+) Rhombohedron



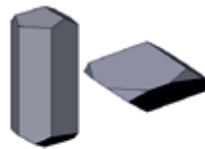
negative Rhombohedron



Prism (-) Rhombohedron

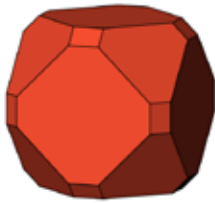


pos. Rhombohedron
Prism
Scalenohedron



Prism (+) Rhombohedron

Chemical composition : CaCO_3
 Crystal system : trigonal, -3m
 Elementary cell : $a_0 = 4.99, c_0 = 17.06$
 Space group : R-3c
 Forms : Rock forming mineral, compact, oolitic, massive spaty, xx very different xx (different shapes of this Mineral!), example Rhombohedra and Scalenohedra, columnar xx a.o.

Cuprit:

Hexaeder
Oktaeder
Rhombendodekaeder

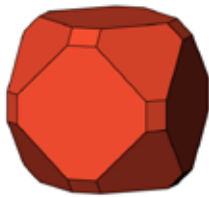


Hexaeder
Oktaeder
Rhombendodekaeder



Hexaeder
Oktaeder
Pentagonikositetraeder

Chemische Zusammensetzung	: Cu ₂ O
Kristallsystem	: Kubisch
Elementarzelle	: a ₀ =4.27 Å
Raumgruppe	: Pn3m
Ausbildung	: Derb. eingesprengt, Würfel, Oktaeder, nadelige

Cuprite:

Hexahedron
Octahedron
Rhombic dodecahedron

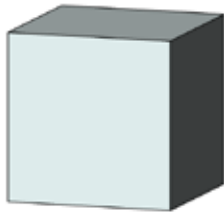


Hexahedron
Octahedron
Rhombic dodecahedron



Hexahedron
Octahedron
Pentagonal icositetrahedron

Chemical composition	: Cu_2O
Crystal system	: cubic
Elementary cell	: $a_0=4.27 \text{ \AA}$
Space group	: $\text{Pn}3\text{m}$
Forms	: passive, disseminated, cubes, octahedra, needle x

Diamant:

Hexaeder



Oktaeder



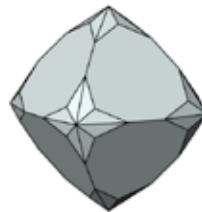
Rhombendodekaeder



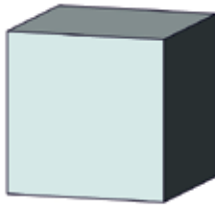
Tetrakishexaeder



Hexakisoktaeder

Oktaeder
Hexakisoktaeder

Chemische Zusammensetzung	: C
Kristallsystem	: Kubisch
Elementarzelle	: $a_0 = 3.559 \text{ \AA}$
Raumgruppe	: F d3m
Ausbildung	: Oktaeder, Rhombendodekaeder, Würfel, meist abgerundet

Diamond:

Hexahedron



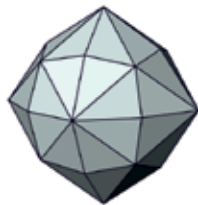
Octahedron



Rhombic dodecahedron



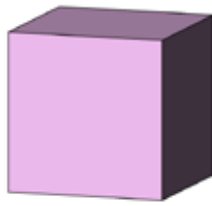
Tetrakis hexaeder



Hexakis octahedron

Octahedron
Hexakis octahedron

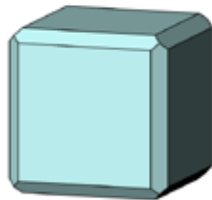
Chemical composition	: C
Crystal system	: cubic
Elementary cell	: $a_0 = 3.559 \text{ \AA}$
Space group	: F d3m
Forms	: octahedra, rhombic dodecaedron, cubes, xx mostly rounded

Fluorit:

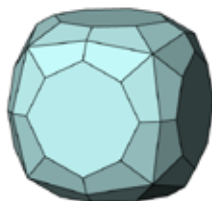
Hexaeder



Oktaeder

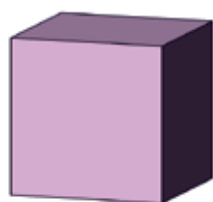
Hexaeder
OktaederOktaeder
RhombendodekaederHexaeder
Rhombendodekaeder

Tetrakisohedraeder

Hexaeder
Hexakisoktaeder

Chemische Zusammensetzung
 Kristallsystem
 Elementarzelle
 Raumgruppe
 Ausbildung

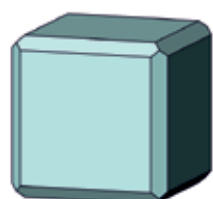
: CaF_2
 : Kubisch
 : $a_0 = 5.46 \text{ \AA}$
 : $F m\bar{3}m$
 : Derb, grobspätig, Würfel, Oktaeder,
 Rhombendodekaeder, oft grobkristallin

Fluorite:

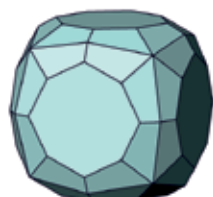
Hexahedron



Octahedron

Hexahedron
OctahedronOctahedron
Rhombic dodecahedronHexahedron
Rhombic dodecahedron

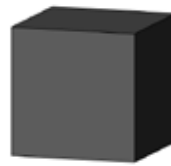
Tetrakis hexahedron

Hexahedron
Hexakis octahedron

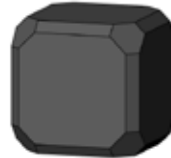
Chemical composition	: CaF ₂
Crystal system	: cubic
Elementary cell	: a ₀ = 5.46 Å
Space group	: F m3m
Forms	: massive, grobspätig, cubes, octahedra, Rhombic dodecahedron, often cristalline.

Galenit:

Oktaeder



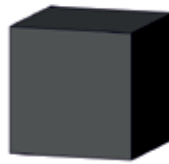
Würfel

div.
Kub-OktaederWürfel
Oktaeder
Rhombendodekaeder

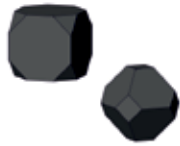
Chemische Zusammensetzung	: PbS
Kristallsystem	: Kubisch, NaCl-Gitter
Elementarzelle	: $a_0 = 5.94 \text{ \AA}$
Raumgruppe	: F m 3 m
Ausbildung	: Derb, eingesprengt, Würfel, Oktaeder, Rhombendodekaeder. oft Kombinationen

Galenite:

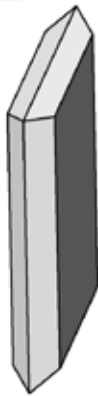
Octahedron



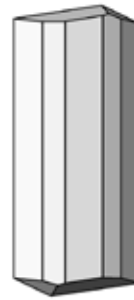
Cube

various
Cube-OctahedronCube
Octahedron
Rhombic dodecahedron

Chemical composition : PbS
 Crystal system : cubic, NaCl-lattice
 Elementary cell : $a_0 = 5.94 \text{ \AA}$
 Space group : F m 3 m
 Forms : massive, disseminated, cubes, octahedra,
 rhombic dodecahedron, often combinations

Gips:

Prisma
Prisma
Pinakoid



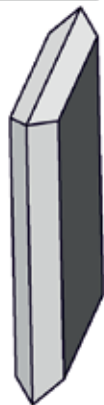
Prisma
Prisma
Prisma
Prisma
Pinakoid



Prisma
Basispinakoid
Pinakoid

Chemische Zusammensetzung	: $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$
Kristallsystem	: Monoklin, $2/m$
Elementarzelle	: $a_0 = 5.68\text{\AA}$, $b_0 = 15.18\text{\AA}$, $c_0 = 6.89\text{\AA}$, $\beta = 113.83$
Raumgruppe	: $A 2/a$
Ausbildung	: Dicht, feinkörnig, faserig, xx nadelig, primatisch, tafelig, oft Schwalbenschwanzzwillinge

Gypsum:



Prism
Prism
Pinacoid



Prism
Prism
Prism
Prism
Pinacoid

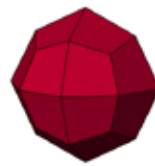


Prism
Basispinacoid
Pinacoid

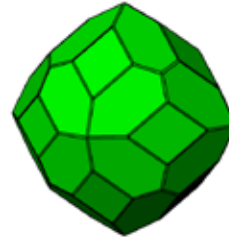
Chemical composition	: $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$
Crystal system	: monoclinic, $2/m$
Elementary cell	: $a_0 = 5.68\text{\AA}$, $b_0 = 15.18\text{\AA}$, $c_0 = 6.89\text{\AA}$, $\beta = 113.83$
Space group	: $A 2/a$
Forms	: compact, finely granular, fibrous, xx needles, prismatic, tabular, often dove tail twins

Granat:

Rhombendodekaeder



Deltoidikositetraeder

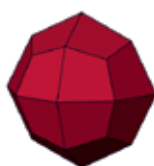
Rhombendodekaeder
DeltoidikositetraederRhombendodekaeder
Deltoidikositetraeder
Tetrakisheptaeder
Hexakisoktaeder
Trisoktaeder

Chemische Zusammensetzung	: $X_3Y_2[SiO_4]_3$ X: Mg, Fe ²⁺ , Mn ²⁺ , Ca (8er Koordinat.) Y: Al, Fe ³⁺ , Cr ³⁺ , V ³⁺ (6er Koordinat) Für (SiO ₄) auch (AlO ₄) oder (OH)
Kristallsystem	: Kubisch
Ausbildung	: Rhombendodekaeder (110), Deltoidikositetraeder (211)

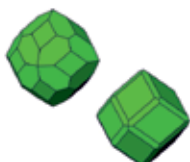
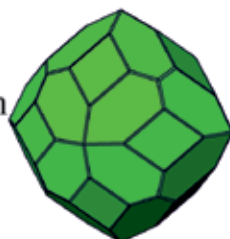
„Pyralspite“Pyrop: Mg₃Al₂[SiO₄]₃Almandin: Fe₃Al₂[SiO₄]₃Spreßartin: Mn₃Al₂[SiO₄]₃**„Ugrandite“**Uwarowit: Ca₃Cr₂[SiO₄]₃Grossular: Ca₃Al₂[SiO₄]₃Andradit: Ca₃Fe₂[SiO₄]₃

Garnet:

Rhombic dodecahedron



Deltoidal icositetrahedron

Rhombic dodecahedron
Deltoidal
icositetrahedronRhombic dodecahedron
Deltoidal icositetrahedron
Tetrakis hexahedron
Hexakis octahedron
Trisoctahedron

Chemical composition	: $X_3Y_2[SiO_4]_3$ X: Mg, Fe ²⁺ , Mn ²⁺ , Ca (8 coord.) Y: Al, Fe ³⁺ , Cr ³⁺ , V ³⁺ (6 coord.) (SiO ₄) replaced by (AlO ₄) or (OH)
Crystal system	: cubic
Forms	: Rhombic dodecahedron (110), Deltoidalicositetrahedron (211)

„Pyralspite“Pyrop: Mg₃Al₂[SiO₄]₃Almandin: Fe₃Al₂[SiO₄]₃Spessartine: Mn₃Al₂[SiO₄]₃**“Ugrandite”**Uwarowite: Ca₃Cr₂[SiO₄]₃Grossular: Ca₃Al₂[SiO₄]₃Andradite: Ca₃Fe₂[SiO₄]₃

Hämatit:

Rhomboeder
Basispinakoid



Rhomboeder
Basispinakoid



Dipyramide
Basispinakoid



Rhomboeder
Dipyramide



Rhomboeder
Dipyramide
Rhomboeder

Chemische Zusammensetzung	: Fe_2O_3
Kristallsystem	: Trigonal
Elementarzelle	: $a_0 = 5.03\text{\AA}$, $c_0 = 13.74\text{\AA}$
Raumgruppe	: R -3c
Ausbildung	: Erdig, schuppig, radialstrahlig, knollig, nierig, xx tafelig, plattig

Hematite:

Rhombohedron
Basispinacoid =



Rhombohedron
Basispinacoid



Dipyramid
Basispinacoid



Rhombohedron
Dipyramid



Rhombohedron
Dipyramid
Rhombohedron

Chemical composition	: Fe_2O_3
Crystal system	: trigonal
Elementary cell	: $a_0 = 5.03\text{\AA}$, $c_0 = 13.74\text{\AA}$
Space group	: R -3c
Forms	: earthy, flaky, radial, nodular, nierig, xx tabular, platy

Korund:

Prisma
Basispinakoid



Prisma
Basispinakoid
Dipyramide



Prisma
Basispinakoid
Dipyramide
Dipyramide
Rhomböeder

Chemische Zusammensetzung	: Al_2O_3
Kristallsystem	: Trigonal
Elementarzelle	: $a_0=4.75 \text{ \AA}$, $c_0=12.98 \text{ \AA}$
Raumgruppe	: R-3c
Ausbildung	: Eingesprengt, säulige bis tonnenförmig, meist mit Flächenstreifung

Corundum:

Prism
Basispinacoid

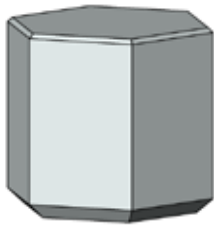


Prism
Basispinacoid
Dipyramid

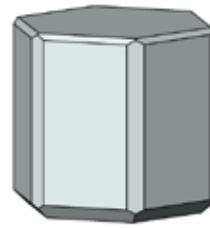


Prism
Basispinacoid
Dipyramid
Dipyramid
Rhombohedron

Chemical composition	: Al_2O_3
Crystal system	: Trigonal
Elementary cell	: $a_0=4.75 \text{ \AA}$, $c_0= 12.98 \text{ \AA}$
Space group	: R-3c
Forms	: disseminated, columnar to barrel like forms xx, mostly with streaking

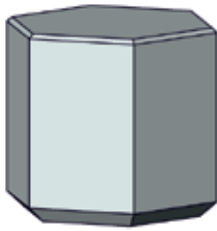
Nephelin:

Prisma
 Basispedion
 Basispedion
 Pyramide
 Pyramide

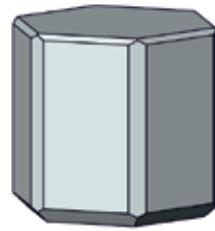


Prisma
 Basispedion
 Basispedion
 Pyramide
 Pyramide
 Prisma

Chemische Zusammensetzung	: $\text{KNa}_3[\text{Al SiO}_4]_4$
Kristallsystem	: Hexagonal
Elementarzelle	: $a_0 = 9.99 \text{ \AA}$, $c_0 = 8.37 \text{ \AA}$
Raumgruppe	: $P 6_3$
Ausbildung	: Derb. selten säulig

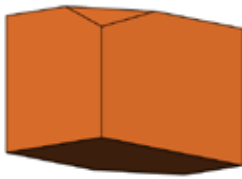
Nepheline:

Prism
Basispedia
Basispedia
Pyramid
Pyramid



Prism
Basispedia
Basispedia
Pyramid
Pyramid
Prism

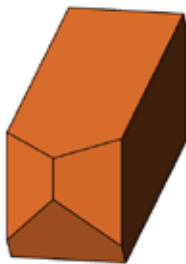
Chemical composition	: $\text{KNa}_3[\text{Al SiO}_4]_4$
Crystal system	: hexagonal
Elementary cell	: $a_0 = 9.99 \text{ \AA}$, $c_0 = 8.37 \text{ \AA}$
Space group	: $P 6_3$
Forms	: massive, rare column crystals xx

Orthoklas:

Prisma
Pinakoid
Pinakoid

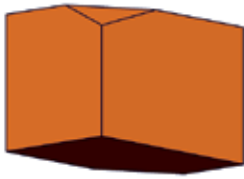


Prisma
Basispinakoid
Pinakoid
Pinakoid



Prisma
Pinakoid
Basispinakoid
Pinakoid

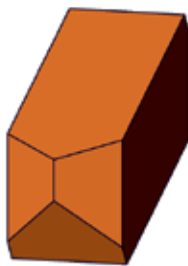
Chemische Zusammensetzung	: $K[Al Si_3 O_8]$ „Kalifeldspat“
Kristallsystem	: Monoklin
Elementarzelle	: $a_0 = 8.62 \text{ \AA}$, $b_0 = 12.99 \text{ \AA}$, $c_0 = 7.19 \text{ \AA}$, $\beta = 116.02^\circ$
Raumgruppe	: $C 2/m$
Ausbildung	: Gesteinsbildend, tafelig, prismatisch, häufig Zwillinge (Karlsbader, Manebacher, Bavenoer Gesetz)

Orthoclase:

Prism
Pinacoid
Pinacoid



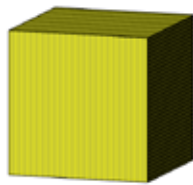
Prism
Basispinacoid
Pinacoid
Pinacoid



Prism
Pinacoid
Basispinacoid
Pinacoid

Chemical composition
Crystal system
Elementary cell
Space group
Forms

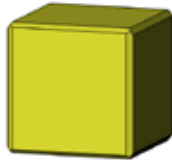
: $K[Al Si_3O_8]$ „Kalifeldspat“
: monoclinic
: $a_0 = 8.62 \text{ \AA}$, $b_0 = 12.99 \text{ \AA}$, $c_0 = 7.19 \text{ \AA}$, $\beta = 116.02^\circ$
: C 2/m
: rockforming mineral, xx tabular, prismatic, common
twins (Karlsbader, Manebacher, Bavenoer twin law)

Pyrit:

Würfel (Hexaeder) mit
Flächenstreifung



Pentagondodekaeder



Pentagondodekaeder
Würfel (Hexaeder)



Pentagondodekaeder
Würfel (Hexaeder)



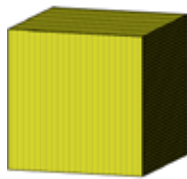
Pentagondodekaeder
Oktaeder
= "Pseudo-Ikosaeder"



Pentagondodekaeder
Oktaeder

Chemische Zusammensetzung	: FeS ₂
Kristallsystem	: Kubisch; Struktur: ähnlich NaCl, S ₂ -Hanteln liegen parallel (1 1 1)
Elementarzelle	: a ₀ = 5.42 Å
Raumgruppe	: P a 3
Ausbildung	: Derb, eingesprengt, Würfel (oft mit Flächenstreifung), Oktaeder, Pentagondodekaeder

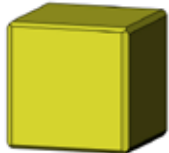
Pyrite:



Cube (hexahedron) with surface stripping



Pentagonal dodecahedron



Pentagonal dodecahedron
Cube (hexahedron)



Pentagonal dodecahedron
Cube (hexahedron)

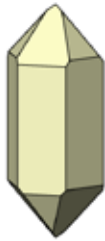


Pentagonal dodecahedron
Octahedron
= "Pseudo-icosahedron"

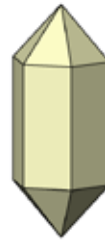


Pentagonal dodecahedron
Octahedron

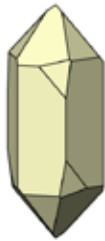
Chemical composition	: FeS ₂
Crystal system	: cubic Structure; similar to NaCl, S ₂ -barbells parallel (1 1 1)
Elementary cell	: a ₀ = 5.42 Å
Space group	: P a 3
Forms	: massive, disseminated, cubes, (often with plane striations), octahedron, pentagonal dodecahedron

Quarz:

Hexagonales Prisma
pos. Rhomboeder
neg. Rhomboeder



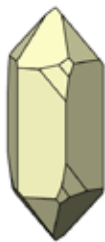
Hexagonales Prisma
Hex. Dipyramide
Hochquarz!
(andere Symetrie)



Hexagonales Prisma
pos. Rhomboeder
neg. Rhomboeder
rechter Trapezoeder



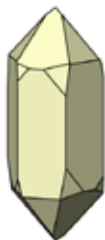
Hexagonales Prisma
pos. Rhomboeder
neg. Rhomboeder
linker Trapezoeder



Hexagonales Prisma
pos. Rhomboeder
neg. Rhomboeder
rechter Trapezoeder
rechte Dipyramide



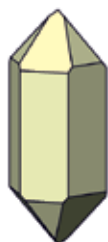
Hexagonales Prisma
pos. Rhomboeder
neg. Rhomboeder
linker Trapezoeder
linke Dipyramide



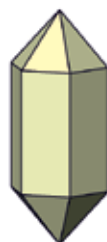
Hexagonales Prisma
pos. Rhomboeder
neg. Rhomboeder
"linker"
"rechter" Trapezoeder

Chemische Zusammensetzung	: SiO ₂
Kristallsystem	: Trigonal
Elementarzelle	: a ₀ = 4.91 Å, c ₀ = 5.40 Å
Raumgruppe	: P 3 ₁ 2 1, P 3 ₂ 2 1
Ausbildung	: derb. schöne Kristalle. z.T. wasserklar

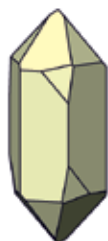
Quartz:



Hexagonal prism
pos. rhombohedron
neg. rhombohedron



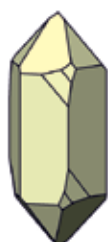
Hexagonal prism
Hex. dipyramid
high quartz!
(other symmetry)



Hexagonal prism
pos. rhombohedron
neg. rhombohedron
right trapezohedron



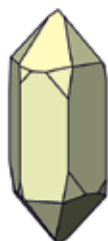
Hexagonal prism
pos. rhombohedron
neg. rhombohedron
left trapezohedron



Hexagonal prism
pos. rhombohedron
neg. rhombohedron
right trapezohedron
right dipyramid



Hexagonal prism
pos. rhombohedron
neg. rhombohedron
left trapezohedron
left dipyramid

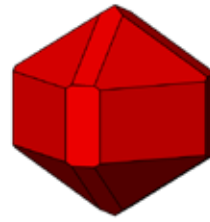


Hexagonal prism
pos. rhombohedron
neg. rhombohedron
"left" Trapezohedron
"right"

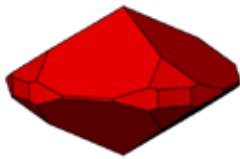
Chemical composition	: SiO ₂
Crystal system	: trigonal
Elementary cell	: a ₀ = 4.91 Å, c ₀ = 5.40 Å
Space group	: P 3 ₁ 2 1, P 3 ₂ 2 1
Forms	: massive, nice crystals, partially clear transparent

Rutil:

Prisma
Prisma
Dipyramide



Prisma
Prisma
Dipyramide
Dipyramide

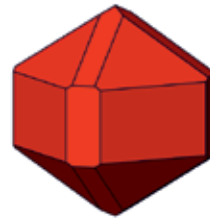


Prisma
Prisma
Dipyramide
Dipyramide
ditetragonales Prisma

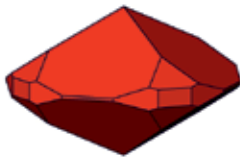
Chemische Zusammensetzung	: TiO_2
Kristallsystem	: Tetragonal
Elementarzelle	: $a_0=4.594\text{\AA}$, $c_0=2.958\text{\AA}$
Raumgruppe	: $P 4/ m n m$
Ausbildung	: Derb, nadelige, stengelig, häufig Verzwilligung (Winkel 120°)

Rutile:

Prism
Prism
Dipyramid



Prism
Prism
Dipyramid
Dipyramid

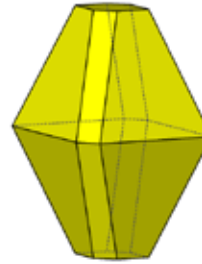


Prism
Prism
Dipyramid
Dipyramid
Ditetragonal prism

Chemical composition	: TiO ₂
Crystal system	: tetragonal
Elementary cell	: a ₀ =4.594Å, c ₀ =2.958Å
Space group	: P 4/ m n m
Forms	: massive, nadelige, columnar, common twins (angle 120°), gratlike aggregates

Schwefel:

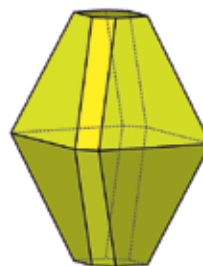
Dipyramide

Dipyramide
Basispinakoid
PrismaDipyramide
Dipyramide
Basispinakoid
Prisma**Kopfbild**
Dipyramide
Dipyramide
Basispinakoid
Prisma

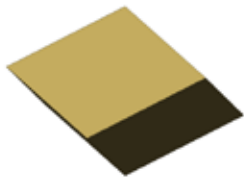
Chemische Zusammensetzung	: S
Kristallsystem	: orthorhombisch
Elementarzelle	: $a_0 = 10.45 \text{ \AA}$, $b_0 = 12.85 \text{ \AA}$, $c_0 = 24.46 \text{ \AA}$
Raumgruppe	: F ddd
Besonderheiten	: bis 95° orthorhombisch, ab 95° monoklin, Schmelzpunkt $119,2^\circ \text{C}$ Ringförmige S_8 -Moleküle, zwischen den Molekülen nur Van-der-Waalsche Bindung
Ausbildung	: Derbe, dichte, erdige Aggregate, pyramidale und sphenoidische xx

Sulfur:

Dipyramid

Dipyramid
Basispinacoid
PrismDipyramid
Dipyramid
Basispinacoid
Prism**Headshot**
Dipyramid
Dipyramid
Basispinacoid
Prism

Chemical composition	: S
Crystal system	: orthorhombic
Elementary cell	: $a_0 = 10.45 \text{ \AA}$, $b_0 = 12.85 \text{ \AA}$, $c_0 = 24.46 \text{ \AA}$
Space group	: F ddd
Specials	: up to 95° orthorhombic, ab 95° monoclinic, Melting point $119,2^\circ \text{C}$ Ringlike S_8 -Molecules, between molecules only Van-der-Waalsche Bonding
Forms	: massive, compact, earthy aggregates, pyramidal and sphenoidic xx

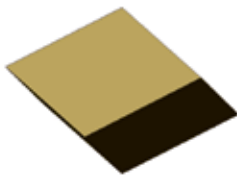
Siderit:

Rhomboeder

Rhomboeder
SkalenoederRhomboeder
Skalenoeder
Skalenoeder
RhomboiderRhomboeder
BasispinakoidRhomboeder
Basispinakoid
Rhomboider
Skalenoeder

Chemische Zusammensetzung	: FeCO_3
Kristallsystem	: Trigonal, $-3m$
Elementarzelle	: $a_0 = 4.72, c_0 = 15.46$
Raumgruppe	: $R - 3c$
Ausbildung	: Feinkörnig bis grobspätig, rhomboedrische xx

Siderite:



Rhombohedron



Rhombohedron
Scalenohedron



Rhombohedron
Scalenohedron
Scalenohedron
Rhombohedron



Rhombohedron
Basispinacoid



Rhombohedron
Basispinacoid
Rhombohedron
Scalenohedron

Chemical composition	: FeCO ₃
Crystal system	: trigonal, -3m
Elementary cell	: a ₀ = 4.72, c ₀ = 15.46
Space group	: R -3c
Forms	: Fine granular, spaty, rhomboedric xx

Staurolith:

Prisma
Basispinakoid
Pinakoid
Prisma



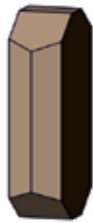
Prisma
Basispinakoid
Pinakoid
Prisma



Prisma
Basispinakoid
Pinakoid
Prisma

Chemische Zusammensetzung	: $(\text{Fe,Mg})_2\text{Al}_9(\text{Si,Al})_4\text{O}_{20}(\text{O,OH})_2$
Kristallsystem	: Monoklin
Elementarzelle	: $a_0 = 7.863 \text{ \AA}$, $b_0 = 16.61 \text{ \AA}$, $c_0 = 5.65 \text{ \AA}$, $\beta = 90-93^\circ$
Raumgruppe	: C 2/m
Ausbildung	: Gesteinsbildend, tafelig, prismatisch, häufig

Staurolite:



Prism
Basispinacoid
Pinacoid
Prism



Prism
Basispinacoid
Pinacoid
Prism



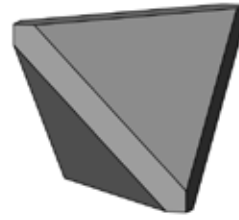
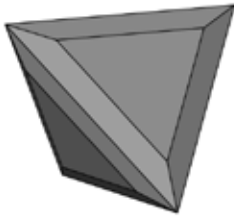
Prism
Basispinacoid
Pinacoid
Prism

Chemical composition
Crystal system
Elementary cell
Space group
Forms

: Staurolite $(\text{Fe,Mg})_2\text{Al}_9(\text{Si,Al})_4\text{O}_{20}(\text{O,OH})_2$
: monoclinic
: $a_0= 7.863 \text{ \AA}$, $b_0= 16.61 \text{ \AA}$, $c_0= 5.65 \text{ \AA}$, $\beta = 90-93^\circ$
: C 2/m
: Rock forming mineral, xx tabular, prismatic, common

Tetraedrit:

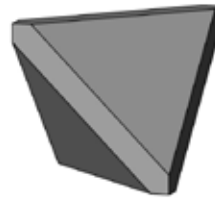
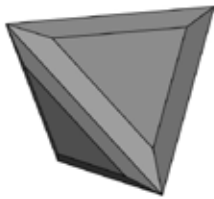
Tetraeder

Tetraeder
HexaederTetraeder
TristetraederTetraeder
Tristetraeder
Rhomboeder

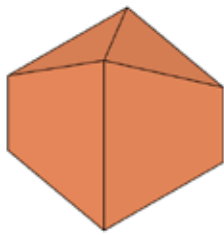
Chemische Zusammensetzung	: $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$, z. T. mit Ag-, Hg, Zn-Gehalten
Kristallsystem	: kubisch
Elementarzelle	: $a_0 = 10.36 \text{ \AA}$
Raumgruppe	: I-4 3 m
Ausbildung	: derb, tetraedr. Ausbildung, mit Calcit, Quarz verwachsen

Tetraedrite:

Tetrahedron

Tetrahedron
HexahedronTetrahedron
TristetrahedronTetrahedron
Tristetrahedron
Rhombohedron

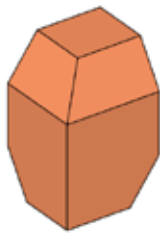
Chemical composition	: $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$, z. T. with Ag-, Hg, Zn-Content
Crystal system	: cubic
Elementary cell	: $a_0 = 10.36 \text{ \AA}$
Space group	: I -4 3 m
Forms	: massive, tetraedr. Forms, with calcite, quartz, intergrown

Titanit:

Prisma
Prisma

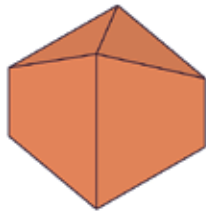


Prisma
Prisma



Prisma
Basispinakoid
Prisma

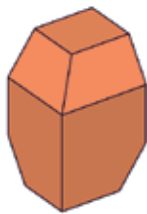
Chemische Zusammensetzung	: CaTi [O / SiO ₄]
Kristallsystem	: Monoklin
Elementarzelle	: $a_0=7.06\text{\AA}$, $b_0=8.71\text{\AA}$, $c_0=6.56\text{\AA}$, $\beta =113.8^\circ$
Raumgruppe	: P2 ₁ /a
Ausbildung	: Eingesprengter, oft Briefkuvertförmig, stengelig, tafelig

Titanite:

Prism
Prism

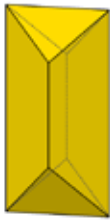


Prism
Prism

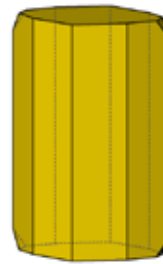


Prism
Basispinacoid
Prism

Chemical composition	: CaTi [O / SiO ₄]
Crystal system	: monoclinic
Elementary cell	: a ₀ =7.06Å, b ₀ =8,71 Å, c ₀ =6.56Å, β =113.8 °
Space group	: P2 ₁ /a
Forms	: disseminated, xx often envelope shape, columnar, tabular

Topas:

Prisma
Basispinakoid



Prisma
Prisma
Basispinakoid
Prisma



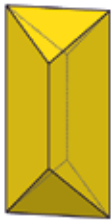
Prisma
Prisma
Basispinakoid
Prisma
Pinakoid



Prisma
Prisma
Basispinakoid
Prisma
Pinakoid

Chemische Zusammensetzung	: $\text{Al}_2 [\text{F}_2 / \text{SiO}_4]$
Kristallsystem	: Orthorhombisch
Elementarzelle	: $a_0 = 4.65 \text{ \AA}$, $b_0 = 8.8 \text{ \AA}$, $c_0 = 8.4 \text{ \AA}$
Raumgruppe	: $P b n m$
Ausbildung	: Körnig, stengelig, prismatisch, meißelförmig

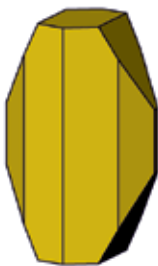
Topaz:



Prism
Basispinacoid



Prism
Prism
Basispinacoid
Prism



Prism
Prism
Basispinacoid
Prism
Pinacoid

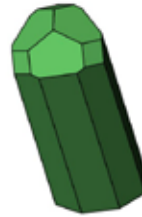


Prism
Prism
Basispinacoid
Prism
Pinacoid

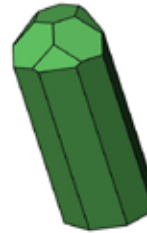
Chemical composition	: Al ₂ [F ₂ / SiO ₄]
Crystal system	: orthorhombic
Elementary cell	: a ₀ = 4.65Å, b ₀ = 8.8Å, c ₀ = 8.4Å
Space group	: P b n m
Forms	: grains, xx columnar, prismatic, chisel forms

Turmalin:*versch. Köpfe:*

Basispedion
Trigonales Prisma
Hexagonales Prisma



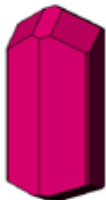
Pyramide
Pyramide
Trigonales Prisma
Hexagonales Prisma



Pyramide
Pyramide
Trigonales Prisma
Hexagonales Prisma



Pyramide
Trigonales Prisma
Pyramide
Pyramide



Pyramide
Pyramide
Trigonales Prisma
Pyramide
Pyramide



Pyramide
Trigonales Prisma
Hexagonales Prisma
Pyramide
Basispedion

Chemische Zusammensetzung : $XY_3Z_6 [(OH, F)_4 / (BO_3)_3 / Si_6O_{18}]$

X: Na, Ca;

Y: Mg, Li, Fe, Mn, Al;

Z: Al, Fe, Cr

Kristallsystem : Trigonal, polare c-Achse

Elementarzelle : $a_0 = 15.92\text{\AA}$, $c_0 = 7.19\text{\AA}$

Raumgruppe : R 3 m

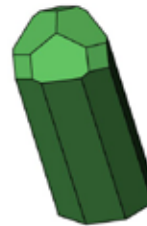
Ausbildung : Säulige, Stängelige, nadelige xx, radialstrahlig
deutl. erkennbare Längsstreifung

Tourmaline:

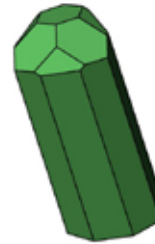
various heads:



Basispedia
Trigonal prism
Hexagonal prism



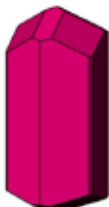
Pyramid
Pyramid
Trigonal prism
Hexagonal prism



Pyramid
Pyramid
Trigonal prism
Hexagonal prism



Pyramid
Trigonal prism
Pyramid
Pyramid



Pyramid
Pyramid
Trigonal prism
Pyramid
Pyramid



Pyramid
Trigonal prism
Hexagonal prism
Pyramid
Basispedia

Chemical composition

: $XY_3Z_6 [(OH, F)_4 / (BO_3)_3 / Si_6O_{18}]$

X: Na, Ca;

Y: Mg, Li, Fe, Mn, Al;

Z: Al, Fe, Cr

Crystal system

: Trigonal, polare c-Achse

Elementary cell

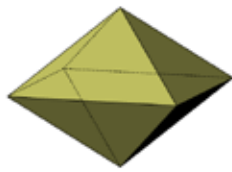
: $a_0 = 15.92\text{\AA}$, $c_0 = 7.19\text{\AA}$

Space group

: R 3 m

Forms

: columnar, needles, radial, perfect columnar, needles, radial

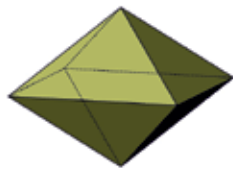
Zirkon:

Dipyramide

Dipyramide
und Prisma
II. StellungDipyramide
und Prisma
I. StellungDipyramide
Prisma I. St.
Prisma II. St.

Chemische Zusammensetzung
Kristallsystem
Elementarzelle
Raumgruppe
Ausbildung

: ZrSiO_4 , enthält diadoch Hf, SEE
: Tetragonal
: $a_0 = 6.60 \text{ \AA}$, $c_0 = 5.98 \text{ \AA}$
: $I 4_1/a m d$
: Körner, häufig idiomorph, kurzsäulig, prismatisch

Zircon:

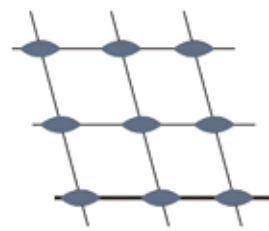
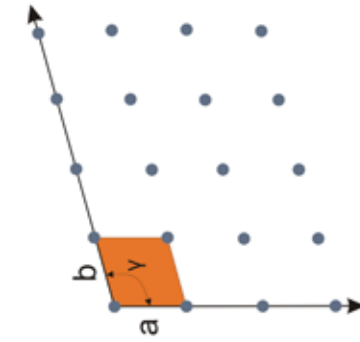
Dipyramid

Dipyramid
and prism
I. positionDipyramid
and prism
II. positionDipyramid
Prism I. pos.
Prism II. pos.

Chemical composition
Crystal system
Elementary cell
Space group
Forms

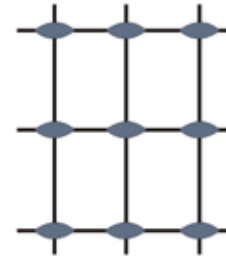
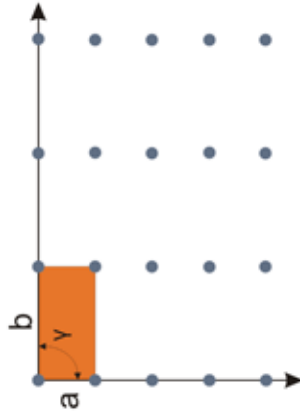
: $Zr[SiO_4]$, contains diadochic Hf, REE
: tetragonal
: $a_0 = 6.60 \text{ \AA}$, $c_0 = 5.98 \text{ \AA}$
: $I 4_1/a m d$
: grains, common idiomorphic xx, short columnar,
prismatic

Schiefes Netz
 $a \neq b$
 $\gamma \neq 90^\circ$



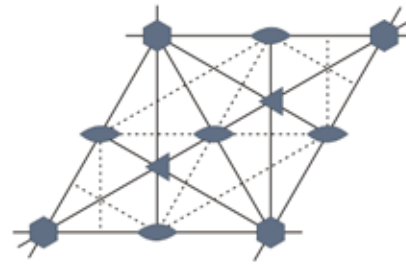
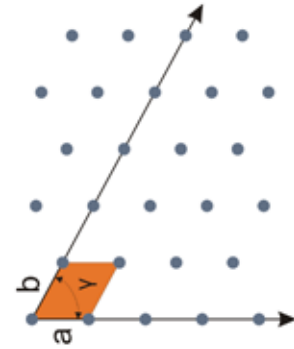
p2

Rechtwinkliges Netz
 $a \neq b$
 $\gamma = 90^\circ$



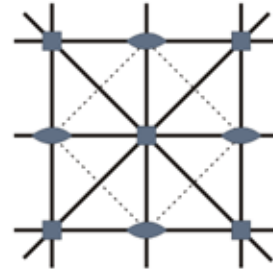
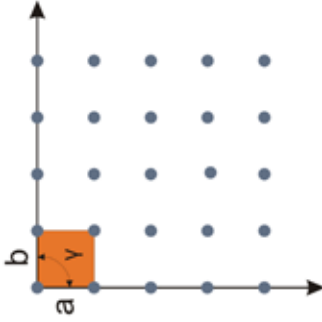
p2mm

Hexagonales Netz
 $a = b$
 $\gamma = 60^\circ$



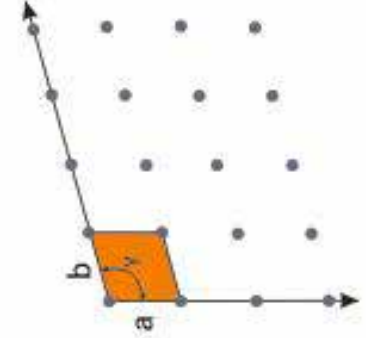
p6mm

Quadratisches Netz
 $a = b$
 $\gamma = 90^\circ$



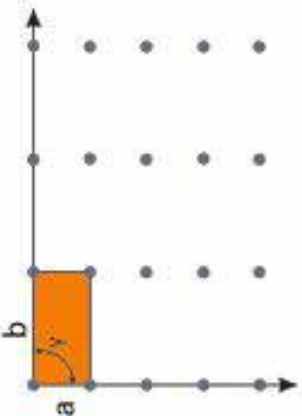
p4mm

skewed grid
 $a \neq b$
 $\gamma \neq 90^\circ$



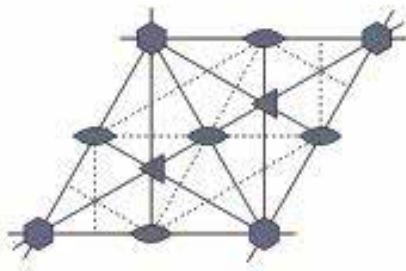
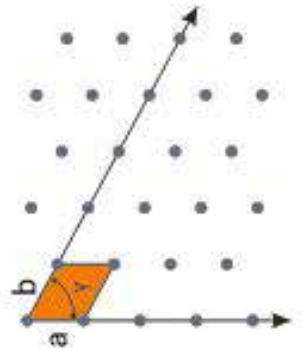
p2

rectangular grid
 $a \neq b$
 $\gamma = 90^\circ$



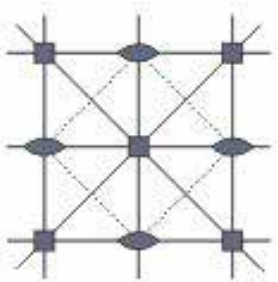
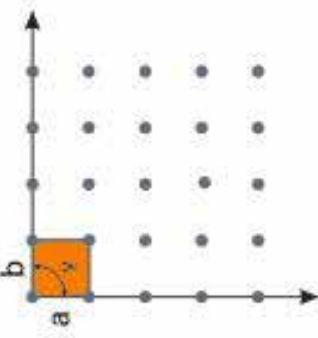
p2mm

hexagonal grid
 $a = b$
 $\gamma = 120^\circ$



p6mm

quadratic grid
 $a = b$
 $\gamma = 90^\circ$



p4mm

Symbolik der Symmetrieebenen

Symmetrieoperation	Symbol	Symbol zur Zeichenebene		Bemerkung
		senkrecht	parallel	
Spiegelung	m			falls die Spiegelebene über der Zeichenebene liegt, wird die Höhe in Bruchteilen der Gitterkonstante angegeben 1/4 = Spiegelebene liegt um 1/4 über der Zeichenebene in der Elementarzelle
Gleit- spiegelung, achsial	a, b			Gleit- spiegelung um $\bar{a}/2 \parallel$ a-Achse $\bar{b}/2 \parallel$ b-Achse $\bar{c}/2 \parallel$ c-Achse falls die Spiegelebene über d. Zeichenebene liegt, wird die Höhe in Bruchteilen der Gitterkonstante angegeben
	c		keines	
Gleit- spiegelung, diagonal	n			$\bar{t}(\bar{a}+\bar{b})/2 \parallel (001)$ $\bar{t}(\bar{a}+\bar{c})/2 \parallel (010)$ $\bar{t}(\bar{b}+\bar{c})/2 \parallel (100)$ im tetragonalen und kubischen Fall $\bar{t}(\bar{a}+\bar{b}+\bar{c})/2 \parallel (111)$
Diamant- gleit- spiegelung	d			$\bar{t}(\bar{a}+\bar{b})/4$ $\bar{t}(\bar{a}+\bar{c})/4$ $\bar{t}(\bar{b}+\bar{c})/4$ im tetragonalen und kubischen Fall $\bar{t}(\bar{a}+\bar{b}+\bar{c})/4$



Wirkung einer Spiegelebene \perp zur Projektionsebene:
Original- und Bildpunkt werden durch ein Komma voneinander unterschieden. Die Lage der Projektionsebene wird durch + (oberhalb) oder - (unterhalb) beschrieben.

Symbolism of symmetry planes

Symmetry-operation	Symbol	vertical lto drawing plane	parallel	remarks
mirror plane	m			when mirror plane is above drawing plane, the height is given in fractions of the lattice parameter $\frac{1}{4}$ = mirror plane lies in $\frac{1}{4}$ above the drawing plane in elementary cell
Glide plane, axial	a, b			Glide plane in $\bar{a}/2 \parallel$ a-axis $\bar{b}/2 \parallel$ b-axis $\bar{c}/2 \parallel$ c-axis when glide plane is above drawing plane the height is given in fractions of the lattice parameters
	c		keines	
Glide plane, diagonal	n			$\bar{t}(\bar{a}+\bar{b})/2 \parallel (001)$ $\bar{t}(\bar{a}+\bar{c})/2 \parallel (010)$ $\bar{t}(\bar{b}+\bar{c})/2 \parallel (100)$ in tetragonal and cubic case $\bar{t}(\bar{a}+\bar{b}+\bar{c})/2 \parallel (111)$
Diamond-glide-plane	d			$\bar{t}(\bar{a}+\bar{b})/4$ $\bar{t}(\bar{a}+\bar{c})/4$ $\bar{t}(\bar{b}+\bar{c})/4$ in tetragonal and cubic case $\bar{t}(\bar{a}+\bar{b}+\bar{c})/4$



Effect of mirrorplane \perp to the projection plane:
Original- and image point are separated by commas. The position of the projection plane is given by + (above) or - (below).

Symbolik der Symmetrieachsen

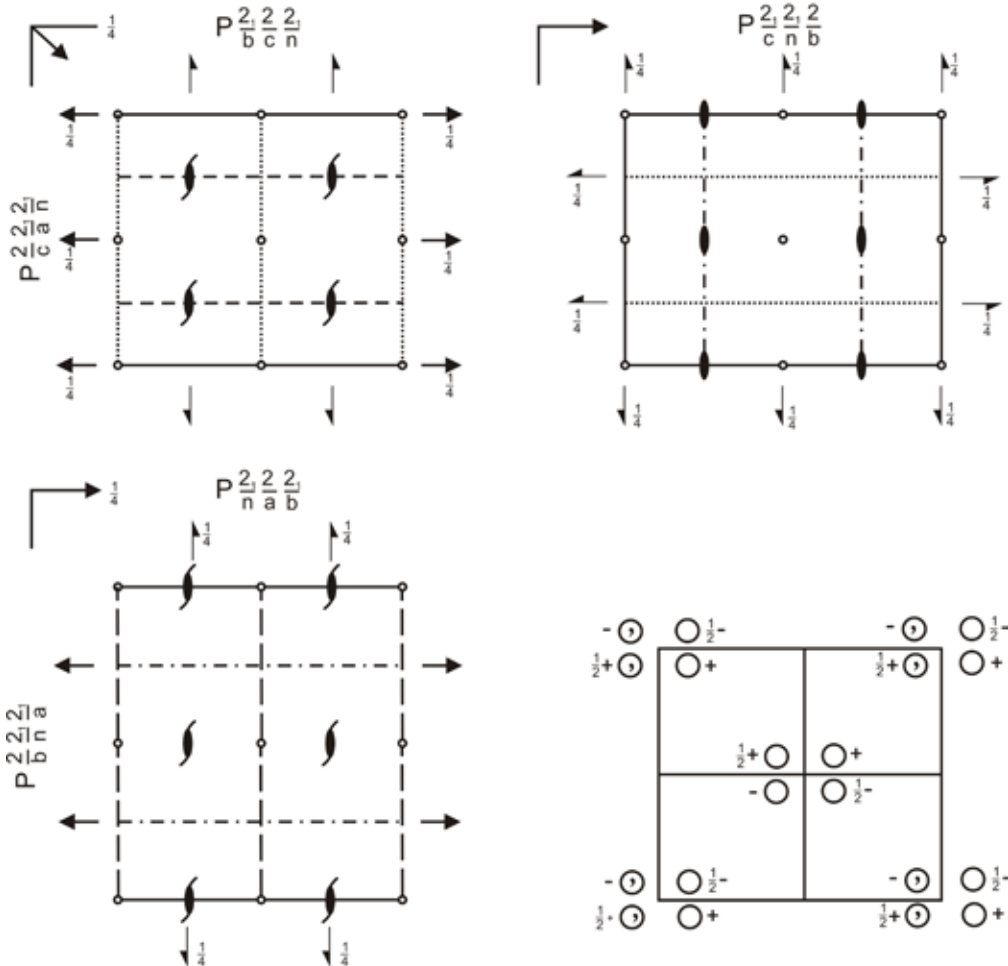
Symmetrieachse	Symbol	graphisches Symbol	Translation Symmetrieachse bei rechtshändiger Symmetrieoperation
Symmetrieachse	1		keine
Inversionszentrum	$\bar{1}$	○	keine
2-zählige Drehung	2	∩	keine
2-zählige Schraubung	2_1	∩	1/2
3-zählige Drehung	3	▲	keine
3-zählige Schraubung	3_1	▲	1/3
	3_2	▲	2/3
3-zählige Inversionsdrehachse	$\bar{3}$	△	keine
4-zählige Drehung	4	◆	keine
4-zählige Schraubung	4_1	◆	1/4
	4_2	◆	1/2
	4_3	◆	3/4
4-zählige Inversionsdrehachse	$\bar{4}$	◇	keine
6-zählige Drehung	6	⬠	keine
6-zählige Schraubung	6_1	⬠	1/6
	6_2	⬠	2/6
	6_3	⬠	3/6
	6_4	⬠	4/6
	6_5	⬠	5/6
6-zählige Inversionsdrehachse	$\bar{6}$	⬠	keine

Symbolism of the symmetry axes

Symmetry axes	Symbol	graphic symbol	Translation parallel to symmetry axis of right-handed symmetry operation
Symmetry axis	1		none
Inversion center	$\bar{1}$	○	none
2-fold rotation	2		none
2-fold screw	2_1		1/2
3-fold rotation	3	▲	none
3-fold screw	3_1		1/3
	3_2		2/3
3-fold inversion axis	$\bar{3}$	△	none
4-fold rotation	4	◆	none
4-fold screw	4_1		1/4
	4_2		1/2
	4_3		3/4
4-fold inversion axis	$\bar{4}$	◊	none
6-fold rotation	6	⬠	none
6-fold screw	6_1		1/6
	6_2		2/6
	6_3		3/6
	6_4		4/6
	6_5		5/6
6-fold inversion axis	$\bar{6}$	⬡	none

Beispiel einer Raumgruppe **Pbcn**

Pbcn D_{2h}^{14} mmm Orthorhombisches Kristallsystem
 P $2_1/b$ $2/c$ $2_1/n$ Patterson-Symmetrie Pmmm



Ursprung bei $\bar{1}$ in $1\ c\ 1$

Asymmetrische Einheit $0 \leq x \leq 1/2$ $0 \leq y \leq 1/2$ $0 \leq z \leq 1/2$

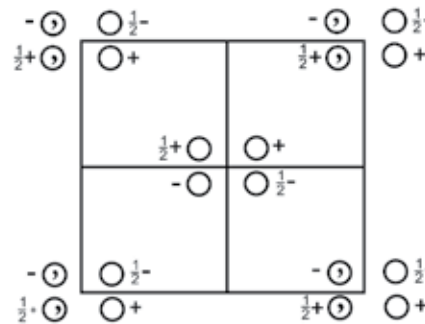
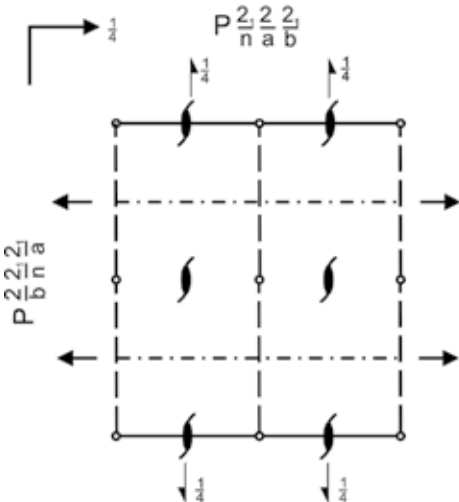
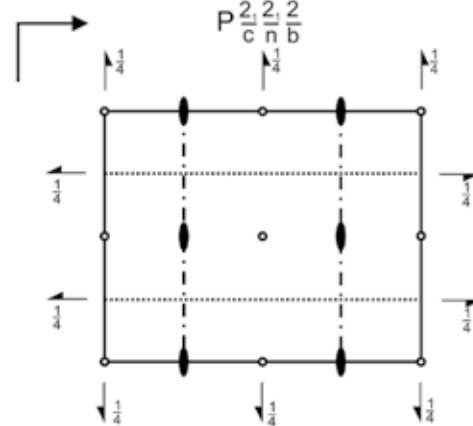
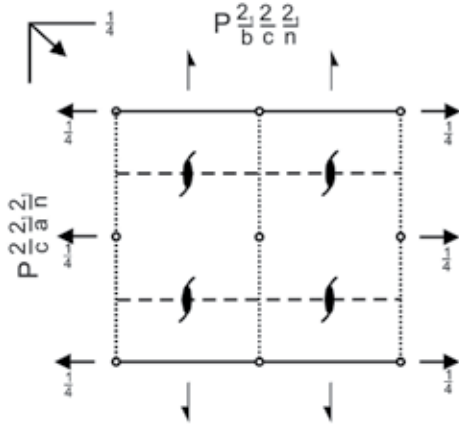
Symmetrie Operationen

- | | | |
|--------------------------------|--------------------------------|----------------------------------|
| (1) 1 | (2) $2(0,0,1/2)$ $1/4, 1/4, z$ | (3) $2\ 0, y, 1/4$ |
| (4) $2(1/2, 0, 0)$ $x, 1/4, 0$ | (5) $\bar{1}\ 0, 0, 0$ | (6) $n(1/2, 1/2, 0)$ $x, y, 1/4$ |
| (7) $c\ x, 0, z$ | (8) $b\ 1/4, y, z$ | |

Example of a space group **Pbcn**

Pbcn D_{2h}^{14} mmm
 $P 2_1/b 2_1/c 2_1/n$

Orthorhombic crystal system
 Patterson symmetry Pmmm



Origin at $\bar{1}$ in $1 c 1$

Asymmetric unit $0 \leq x \leq 1/2$ $0 \leq y \leq 1/2$ $0 \leq z \leq 1/2$

Symmetry operations

- | | | |
|----------------------------|------------------------------|------------------------------|
| (1) 1 | (2) $2(0,0,1/2)$ $1/4,1/4,z$ | (3) $2 0,y,1/4$ |
| (4) $2(1/2,0,0)$ $x,1/4,0$ | (5) $\bar{1} 0,0,0$ | (6) $n(1/2,1/2,0)$ $x,y,1/4$ |
| (7) $c x,0,z$ | (8) $b 1/4,y,z$ | |

Die 230 Raumgruppen

Kristallsystem	Punktgruppe	Raumgruppen				
Triklin	1	P1				
	-1	P1				
Monoklin	2	P2	P2 ₁	C2		
	M	Pm	Pc	Cm	Cc	
	2/m	P2/m	P2 ₁ /m	C2/m	P2/c	
		P2 ₁ /c	C2/c			
Orthorhombisch	222	P222	P222 ₁	P2 ₁ 2 ₁ 2	P2 ₁ 2 ₁ 2 ₁	
		C222 ₁	C222	F222	I222	
	mm2	I2 ₁ 2 ₁ 2 ₁				
		Pmm2	Pmc2 ₁	Pcc2	Pma2	
		Pca2 ₁	Pnc2	Pmn2 ₁	Pba2	
		Pna2 ₁	Pnn2	Cmm2	Cmc2 ₁	
		Ccc2	Amm2	Abm2	Ama2	
		Aba2	Fmm2	Fdd2	Imm2	
		Iba2	Ima2			
		Pmmm	Pnnn	Pccm	Pban	
	mmm	Pmma	Pnna	Pmna	Pcca	
		Pbam	Pccn	Pbcm	Pnnm	
		Pmmn	Pbcn	Pbca	Pnma	
		Cmcm	Cmca	Cmmm	Ccem	
		Cmma	Ccca	Fmmm	Fddd	
		Immm	Ibam	Ibca	Imma	
Tetragonal	4	P4	P4 ₁	P4 ₂	P4 ₃	
	-4	I4	I4 ₁			
	4/m	P-4	I-4			
	422	P4/m	P4 ₂ m	P4/n	P4 ₂ /n	
		I4/m	I4 ₁ /a			
		P422	P4 ₂ 1 ₂	P4 ₁ 22	P4 ₁ 2 ₁ 2	
		P4 ₂ 22	P4 ₂ 2 ₁ 2	P4 ₃ 22	P4 ₃ 2 ₁ 2	
	4mm	I422	I4 ₁ 22			
		P4mm	P4bm	P4 ₂ cm	P4 ₂ nm	
		P4cc	P4nc	P4 ₂ mc	P4 ₂ bc	
		I4mm	I4cm	I4 ₁ md	I4 ₁ cd	
	-42m	P-42m	P-42c	P-42 ₁ m	P-42 ₁ c	
		P-4m2	P-4c2	P-4b2	P-4n2	
		I-4m2	I-4c2	I-42m	I-42d	
		P4/mmm	P4/mcc	P4/nbm	P4/nnc	
	4/mmm	P4/mbm	P4/mnc	P4/nmm	P4/ncc	
P4 ₂ /mmc		P4 ₂ /mcm	P4 ₂ /nbc	P4 ₂ /nnm		
P4 ₂ /mbc		P4 ₂ /mnm	P4 ₂ /nmc	P4 ₂ /ncm		
I4/mmm		I4/mcm	I4 ₁ /amd	I4 ₁ /acd		
Trigonal	3	P3	P3 ₁	P3 ₂	R3	
	-3	P-3	R-3			
	32	P312	P321	P3 ₁ 12	P3 ₁ 21	
		P3 ₂ 12	P3 ₂ 21	R32		
	3m	P3m1	P31m	P3c1	P31c	
		R3m	R3c			
	-3m	P-31m	P-31c	P-3m1	P-3c1	
		R-3m	R-3c			
Hexagonal	6	P6	P6 ₁	P6 ₅	P6 ₂	
	-6	P6 ₄	P6 ₃			
	6/m	P-6				
	622	P6/m	P6 ₃ /m			
		P622	P6 ₁ 22	P6 ₅ 22	P6 ₂ 22	
		P6 ₄ 22	P6 ₃ 22			
	6mm	P6mm	P6cc	P6 ₃ cm	P6 ₃ mc	
	-6m2	P-6m2	P-6c2	P-62m	P-62c	
6/mmm	P6/mmm	P6/mcc	P6 ₃ /mm	P6 ₃ /mmc		
Kubisch	23	P23	F23	I23	P2 ₁ 3	
	m-3	I2 ₁ 3				
		Pm-3	Pn-3	Fm-3	Fd-3	
		Im-3	Pa-3	Ia-3		
	432	P432	P4 ₂ 32	F432	F4 ₁ 32	
		I432	P4 ₃ 32	P4 ₁ 32	I4 ₁ 32	
	-43m	P-43m	F-43m	I-43m	P-43n	
		F-43c	I-43d			
	m-3m	Pm-3m	Pn-3n	Pm-3n	Pn-3m	
		Fm-3m	Fm-3c	Fd-3m	Fd-3c	
Im-3m		Ia-3d				

The 230 space groups

Crystal system	Point group	Space group					
triclinic	1	P1					
	-1	P1					
monoclinic	2	P2	P2 ₁	C2			
	M	Pm	Pc	Cm	Cc		
	2/m	P2/m	P2 ₁ /m	C2/m	P2/c		
		P2 ₁ /c	C2/c				
orthorhombic	222	P222	P222 ₁	P2 ₁ 2 ₁ 2	P2 ₁ 2 ₁ 2 ₁		
		C222 ₁	C222	F222	I222		
		I2 ₁ 2 ₁ 2 ₁					
	mm2		Pmm2	Pmc2 ₁	Pcc2	Pma2	
			Pca2 ₁	Pnc2	Pmn2 ₁	Pba2	
			Pna2 ₁	Pnn2	Cmm2	Cmc2 ₁	
			Ccc2	Amm2	Abm2	Ama2	
			Aba2	Fmm2	Fdd2	Imm2	
			Iba2	Ima2			
			Pmmm	Pnnn	Pccm	Pban	
	mmm		Pmma	Pnna	Pmna	Pcca	
			Pbam	Pvvn	Pbcm	Pnnm	
			Pmmn	Pbcn	Pbca	Pnma	
			Cmcm	Cmca	Cmmm	Ccem	
			Cmma	Ccca	Fmmm	Fddd	
			Immm	Ibam	Ibca	Imma	
		tetragonal	4	P4	P4 ₁	P4 ₂	P4 ₃
				I4	I4 ₁		
				P-4	I-4		
			P4/m	P4 ₂ m	P4/n	P4 ₂ /n	
	I4/m		I4 ₁ /a				
	P422		P4 ₂ 2 ₁ 2	P4 ₁ 22	P4 ₁ 2 ₁ 2		
	422		P4 ₂ 22	P4 ₂ 2 ₁ 2	P4 ₃ 22		
	I422		I4 ₁ 22				
4mm			P4mm	P4bm	P4 ₂ cm	P4 ₂ nm	
			P4cc	P4nc	P4 ₂ mc	P4 ₂ bc	
		I4mm	I4cm	I4 ₁ md	I4 ₁ cd		
		P-42m	P-42c	P-42 ₁ m	P-42 ₁ c		
	-42m	P-4m2	P-4c2	P-4b2	P-4n2		
4/mmm		I-4m2	I-4c2	I-42m	I-42d		
		P4/mmmm	P4/mcc	P4/nbm	P4/nnc		
		P4/mbm	P4/mnc	P4/nmm	P4/ncc		
		P4 ₂ /mmc	P4 ₂ /mcm	P4 ₂ /nbc	P4 ₂ /nmm		
		P4 ₂ /mbc	P4 ₂ /mnm	P4 ₂ /nmc	P4 ₂ /ncm		
		I4/mmm	I4/mcm	I4 ₁ /amd	I4 ₁ /acd		
trigonal	3	P3	P3 ₁	P3 ₂	R3		
		P-3	R-3				
	32	P312	P321	P3 ₁ 12	P3 ₁ 21		
		P3 ₂ 12	P3 ₂ 21	R32			
	3m	P3m1	P31m	P3c1	P31c		
		R3m	R3c				
	-3m	P-31m	P-31c	P-3m1	P-3c1		
		R-3m	R-3c				
hexagonal	6	P6	P6 ₁	P6 ₅	P6 ₂		
		P6 ₄	P6 ₃				
		P-6					
	6/m	P6/m	P6 ₃ /m				
	622	P622	P6 ₁ 22	P6 ₅ 22	P6 ₂ 22		
		P6 ₄ 22	P6 ₃ 22				
	6mm	P6mm	P6cc	P6 ₃ cm	P6 ₃ mc		
		P-6m2	P-6c2	P-62m	P-62c		
	6/mmm	P6/mmm	P6 ₃ /m	P6 ₃ /mmc			
Cubic	23	P23	F23	I23	P2 ₁ 3		
		I2 ₁ 3					
	m-3		Pm-3	Pn-3	Fm-3	Fd-3	
			Im-3	Pa-3	Ia-3		
	432	P432	P4 ₂ 32	F432	F4 ₁ 32		
		I432	P4 ₃ 32	P4 ₁ 32	I4 ₁ 32		
	-43m		P-43m	F-43m	I-43m	P-43n	
			F-43c	I-43d			
	m-3m		Pm-3m	Pn-3n	Pm-3n	Pn-3m	
			Fm-3m	Fm-3c	Fd-3m	Fd-3c	
		Im-3m	Ia-3d				

Literatur/Literature

- Aroyo, M. I. & Hahn T. (2013): International Tables for Crystallography: Space-Group Symmetry. Brief Teaching Edition of Volume A, Wiley-Blackwell.
- Aslanov, L.A, Fetisov, G.V. & Howard, J.A. K. (1998): Crystallographic Instrumentation. OUP/ International Union of Crystallography.
- Backhaus, K.-O. (1972): Wörterbuch Kristallografie. Englisch-Deutsch-Französisch-Russisch. Verlag Technik, Berlin.
- Bernstein, J. (2008): Polymorphism in Molecular Crystals. Oxford University Press, Oxford.
- Berger, M. & Ehrenberg, L. (1981): Theorie und Anwendung der Symmetriegruppen. BSB B.G. Teubner Verlagsgesellsch., Leipzig.
- Berry, L.G. & Mason, B. (1959): Mineralogy. W.H. Freeman.
- Bloss, F.D. (1971): Crystallography and Crystal Chemistry. Holt Rinehart & Winston, New York.
- Boldyreva, E. (2010): High-Pressure Crystallography: From Fundamental Phenomena to Technological Applications. NATO Science for Peace and Security Series - B: Physics and Biophysics, Springer Verlag.
- Borchardt-Ott, W. (1995): Kristallographie. Springer Verlag.
- Borges, F.S. (1980): Elementos de cristalografia. Fundação Calouste Gulbenkian, Lisboa.
- Bruhns, W. & Ramdohr, P. (1965): Kristallographie. 6. Aufl., Walter de Gruyter, Berlin.
- Brown, I.D. (2006): The Chemical Bond in Inorganic Chemistry: The Bond Valence Model. OUP Oxford.
- Bunn, C.W. (1961): Chemical Crystallography. 2. Aufl., Clarendon Press, Oxford.
- Burkhardt, J.J. (1947): Die Bewegungsgruppen der Kristallographie. Birkhäuser Verlag, Basel.
- Burkhardt, J.J. (1988): Die Symmetrie der Kristalle. Birkhäuser Verlag, Basel.
- Burns, G. & Glasser, A.M. (1978): Space Groups for Solid State Scientists. Academic Press, New York.
- Burzlauff, H. & Zimmermann, H. (1986): Kristallsymmetrie, Kristallstruktur. R. Merkel Verlag, Erlangen.
- Chatterjee, S.K. (2008): Crystallography and the Worlds of Symmetry. Berlin, Springer.
- Clegg, W., Blake, A.J., Gould, R.O. & Main, P. (2009): Crystal Structure Analysis, Principles and Practice. Oxford Science, IUCR.
- Deer, W.A., Howie, R.A. & Zussman, J. (1992): An introduction to the rock-forming minerals. Essex, Longman Scientific & Technical.
- Dent Glasser, L.S. (1977): Crystallography and its Applications. New York, Van Nostrand.
- De Graef, M. & McHenry, M. E. (2012): Structure of Materials: An Introduction to Crystallography, Diffraction and Symmetry. Cambridge University Press.
- Donaldson, J.D. & Ross, S.D. (1972): Symmetrie und Stereochemie. London.
- Dorain, P.B. (1972): Symmetrie und anorganische Strukturchemie. Berlin Akademie-Verlag.
- Dyar, M.D. & Gunter, M.E. (2008): Mineralogy and optical mineralogy. Mineralogical Society of America, CD-ROM.

- Eckert, E. & Lindner, J.H. (1976): Strukturen der Materie und ihre Symmetrie. Bildstudio, Nieder-Ramstadt.
- Fabian, E. (1986): Die Entdeckung der Kristalle. Deutscher Verlag für Grundstoffindustrie, Leipzig.
- Ferraris, G., Makovicky, E. & Merlino S. (2008): Crystallography of Modular Materials. IUCR 15.
- Gay, P. (1972): The Crystalline State. Oliver and Boyd, Edinburgh.
- Giacovazzo, C. (2011, Ed.): Fundamentals of Crystallography. Oxford University Press, Oxford.
- Girolami, S. G. (2013): X-Ray Crystallography. University Science Books U.S.
- Glusker, J. P. & Trueblood, K. N. (2010): Crystal Structure Analysis. Oxford University Press, Oxford.
- Gottstein, G. (2007): Physikalische Grundlagen der Materialkunde. Springer Verlag, Berlin.
- Groth, P.H. von (2007): Tabellarische Übersicht der Mineralien nach ihren kristallographisch-chemischen Beziehungen. VDM Verlag Dr. Müller, Saarbrücken.
- Groth, P.H. von (1966): Entwicklungsgeschichte der mineralogischen Wissenschaften (Nachdruck d. Ausgabe von 1926). Springer Verlag, Berlin.
- Hahn, T. (2005): Complete Printed Set of International Tables for Crystallography: International Tables for Crystallography, Volume A: Space Group Symmetry.
- Hammond, C. (2009): The basics of crystallography and diffraction. 3. edition. International Union of Crystallography, Oxford Science Publications, Oxford University Press, Oxford.
- Hochleitner, R., Philipsborn, H. & Weiner, K.L. (1996): Minerale. Bestimmen nach äußeren Kennzeichen. Schweizerbart, Stuttgart.
- Jaffe, H.H. & Orchin, M. (1973): Symmetrie in der Chemie. 2. Aufl., Hüthig-Verlag, Heidelberg.
- Janssen, T., Chapuis, G. & de Boissieu, M. (2007): Aperiodic Crystals From Modulated Phases to Quasicrystals. IUCR 20.
- Julian, M.M. (2008): Foundations of Crystallography with Computer Applications. Crc Pr Inc.
- Kelly, A.A. & Knowles, K.M. (2012): Crystallography and Crystal Defects. John Wiley & Sons, New York.
- Kleber, W., Bautsch, H.-J. & Bohm, J. (1998): Einführung in die Kristallographie. 18. Auflage. Verlag Technik, Berlin.
- Klein, C. & Dutrow, B. (2007): Manual of Mineral Science. 23rd Edition. John Wiley & Sons, New York.
- Klein, C. (1989): Minerals and rocks exercises in crystallography, mineralogy and hand specimen petrology. Revised edition, John Wiley & Sons, New York.
- Klemm, M. (1982): Symmetrien von Ornamenten und Kristallen. Springer Verlag, Berlin.
- Knox, R.S. & Gold, A. (1964): Symmetry in the Solid State. W.A. Benjamin, New York Amsterdam.
- Krivovichev, S.V. (2009): Krivovichev Structural Crystallography of Inorganic Oxysalts. IUCR 22.
- Loeb, A.L. (1971): Color and Symmetry. John Wiley & Sons, New York.

- Ludwig, W. & Falter, C. (1988): Symmetries in Physics. Group Theory Applied to Physical Problems. Springer-Verlag, Berlin.
- McKie, S. & McKie, Chr. (1974): Crystalline Solids. Thomas Nelson, London.
- Muller, O. & Roy, R. (1974): Crystal chemistry of non-metallic materials. Springer Verlag, Berlin.
- Nicolle, J. (1954): Die Symmetrie und ihre Anwendung. Deutscher Verlag der Wissenschaften, Berlin.
- Niimura, N. & Podjarny, N. (2011): Neutron Protein Crystallography Hydrogen, Protons, and Hydration in Bio-macromolecules. Oxford University Press, Oxford.
- Oppermann, E. (2004): Kristalle und ihre Formen. Band 1, Kristallografik Verlag, Achberg.
- Oppermann, E. (2004): Kristalle und ihre Formen. Band 2, KristalloGrafik Verlag, Achberg.
- Oppermann, E. (2009): Kristalle und ihre Formen. Band 3, KristalloGrafik Verlag, Achberg.
- Pecharsky, V.K. & Zavalij, P. (2009): Fundamentals of Powder Diffraction and Structural Characterization of Materials. 2. Auflage, Springer Verlag, New York.
- Phillips, F.C. (1971): An Introduction to Crystallography. 4. Aufl., Oliver and Boyd, Edinburgh.
- Pöllmann, H., DaCosta, M.: Symmetrien, symetria, die 32 Punktgruppen, os 32 grupos de pontos, Minerale und ihre Formen, minerais e suas formas, Hallisches Jahrbuch für Geowissenschaften, Beiheft 30, (2013)
- Prince, E. (1982): Mathematical Techniques in Crystallography and Materials Science. Springer-Verlag, Berlin.
- Putnis, A. (1993): Introduction to mineral sciences. Cambridge University Press, New York.
- Quenstedt, F.A. (2007): Methode der Kristallographie. VDM Verlag Dr. Müller, Saarbrücken.
- Rath, R. (1965): Kristallographie. Philips, Eindhoven.
- Rohdes, G. (2006): Crystallography Made Crystal Clear. A Guide for Users of Macromolecular Models. Academic Press.
- Rosenfeld, B.A. & Sergejewa, N.D. (1978): Stereographic Projection. Izd. Mir ,Moskva.
- Rösler, H.-J. (1985): Lehrbuch der Mineralogie. 3. Aufl., Deutscher Verlag für Grundstoffindustrie, Leipzig.
- Rösler, H.-J. (1991): Lehrbuch der Mineralogie. 5. Aufl., Spectrum Akademischer Verlag, Berlin.
- Rupp B. (2009): Biomolecular Crystallography: Principles, Practice, and Application to Structural Biology. Taylor & Francis Ltd.
- Sands, D.E. (1994): Introduction to Crystallography. New York: W. A. Benjamin 1969. Reprint: Dover Pub. Inc.
- Scholz, E. (1990): Symmetrie, Gruppe, Dualität. Birkhäuser Verlag, Basel.
- Schroeder, R. (1950): Krystallometrisches Praktikum. Springer-Verlag, Berlin.
- Schumann, H. (1980): Kristallgeometrie. Deutscher Verlag für Grundstoffindustrie, Leipzig.
- Schwarzenbach, D. (1997): Crystallography. John Wiley & Sons.
- Schwarzenberger, R.L.E. (1980): N-dimensional Crystallography. Pitman, San Francisco.

- Shmueli, U. & Weiss G.H. (1995): Introduction to Crystallographic Statistics. OUP/International Union of Crystallography.
- Sommerfeldt, E. (1906): Geometrische Kristallographie. Wilhelm Engelmann, Leipzig.
- Steadman, R. (1982): Crystallography. Van Nostrand, New York.
- Strunz, H. & Nickel, E.H. (2001): Mineralogical Tables. Chemical-Structural Mineral Classification System. 9. Edition, Schweizerbart, Stuttgart.
- Tertsch, H. (1954): Die stereographische Projektion in der Kristallkunde. Verlag für Angew. Wissensch., Wiesbaden.
- Turowski, S. & Borchardt, R. (1999): Symmetriellehre der Kristallographie. Oldenbourg-Verlag, München.
- Vainshtein, B.K. (1983, Ed.): Modern Crystallography. Vol. 2: Structure of Crystals. Springer-Verlag, Berlin.
- Verma, A.R. & Krishna, P. (1966): Polymorphism and Polytypism in Crystals. John Wiley & Sons, New York.
- Van Smaalen, S. (2007): Incommensurate Crystallography. IUCR 21.
- Li W., Zhou G. & Mak T. (2008): Advanced Structural Inorganic Chemistry. Oxford Univ. Press, Oxford.
- Weiss, A. & Witte, U. (1983): Kristallstruktur und chemische Bindung. Verlag Chemie, Weinheim/Bergstr.
- Weyl, H. (1952): Symmetrie. University Press, Princeton.
- Wondratschek, H. & Müller, U. (2010): International Tables for Crystallography: Volume A1: Symmetry Relations between Space Groups. John Wiley & Sons, New York.
- Wyckoff, R.W.G. (1962-1968): Crystal Structures. Vol. 1-6, 2. Aufl., Interscience Publishers, New York.
- Zolotoyabko, E. (2011): Basic Concepts of Crystallography. Wiley-VCH Verlag, Weinheim.

